

# Diffusion barriers block defect occupation on reduced CeO<sub>2</sub>(111)

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Figure S1 shows the clean and defective ceria (111) surfaces with the atom color scheme used in this work.

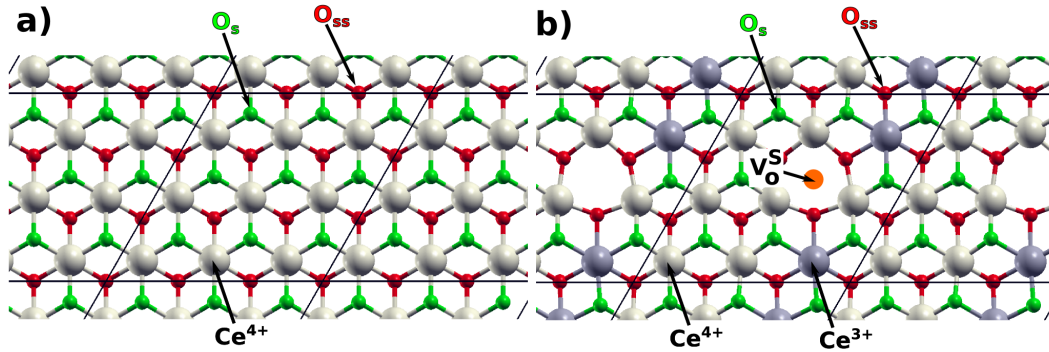


FIG. S1: a) Clean CeO<sub>2</sub> (111) surface and b) defective CeO<sub>2-x</sub> (111) surface with one surface oxygen vacancy ( $V_o^s$ ). The Ce<sup>4+</sup> and Ce<sup>3+</sup> ions and the surface and subsurface O atoms are depicted in white, gray, green and red, respectively.

## A. Au Atom Adsorption

Tables S1 and S2 show the calculated Au adsorption energy and its charge state, using the PBE+ $U$  approach and the HSE06 functional, for Au on various positions, and for different configurations of the Ce<sup>3+</sup> ions. In all the cases, a  $2 \times 2 \times 1$   $k$ -point mesh has been employed. With PBE+ $U$ , two cases have been considered: *i*) optimization of the Au atom coordinates, keeping the initially relaxed surfaces *frozen*, and *ii*) full geometry optimization, including Au and the atoms of the outermost oxide trilayer. With HSE, only full optimization has been carried out. The corresponding structures are shown in Figs. S2 and S3.

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TABLE S1: PBE+U and HSE06 adsorption energies (in eV) for one gold atom on the clean CeO<sub>2</sub> (111) surface. Also, the number of Ce<sup>3+</sup> ions and the Au oxidation state are shown. The symbol \* denotes unstable states.

Adsorption Site	PBE+U						HSE06		
	Frozen Surface <sup>a</sup>			1° Trilayer Relaxed			1° Trilayer Relaxed		
	No. Ce <sup>3+</sup>	Au State	E <sub>ads</sub> (eV)	No. Ce <sup>3+</sup>	Au State	E <sub>ads</sub> (eV)	No. Ce <sup>3+</sup>	Au State	E <sub>ads</sub> (eV)
O-bridge.a	0	Au <sup>0</sup>	-0.72	1	Au <sup>1+</sup>	-1.05	1	Au <sup>1+</sup>	-0.42
O-bridge.b	0	Au <sup>0</sup>	-0.72	1	Au <sup>1+</sup>	-0.97	-	-	-
O-bridge.c	0	Au <sup>0</sup>	-0.72	1	Au <sup>1+</sup>	-0.88	-	-	-
O-hollow	0	Au <sup>0</sup>	-0.60	1	Au <sup>1+</sup>	-0.90	1	Au <sup>1+</sup>	-0.29
O-top.a	0	Au <sup>0</sup>	-0.68	1	Au <sup>1+</sup>	-0.87	1	Au <sup>1+</sup>	-0.35
O-top.b	0	Au <sup>0</sup>	-0.68	1	Au <sup>1+</sup>	-0.84	1	Au <sup>1+</sup>	-0.35
O-top.c	0	Au <sup>0</sup>	-0.68	1	Au <sup>1+</sup>	-0.98	-	-	-
O-top.d	0	Au <sup>0</sup>	-0.68	1	Au <sup>1+</sup>	-0.98	-	-	-
Ce-top.a	-	-	-	0	Au <sup>0</sup>	-0.40	-	-	-
Ce-top.b	-	-	-	0	Au <sup>0</sup>	-0.40	-	-	-
OCe-bridge.a	0	Au <sup>0</sup>	-0.65	1	Au <sup>1+</sup>	-0.80	-	-	-
OCe-bridge.b	0	-	-	1	Au <sup>1+</sup>	-0.95	-	-	-
Ce-bridge	-	Au <sup>0</sup>	-0.72	1	Au <sup>1+</sup>	-0.80	-	-	-
O-dhollow	*	*	*	*	*	*	-	-	-

