Atomic multiplet and charge transfer screening effects in 1s and 2p core-level x-ray photoelectron spectra of early 3d transition-metal oxides

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We present a comparative analysis of 1s and 2p core-level hard x-ray photoelectron spectroscopy (HAXPES) spectra of metallic VO₂ and CrO₂. Even though the V 1s and 2p spectra in VO₂ show similar line shapes except for the absence/presence of a spin-orbit splitting in the core levels, the Cr 1s and 2p spectra exhibit distinct main-line shapes. The experimental HAXPES spectra are analyzed by the Anderson impurity model based on the density functional theory + dynamical mean-field theory and a conventional MO₆ cluster model. We examine the interplay between the formation of the intra-atomic multiplet between a core hole and valence electrons and the charge-transfer effect on the chemical bonding followed by the 1s and 2p core electron excitations. We demonstrate the advantage of 1s HAXPES to the routinely employed 2p one for distinguishing contributions of a metal-ligand and metallic screening from a state near the Fermi level in metallic early 3d transition-metal oxides.

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I. INTRODUCTION

Transition metal oxides (TMOs) exhibit a variety of functional properties ranging from unconventional superconductivity with high $T_{\rm c}$ to exotic phase transitions, including metal-insulator transitions [1,2]. These properties often arise from the complex interplay between local electron-electron interaction and interatomic hybridization. The formation of intra-atomic multiplet introduces additional complexity in multiorbital cases, resulting in nontrivial correlated states, even in metallic systems, such as Hund's metal [3–5]. Corelevel x-ray photoelectron spectroscopy (XPS) is a powerful tool to investigate TMOs [6]. In XPS, a highly localized core hole created by x-ray irradiation strongly interacts with the d electrons on the same transition metal (TM) ion, leading to atomic multiplet fine structures in the XPS spectrum. Besides, the sudden appearance of the core hole induces a dynamical charge response from surrounding ions to the excited ion, traditionally called charge-transfer (CT) screening [7–9]. The CT screening often generates satellite peaks in the core-level spectra, providing valued insights into the chemical bonding properties. The development of hard x-ray photoelectron spectroscopy (HAXPES) has enabled the exploration of bulk properties and deeper core levels of materials compared to conventional soft x-rays [6,10–14].

Recently, TM 1s HAXPES spectra were examined for insulating late (NiO, Fe₂O₃) [9,15] and early 3d TMOs (TiO₂, SrTiO₃) [16–18]. For the band-insulating Ti oxides with a d^0 configuration in the formal valence, Ti 1s HAXPES revealed low-energy satellites that are not seen in the Ti $2p_{3/2}$ one, since the satellites are positioned at the same binding energies E_B as the Ti $2p_{1/2}$ spin-orbit-partner. In Fe₂O₃ (Mott insulator) and NiO (CT insulator), the 1s spectra clearly exhibit CT screening features, including a nonlocal CT channel from the distant Ni or Fe ions beyond the nearest-neighboring ligands [7,8,11,19], despite a larger lifetime broadening than the 2p core-level excitation [9,15]. These findings suggest the 1s core-level excitation as a complementary tool to the routinely used 2p excitation for studying 3d TMOs. However, the previous studies have exclusively focused on the insulators. In metallic TMOs, 2p HAXPES spectra often exhibit a metallic nonlocal screening feature (edge singularity) from a conducting state at the Fermi level, providing insights into the correlated metallic properties [11,20-28]. Yet, the metallic screening effect in 1s spectra has not been explored.

Another aspect in the 1s core-level excitation is the absence of the core-valence Coulomb multiplet effect, which would eliminate fine features and thus allow an easy access to the valence and chemical bonding information. However, this potential benefit of 1s HAXPES is not emphasized in the previous studies, primarily because the TM 3d shell is almost empty in SrTiO₃ and TiO₂, and largely occupied in NiO. In such cases, the core-valence multiplet effect is weak in the 2p XPS spectra. Thus, survey for other TM oxides with rich core-valence multiplet effect in the 2p spectra will be valuable to examine the benefit of the multiplet-free 1s excitation.

