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Intrinsic electron traps in atomic-layer deposited HfO₂ insulators

F. Cerbu,¹ O. Madia,¹ D. V. Andreev,^{1,2} S. Fadida,³ M. Eizenberg,³ L. Breuil,⁴ J. G. Lisoni,^{4,5} J. A. Kittl,^{1,6} J. Strand,⁷ A. L. Shluger,⁷ V. V. Afanas'ev,¹ M. Houssa,¹ and A. Stesmans¹

¹Laboratory of Semiconductor Physics, Department of Physics and Astronomy, University of Leuven, 3001 Leuven, Belgium

²Bauman Moscow State Technical University—Kaluga Branch, 248000 Kaluga, Moscow obl., Russian Federation

³Department of Materials Science and Engineering, Technion-Israel Institute of Technology, 32000 Haifa, Israel

⁴imec, 3001 Leuven, Belgium

⁵Institute of Physics and Mathematics, Faculty of Science, Universidad Austral de Chile, Valdivia, Chile

⁶Advanced Logic Lab, Samsung Semiconductor, Inc., Austin, 78754 Texas, USA

⁷Department of Physics and Astronomy, University College London, London WC1E 6BT, United Kingdom

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Analysis of photodepopulation of electron traps in HfO₂ films grown by atomic layer deposition is shown to provide the trap energy distribution across the entire oxide bandgap. The presence is revealed of two kinds of deep electron traps energetically distributed at around $E_t \approx 2.0$ eV and $E_t \approx 3.0$ eV below the oxide conduction band. Comparison of the trapped electron energy distributions in HfO₂ layers prepared using different precursors or subjected to thermal treatment suggests that these centers are intrinsic in origin. However, the common assumption that these would implicate O vacancies cannot explain the charging behavior of HfO₂, suggesting that alternative defect models should be considered. *Published by AIP Publishing.* [<http://dx.doi.org/10.1063/1.4952718>]

In recent years, hafnium dioxide (HfO₂) based insulators have emerged as the primary contenders to replace traditional SiO₂ in a variety of nano-electronic devices ranging from deep-scaled transistors to DRAM^{1,2} and non-volatile memory cells.^{3,4} Furthermore, thanks to the ferroelectricity property of both doped^{5,6} and pure⁷ HfO₂, novel applications including memories⁸ and high sub-threshold slope transistors⁹ are envisioned. However, the reliability of the dielectric associated with electron trapping appears to be the “show stopper”: It has already been shown that the positive bias-temperature instability (PBTI) limits the gate oxide scaling in metal-HfO₂-Si transistors^{10–13} while in flash cells, electron trapping in the intergate HfO₂ insulator degrades the program/erase window, retention, and endurance.^{14,15} Moreover, in ferroelectric applications, trapping of electrons within HfO₂ will screen the electric field at the surface of the semiconductor channel, thus directly impairing the device functionality. In particular, electron injection and trapping are expected to be an issue because of the high coercive field strength typical for ferroelectric HfO₂.¹⁶ This signifies the importance of developing proper methods for quantifying and analyzing electron traps aiming at their atomic identification to eventually allow one to eliminate or limit their detrimental impact.

Yet, despite numerous indications of electron traps in HfO₂, little is still known about their atomic origin and the distribution of energy levels. Electron traps in HfO₂ are often associated with oxygen vacancy defects¹⁰ on the basis of electron energy levels found in the energy range 1.2–1.8 eV below the conduction band (CB) bottom edge.^{10,14} However, since theoretical calculations consistently predict that the O vacancy in HfO₂ can act also as a hole trap,^{17–21} this assignment cannot be reconciled with the absence of hole trapping from the valence band (VB) inside the high-k layer.²² The observed trapped hole charge in the SiO₂/HfO₂ stacks appears to be

insensitive to the HfO₂ layer thickness ranging from 5 to >100 nm. Rather, the strong electron trapping in HfO₂ films prepared from the nitrate precursor Hf(NO₃)₄ can be correlated with the presence of N-related defects revealed by electron spin resonance (ESR)^{22,23} suggesting an impurity-related electron trapping. Since then, numerous ESR experiments on the production-grade HfO₂ layers have failed to reveal any feasible signal, prompting the search for alternative trap characterization method(s).

