



# Article Modelling of SO<sub>2</sub> and NO<sub>x</sub> Emissions from Coal and Biomass Combustion in Air-Firing, Oxyfuel, iG-CLC, and CLOU Conditions by Fuzzy Logic Approach

Jaroslaw Krzywanski <sup>1,\*</sup>, Tomasz Czakiert <sup>2</sup>, Anna Zylka <sup>1,\*</sup>, Wojciech Nowak <sup>3</sup>, Marcin Sosnowski <sup>1</sup>, Karolina Grabowska <sup>1</sup>, Dorian Skrobek <sup>1</sup>, Karol Sztekler <sup>3</sup>, Anna Kulakowska <sup>1</sup>, Waqar Muhammad Ashraf <sup>4,5</sup> and Yunfei Gao <sup>6</sup>

- <sup>1</sup> Faculty of Science and Technology, Jan Dlugosz University in Czestochowa, A. Krajowej 13/15, 42-200 Czestochowa, Poland
- <sup>2</sup> Department of Advanced Energy Technologies, Faculty of Infrastructure and Environment, Czestochowa University of Technology, Dabrowskiego 73, 42-200 Czestochowa, Poland
- <sup>3</sup> Faculty of Energy and Fuels, AGH University of Science and Technology, Mickiewicza 30, 30-059 Krakow, Poland
- <sup>4</sup> Centre for Process Systems Engineering, Department of Chemical Engineering, University College London, Gower Street, London WC1E 6BT, UK
- <sup>5</sup> Department of Mechanical Engineering, University of Engineering and Technology, University of Engineering & Technology, Lahore 54890, Punjab, Pakistan
- <sup>6</sup> Department of Chemical and Biomolecular Engineering, North Carolina State University, Raleigh, NC 27607, USA
- \* Correspondence: j.krzywanski@ujd.edu.pl (J.K.); a.zylka@ujd.edu.pl (A.Z.); Tel.: +00-48-343615970 (J.K.)

**Abstract:** Chemical looping combustion (CLC) is one of the most advanced technologies allowing for the reduction in CO<sub>2</sub> emissions during the combustion of solid fuels. The modified method combines chemical looping with oxygen uncoupling (CLOU) and in situ gasification chemical looping combustion (iG-CLC). As a result, an innovative hybrid chemical looping combustion came into existence, making the above two technologies complementary. Since the complexity of the CLC is still not sufficiently recognized, the study of this process is of a practical significance. The paper describes the experiences in the modelling of complex geometry CLC equipment. The experimental facility consists of two reactors: an air reactor and a fuel reactor. The paper introduces the fuzzy logic (FL) method as an artificial intelligence (AI) approach for the prediction of SO<sub>2</sub> and NO<sub>x</sub> (i.e., NO + NO<sub>2</sub>) emissions from coal and biomass combustion carried out in air-firing; oxyfuel; iG-CLC; and CLOU conditions. The developed model has been successfully validated on a 5 kW<sub>th</sub> research unit called the dual fluidized bed chemical looping combustion of solid fuels (DFB-CLC-SF).

Keywords: SO<sub>2</sub>; NO<sub>x</sub>; oxyfuel; iG-CLC; CLOU; artificial intelligence; fuzzy logic

# 1. Introduction

The fluidized bed combustion of coal or a co-combustion with the biomass under the oxyfuel conditions, as well as chemical looping combustion (CLC) or chemical-looping with oxygen uncoupling combustion processes (CLOU), seem to be the most promising technologies, with a high application potential which meets the requirements for the reduction in gaseous pollutant emissions [1–3]. Carbon dioxide sequestration (CCS) technologies reduce CO<sub>2</sub> emissions to the atmosphere. The following are distinguished among CCS technologies: pre-combustion capture, oxyfuel combustion, and post-combustion capture. The chemical looping combustion is classified as a CCS technology. Due to the high concentration of CO<sub>2</sub> in the exhaust gas, the CLC process is less energy-consuming and more effective than other CCS technologies. This is because there is no need to build additional installations for CO<sub>2</sub> separation [4,5]. Two processes are distinguished depending on the



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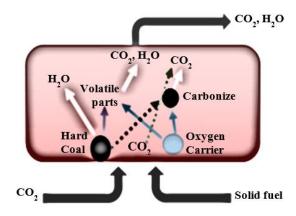
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**Copyright:** © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). type of solid oxygen carrier: CLOU (chemical looping with oxygen uncoupling) and iG-CLC (in situ gasification chemical looping combustion). The CLOU-type oxygen carriers do not require a direct contact with the fuel to release oxygen (Figure 1) [6]. Under the operating conditions of a fuel reactor, they can release gaseous oxygen. As a result of the reduction in the oxygen carrier in the fuel reactor, oxygen appears in addition to carbon dioxide (a gas mixture consisting of  $CO_2$  and  $O_2$  is formed) [7,8].



**Figure 1.** The scheme of the oxygen release process from the CLOU type of the oxygen carrier in the fuel reactor.

However, in the iG-CLC process, which involves combustion in chemical looping with a fuel gasification in a fuel reactor, the oxygen carrier requires a direct contact with the fuel (Figure 2) [6].

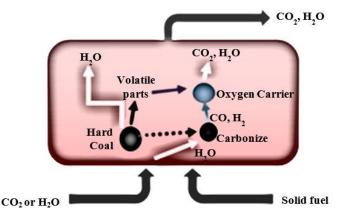
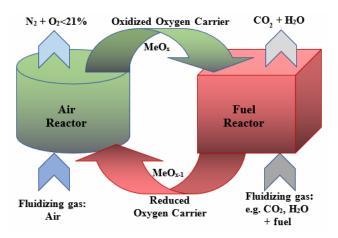


Figure 2. The scheme of oxygen release in the fuel reactor in the iG-CLC process.

Therefore, in the first step, the metal oxide is mixed with the fuel after it enters the combustion chamber. Only then does the oxygen carrier react with the volatile components and gaseous products resulting from the gasification of the fuel. The gasifying agent, and gas fluidising the fuel reactor bed, is  $CO_2$  or  $H_2O$ . However, carbon dioxide is often used for economic reasons, which can also be recirculated from flue gas [9].