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In this study, we present a systematic comparison of 1sand 2p HAXPES spectra of metallic vanadium dioxide (VO₂) and chromium dioxide (CrO_2) . We examine the sensitivity of the 1s excitation to the metallic nonlocal screening and demonstrate difference in the 1s and 2p HAXPES spectra, where the core-valence atomic multiplet effect is not negligible in the 2p excitation.VO₂ exhibits a metallic behavior in the high-temperature rutile structure phase. The V 2p HAXPES spectrum displays a shoulder feature associated with the V $2p_{3/2}$ main line due to a metallic CT screening [23–25]. In the previous studies [23,24], the V 1s HAXPES spectrum exhibits a similar shoulder feature. CrO₂ is a half-metallic compound. Sperlich et al. [29] reported Cr 2p HAXPES of CrO₂, revealing a characteristic double-peak structure in the Cr $2p_{3/2}$ main line. The low- E_B feature in the double-peak structure is suppressed in a surface-sensitive soft x-ray spectrum, suggesting that a metallic CT screening is active in the 2p excitation. In this study, we conduct a HAXPES measurement of the Cr 1s core-level in CrO₂, and compare the 1s and 2p spectra. Though both oxides are textbook examples of correlated metals, recent advances in, e.g., heterostructure and thin-film engineering have promoted renewed interest in their electronic structure and core-level HAXPES spectra [30–33].

We simulate the experimental spectra using the Anderson impurity model (AIM) based on the local density approximation (LDA) + dynamical mean-field theory (DMFT) and conventional MO₆ cluster model. The LDA+DMFT method can be viewed as an extension of the MO₆ cluster model for computing the core-level XPS spectra. In this approach, the discrete ligand orbitals of the cluster model are replaced by a continuous metallic bath optimized within the LDA+DMFT scheme, which contains information about the entire lattice. Thus, the LDA+DMFT AIM represents the CT screening responses, including the nonlocal CT channels [9,19,34]. The cluster model considers only the CT screening from the nearest-neighboring oxygens, traditionally called local screening [6]. Thus, a comparison of the two models enables us to assess the contributions of the metallic nonlocal screening in the 1s and 2p spectra. The impurity model description allows us to consider the local core-valence multiplet interaction explicitly.

II. METHODS

The Cr 1s HAXPES of CrO₂ was measured at the HIKE station of the KMC-1 beamline at BESSY. A photon energy of 8.95 keV was used, and the energy resolution is \sim 0.5 eV [9]. The CrO₂ sample was a commercially available crystalline CrO₂ powder (Magtrieve) [35]. The powder was ground using a mortar and pressed into a pill. The V 1s HAXPES spectrum of VO₂ was reproduced from the previous study [23].

The simulation of Cr (V) 1s and 2p HAXPES for CrO₂ (VO₂) starts with a standard LDA+DMFT calculation [36–38]. Both CrO₂ and VO₂ (metallic phase) crystallize in a rutile structure with space group $P4_2/mnm$ (No. 136). The experimental lattice parameters of a = 4.421 Å(4.556 Å) and c = 2.916 Å(2.859 Å) in CrO₂ (VO₂) [39,40] are adopted in this work. We perform an LDA calculation for the experimental crystal structure and subsequently construct a tight-binding model spanning both Cr (V) 3d and O 2p states from the LDA

bands using Wien2K and wannier90 packages [41-43]. The tight-binding model is augmented with local electron-electron interaction within the TM 3d shell that is determined by Hubbard U and Hund's J parameters. We set these values to (U, J) = (5.0 eV, 1.0 eV) and (6.0 eV, 1.0 eV) for the Cr and V cases, respectively, consulting with previous density functional theory (DFT)-based and spectroscopy studies for chromium and vanadium oxides [44,45]. The strong-coupling continuous-time quantum Monte Carlo (CT-QMC) impurity solver [46-49] with density-density approximation to the on-site electron-electron interaction is used to obtain the selfenergies $\Sigma(i\omega_n)$ of Cr (V) 3d electrons from the AIM. The double-counting correction μ_{dc} , which is introduced to subtract the d-d interaction effect present already in LDA step [37,50], is treated as an adjustable parameter and fixed to reproduce the experimental valence and core-level spectra. The μ_{dc} dependence of the calculated spectra can be found in the Supplementary Material (SM) [51]. To compute valence spectra and hybridization densities $\Delta(\varepsilon)$ on the real-frequency axis, analytically continued $\Sigma(\varepsilon)$ in the real-frequency domain is obtained using the maximum entropy method [52]. We do not implement the charge self-consistency (updating the DFT charge) in the present LDA+DMFT calculation. Experimentally, VO₂ is a paramagnetic metal in a high-temperature phase [53] and CrO₂ is a ferromagnetic metal with a Curie temperature of approximately 390 K [54]. Thus, the calculations are performed for the paramagnetic phase for VO₂ and the ferromagnetic phase for CrO₂.