In the present work, we demonstrate that the exhaustive photo-depopulation spectroscopy (EPDS) allows one to determine the energy distribution of trapped electrons in the HfO₂ bandgap. Studies on HfO₂ layers prepared using three different production-grade atomic layer deposition (ALD) routes have demonstrated that at least two different components in the electron trap spectrum are sensitive both to the deposition chemistry and to the subsequent thermal processing. On the basis of these observations, we argue that these traps are associated with intrinsic defects in HfO₂, sensitive to the HfO₂ phase [amorphous (a-) vs. crystalline]. However, upon re-evaluation of the energy levels pertaining to the O vacancy in a-hafnia, we conclude that this model cannot account for the revealed charging behavior and that alternative atomic configurations are at the origin of the electron trapping sites.

Samples were prepared by ALD of 19–20 nm thick HfO₂ on a 5- or 7.5-nm thick SiO₂ layer thermally grown on (100)Si substrates. This SiO₂ layer serves as the tunnel oxide enabling electron injection from Si into the HfO₂ films prepared using three production-grade ALD processes, labeled as A, B, and C: A—using HfCl₄ + H₂O precursors at 300 °C; B—using tetrakis(dimethylamido)hafnium (TDMA-Hf) and H₂O at 300 °C; C—state-of-the-art industrial Cl-free ALD process. For the sake of comparison, process A was also used to synthesize 19-nm thick Hf_{0.8}Al_{0.2}O_x (20 at. % Al nominal)

layers by combining HfCl_4 and $\text{Al}(\text{CH}_3)_3$ pulses. In addition to the as-deposited films, samples annealed for 15 min in N_2 (1 atm) at 600, 800, or 1000 °C were also analyzed. Metal-oxide-Si (MOS) capacitors were completed by thermoresistive evaporation of semitransparent electrodes (13 nm Au) of 1 mm² area. Alternatively, 10 nm TiN/2 nm Si electrodes were formed by physical vapor deposition followed by a 30 min anneal in forming gas (10% H_2 + N_2) at 400 °C.

The EPDS, carried out under application of small positive metal bias (+1 or +2 V), allows the photo-depopulation of electron states in the oxide to reach saturation at a given photon energy $h\nu$ and quantifies the corresponding charge variation from the observed capacitance-voltage (CV) curve shift.^{24–26} The saturation of de-trapping means that all charge carriers available for photo-excitation at this $h\nu$ value are removed (the charge reservoir is exhausted), i.e., there are no electrons left in the states with the optical energy depth $E_t < h\nu$ below the CB bottom. On starting the illumination from a low photon energy $h\nu$ and then, after having reached saturation, increasing $h\nu$ by a small energy step $\delta h\nu$, the next saturation of the de-trapping kinetics would correspond to the complete removal of charge carriers from the traps in the energy interval $\{h\nu; h\nu + \delta h\nu\}$. Then, by performing the depopulation at incremental photon energies, one can find the distribution of the electron states as a function of E_t across the insulator bandgap. Importantly, unlike in the photocurrent measurements, charge monitoring allows the determination of its sign thus enabling discrimination between the excitation of an electron from the insulator valence band (VB) to the empty defect level (negative charging) or the removal of an electron from the gap state to the CB (positive charging). The remaining task consists in the separation of donor and acceptor states corresponding to (0/+ or +/0) and (0/– or –/0) transitions, respectively. The latter is achieved by correlating the density of states inferred from EPDS to the initial net charge in the insulating stack. This charge can be controllably varied by injecting electrons or holes from the Si substrate (tunnelling through the SiO_2 barrier) or e-h pair generation in the insulator under UV illumination through the semitransparent metal

gate.^{22,24} Technical details of the measurements are summarized in the supplementary material.²⁷