Moreover, with its fuel flexibility feature, fluidised bed technology is suitable for co-firing the coal and biomass, which is considered to be a renewable energy source [10,11]. Finally, different combustion atmospheres can be applied in fluidized bed systems, i.e., the air-firing mode, when the combustion runs in air or in oxy-combustion conditions; that is, the mixture consists of oxygen and carbon dioxide or oxygen with the flue gas mixture. The exclusion of nitrogen from inlet gas in oxyfuel combustion generates exhaust gas consisting mostly of  $CO_2$  and  $H_2O$ . This exhaust composition is suitable for industrial reuse or geological storage [12]. Similar results are obtained with the CLC technology, which uses oxygen transported by solid oxygen carriers [6]. The operation of such systems uses the cyclic



processes of oxidation and reduction in solid oxygen carriers (metal oxides). The oxidation and reduction processes occur in the air and fuel reactor, respectively (Figure 3) [13].

Figure 3. The scheme of the oxidation and reduction in the oxygen carrier in the reactors.

The above-listed combustion methods belong to so-called clean combustion technologies. They can offer combustion processes with a reduced fuel consumption and zero or almost zero unwanted pollutants [14]. The chemical looping combustion process takes place in the fluidized bed regime; therefore, this combustion does not require such high temperatures as conventional combustion. Usually, the chemical looping combustion process occurs at temperatures of 800–900 °C. This prevents the formation of nitrogen oxides from the atmospheric air. The exhaust gases have a high CO<sub>2</sub> concentration (above 95%) and are not diluted with air, as in the case of conventional combustion. Therefore, the flue gases from the CLC process can be directly stored or used in industry after the steam condensation. In addition, the operating parameters of the CLC system need to be adequately selected to ensure that the fuel is completely burned, reducing its consumption in the combustion process [9].

The composition of solid fuels includes sulphur and nitrogen, which are the sources of  $NO_x$  (i.e.,  $NO + NO_2$ ) and  $SO_2$  emissions from combustion, i.e., the major gaseous pollutants, and their emissions should be considered when a combustion technology is implemented. Many factors influence  $NO_x$  and  $SO_2$  formation, including the fuel properties, combustion conditions, and construction of the combustor [15]. The CLC technology produces  $NO_x$  from the fuel, which means that this technology does not produce thermal nitrogen oxides from the air, as in conventional combustion. Additionally, other factors, such as the temperature and amount of oxygen transported by oxygen carriers, impact fuel combustion. A too low temperature and small amounts of oxygen in the fuel reactor result in incomplete combustion of the fuel, which results in lower  $NO_x$  and  $SO_x$  emissions [12].

Many papers have dealt with the topic of solid fuel combustion in fluidized bed boilers [8]. The modelling methods can generally be classified into programmed computing and artificial intelligence (AI) approaches. The programmed computing approach is based on mathematical descriptions of the considered processes [10,16–19]. A review of the CFB combustor models can be found in [17]. The author defined the two main approaches to performance modelling. The first one, i.e., the furnace approach, describes what happens in the furnace, but the second, the so-called system approach, is focused on system integration. In the furnace approach, the author distinguishes three levels of the model advancement: level 1:1D is the plug flow/stirred tank, using simple mass and energy balance; level 2:1.5D is the core-annulus models; and level 3:3D is the models which are based on the Navier–Stokes Equation with chemical kinetics and physical processes. A similar classification of the existing models was provided by Gungor and Eskin [20].

Adanez and Abad [21] selected the mechanistic and process models as the two main approaches in CLC modelling. In the first one, the fluid dynamics, the reactions' kinetics,

and the heat transfer in each reactor can be considered separately. The fluid dynamics modelling works can be further classified into two categories: computational fluid dynamic (CFD) models and macroscopic ones. The first one is the most complex and computationally advanced, requiring an extensive computation effort. On the other hand, the macroscopic models employ empirical equations and/or correlations describing the gas and particles' characteristic flows in the fluidized beds, making them, therefore, simpler to use but still making them effective and reliable.

The process models allow for defining the design and operating conditions, so they are valuable tools in optimising the CLC performance processes [21].

Finally, the AI approach employs nature-inspired artificial intelligence methods. Comprehensive literature reviews of the AI methods in the Energy sector can be found in [22–24]. Among other main AI modelling methods are artificial neural networks, fuzzy logic, and genetic algorithms.

Even though much work has been done, the complex mechanisms of the formation of and the reduction in  $NO_x$  and  $SO_2$  emissions during combustion in oxyfuel, iG-CLC, and CLOU technologies are still not sufficiently recognized [21].

Some recent studies reported in the literature have demonstrated the reliable and effective utilization of machine learning models for the modelling, predicting, and optimising of coal-based energy devices and power generation systems. An artificial intelligence modelling-based operational strategy was built to optimize the SO<sub>2</sub>, Hg, NO<sub>x</sub>, and dust emissions from the wet flue gas desulphurization system simultaneously. Similarly, comprehensive utilization of the machine learning models like artificial neural networks and support vector machines was presented for optimizing the coal-fired power plant performance parameters from the component, system, and strategic levels. Recently, artificial intelligence-based models were developed for the effective power production from a 660 MW power plant for optimizing the efficiency of industrial-scale steam turbines that support the net-zero goal in terms of a higher energy efficiency and a reduced emissions load [25].

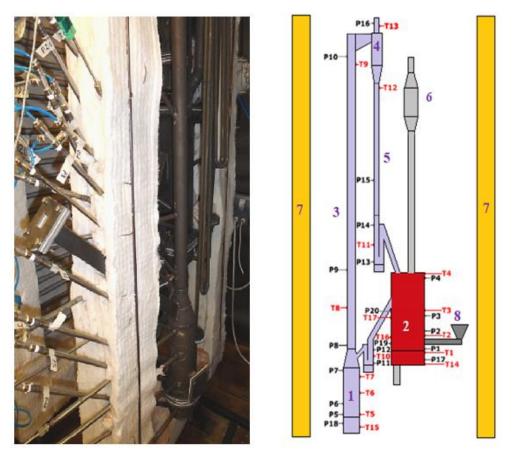
This paper introduces the fuzzy logic (FL) method as an artificial intelligence (AI) approach to predict the  $NO_x$  and  $SO_2$  emissions from the biomass and bituminous coal combustion in air-firing, oxyfuel, iG-CLC, CLOU, and air-firing conditions (as a reference). The proposed model has been validated using the experimental data from the existing combustion facility.