The Cr (V) 1s and 2p HAXPES spectra are computed from the AIM implementing the DMFT hybridization densities $\Delta(\varepsilon)$, following methods detailed in Refs. [9,19,55]. In this step, the AIM is extended to include the core orbitals and their interaction with the valence electrons. Though the core-valence interaction value U_{dc} is an adjustable parameter in this approach, an empirical relation $U_{dc} = 1.2 \sim 1.3 \times U_{dd}$ is well established for 3d TMOs, where U_{dd} represents the configuration-averaged interaction on the TM 3d shell [9,19]. In SM [51], we verified that varying U_{dc} within this range does not affect the spectra quantitatively, and thus it is set to $U_{dc} = 1.2 \times U_{dd}$ in the main result. The multipole term in the core-valence interaction is parameterized by the higher-order Slater integrals obtained from the atomic Hartree-Fock code. The atomic Slater integral values are reduced to 80% from the bare values to simulate the effect of intraatomic configuration interaction from higher basis configurations neglected in the atomic calculation. This treatment was successfully applied in modeling XPS as well as core-level x-ray absorption spectra in a wide range of 3d TM compounds [56–59]. For CrO₂, the simulations implementing different scaling factors (R_{2p-3d}) to the bare values are performed. The small spin-orbit coupling within the TM 3d shell ($\xi_{soc} \sim 0.04$ eV) is ignored in the LDA+DMFT self-consistent calculations to avoid complex off-diagonal elements in $\Delta(\varepsilon)$ in the CT-QMC calculation for the AIM, while it is considered explicitly when computing the core-level HAXPES spectra. The spin-orbit coupling in the TM 2p shell is also considered, and the spin-orbit coupling constants are taken from the atomic calculation [9, 19].

We employ the configuration-interaction (CI) solver for the AIM with the same implementation as described in Refs. [19,55,60]. In the MO_6 cluster-model calculation, the



FIG. 1. The LDA+DMFT AIM (blue) and the VO₆ cluster model (red) calculations for (a) V 1s XPS and (b) V 2p XPS spectra for metallic VO₂ in the high-temperature rutile structure phase. The experimental HAXPES data are taken from Ref. [23]. The spectral intensities are convoluted with a Gaussian of 200 meV and a Lorentzian of 650 meV in the half width at half maximum (HWHM) for both 1s and 2p spectra. The main line (ML) and the satellite (ST) are indicated.

electron hopping amplitudes with the nearest-neighboring ligands are extracted from the tight-binding model constructed from the LDA bands mentioned above and summarized in Appendix B. The cluster model implements the same local Hamiltonian for the x-ray excited Cr (V) ion with the LDA+DMFT AIM. As noted in Refs. [18,55], care must be taken for the number of electronic configurations included to represent the XPS initial and final states within the CI scheme for a highly covalent system. We checked the convergence of the XPS spectral intensities as a function of the number of configurations in the studied compounds in Appendix B.

III. RESULTS

Figure 1 shows the V 1s and 2p XPS spectra of metallic VO₂ calculated by the LDA+DMFT AIM method and the VO₆ cluster model. The experimental 1s and 2p XPS data are reproduced from Ref. [23]. The LDA+DMFT valenceband spectra can be found in Appendix A. Apart from the presence of the V $2p_{1/2}$ spin-orbit component and a different lifetime broadening, the V 1s and $2p_{3/2}$ experimental spectra exhibit similar line shapes. A satellite (ST) feature is located



FIG. 2. V 1s XPS spectra simulated by the VO₆ cluster model with varying metal-ligand (V 3d-O 2p) hybridization strength. 100% corresponds to the hybridization strength derived from the LDA bands for the experimental rutile structure of VO₂ in Appendix B. The spectral intensities are convoluted with a Gaussian of 100 meV and a Lorentzian of 100 meV (HWHM).

approximately 13 eV above the main line (ML), as indicated by a dashed line.