<sup>a</sup>The 1° trilayer was optimized before Au adsorption

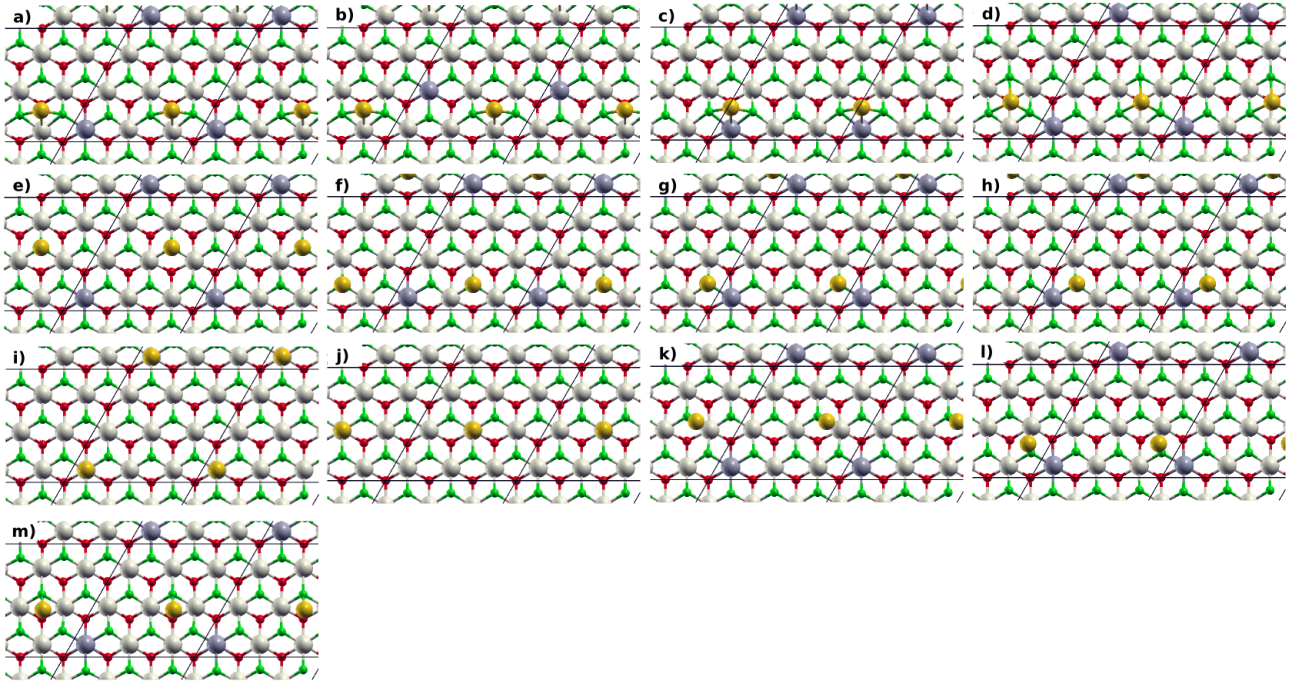


FIG. S2: Au adsorption configurations on CeO<sub>2</sub>(111): a) O-bridge.a, b) O-bridge.b, c) O-bridge.c, d) O-hollow, e) O-top.a, f) O-top.b, g) O-top.c, h) O-top.d, i) Ce-top.a, j) Ce-top.b, k) OCe-bridge.a, l) OCe-bridge.b, and and m) Ce-bridge.

TABLE S2: PBE+ $U$  and HSE06 adsorption energies (in eV) for one gold atom on  $\text{CeO}_{2-x}$  (111) with one surface oxygen vacancy. Also, the number of  $\text{Ce}^{3+}$  ions and the Au oxidation state are shown. The symbol \* denotes unstable states.