Figure 1 compares the illumination-induced charge Q_{stack} variation (a), (b) and the of these derived spectral charge density (SCD) (c), (d) spectra in the pristine and electron-injected (charging voltage pulse $V_g = +15$ V applied to the metal electrode) samples with 19-nm thick HfO_2 and $\text{Hf}_{0.8}\text{Al}_{0.2}\text{O}_x$ insulators (panels a/c and b/d, respectively) fabricated using the HfCl_4 -based process A. One can distinguish three spectral ranges with different electron transitions dominating the charging process. First, for $h\nu < 4$ eV, electrons are excited from the energy levels in the oxide gap [cf. transition 1 in the inset in Fig. 1(a)] leading to a slight “positive” charging in the pristine HfO_2 film or, else, to the removal of electrons captured in the HfO_2 or $\text{Hf}_{0.8}\text{Al}_{0.2}\text{O}_x$ layer upon electron tunnelling. Two important features are worth noting: (1) in the electron-injected samples, all the trapped electrons can be de-trapped under illumination in the spectral range $h\nu < 4$ eV; (2) the charging spectrum of the pristine HfO_2 layer fits in with that of the electron de-trapping suggesting that the “positive” charging of the as-deposited HfO_2 layer is also due to the de-trapping of electrons from the acceptor states filled by electrons during ALD.

Second, when $h\nu$ exceeds 4 eV, the sign of the charging changes to negative. The same effect has been observed earlier in $\text{SiO}_2/\text{Y}_2\text{O}_3$ ²⁵ and $\text{SiO}_2/\text{Al}_2\text{O}_3$ stacks.²⁶ It is related to photo-injection of electrons from Si into the SiO_2 barrier layer and the subsequent trapping of these in HfO_2 [cf. transition 2 in Fig. 1(a)]. The saturation negative charge density detected in the EPDS corresponds to the balance between the trapping of the electrons photo-injected from Si and their de-trapping by the optical transitions of type 1. Third and finally, when $h\nu$ reaches the threshold of intrinsic photoconductivity of HfO_2 (the lowest bandgap $E_g = 5.6$ eV corresponds to the monoclinic phase, see Ref. 28), the generation of electron-hole pairs within the HfO_2 layer leads to the annihilation of trapped electrons.²² Since the final charge state of the oxide stack after exposure to photons with $h\nu > 5.6$ eV is indistinguishable from that of the as-prepared (neutral) oxide

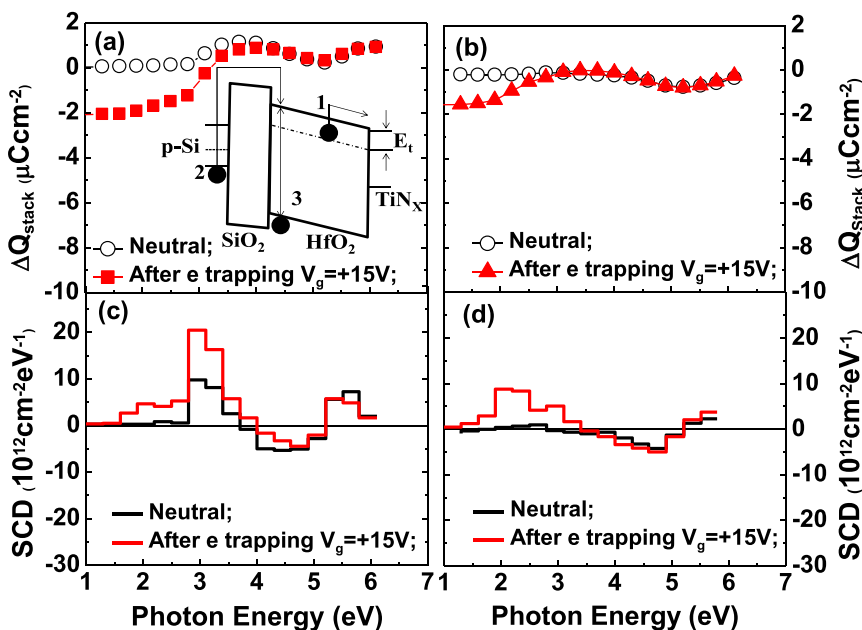


FIG. 1. Illumination-induced charge variations observed under +1 V bias on metal electrode (a), (b) and the inferred SCD distributions (c), (d) for samples with 19-nm thick HfO_2 (a), (c) or $\text{Hf}_{0.8}\text{Al}_{0.2}\text{O}_x$ (b), (d) insulators. Results are shown for both the as-fabricated neutral samples (○) and for the samples injected with electrons by tunneling out of silicon (red online). The inset illustrates the schematics of electron transitions during the EPDS experiment: photoionization of traps with energy E_t (1), electron photoinjection from Si followed by trapping in HfO_2 (2), and electron-hole pair generation in HfO_2 (3).

stack, we conclude that there is neither present a detectable density of hole traps nor donor-type gap states in the HfO_2 or $\text{Hf}_{0.8}\text{Al}_{0.2}\text{O}_x$ layers.