The VO₆ cluster model represents the overall features in the V 1s and 2p spectra. In Fig. 2, the 1s spectra computed with varying the metal-ligand hybridization in the cluster model are shown. The diagonal energies of the electronic configurations in the final states are listed in Table I; see Appendix B for those in the initial state. Note that the screened $|\underline{c}d^{2}\underline{L}\rangle$ configuration has an energy lower than that of $|\underline{c}d^{1}\rangle$ due to the core-valence interaction U_{dc} for an excess d electron. The effective hybridization V_{eff} with the ligands can be given as $V_{eff} = [(4 - N_{e_g}) \times V_{e_g}^2 + (6 - N_{t_{2g}}) \times V_{t_{2g}}^2]^{1/2}$, where $N_{e_g} (N_{t_{2g}})$ and $V_{e_g} (V_{t_{2g}})$ denote occupation of the $e_g (t_{2g})$ states and hybridization strength of the ligand and $e_g (t_{2g})$ orbitals [6,9,18,61]. For VO₂ (with $N_{t_{2g}} = 1$ and $N_{e_g} = 0$) in a formal valence state, V_{eff} accounts for 9.1 eV (Appendix B). The large hybridization leads to well-split bondinglike and antibondinglike states of the $|\underline{c}d^{1}\rangle$ and $|\underline{c}d^{2}\underline{L}\rangle$ states, forming the

TABLE I. The configuration (diagonal) energies in the V 1s XPS final states within the VO₆ cluster-model Hamiltonian in eV. Here, the charge-transfer energy is defined for a V⁴⁺ (d^1) electronic configuration as $\Delta_{\text{CT}} = E(d^2\underline{L}) - E(d^1) = \varepsilon_d - \varepsilon_L + U_{dd}$, where ε_d (ε_L) denotes the averaged V 3d (ligand 2p) energy. The values of charge-transfer energy Δ_{CT} , 3d - 3d interaction U_{dd} , and 2p-3d core-valence interaction U_{dc} are found in Appendix B.

Conf.	Diagonal energies	Value		
$ cd^1\rangle$	0	0.0		
$ \underline{c}d^2\underline{L}^1\rangle$	$\Delta_{ m CT} - U_{dc}$	-4.7		
$ \underline{c}d^3\underline{L}^2\rangle$	$2\Delta_{\mathrm{CT}} + U_{dd} - 2U_{dc}$	-3.8		
$ \underline{c}d^{4}\underline{L}^{3}\rangle$	$3\Delta_{\rm CT} + 3U_{dd} - 3U_{dc}$	2.7		
$ cd^5L^4\rangle$	$4\Delta_{\rm CT}+6U_{dd}-4U_{dc}$	14.7		
$ cd^6L^5\rangle$	$5\Delta_{\rm CT} + 10U_{dd} - 5U_{dc}$	32.3		
$ \underline{c}d^7\underline{L}^6\rangle$	$6\Delta_{\mathrm{CT}} + 15U_{dd} - 6U_{dc}$	55.4		



FIG. 3. The experimental HAXPES, the LDA+DMFT AIM (blue), and the CrO_6 cluster-model (red) calculations for (a) Cr 1s XPS and (b) Cr 2p XPS spectra. The spectral intensities are convoluted with a Gaussian and a Lorentzian of 300 and 250 meV (HWHM), respectively, for both 1s and 2p spectra. The experimental Cr 2p HAXPES data are taken from Ref. [29]. The main line (ML) and the satellite (ST) are indicated.

main line and satellite, respectively. Similar to Ti oxides [18], higher configurations up to $|\underline{c}d^5\underline{L}^4\rangle$ are necessary for accurate description of the satellites, as demonstrated in Appendix B.