Adsorption Site	PBE+ $U$						HSE06		
	Frozen Surface <sup>a</sup>			1° Trilayer Relaxed			1° Trilayer Relaxed		
	No. $\text{Ce}^{3+}$	Au State	$E_{ads}$ (eV)	No. $\text{Ce}^{3+}$	Au State	$E_{ads}$ (eV)	No. $\text{Ce}^{3+}$	Au State	$E_{ads}$ (eV)
Ov-hollow.a	2	$\text{Au}^0$	-1.39	1	$\text{Au}^{1-}$	-2.37	1	$\text{Au}^{1-}$	-2.39
Ov-hollow.b	2	$\text{Au}^0$	-1.39	1	$\text{Au}^{1-}$	-2.29	-	-	-
Ov-hollow.c	2	$\text{Au}^0$	-1.39	1	$\text{Au}^{1-}$	-2.22	-	-	-
O-top.a	*	*	*	3	$\text{Au}^{1+}$	-0.97	3	$\text{Au}^{1+}$	-0.42
O-top.b	2	$\text{Au}^0$	-0.66	3	$\text{Au}^{1+}$	-0.89	-	-	-
O-top.c	*	*	*	3	$\text{Au}^{1+}$	-0.86	-	-	-
O-top.d	*	*	*	2	$\text{Au}^0$	-0.71	-	-	-
O-top.e	*	*	*	3	$\text{Au}^{1+}$	-0.70	-	-	-
O-top.f	*	*	*	3	$\text{Au}^{1+}$	-0.85	-	-	-
O-dhollow.a	2	$\text{Au}^0$	-0.90	2	$\text{Au}^0$	-1.06	2	$\text{Au}^0$	-0.80
O-dhollow.b	2	$\text{Au}^0$	-0.90	2	$\text{Au}^0$	-1.05	2	$\text{Au}^0$	-0.81
O-dhollow.c	2	$\text{Au}^0$	-0.89	2	$\text{Au}^0$	-0.95	-	-	-
O-dhollow.d	-	-	-	2	$\text{Au}^0$	-1.01	-	-	-
O-hollow	*	*	*	2	$\text{Au}^0$	-0.93	2	$\text{Au}^0$	-0.58
Ce-bridge	2	$\text{Au}^0$	-0.68	3	$\text{Au}^{1+}$	-0.87	-	-	-
O-bridge.a	2	$\text{Au}^0$	-0.68	3	$\text{Au}^{1+}$	-1.00	3	$\text{Au}^{1+}$	-0.32
O-bridge.b	*	*	*	3	$\text{Au}^{1+}$	-1.00	-	-	-
O-bridge.c	2	$\text{Au}^0$	-0.51	3	$\text{Au}^{1+}$	-0.95	-	-	-
O-bridge.d	*	*	*	3	$\text{Au}^{1+}$	-0.90	3	$\text{Au}^{1+}$	-0.20
O-bridge.e	2	$\text{Au}^0$	-0.67	3	$\text{Au}^{1+}$	-0.85	-	-	-
O-bridge.f	2	$\text{Au}^0$	-0.68	3	$\text{Au}^{1+}$	-0.83	-	-	-
O-bridge.g	*	*	*	3	$\text{Au}^{1+}$	-0.82	3	$\text{Au}^{1+}$	-0.11
O-bridge.h	*	*	*	3	$\text{Au}^{1+}$	-0.85	3	$\text{Au}^{1+}$	-0.11