Since the excitation of electrons from the gap states into the HfO_2 CB represents the dominant (dis)-charging mechanism in the range $h\nu < 4\text{ eV}$, the SCD directly reflects the energy distribution of the initial electron states. The spectral plots in Fig. 1 show that there are at least two main components in the trapped electron density—one shallow ($2\text{ eV} < E_t < 3\text{ eV}$) found both in HfO_2 and $\text{Hf}_{0.8}\text{Al}_{0.2}\text{O}_x$ layers and a deeper one ($3\text{ eV} < E_t < 3.5\text{ eV}$) present in HfO_2 . Furthermore, comparison of the SCD distributions observed in ALD HfO_2 layers grown from different precursors (processes A-C), as shown in Fig. 2, panels (a)–(c), respectively, reveals a much lower density of the deep

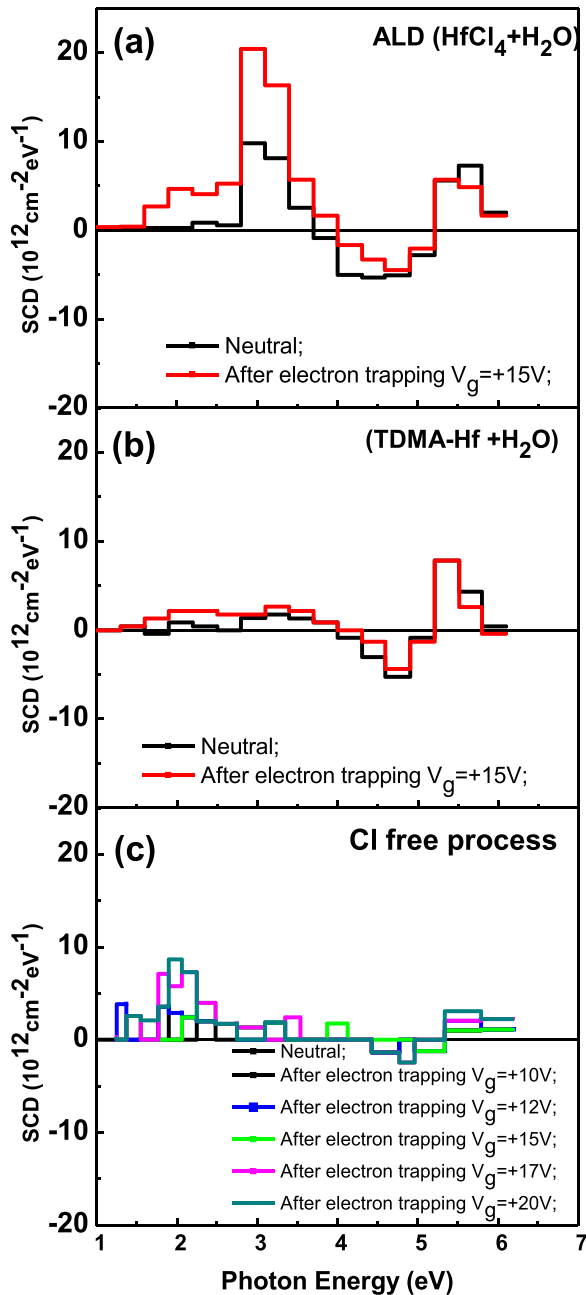


FIG. 2. SCD spectra measured on $\text{Si}/\text{SiO}_2/\text{HfO}_2/\text{Au}$ samples with HfO_2 layers deposited using different ALD processes: A, B, C—corresponding panels (a)–(c), respectively.

electron traps in the Cl-free sample B while in the sample C (also Cl-free) the shallow traps become dominant. The observation of shallow traps both in the carbon-free HfO_2 films (process A) and in the Cl-free layers grown using metallo-organic precursors (processes B and C) suggests that these defects are not impurity-related, but likely intrinsic.