Both the V 1s and $2p_{3/2}$ main lines exhibit a shoulder feature originating from metallic nonlocal CT screening at $E_B = 5467 \text{ eV}$ in 1s and $E_B = 514 \text{ eV}$ in $2p_{3/2}$, as indicated by the dotted lines in Fig. 1. The LDA+DMFT AIM reproduces the shoulder feature, while it is missing in the cluster-model result, as expected given that the former model includes a metallic CT screening, while the latter model considers only the metal-ligand one. Furthermore, the LDA+DMFT AIM yields a broad band feature around $E_B = 5472 \text{ eV}$ in the 1s spectrum, which shares similarities with the low- E_B satellites in Ti 1s HAXPES spectra of TiO₂ and SrTiO₃ [16–18]. Due to its overlap with the broad tail of the V $2p_{1/2}$, this feature is not clearly seen in the V $2p_{3/2}$ spectrum [Fig. 1(b)].

Next, we examine ferromagnetic metallic CrO₂. Figure 3 presents the Cr 1*s* and 2*p* XPS spectra calculated by the LDA+DMFT AIM and the CrO₆ cluster model. The Cr 2*p* experimental HAXPES spectrum is taken from Ref. [29]. The LDA+DMFT valence-band spectra can be found in Appendix A. The Cr 2*p*_{3/2} main line ($E_B \sim 576$ eV) exhibits the double-peak feature. Interestingly, the Cr 1*s* main line



FIG. 4. (a) Cr 2*p* XPS spectral intensities calculated by the LDA+DMFT AIM method with reducing strength of the 2*p*-3*d* core-valence Coulomb multiplet interaction from the ionic values. (b) The calculated spectra without the spin-orbit coupling ($\xi_{SOC}^{2p} = 0$) on the 2*p* core orbitals. The spectral intensities are convoluted with a Gaussian and a Lorentzian of 300 and 250 meV (HWHM), respectively.

 $(E_B \sim 5968.5 \text{ eV})$ shows an asymmetric line shape, but not a double-peak feature as in the Cr $2p_{3/2}$ spectrum. This is a notable contrast with VO₂ (Fig. 1), as the V 1s and $2p_{3/2}$ spectra resemble each other. The LDA+DMFT results reproduce both the Cr 1s and 2p spectra reasonably well. The cluster model incorrectly yields a symmetric single-peak 1s main line and does not reproduce the double-peak feature in the Cr $2p_{3/2}$ main line (Fig. 3). This suggests the presence of metallic CT contributions in both the Cr 1s and 2p spectra of CrO₂ as in the VO₂ case. A detailed analysis of the spectra within the CrO₆ cluster model can be found in Appendix C.

To get insight into the different shapes in the 1s and 2p main line in CrO₂, we calculate the 2p spectra with modulating the multiplet coupling between the Cr 2p and 3d electrons in Fig. 4(a). Here, R_{2p-3d} denotes a scaling factor for the multipole part in the Cr 2p-3d interaction in the LDA+DMFT AIM. The 2p-3d core-valence multiplet interaction significantly influences the $2p_{3/2}$ main-line shape. The 2p spectrum with $R_{2p-3d} = 0\%$ closely resembles the Cr 1s spectrum in Fig. 3(a). This indicates that the CT screening to the x-ray excited Cr from the rest of the crystal is essentially the same in the 1s and 2p excitations, which is not surprising, given that the hybridization strength of the Cr 3d states with the electron bath is determined in the valence electronic structure

when a core hole is absent. However, the 2p-3d multiplet interaction at the x-ray excited Cr site mixes the screened states via different CT channels substantially, and consequently the intrinsic CT spectral peaks are distributed in a complex way in the 2p spectrum in the case of CrO₂. The 1*s* excitation, with negligibly weak core-valence interaction, is thus better suited for examining and quantifying the CT effect.

In the 1s spectrum, the prominent peak at the low- E_B side (~5968.5 eV, dashed line) dominantly consists of the metallic-screening states, while the high- E_B feature (~5969.5 eV, solid bar) contains mostly the local-screening states from ligands. Only the latter CT channel can be captured in the CrO₆ cluster model [see Fig. 3(a)]. The LDA+DMFT AIM result indicates that the intrinsic spectral intensity of the metallic screening is substantial compared to the local screening one upon the core-level excitation, manifesting in the observed asymmetric line shape in the experimental 1s data. However, the intensities of the two CT features are largely modulated due to the strong 2p-3d multiplet interaction in the 2p spectrum. Thus, the double peaks in the $2p_{3/2}$ spectrum do not serve as a good measure of the two CT channels.