### 2. Experimental Procedures and Methods

The measurements of the gas emissions (NO<sub>x</sub> and SO<sub>2</sub>) were carried out on the existing 5 kWth dual fluidized bed chemical looping combustion of solid fuels (DFB-CLC-SF) facility located at the Czestochowa University of Technology, Poland. In the Fuel Reactor (FR), the oxygen carriers (OC) release oxygen, resulting in the combustion process. Then, the reduced metal oxides are transported to the air reactor (AR) for the oxidation processes [15]. A schematic diagram of the FB-CLC-SF system is shown in Figure 4.

The total height of the air reactor is 2.52 m, and the fuel reactor is 0.5 m, while the air reactor's and the riser's diameters are 0.098 m and 0.04 m, respectively. A more detailed description of the discussed research unit can be found in other papers [12].

The experimental tests were carried out on a research unit in the temperature range from 1079 to 1157 K. Two different solid fuels were used during these investigations, i.e., wood chips and bituminous coal from a Polish coal mine (Table 1).



**Figure 4.** The hot FB-CLC-SF facility; 1—air reactor, 2—fuel reactor, 3—riser, 4—cyclone, 5—downcomer, 6—particle's collector, 7—heaters, 8—screw feeder, P1—P19—pressure transducers, and T1–T16—thermocouples [15].

Table 1. Properties of fuels (as received).

De	Coal	Biomass	
	26.16	17.25	
	Moisture content M	6.6	6.2
Proximate analysis	Volatile matter content VM	35.7	77.0
(% wt.)	Ash content, A	5.5	1.4
	Fixed carbon content, FC <sup>by diff.</sup> *	52.2	15.4
	Carbon content C	68.2	47.7
T Il time e tra a malanzia	Hydrogen content, H	4.90	5.47
Ultimate analysis	Sulphur content, S	1.02	0.11
(% wt.)	Nitrogen content, N	1.01	0.27
	Oxygen content, O <sup>by diff.</sup> *	12.77	38.85

\* by diff.—by difference.

A wide variety of working conditions were used in the present study. Seven tests involving different fuels, gaseous atmospheres in the FR, and combustion modes were carried out. The first one, called Test 0, considers a conventional coal combustion in the air in the atmosphere. Test 1 corresponds to oxyfuel conditions when the combustion process occurs in a suitable gas mixture of  $CO_2$  and  $O_2$ .

Solid oxygen carriers (OCs) acting as oxygen sources, and  $CO_2$  as fluidizing gas, were applied during the other five tests 2–6. Three different types of OCs were considered: calcined ilmenite (OC1) in tests 2, 5, and 6, CuO (60% wt.) enriched with the copper tailing support (OC2) in test 3, as well as CuO (60% wt.) with the support of ilmenite (20% wt.) and fly ash (OC3) in test 4. Ilmenite is a natural mineral and belongs to the group

of iG-CLC oxygen carriers, while the CuO-based oxygen carriers belong to the CLOU functional group. A detailed description of all the oxygen carriers used in this work can be found in [12,15]. The flue gas components were measured at the outlet of the fuel reactor of the FB-CLC-SF facility using gas analysers based on infrared detection (GASMET DX-4000). The measurement technique is FTIR (Fourier transform infrared spectroscopy). A dedicated software Calcmet was used for the data acquisition and further analyses. The sampling frequency was set as 1 Hz. The measurement ranges were 0–1000 ppm for SO<sub>2</sub>, 0–1000 ppm for NO, and 0–100 ppm for NO<sub>2</sub>, respectively. The NO and NO<sub>2</sub> concentrations were recalculated into the NO<sub>x</sub> in this study. The measurement accuracy is 1% of the measurement range for all the flue gas components. The detailed operating conditions and results of the measurements are summarized in Table 2.

Table 2. The operating parameters	(ad-air-dried basis)
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Fuel		ID <sub>mode</sub>	Oxygen Excess	Avg. Temp. in FR	EC. <sup>ad</sup> /VM <sup>ad</sup>	N <sup>ad</sup> /C <sup>ad</sup>	S <sup>ad</sup>	A <sup>ad</sup>	$SO_2$	NO <sub>x</sub>
	Test/Combustion Atmosphere/OC		OE	Т					exp.	
		-	%	К		%			ppm	
	Test 0/Air/-	Air-fired conditions	40	1147					627	393
	Test 1/O <sub>2</sub> (21%)/CO <sub>2</sub> (79%)/-	Oxyfuel conditions	50	1135					843	379
	Test 2/100% CO <sub>2</sub> /OC1 (ilmenite)		30	1157					85	56
Sobieski bituminous coal	Test $3/100\%$ CO <sub>2</sub> /OC2 (copper oxide (60% wt.) with the support of carbonate waste from ore flotation)	CLC &	50	1085	1.5	0.015	1.09	5.9	40	116
	Test 4/100% CO <sub>2</sub> /OC3 (copper oxide (60% wt.) with the support of ilmenite (20% wt.) and fly ash (20% wt.)	CLOU combustion	10	1087					61	176
Wood chips	Test 5/100% $CO_2/OC1$ (ilmenite) Test 6/100% $CO_2/OC1$ (ilmenite)		10	1079 1106	0.2	0.006	0.12	1.5	21 8	57 125

Since the complexity of the CLC process is still not sufficiently recognized, especially in such CLC equipment of a sophisticated geometry, the fuzzy logic (FL) approach, as one of the main artificial intelligence (AI) methods, was introduced to the research presented in this paper [22]. Together with artificial neural networks and evolutionary algorithms, the FL approach belongs to the main AI techniques [26]. The fuzzy logic-based methods employ linguistic variables and fuzzy sets to evaluate the considered process, enabling a qualitative judgment to be applied to the quantitative parameters and dealing with imprecise, vague, and uncertain information [27,28]. The following four components make an FL model: the fuzzifier, a fuzzy rule base, an inference engine, and a defuzzifier [29–31]. The first stage of building a model consists of expressing the input variables by fuzzy sets, where a numeric value input is assigned to a value of a membership function (ranging from 0 to 1) [30]. This is the fuzzification stage since transforming a crisp input value into a fuzzy set occurs. The IF-THEN fuzzy rule base allows the system to generate fuzzy outputs. Finally, the defuzzification process produces crisp outputs corresponding to the crisp inputs supplied to the model. The Qtfuzzylite fuzzy logic control application [32] is used for this research to predict the  $SO_2$  and  $NO_x$  emissions from coal and biomass combustion in air-firing, oxyfuel, iG-CLC, and CLOU conditions in a complex geometry fluidized bed furnace.