<sup>a</sup>The 1° trilayer was optimized before Au adsorption

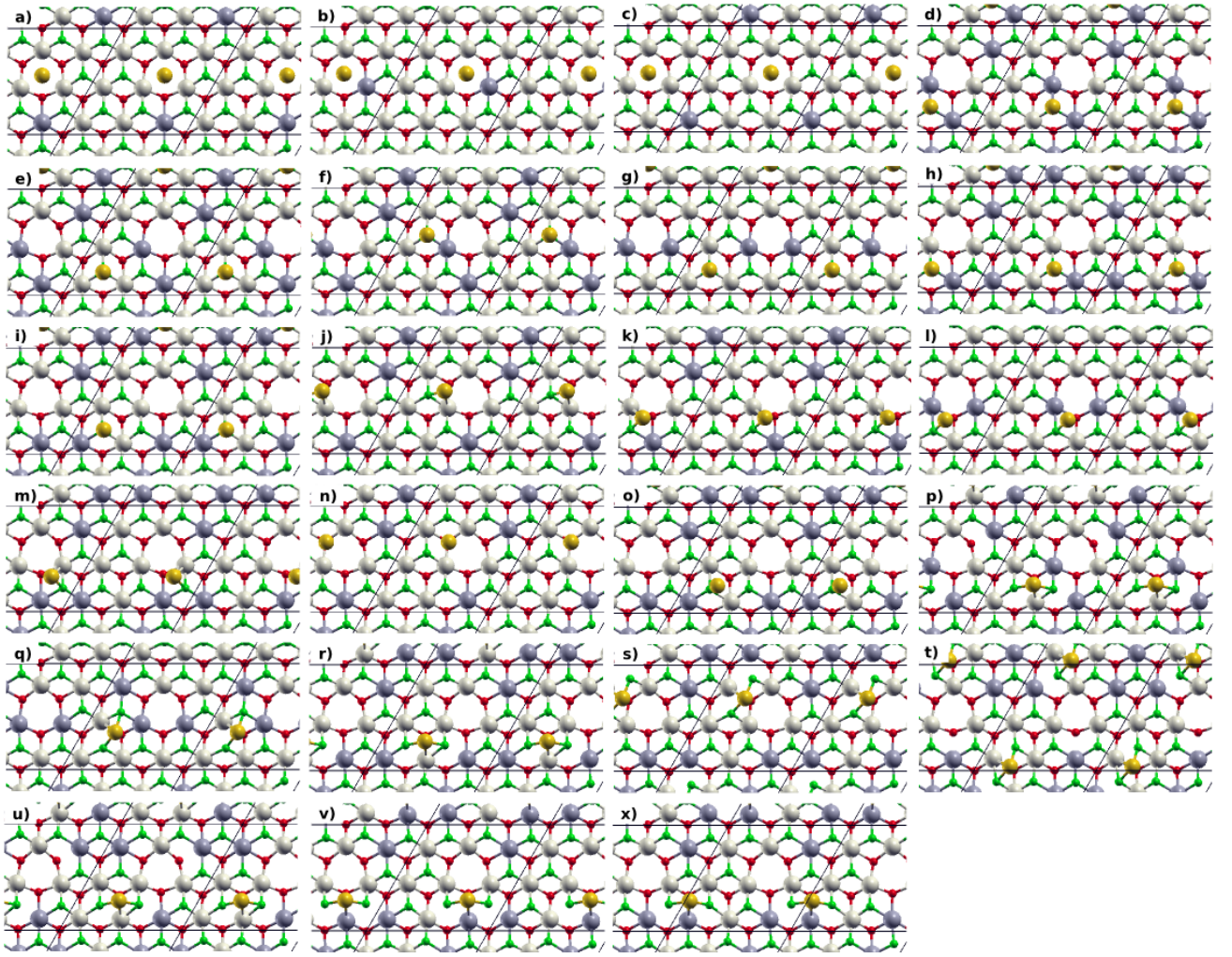


FIG. S3: Au adsorption configurations on  $\text{CeO}_{2-x}(111)$  with one surface oxygen vacancy: a)  $V_5^s$ -hollow.a, b)  $V_5^s$ -hollow.b, c)  $V_5^s$ -hollow.c, d) O-top.a, e) O-top.b, f) O-top.c, g) O-top.d, h) O-top.f, i) O-dhollow.a, j) O-dhollow.b, k) O-dhollow.c, l) O-dhollow.d, m) O-hollow, o) Ce-bridge, p) O-bridge.a, q) O-bridge.b, r) O-bridge.c, s) O-bridge.d, t) O-bridge.e, u) O-bridge.f, v) O-bridge.g, and x) O-bridge.h.

## B. Au Diffusion Pathways

### 1. Clean $\text{CeO}_2(111)$ Surface

The Au diffusion pathways on the clean  $\text{CeO}_2(111)$  surface are shown in Fig. S4. Two scenarios were considered, namely, adiabatic (red energy profile in the lower panel) and diabatic diffusion. For the adiabatic case, relaxations include the Au atom and the atoms of the outermost oxide trilayer. For the diabatic case, the ceria lattice and the Au charge state were kept *frozen* along the path, and two configurations were considered: *i*) diffusion on the pristine (relaxed) ceria surface (upper panel), and *ii*) diffusion on a surface that corresponds to the relaxed structure for Au adsorption on a particular site (blue, green, yellow, and magenta energy profiles in the lower panel).