To get further insight into the origin of the electron traps in HfO_2 , we analyzed the effect of annealing on the samples A and C, which exhibit different energy distributions of trapped electrons in the as-deposited state. Figures 3 and 4 show the SCD spectra obtained on the samples A and C, respectively, after 15 min anneal in N_2 at the indicated temperature. In sample type A, the dominant deep traps at around $E_t = 3\text{ eV}$

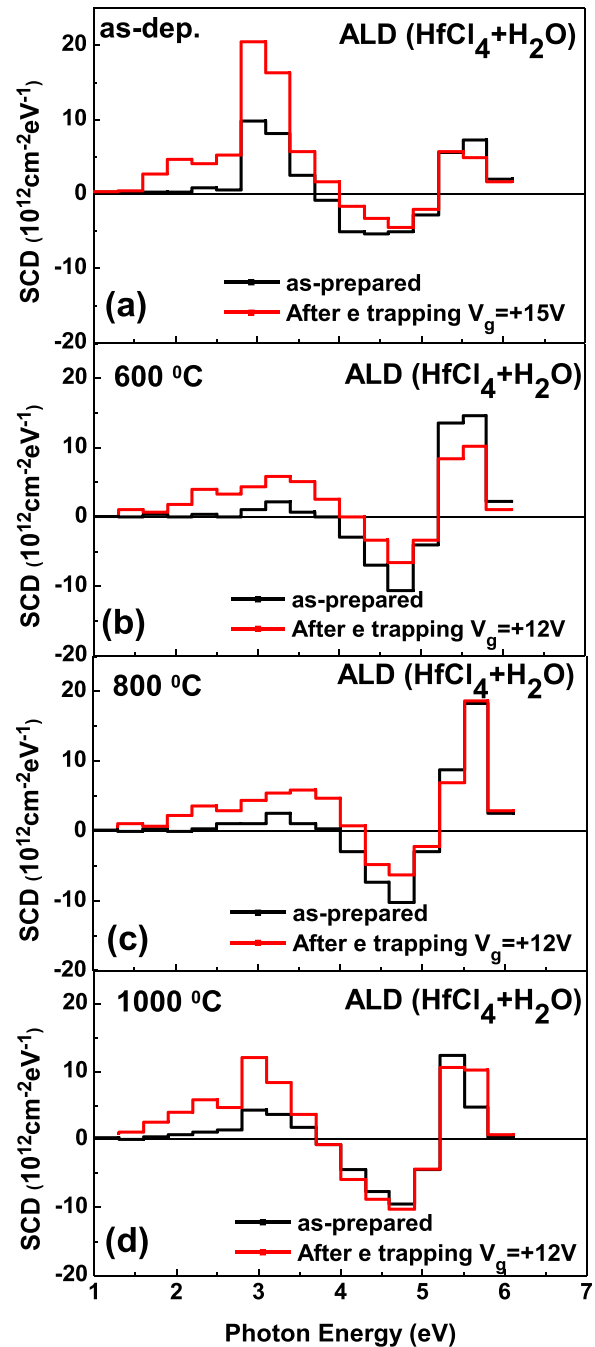


FIG. 3. SCD spectra as measured on $\text{Si}/\text{SiO}_2/\text{HfO}_2$ samples with HfO_2 layers deposited using the ALD process A and subjected to post-deposition 15 min annealing in N_2 at the indicated temperature.