This approach is a valuable modelling technique suitable for describing complex systems, which allows for overcoming the shortcomings of the programmed computing approach and measurements during the experimental research [33]. The introduced strategy can be considered an alternative to the above data processing techniques, taking into account the complexity of analytical and numerical methods and the high cost of empirical experiments [34]. The FL approach is practical, especially when the subjective knowledge and experience of the expert are significant in defining the objective function and decision variables [33,35]. It is also an approach to map the input into the output space, involving the so-called membership functions to define how each input value is mapped to a membership one [35].

On the other hand, the low data availability of expert knowledge and a small number of input variables are the main limitations of this technique. Intelligent hybrid neuro-fuzzy systems may overcome these shortcomings [34].

#### 3. Results and Discussion

The Qtfuzzylite fuzzy logic control application [32] was used as a computational tool to make the presented model. Nine inputs influencing the SO<sub>2</sub> and NO<sub>x</sub> emissions were employed to establish the model, i.e., (1) the ID<sub>mode</sub> tag defining the combustion mode (Air1, Air2 for air-fired combustion, Oxy for oxyfuel conditions, and CLC\_CLOU for iG-CLC and CLOU functionalities), (2) the kind of oxygen carrier OC (symbols OC0 and OC4 for air-fired conditions, indicating that no oxygen carriers are used, OC1 for ilmenite, OC2 for CuO (60% wt.) enriched with the copper tailing support, and OC3 for CuO (60% wt.) enriched with the copper tailing support, and OC3 for CuO (60% wt.) enriched with ilmenite (20% wt.) and fly ash (20% wt.) supports, (3) excess oxygen OE in the fuel reactor, (4) the average fuel reactor temperature T, (5) the FC<sup>ad</sup>./VM<sup>ad</sup> ratio (ad—air-dried basis), (6) the N<sup>ad</sup>/C<sup>ad</sup> molar ratio, (7) the sulphur content S<sup>ad</sup> and (8) ash content A<sup>ad</sup> in the fuel (coal and biomass), and 9. the ID<sub>fuel</sub> tag defines the fuel type, 1 being for coal and 2 being for wood chips (Table 3).

Table 3. Parameters of entry conditions.

	Description (Units)	Value		
	1. Combustion mode tag ID <sub>atm</sub> <sup>a</sup> (-)	Air1, Air2, Oxy, CLC_CLOU		
	2. Kind of oxygen carriers OC <sup>b</sup> (-)	OC0 OC1, OC2, OC3, OC4		
	3. Excess Oxygen, OE (%)	10-50		
	4. Average fuel reactor temperature T, (K)	1079–1157		
Inputs	5. $FC^{ad}/VM^{ad}$ ratio (-) <sup>c</sup>	0.20-1.50		
	6. $N^{ad}/C^{ad}$ molar ratio (-) <sup>d</sup>	0.006-0.015		
	7. Sulphur content S <sup>ad</sup> . (% wt.)	0.12-1.09		
	8. Ash content A <sup>ad</sup> (% wt.)	1.5–5.9		
	9. Kind of fuel tag ID <sub>fuel</sub> (-)	1–2		
Outputs	1. SO <sub>2</sub> emission (ppm)	8–843		
Outputs	2. $NO_x$ emission (ppm)	56–393		

<sup>a</sup> Air1, Air2 stands for air-fired combustion, Oxy—oxyfuel conditions, CLC\_CLOU—iG-CLC and CLOU systems, <sup>b</sup> OC1, OC4 stands for ilmenite, OC2—CuO (60% wt.) enriched with the copper tailing support, OC3—CuO (60% wt.) enriched with ilmenite (20% wt.) and fly ash (20% wt.) supports. <sup>c</sup> FC<sup>ad</sup>/VM<sup>ad</sup>—fixed carbon/ volatile matter content in the fuel material (on air-dried basis). <sup>d</sup> N<sup>ad</sup>/C<sup>ad</sup>—nitrogen/ carbon content in the fuel material (on air-dried basis).

Thus, the model has the unique ability to consider the diverse combinations of solid fuels combustion modes and atmospheres as well as various OCs mixtures, which is very effective from a practical point of view [12,15]. The developed model can generalize the information about the process behaviour. This comprehensive tool provides innovative abilities for design considerations and the performance of iG-CLC and CLOU systems. Such selected inputs allowed for the description of the outputs, i.e., the SO<sub>2</sub> and NO<sub>x</sub> emissions from the FB-CLC-SF unit. The model uses triangular and constant terms for both the inputs and outputs, allowing for a mapping of the inputs to the outputs domains. Sugeno-type fuzzy inference systems were applied in this study [35].

The screenshot of the developed system is shown in Figure 5, whereas the IF-THEN rule-base is depicted in Table 4.

The developed model may serve as a novel and convenient tool to predict the influence of the selected input parameters on the  $SO_2$  and  $NO_x$  emissions from advanced biomass and bituminous coal combustion under oxyfuel, iG-CLC, and CLOU conditions as well as from conventional air-firing, which was carried out with complex geometry fluidized bed equipment.

The validation procedure of the presented model was successfully carried out using the experimental data from the 5 kWth FB-CLC-SF facility (Figure 6).

The results depicted in Figure 6 correspond to a variety of operating parameters. A good accuracy of the developed model was achieved. The maximum relative percentage error between the measured values and predicted emissions of  $SO_2$  and  $NO_x$  is lower than 10% and hence stands for a reliable basis for the possible use of the developed system in practice.

The SO<sub>2</sub> and NO<sub>x</sub> emissions from wood chips and bituminous coal combustion under different combustion modes for OE = 10% and T = 1080 K, predicted by the model, are shown in Figure 7.