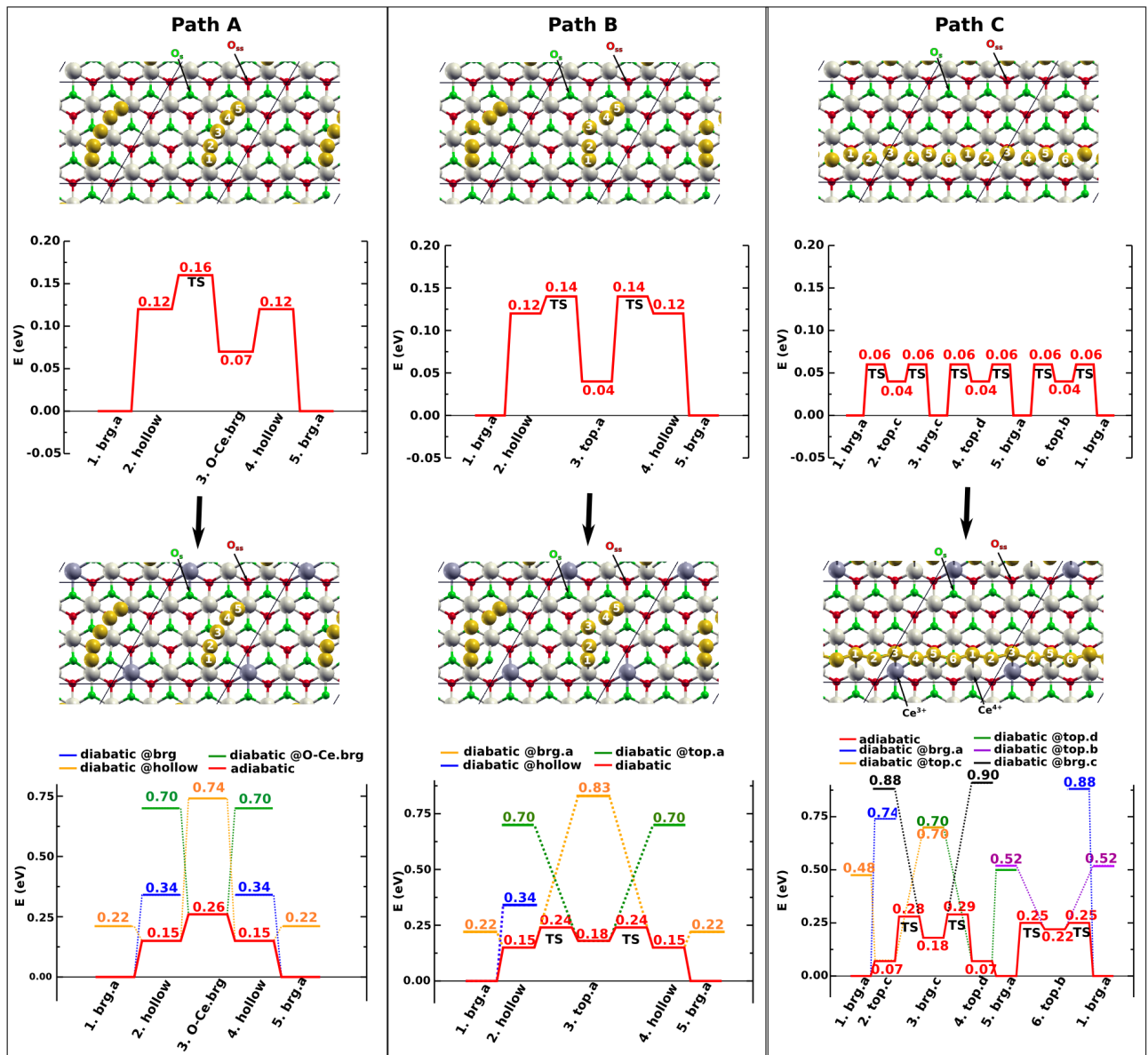


FIG. S4: Diffusion pathways for Au on  $\text{CeO}_2(111)$ : diabatic (upper panel) on the frozen pristine (relaxed) ceria surface, adiabatic (lower panel, red profile), and diabatic (lower panel, blue, green, yellow, and magenta profiles) on a frozen surface that corresponds to the relaxed structure for a particular Au adsorption site.