Figure 4(b) shows the 2*p* XPS spectra calculated with eliminating the spin-orbit interaction on the Cr 2*p* shell in the LDA+DMFT AIM Hamiltonian. The spin-orbit interaction plays a marginal role for the $2p_{3/2}$ main-line shape and does not affect the spectra in the absence of the 2p-3*d* interaction ($R_{2p-3d} = 0\%$). Thus, the 2*p*-3*d* core-valence multiplet interaction is the decisive factor for the difference in the Cr 1*s* and 2*p* spectra.

IV. CONCLUDING REMARKS

In this study, we have analyzed 1s and 2p core-level HAX-PES spectra of metallic VO₂ and CrO₂, archetypal correlated metallic 3d transition-metal oxides. We conducted a HAXPES measurement for the 1s core level of CrO₂. While the V 1s spectrum in VO₂ closely resembles the $2p_{3/2}$ spectrum in the literature, we find the Cr 1s main line in CrO₂, exhibiting asymmetric line shape, substantially differs from the Cr $2p_{3/2}$ main line with the double peaks. Through a comprehensive computational analysis for the 1s and 2p spectra of VO_2 and CrO₂ using the LDA+DMFT AIM method and conventional MO₆ cluster model, we demonstrated that the difference in the CrO₂ and VO₂ spectra originates from a strong corevalence atomic multiplet effect in the 2p excitation of CrO₂. The double peaks in the Cr $2p_{3/2}$ main line do not directly represent either of the metallic-screening or metal-ligand CT final states, as these CT final states are mixed by the multiplet interaction acting locally at the x-ray excited Cr site. The core-valence multiplet-free 1s spectrum unveils intrinsic spectral features of the metallic and metal-ligand CT screenings upon the core-level excitation. Our LDA+DMFT AIM calculation for the 1s spectra reveals that the spectral intensity of the metallic screening dominates that of the metal-ligand screening, highlighting the advantage of 1s excitation in characterizing correlated metallic 3d TMOs, especially when the atomic multiplet effect is not negligible in the 2p spectra. Future systematic investigations of the 1s HAXPES spectra



FIG. 5. The LDA+DMFT valence spectral intensities for metallic bulk VO₂. The experimental valence-band XPS spectrum measured at 320 K is taken from Ref. [23].

for 3d TMOs with the active 2p-3d multiplets, such as Co or Mn oxides, would be of interest.

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APPENDIX A: LDA+DMFT VALENCE SPECTRA

Figure 5 shows valence-band spectra of metallic VO₂ calculated by the LDA+DMFT method, together with the experimental valence-band XPS spectrum taken from Ref. [23]. As previously discussed in Ref. [62], while the single-site DMFT lacks the V-V intersite self-energy needed for describing the dimerized insulating phase with a monoclinic structure, it provides a reasonable description for the correlated metallic phase of the high-temperature rutile-structure phase. Our DMFT valence-band spectrum is similar to the cluster DMFT



FIG. 6. The LDA+DMFT valence spectral intensities for bulk CrO₂. The experimental valence-band XPS spectrum (dashed) measured at 150 K is taken from Ref. [29].

TABLE II. The parameter values for (configuration-averaged) 3d-3d (U_{dd}) and 2p-3d (U_{dc}) interactions and the metal-ligand hybridization amplitude adopted in the VO₆ and CrO₆ cluster models in eV unit.

	U_{dd}	U_{dc}	$V_{B_{1g}}$	V_{A_g}	$V_{A_g'}$	$V_{B_{2g}}$	$V_{B_{3g}}$
VO ₂	5.56	6.67	3.80	3.83	2.08	2.10	2.23
CrO ₂	4.56	5.47	3.58	3.72	1.93	1.70	1.90

result for the metallic phase [45,62]. The LDA+DMFT spectrum exhibits a lower Hubbard-band feature around -1 eVand a quasiparticle feature near the Fermi energy E_F . The calculated binding energies E_B of the O 2p states with respect to the V 3*d* states match well with the experimental data, supporting the used value of the double-counting correction μ_{dc} in the LDA+DMFT calculation. In Fig. 6, the LDA+DMFT valence-band spectra of CrO₂ together with the experimental valence-band XPS spectrum taken from Ref. [29] is also shown. The double-counting dependence of the valence spectra of VO₂ and CrO₂ can be found in the SM [51].