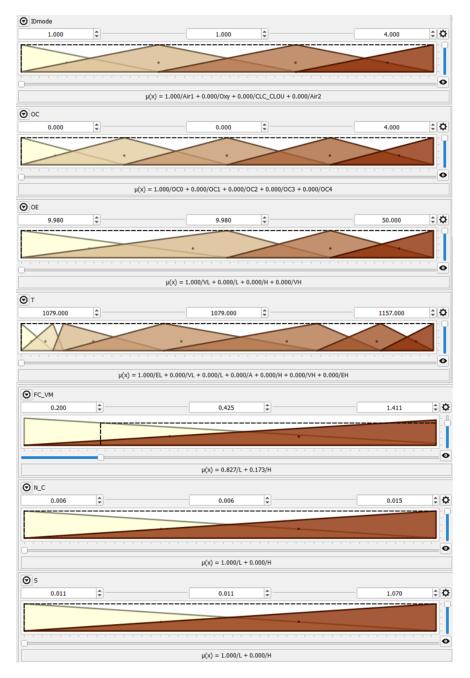
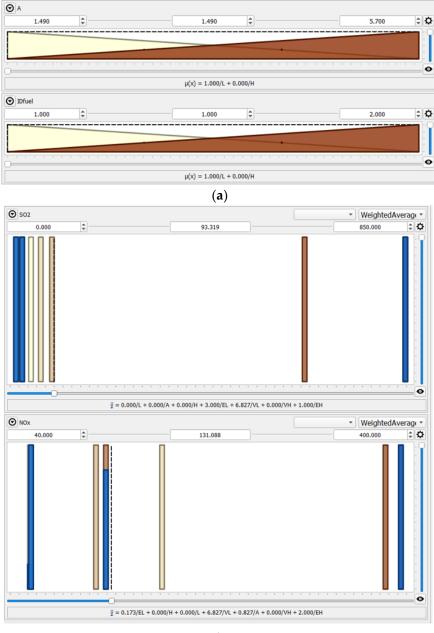


Figure 5. Cont.



(**b**)

Figure 5. The FL model with graphical representations of input (a) and output (b) variables.

Table 4. The IF-THEN rule-base of the model for (a)  $SO_2$  and (b)  $NO_x$ .

(a)							
SO <sub>2</sub>	EL	VL	L	А	Н	VH	EH
ID <sub>mode</sub> OC OE T FC/VM N/C S	CLC_CLOU OC1/OC4 VL A L L L L	CLC_CLOU OC1/OC4 VL EL L L L	CLC_CLOU OC2 VH VL H H H	CLC_CLOU OC3 VH L H H H	CLC_CLOU OC1/OC4 L EH H H H	Air1/Air2 OC0 H VH H H H	Oxy OC0 VH H H H H
A ID <sub>fuel</sub>	L H	L H	H L	H L	H L	H L	H L

		Table 4. Cont.							
	(b)								
NO <sub>x</sub>	EL	VL	L	А	Н	VH	EH		
ID <sub>mode</sub> OC OE T FC/VM N/C S A	CLC_CLOU OC1/OC4 L EH H H H H H	CLC_CLOU OC1/OC4 VL EL L L L L L	CLC_CLOU OC2 VH VL H H H H H	CLC_CLOU OC1/OC4 VL A L L L L L	CLC_CLOU OC3 VH L H H H H H	Oxy OC0 VH H H H H H	Air1/Air2 OC0 H VH H H H H H		
ID <sub>fuel</sub>	L	Н	L	Н	L	L			

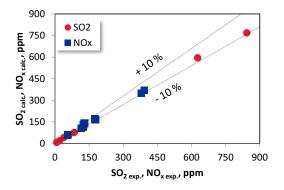
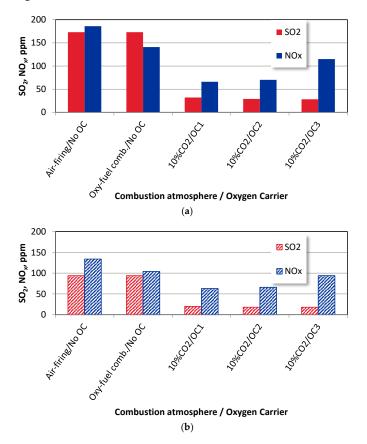


Figure 6. Validation of the model.



**Figure 7.** The SO<sub>2</sub> and NO<sub>x</sub> emissions from (**a**) bituminous coal and (**b**) biomass combustion under different combustion modes, where OE = 10% and T = 1080 K.

The SO<sub>2</sub> and NO<sub>x</sub> emissions are the lowest for the CLOU and iG-CLC combustion modes, with OC1 and OC2 oxygen carriers. This behaviour corresponds to the first observations reported during the experiments [12]. Since the use of ilmenite (OC1) was accompanied by the relatively low efficiency of the combustion process, the emissions of SO<sub>2</sub> and NO<sub>x</sub> are also low. The rather good oxygen transport abilities of OC2 at the beginning (iG-CLC feature), reported in [12], were gradually worsened during the experiments.

Moreover, the numerous sinters observed in the fuel reactor may contribute to a further deterioration of the oxygen transport. This oxygen carrier expressed only about half of its oxygen transport capabilities [12]. Such a behaviour resulted in low  $SO_2$  and  $NO_x$  concentrations in the exhaust gas.

These results also comply with the conclusions withdrawn from the comparison of the  $SO_2$  and  $NO_x$  emissions during the coal and biomass combustion, shown in Figure 7a,b. Since the sulphur and nitrogen contents in the bituminous coal are higher than those for the biomass, the  $SO_2$  and  $NO_x$  emissions from the coal combustion are also higher. Moreover, the DeNO<sub>x</sub> mechanism is favoured by H and OH radicals from the higher volume of volatiles released from the biomass. This can be reported for all tests, including the airfiring and oxyfuel combustion modes and CLOU and iG-CLC combustion with OC1, OC2, and OC3 oxygen carriers.

As the fuel-bound sulphur is the only source of  $SO_2$  in this study, replacing the  $O_2/N_2$  (air-firing conditions) with an  $O_2/CO_2$  (oxyfuel) atmosphere results in the same  $SO_2$  emissions when burning the same fuel. On the other hand, a high concentration of  $CO_2$  under the oxyfuel mode favours a higher CO content in the fuel reactor's atmosphere due to the Bouduoard reaction, leading to decreased  $NO_x$  emissions which are observed when comparing Figure 7a,b [2]. In addition, CO oxidation inhibition occurs in the presence of high concentrations of  $SO_2$  [15]. The increase in the carbon conversion to CO in the presence of  $SO_2$  [9] also decreases the NO emissions.

The lower fuel burn-out arising from poor oxygen transport by the OCs results in comparable low levels of  $SO_2$  emissions [36].