TABLE S3: Values of the imaginary frequency vibration of the transition states along the Diabatic and Adiabatic diffusion pathways shown in the upper and lower panel, respectively, of Fig. S4.

Clean Surface		
Reaction Pathway	Diabatic (Unrelax Surf.)	Adiabatic
path.A	$40 \text{ cm}^{-1}$	–
path.B	$39 \text{ cm}^{-1}$	$43 \text{ cm}^{-1}$
	$39 \text{ cm}^{-1}$	$43 \text{ cm}^{-1}$
path.C	$38 \text{ cm}^{-1}$	$49 \text{ cm}^{-1}$
	$38 \text{ cm}^{-1}$	$51 \text{ cm}^{-1}$
	$38 \text{ cm}^{-1}$	$46 \text{ cm}^{-1}$
	$38 \text{ cm}^{-1}$	$46 \text{ cm}^{-1}$

## 2. Defective $\text{CeO}_{2-x}(111)$ Surface

For the defective surface, adiabatic and diabatic diffusion pathways were considered. Figure S5 shows the diabatic path for Au diffusion on the frozen defective surface with an oxygen vacancy. Here, surface atoms were not optimized as the Au atom moved. For this reason, the oxidation state of Au is zero along the path. This path shows two transition states with energies of 0.06 and -0.16 eV that are characterized by having the same imaginary frequency of  $40 \text{ cm}^{-1}$ .

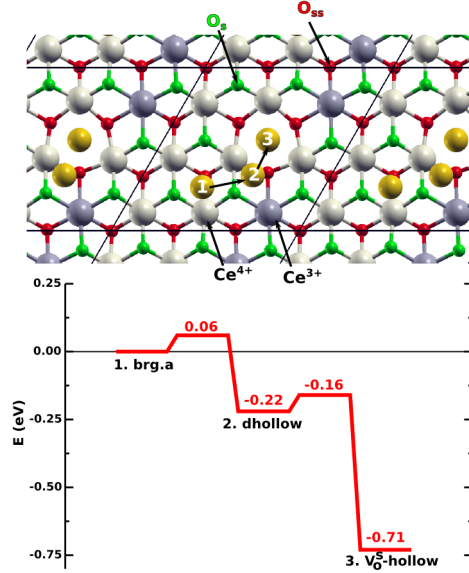


FIG. S5: Diabatic diffusion pathway for Au on the frozen (relaxed) ceria surface with one  $\text{V}_\text{O}^s$ .

Figure S6 shows energy profiles for five adiabatic diffusion pathways (red), and various diabatic pathways on surfaces that correspond to relaxed structures for Au adsorption on a particular site (blue, green, yellow, and magenta energy profiles). Path.A has a transition state (TS) with an energy of 0.23 eV and imaginary frequency of  $46 \text{ cm}^{-1}$  and for Path.B these values are 0.22 eV and  $46 \text{ cm}^{-1}$ , respectively. Path.C has two TS with an energy of 0.16 eV and 0.10 eV and imaginary frequencies of  $50 \text{ cm}^{-1}$  and  $41 \text{ cm}^{-1}$ , respectively. Path.D has a TS with an energy equal 0.31 eV and an imaginary frequency of  $42 \text{ cm}^{-1}$ . Finally, Path.E has six transition states with energies: 0.22, 0.12, 0.10, 0.00, 0.15 and 0.21 eV, characterized by an imaginary frequency of 47, 44, 42, 40, 45 and  $46 \text{ cm}^{-1}$ , respectively.