APPENDIX B: MODEL PARAMETERS IN THE CLUSTER-MODEL CALCULATION

Table II summarizes the parameter values employed in our cluster-model calculations for VO₂ and CrO₂. The interaction parameters U_{dd} and U_{dc} are set to the same values as the ones in the LDA+DMFT AIM Hamiltonian for VO₂ and CrO₂. The one-particle parameters, i.e., electron hopping amplitudes and crystal-field energies, are derived from the tight-binding models constructed from the DFT bands for the experimental structures.

The hopping amplitudes between the V (Cr) *d* orbitals $[B_{1g}(xy), A_g(3z^2 - r^2), A'_g(x^2 - y^2), B_{2g}(zx)$ and $B_{3g}(yz)]$ and the molecular orbitals consisting of the nearest-neighboring ligand 2p orbitals with the same symmetry are given in Table II. The charge-transfer energy Δ_{CT} is taken from Refs. [63,64]. Table III summarizes the diagonal energies of the electronic configurations considered in the VO₆ and CrO₆ cluster model accounting for the interaction U_{dd} and the charge-transfer energy Δ_{CT} up to the $|d^7L^6\rangle$ configuration. Figure 7 shows the V and Cr 1s XPS spectra calculated by

TABLE III. Configuration (diagonal) energies in the initial states of VO_6 (left) and CrO_6 (right) cluster models in eV unit.

Conf.	VO ₂	Val.	Conf.	CrO ₂	Value
$ d^1\rangle$	0	0.0	$ d^2\rangle$	0	0.0
$ d^2 \underline{L}^1\rangle$	Δ_{CT}	2.0	$ d^3\underline{L}^1\rangle$	$\Delta_{\rm CT}$	1.0
$ d^3\underline{L}^2\rangle$	$2\Delta_{\mathrm{CT}} + U_{dd}$	9.6	$ d^4 \underline{L}^2 \rangle$	$2\Delta_{\rm CT} + U_{dd}$	6.6
$ d^4\underline{L}^3\rangle$	$3\Delta_{\rm CT} + 3U_{dd}$	22.7	$ d^5\underline{L}^3\rangle$	$3\Delta_{\rm CT} + 3U_{dd}$	16.7
$ d^5\underline{L}^4\rangle$	$4\Delta_{\rm CT} + 6U_{dd}$	41.4	$ d^{6}\underline{L}^{4}\rangle$	$4\Delta_{\rm CT} + 6U_{dd}$	31.4
$ d^{6}\underline{L}^{5}\rangle$	$5\Delta_{\rm CT} + 10U_{dd}$	65.6	$ d^7 \underline{L}^5\rangle$	$5\Delta_{\rm CT} + 10U_{dd}$	50.6
$ d^7 \underline{L}^6\rangle$	$6\Delta_{\rm CT} + 15U_{dd}$	95.4	$ d^{8}\underline{L}^{6}\rangle$	$6\Delta_{\rm CT} + 15U_{dd}$	74.4

TABLE IV. Weights of the configurations in the ground-state wave function of VO₂ (upper) and CrO₂ (bottom) in the VO₆ and CrO₆ cluster-model simulations for different basis sets (3, 4, 5, 6, and 7 configurations in the basis expansions).

VO_2	3conf.	4conf.	5conf.	6conf.	7conf.
$ d^1\rangle$	0.317	0.248	0.243	0.237	0.237
$ d^2 \underline{L}^1\rangle$	0.522	0.487	0.485	0.477	0.477
$ d^3\underline{L}^2\rangle$	0.161	0.232	0.232	0.240	0.240
$ d^4\underline{L}^3\rangle$	_	0.033	0.038	0.043	0.043
$ d^5 \underline{L}^4\rangle$	_	_	0.002	0.003	0.003
$ d^{6}\underline{L}^{5}\rangle$	_	-	-	0.000	0.000
$ d^7 \underline{L}^6\rangle$	-	-	_	-	0.000
CrO ₂	3conf.	4conf.	5conf.	6conf.	7conf.
$ d^2\rangle$	0.308	0.226	0.212	0.211	0.211
$ d^3\underline{L}^1\rangle$	0.508	0.460	0.446	0.445	0.445
$ d^4 \underline{L}^2\rangle$	0.184	0.270	0.279	0.280	0.280
$ d^5 L^3\rangle$	_	0.044	0.058	0.059	0.059
$ d^6 \overline{L}^4\rangle$	_	_	0.004	0.005	0.005
$ d^7 \underline{L}^5\rangle$	_	-	_	0.000	0.000
$ d^{8}\underline{L}^{6}\rangle$	-	_	_	-	0.000