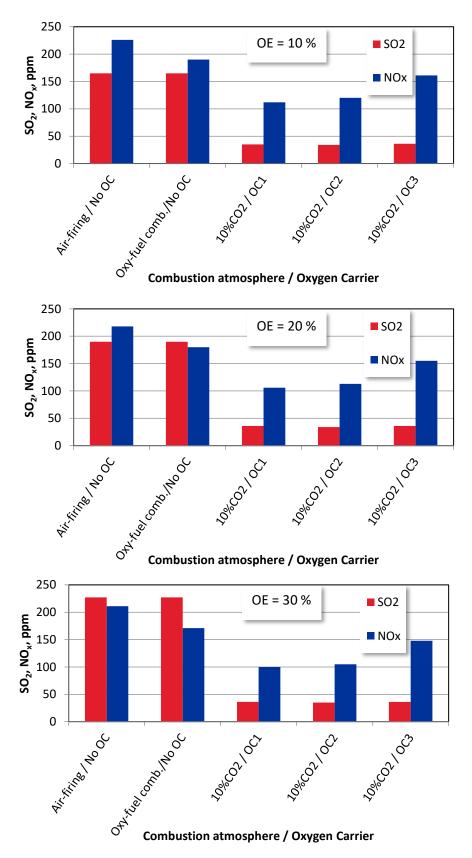
A similar behaviour was reported for the  $NO_x$  emissions with OC1 and OC2, as the reduction character of the atmosphere deteriorates the  $NO_x$  formation mechanisms.

Different observations were made for the fuel combustion with the OC3 oxygen carrier. Relatively good oxygen transport capabilities characterized this oxygen carrier. Thus, the released oxygen deteriorates the  $DeNO_x$  mechanism and allows for converting the volatiles N to  $NO_x$ , limiting the N<sub>2</sub> formation and ultimately leading to an increase in the  $NO_x$  emissions.

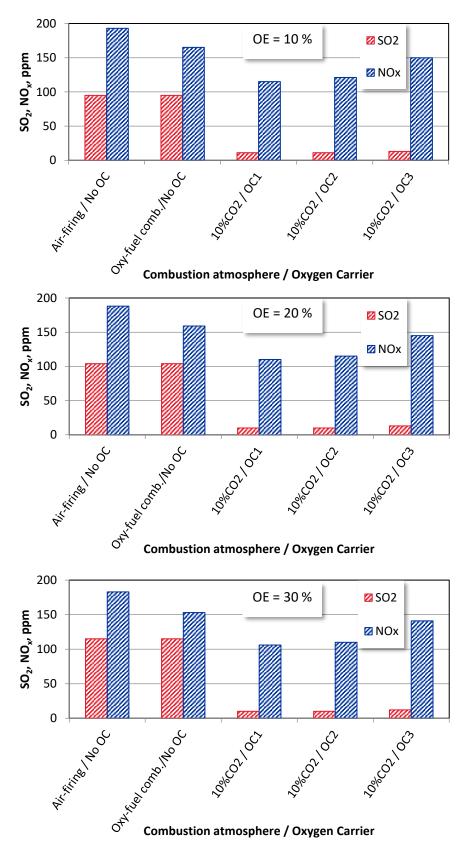
The effect of the oxygen excess on the SO<sub>2</sub> and NO<sub>x</sub> emissions is given in Figure 8.

The increase in the oxygen excess is better expressed in air-firing and oxyfuel combustion atmospheres where no oxygen carriers are employed, and the sulphur evolution is enhanced in such conditions [36]. The increase in the SO<sub>2</sub> concentrations in these combustion conditions leads to a decrease in the NO<sub>x</sub> emissions. This dependence can be explained by the well-known CO oxidation inhibition effect that occurs in the presence of SO<sub>2</sub>, which results in the NO emissions lowering with the SO<sub>2</sub> formation [37].

Due to the replacement of the gaseous oxidized with solid oxygen carriers OC1, OC2, and OC3, the access to oxygen has become limited, which deteriorates the SO<sub>2</sub> and NO<sub>x</sub> formation mechanisms [12]. A similar behaviour can be registered for wood chips combustion (Figure 9). Due to the lack of CLOU functionality, the additional dilution effect makes the lowest emissions for OC1 (ilmenite). The CuO content (OC2 and OC3 carriers), characterized by the CLOU functionality, facilitates the release and transport of oxygen, causing an increase in the NO<sub>x</sub> emissions. At the same time, the dilution effects keep the SO<sub>2</sub> emissions at similar levels.



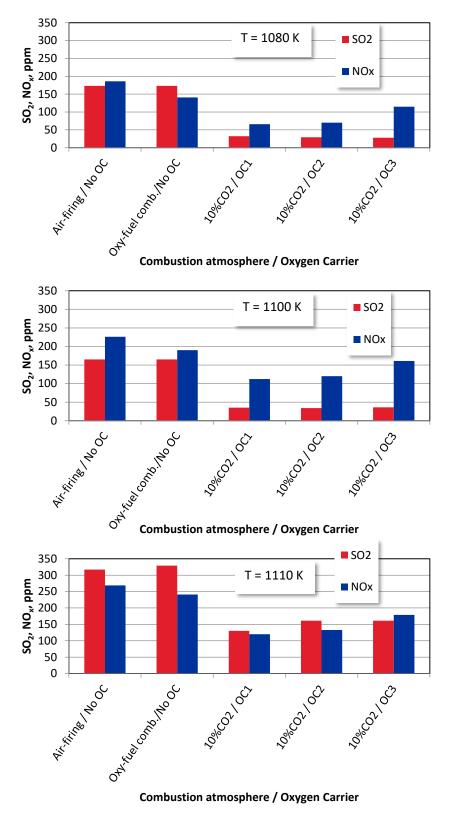
**Figure 8.** The effect of oxygen excess on  $SO_2$  and  $NO_x$  emissions from bituminous coal combustion under different combustion modes (T = 1100 K).