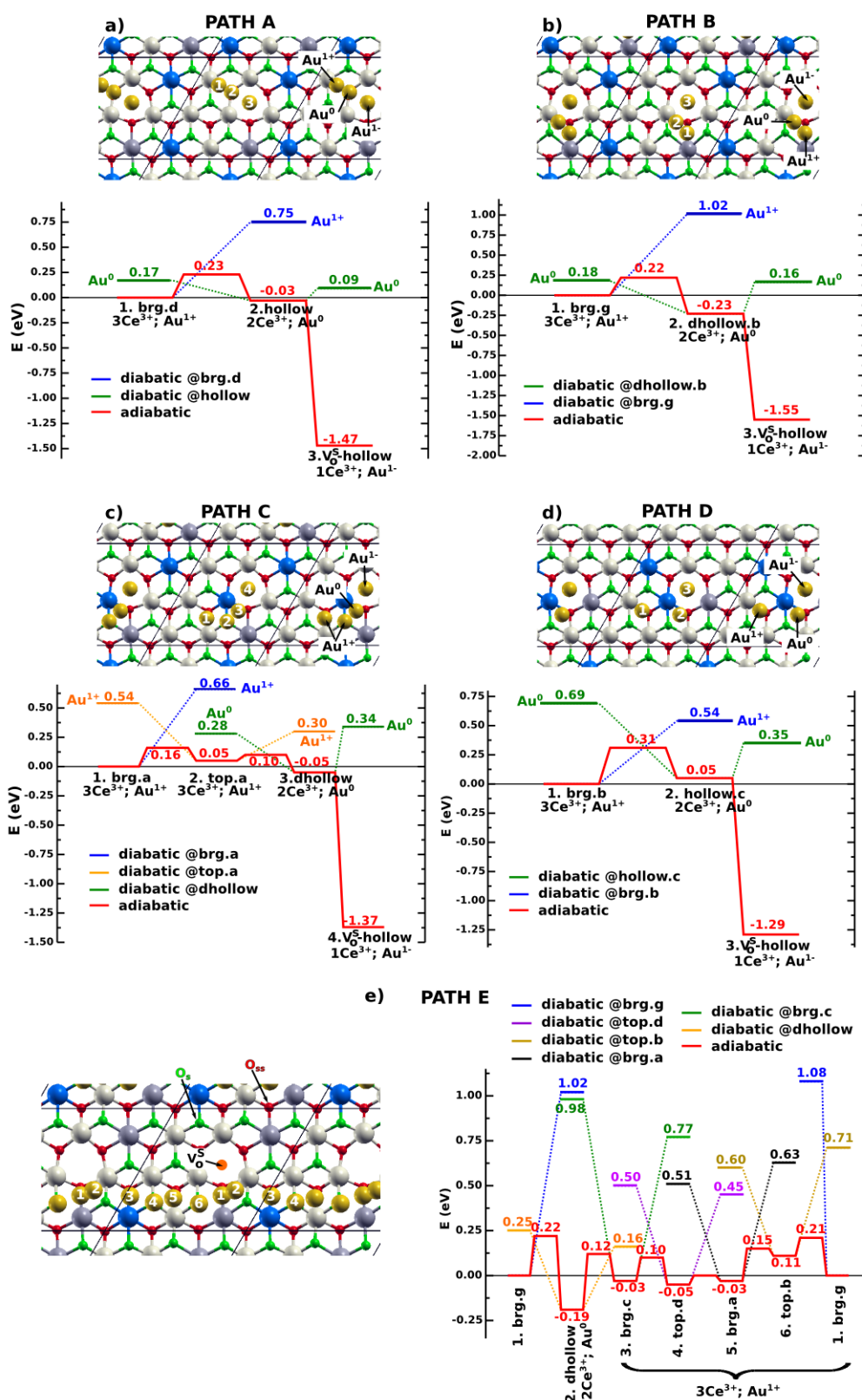


FIG. S6: Adiabatic and diabatic Au diffusion pathways on the defective  $\text{CeO}_{2-x}$ (111) surface with one oxygen vacancy,  $V_o^s$ .

TABLE S4: Values of the imaginary frequency vibration of the transition states along the adiabatic diffusion pathways shown in Fig. S6.

Defective Surface	
Reaction Pathway	Adiabatic
path.A	46 $\text{cm}^{-1}$
path.B	46 $\text{cm}^{-1}$
path.C	50 $\text{cm}^{-1}$
	41 $\text{cm}^{-1}$
path.D	42 $\text{cm}^{-1}$
	42 $\text{cm}^{-1}$
path.D	44 $\text{cm}^{-1}$
	47 $\text{cm}^{-1}$
	46 $\text{cm}^{-1}$
	45 $\text{cm}^{-1}$
	43 $\text{cm}^{-1}$