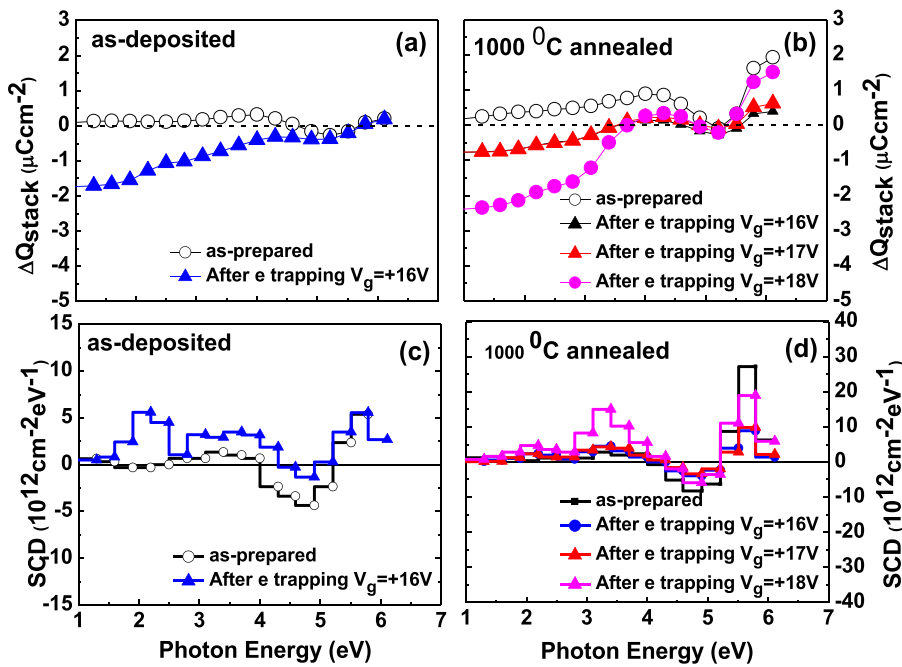


FIG. 4. Illumination-induced charge variation (a), (b) and the inferred SCD spectra (c), (d) as measured on Si/SiO₂/HfO₂ samples with HfO₂ layers deposited using the ALD process C and subjected to post-deposition 15 min annealing in N₂ at 1000 °C. The results of measurements on the electron-injected samples are shown for several charging voltage pulse amplitudes.

(Fig. 3, panel (a)) disappear upon annealing at 600 °C (panel (b)) and 800 °C (panel (c)) but do come back after the anneal at the highest temperature of 1000 °C (d). We also note that the density of positive charge accumulated in the stack upon electron-hole pair generation in HfO₂ ($h\nu > 5.6$ eV) significantly increases as compared to the as-deposited film. The same trends are seen even more clearly in sample type C (cf. Fig. 4) where both the generation of deep electron traps, which are nearly absent in the as-deposited film, and the additional hole trapping leading to the accumulation of net positive charge are observed after heating to 1000 °C. This led us to the hypothesis that the high-temperature (1000 °C) anneal of HfO₂ in the O-free (N₂) ambient may induce oxygen deficiency, prompting us to re-address the O vacancy model.

Since the as-deposited samples are partially crystallized (HfO₂, process A) or amorphous (Hf_{0.8}Al_{0.2}O_x, process A), while crystallization with a significant content of the monoclinic (m-) phase is achieved upon annealing, we consider the O vacancy energy levels both in m-HfO₂ and in a-HfO₂. Extensive theoretical studies of the O vacancy in m-HfO₂^{17–19} demonstrated that it is a deep hole trap resulting in the +1 or +2 charged states while the electron trapping by the neutral vacancy is expected to occur at a level positioned at ≈ 1 eV below the m-HfO₂ CB bottom. The calculated energy positions of electron and hole trapping levels are less reliable in a-HfO₂ because of severe underestimation of the bandgap^{20,21} suggesting to revisit this issue by performing additional *ab initio* simulations of the a-HfO₂ network with O vacancies included using a non-local density functional.

The details of the calculations can be found in the supplementary material.²⁷ The obtained a-HfO₂ structures have a density of 9.6 g/cm³ in good agreement with the experiment. Around 56% of the Hf atoms are bonded to 6 oxygen atoms, 38% are bonded to 7, and 6% to 5 oxygen atoms. On the other hand, 72% of oxygen atoms are bonded to 3 Hf atoms, 22% to 4, and 6% are bonded to 2 Hf atoms. These results are in good agreement with those of other theoretical studies^{29,30} and, importantly, do not depend significantly on the force-field used. The oxide bandgap width is close to

6 eV, to be compared to the 5.9 eV experimental values for a-HfO₂³¹ or a-HfAlO_x.³² Twenty five oxygen vacancies were created in the fully relaxed a-HfO₂ structure by randomly removing O atoms in accordance with the distribution of the number of nearest neighbour Hf ions described above. The structural relaxation and electronic structures were calculated for positively, neutral, and negatively charged states of these vacancies. The vacancy formation energies as a function of the electron chemical potential are plotted in Fig. 5 with the distribution of formation energies for each coordination number for neutral vacancies shown in an inset. Similar distributions were obtained for charged vacancies. These formation energies vary within about 1.5 eV due to the difference in the local environment and resulting relaxation of the amorphous structure, in good agreement with Ref. 21. The spread of the distribution of all formation energies is used in plotting “bands” in Fig. 5, which substitute lines in similar