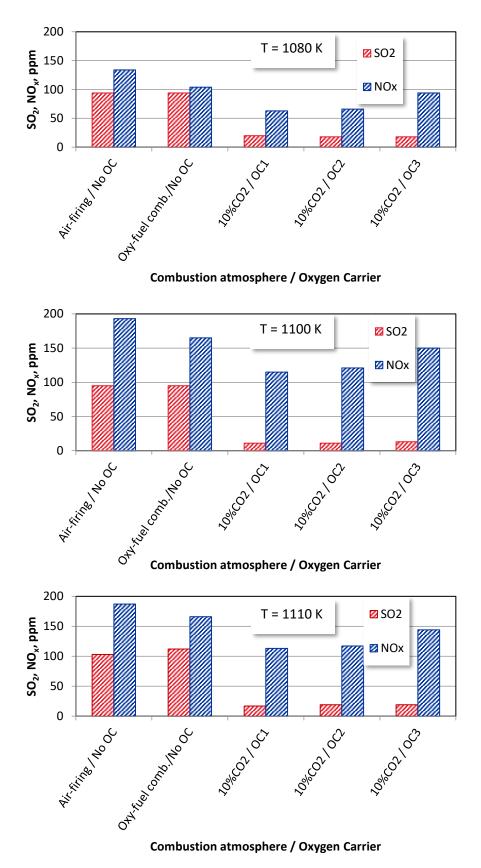
**Figure 9.** The effect of oxygen excess on  $SO_2$  and  $NO_x$  emissions wood chips combustion under different combustion modes (T = 1100 K).

The effects of the average fuel reactor temperatures on the  $SO_2$  and  $NO_x$  emissions from bituminous coal and wood chips under various combustion atmospheres are shown in

Figures 10 and 11. The average fuel reactor temperature is an essential operating parameter strongly influencing the  $SO_2$  and  $NO_x$  emissions. The increase in the temperature leads to the combustion processes intensifications, resulting in the increase in sulphur and nitrogen oxides in all the combustion environments.



**Figure 10.** The average fuel reactor temperature effect on  $SO_2$  and  $NO_x$  emissions from bituminous coal combustion under different combustion modes (OE = 10%).



**Figure 11.** The effect of average fuel reactor temperature on SO<sub>2</sub> and NO<sub>x</sub> emissions from wood chips combustion under different combustion modes (OE = 10%).

Similar behaviour can be observed for biomass combustion in temperatures of 1080 K and 1100 K.

For a temperature of 1110 K, the  $NO_x$  concentrations in the flue gas are lower than for a temperature of 1100 K. It may be caused by high  $SO_2$  emissions inhibiting the oxidation of CO, which is the active agent in the DeNOx mechanisms. Furthermore, the already mentioned  $DeNO_x$  mechanism is favoured by H and OH radicals from the higher volume of volatiles in the biomass material.

## 4. Conclusions

The paper introduces an AI approach for predicting the SO<sub>2</sub> and NO<sub>x</sub> emissions from coal and biomass advanced combustion under oxyfuel, iG-CLC, and CLOU conditions and conventional air firing.

A new fuzzy logic-based model was developed and presented in this article. The model was successfully validated using the experimental data from the existing 5 kW<sub>th</sub> dual fluidized bed chemical looping combustion of solid fuels unit.

The developed system achieved satisfying accuracy. The maximum relative errors between the measured values and predicted by the developed model  $SO_2$  and  $NO_x$  emissions are lower than 10%.

Based on the obtained results, the following main conclusions can be drawn:

- 1. Simulations confirmed that biomass combustion generates lower SO<sub>2</sub> and NO<sub>x</sub> emissions than coal burning for all combustion modes.
- 2. The SO<sub>2</sub> and NO<sub>x</sub> emissions strongly depend on the properties of the oxygen carriers used in the systems and operating parameters.
- 3. The fuzzy logic-based method constitutes a practical AI approach for modelling gaseous pollutants emissions from advanced and complex combustion systems.

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#### References

- Lyngfelt, A. 20-Chemical Looping Combustion (CLC). In *Fluidized Bed Technologies for Near-Zero Emission Combustion and Gasifica*tion; Scala, F., Ed.; Woodhead Publishing: Sawston, UK, 2013; pp. 895–930. ISBN 978-0-85709-541-1.
- Lyngfelt, A.; Linderholm, C. Chemical-Looping Combustion of Solid Fuels—Status and Recent Progress. *Energy Procedia* 2017, 114, 371–386. [CrossRef]
- 3. Imtiaz, Q.; Hosseini, D.; Müller, C.R. Review of Oxygen Carriers for Chemical Looping with Oxygen Uncoupling (CLOU): Thermodynamics, Material Development, and Synthesis. *Energy Technol.* **2013**, *1*, 633–647. [CrossRef]
- Pires, J.C.M.; Martins, F.G.; Alvim-Ferraz, M.C.M.; Simões, M. Recent Developments on Carbon Capture and Storage: An Overview. Chem. Eng. Res. Des. 2011, 89, 1446–1460. [CrossRef]
- 5. Rackley, S.A. Carbon Capture and Storage; Butterworth-Heinemann: Oxford, UK, 2017; ISBN 978-0-12-812042-2.
- Adanez, J.; Abad, A.; Garcia-Labiano, F.; Gayan, P.; de Diego, L.F. Progress in Chemical-Looping Combustion and Reforming Technologies. *Prog. Energy Combust. Sci.* 2012, 38, 215–282. [CrossRef]
- Cormos, C.-C. Chemical Looping with Oxygen Uncoupling (CLOU) Concepts for High Energy Efficient Power Generation with near Total Fuel Decarbonisation. *Appl. Therm. Eng.* 2017, 112, 924–931. [CrossRef]
- 8. de Souza-Santos, M.L. *Solid Fuels Combustion and Gasification: Modeling, Simulation, and Equipment Operations,* 2nd ed.; CRC Press: Boca Raton, FL, USA, 2010; ISBN 978-1-4200-4749-3.