the cluster model with increasing the number of the electronic configurations included in representing the initial and final states. Table IV summarizes the weights of the electronic configurations. The position of the satellites is converged well with the basis set including up to the $|d^5\underline{L}^4\rangle$ configuration for VO₂ and the $|d^6\underline{L}^4\rangle$ configuration for CrO₂. We also checked the convergence of the LDA+DMFT AIM spectra with respect to the number of the electronic configurations included in the configuration-interaction solver evaluation [19,55] of the spectral intensities.



FIG. 7. The V (Cr) 1*s* XPS spectra calculated by the VO₆ (CrO₆) cluster model with varying the number of electronic configurations included in the spectral calculation. Here, the spectral intensities are convoluted with a Gaussian of 200 (300) meV and a Lorenzian of 650 (250) meV for VO₂ (CrO₂) (HWHM).



FIG. 8. (a) Cr 1s and (b) 2p XPS spectra simulated by the CrO₆ cluster model with varying the metal-ligand (Cr 3d-O 2p) hybridization strength. 100% corresponds to the hybridization strength derived from the LDA bands for the experimental crystal structure (see Appendix B). The spectral intensities are convoluted with a Gaussian of 300 meV and a Lorenzian of 250 meV (HWHM).

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TABLE V. The configuration (diagonal) energies in the Cr 1s XPS final states within the CrO₆ cluster-model Hamiltonian in eV. Here, the charge-transfer energy is defined for a Cr⁴⁺ (d^2) electronic configuration as $\Delta_{\text{CT}} = E(d^3\underline{L}) - E(d^2) = \varepsilon_d - \varepsilon_L + 2U_{dd}$, where ε_d (ε_L) denotes the averaged Cr 3d (ligand 2p) energy. The values of charge-transfer energy Δ_{CT} , 3d-3d interaction U_{dd} , and 2p-3d core-valence interaction U_{dc} are found in Appendix B.

Conf.	Diagonal energies	
$ \underline{c}d^2\rangle$	0	0.0
$ \underline{c}d^{3}\underline{L}^{1}\rangle$	$\Delta_{ m CT} - U_{dc}$	-4.5
$ \underline{c}d^{4}\underline{L}^{2}\rangle$	$2\Delta_{\mathrm{CT}} + U_{dd} - 2U_{dc}$	-4.4
$ \underline{c}d^{5}\underline{L}^{3}\rangle$	$3\Delta_{\rm CT} + 3U_{dd} - 3U_{dc}$	0.3
$ \underline{c}d^{6}\underline{L}^{4}\rangle$	$4\Delta_{\mathrm{CT}} + 6U_{dd} - 4U_{dc}$	9.5
$ \underline{c}d^{7}\underline{L}^{5}\rangle$	$5\Delta_{\rm CT} + 10U_{dd} - 5U_{dc}$	23.3
$ \underline{c}d^{8}\underline{L}^{6}\rangle$	$6\Delta_{\mathrm{CT}} + 15U_{dd} - 6U_{dc}$	41.6

APPENDIX C: CrO₆ CLUSTER-MODEL SIMULATION OF THE Cr 2*p* XPS SPECTRA

In Fig. 8, we calculate the Cr 1s and 2p XPS spectra with changing the Cr-O orbital hybridization strength in the CrO₆ cluster model. The configuration energies in the XPS final states within the cluster-model Hamiltonian are summarized in Table V. In the atomic limit (0%), i.e., V = 0, the Cr 2p XPS shows richer multiplet peaks in both the $2p_{3/2}$ and $2p_{1/2}$ components due to the strong Cr 2p-3d multipole interaction. The Cr-O hybridization renormalizes substantially the multiplet peaks [Fig. 3(b)]. The negligibly weak multiplet effect on the 1s excitation yields a quasi-single-peak line in the atomic limit [Fig. 3(a)].

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