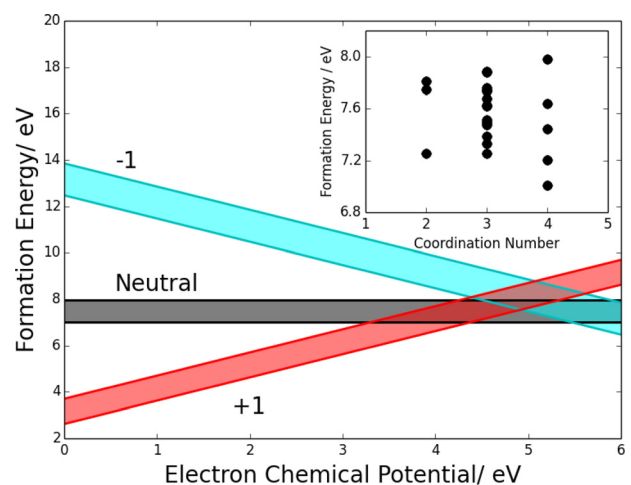


FIG. 5. Calculated formation energies of the singular O vacancy as a function of the electron chemical potential in a-HfO₂, with the origin corresponding to the VB maximum. Charge states are indicated next to the line plots and the width of the line corresponds to the span of formations energies. The inset shows the distribution of formation energies of vacancies as a function of the number of the nearest neighbour Hf ions.

plots for crystals. By using the full width of the distribution shown in the inset of Fig. 5, we assume that the vacancies are created in these sites with equal probability. In thermal equilibrium, vacancies will occupy sites with the lowest formation energies and the distribution will be narrower.

As one can see in Fig. 5, the $+0$ charge transition levels are located far from the VB top meaning that the formation of the positively charged vacancy by hole trapping from the a -HfO₂ VB would require a multi-phonon process. Nevertheless, these positively charged states can be formed in the vicinity of an electrode and may account for the interface barrier lowering caused by oxygen scavenging from HfO₂ by metallic Hf.³³ On the other hand, the $0/-$ levels are too shallow to explain the 2- and 3-eV deep states of the trapped electrons distribution. Thus, even after introducing the effect of disorder, the O-vacancy trapping cannot explain the experimental observations mandating the search for alternative models.

The key experimental observation in this study concerns the demonstration of the vast dominance of acceptor states in all studied HfO₂ samples. This abundance of deep acceptors is consistent with the literature reports^{34,35} on p-type conductivity of O-deficient HfO₂ though in our study the pristine ALD-grown films are stoichiometric within the accuracy of our measurements. Taking into account that the 2- and 3-eV deep acceptors found in HfO₂ cannot be correlated with the presence of typical precursor residuals (C, Cl), they are most likely related to intrinsic defects in the oxide network. Therefore, one can consider two types of candidate sites for the electron trapping: First, some form of O deficiency, different from the commonly assumed isolated O vacancy, may give rise to electron trapping. Clustering of vacancies³⁵ or their interaction with hydrogen, as recently revealed in the case of a -SiO₂,³⁶ represent feasible models deserving further theoretical and experimental analysis. Second, self-trapping of electrons on oxide network sites under strained bonding geometries may also provide deep acceptor states in the amorphous phases³⁷ or inside the grain boundary region in polycrystalline HfO₂. Considerable site-to-site variations in the trapping geometry expected in this case may explain the relatively wide energy distributions of the trapped electrons as well as the mentioned failure of ESR to detect the corresponding paramagnetic state because of excessive line broadening.

In conclusion, the presented EPDS methodology allows for determination of the energy distribution of electron traps across the entire HfO₂ bandgap. Using this technique, we found two kinds of deep electron traps in ALD-grown HfO₂ energetically distributed at around $E_t \approx 2.0$ eV and $E_t \approx 3.0$ eV below the oxide CB bottom. Comparison of the trapped electron energy distributions in HfO₂ layers prepared using different Hf precursors or subjected to post-deposition annealing suggests these centers to be intrinsic in nature. However, the hypothesis of the isolated O vacancy being at the origin cannot explain the charging behavior of a - or m -HfO₂, pointing to the need of alternative defect models.

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