- 9. Czakiert, T.; Krzywanski, J.; Zylka, A.; Nowak, W. Chemical Looping Combustion: A Brief Overview. *Energies* 2022, 15, 1563. [CrossRef]
- 10. Al Asfar, J.J.; AlShwawra, A.; Sakhrieh, A.; Hamdan, M.A. Combustion Characteristics of Solid Waste Biomass, Oil Shale, and Coal. *Energy Sources Part A Recovery Util. Environ. Eff.* **2018**, 40, 335–342. [CrossRef]
- 11. Wang, X.; Gong, Y.; Wang, X.; Jin, B. Experimental and Kinetics Investigations of Separated-Gasification Chemical Looping Combustion of Char with an Iron Ore as the Oxygen Carrier. *Fuel Process. Technol.* **2020**, *210*, 106554. [CrossRef]
- 12. Zylka, A.; Krzywanski, J.; Czakiert, T.; Idziak, K.; Sosnowski, M.; Grabowska, K.; Prauzner, T.; Nowak, W. The 4th Generation of CeSFaMB in Numerical Simulations for CuO-Based Oxygen Carrier in CLC System. *Fuel* **2019**, 255, 115776. [CrossRef]
- 13. Liu, F.; Liu, J.; Yang, Y. Review on the Theoretical Understanding of Oxygen Carrier Development for Chemical-Looping Technologies. *Energy Fuels* **2022**, *36*, 9373–9384. [CrossRef]
- 14. Moghtaderi, B. Review of the Recent Chemical Looping Process Developments for Novel Energy and Fuel Applications. *Energy Fuels* **2012**, *26*, 15–40. [CrossRef]
- Krzywanski, J.; Czakiert, T.; Nowak, W.; Shimizu, T.; Zylka, A.; Idziak, K.; Sosnowski, M.; Grabowska, K. Gaseous Emissions from Advanced CLC and Oxyfuel Fluidized Bed Combustion of Coal and Biomass in a Complex Geometry Facility: A Comprehensive Model. *Energy* 2022, 251, 123896. [CrossRef]
- 16. Basu, P. Circulating Fluidized Bed Boilers; Springer International Publishing: Cham, Switzerland, 2015; ISBN 978-3-319-06172-6.
- 17. Basu, P. Combustion of Coal in Circulating Fluidized-Bed Boilers: A Review. Chem. Eng. Sci. 1999, 54, 5547–5557. [CrossRef]
- 18. Al Asfar, J.; AlShwawra, A.; Shaban, N.A.; Alrbai, M.; Qawasmeh, B.R.; Sakhrieh, A.; Hamdan, M.A.; Odeh, O. Thermodynamic Analysis of a Biomass-Fired Lab-Scale Power Plant. *Energy* **2020**, *194*, 116843. [CrossRef]
- 19. Al Asfar, J.J.; Hammad, A.; Sakhrieh, A.; Hamdan, M.A. Two-Dimensional Numerical Modeling of Combustion of Jordanian Oil Shale. *Energy Sources Part A Recovery Util. Environ. Eff.* **2016**, *38*, 1189–1196. [CrossRef]
- 20. Gungor, A.; Eskin, N. Two-Dimensional Coal Combustion Modeling of CFB. Int. J. Therm. Sci. 2008, 47, 157–174. [CrossRef]
- 21. Adánez, J.; Abad, A. Chemical-Looping Combustion: Status and Research Needs. *Proc. Combust. Inst.* **2019**, *37*, 4303–4317. [CrossRef]
- 22. Lyu, W.; Liu, J. Artificial Intelligence and Emerging Digital Technologies in the Energy Sector. *Appl. Energy* **2021**, 303, 117615. [CrossRef]
- 23. Ahmad, T.; Zhang, D.; Huang, C.; Zhang, H.; Dai, N.; Song, Y.; Chen, H. Artificial Intelligence in Sustainable Energy Industry: Status Quo, Challenges and Opportunities. *J. Clean. Prod.* **2021**, *289*, 125834. [CrossRef]
- 24. Kalogirou, S.A. Artificial Intelligence for the Modeling and Control of Combustion Processes: A Review. *Prog. Energy Combust. Sci.* **2003**, *29*, 515–566. [CrossRef]
- Muhammad Ashraf, W.; Moeen Uddin, G.; Muhammad Arafat, S.; Krzywanski, J.; Wang, X. Strategic-level performance enhancement of a 660 MW<sub>e</sub> supercritical power plant and emissions reduction by AI approach. *Energy Convers. Manag.* 2021, 250, 114913. [CrossRef]
- 26. Esen, H.; Inalli, M.; Sengur, A.; Esen, M. Artificial Neural Networks and Adaptive Neuro-Fuzzy Assessments for Ground-Coupled Heat Pump System. *Energy Build.* **2008**, *40*, 1074–1083. [CrossRef]
- 27. Dragojlovic, Z.; Kaminski, D.A.; Ryoo, J. Tuning of a Fuzzy Rule Set for Controlling Convergence of a CFD Solver in Turbulent Flow. *Int. J. Heat Mass Transf.* 2001, 44, 3811–3822. [CrossRef]
- 28. Wang, G.; Luo, Z.; Zhu, L.; Chen, H.; Zhang, L. Fuzzy Estimation for Temperature Distribution of Furnace Inner Surface. *Int. J. Therm. Sci.* **2012**, *51*, 84–90. [CrossRef]
- 29. Pospíchal, J. Fuzzy Sets and Fuzzy Logic: Theory and Applications. By George J. Klir and Bo Yuan. Prentice Hall: Upper Saddle River, NJ, 1995. 574 pp. ISBN 0-13-101171-5. *J. Chem. Inf. Comput. Sci.* **1996**, *36*, 619. [CrossRef]
- 30. Kucukali, S.; Baris, K. Turkey's Short-Term Gross Annual Electricity Demand Forecast by Fuzzy Logic Approach. *Energy Policy* **2010**, *38*, 2438–2445. [CrossRef]
- 31. Kılıç, B. Optimisation of Refrigeration System with Two-Stage and Intercooler Using Fuzzy Logic and Genetic Algorithm. *Int. J. Eng. Appl. Sci.* 2017, *9*, 42–54. [CrossRef]
- 32. Available online: www.fuzzylite.com (accessed on 26 September 2022).
- 33. Ross, T.J. Fuzzy Logic with Engineering Applications, 3rd ed.; John Wiley: Chichester, UK, 2010; ISBN 978-0-470-74376-8.
- 34. Krzywanski, J. Heat Transfer Performance in a Superheater of an Industrial CFBC Using Fuzzy Logic-Based Methods. *Entropy* **2019**, *21*, 919. [CrossRef]
- 35. Mohd Adnan, M.R.; Sarkheyli, A.; Mohd Zain, A.; Haron, H. Fuzzy Logic for Modeling Machining Process: A Review. *Artif. Intell. Rev.* **2015**, *43*, 345–379. [CrossRef]
- Krzywanski, J.; Blaszczuk, A.; Czakiert, T.; Rajczyk, R.; Nowak, W. Artificial intelligence treatment of NOX emissions from CFBC in air and oxy-fuel conditions. In Proceedings of the CFB-11—11th International Conference on Fluidized Bed Technology, Beijing, China, 14–17 May 2014; pp. 619–624.
- Hildor, F.; Leion, H.; Mattisson, T. Steel Converter Slag as an Oxygen Carrier—Interaction with Sulfur Dioxide. *Energies* 2022, 15, 5922. [CrossRef]