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**Diffuser: a user-friendly program for diffusion chronometry  
with robust uncertainty estimation**

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26 **Abstract**

27 Chemical diffusion in minerals has shown great potential to quantify timescales of  
28 geological processes. The presence of chemical gradients, along with favorable  
29 temperature and time conditions, lead to the formation of measurable diffusion profiles.  
30 Temporal information can be extracted from measured diffusion profiles using either  
31 analytical or numerical solutions of Fick's second law. Currently, there is a lack of  
32 widely adopted programs for diffusion studies. In addition, the uncertainties associated  
33 with timescales derived from diffusion chronometry are critical for geological studies,  
34 but are not always robustly evaluated. In many cases, only uncertainties in curve fitting  
35 parameters and temperature are considered, whereas other uncertainties, such as those  
36 associated with the experimentally determined diffusion coefficients themselves, are  
37 rarely propagated into the calculated timescales. Ignoring these uncertainties reduces  
38 the reproducibility and intercomparability of results. In response to these challenges,  
39 we present Diffuser, a user-friendly program to standardize diffusion chronometry with  
40 transparent and robust propagation of uncertainties. Using analytical and numerical  
41 methods, our program provides an automatic, visual, and efficient curve fit to extract  
42 chronological information from diffusion profiles. The method is complemented by an  
43 algorithm to propagate all uncertainties (i.e., measurement, temperature, curve fitting,  
44 and diffusion coefficient) to derived timescales. Three examples are provided to  
45 highlight how the program can recover timescales with internal consistency, efficient  
46 computing and easy-to-use features. Our freely available and user-friendly program  
47 will hopefully increase the accessibility and consistency of diffusion modeling and

48 thereby to facilitate more high-quality diffusion studies.

49 **Keywords:** diffusion modeling, timescale, Ti in quartz, Ca in olivine, Li in zircon

50

## 51 **1. Introduction**

52 Time is one of the fundamental parameters of earth and planetary sciences, where  
53 age and duration are used to put geological events in chronological order and to quantify  
54 rates of geological processes, respectively. An absolute age is generally determined by  
55 radiometric dating, and two ages can be used to bracket the duration or timescale. This  
56 timescale then is used to quantify the rates of geological processes (e.g., [Borg et al.,](#)  
57 [2017](#); [Li et al., 2017](#); [Schoene et al., 2019, 2021](#); [Sprain et al., 2019](#); [Wang et al., 2021](#)).  
58 Radiometric dating is routinely used to determine timescales on the order of a few  
59 million years and, less frequently, tens to a few thousand years or shorter (e.g., [Burgess](#)  
60 [et al., 2014](#); [Li et al., 2017](#); [Thines et al., 2021](#)). For rapid geological events with shorter  
61 timescales (e.g., hours, days, months, and years), constraining their timescales via  
62 radiometric dating is very challenging (or even impossible) and requires ultra-high  
63 temporal resolution. Complementary to absolute dating, timescales also can be  
64 determined through diffusion modeling in minerals (or other phases) regardless of their  
65 absolute ages (i.e., diffusion chronometry; [Lasaga, 1983](#)), with the potential for  
66 diffusion modeling being a function of temperature and time for a given element in a  
67 given mineral ([Chakraborty, 2008](#); [Costa et al., 2020](#); [Watson and Baxter, 2007](#); [Zhang](#)  
68 [and Cherniak, 2010](#)). The diffusivity of different elements in minerals varies  
69 significantly, hence diffusion chronometry can be used to estimate timescales from a  
70 few seconds to several million years ([Costa et al., 2020](#)). Whilst diffusion is a thermally  
71 activated process and does not require any chemical gradients to operate, the presence  
72 of chemical gradients leads to the formation of measurable chemical diffusion profiles.

73 As soon as a chemical gradient is established, the diffusion clock starts. Element  
74 diffusivity depends strongly on temperature, though other factors such as pressure,  
75 crystallographic orientation, the fugacities of water and oxygen, and major element  
76 activities are important in certain cases (e.g., [Ganguly, 2002](#); [Jollands et al., 2016](#); [Kohn  
77 and Penniston-Dorland, 2017](#)). The power of diffusion chronometry can be  
78 significantly increased by combining multiple elements that simultaneously diffuse  
79 with variable diffusivities in a given mineral or in several minerals within the same rock.  
80 These can be integrated to provide a more comprehensive understanding of processes  
81 that the rock has experienced ([Costa et al., 2020](#); [Dohmen et al., 2017](#)).

82 At the temperature–time conditions relevant to magmatic, metamorphic, and  
83 hydrothermal processes, most diffusion in minerals occurs over small length scales (nm  
84 to mm). Analytical techniques with high spatial resolution are therefore required to  
85 measure changes in composition along the diffusion profiles. In the past few decades,  
86 developments in microanalysis techniques such as Fourier-transform infrared  
87 spectroscopy, scanning electron microscope, electron microprobe, laser-ablation  
88 inductively coupled plasma mass spectrometry, secondary ion mass spectrometry, and  
89 local electrode atom probe tomography have accelerated the availability of high-quality  
90 diffusion profiles (e.g., [Audétat et al., 2021](#); [Bloch et al., 2019](#); [Rubatto et al., 2020](#);  
91 [Tang et al., 2017](#)), which have provided critical information to better quantify the  
92 timescales of magma storage, ascent, and eruption as well as ore formation, and  
93 evolution of metamorphic rocks (e.g., [Chu et al., 2018](#); [Cooper, 2019](#); [Devoir et al.,  
94 2021](#); [Li et al., 2022](#); [Mutch et al., 2019](#)).

95 For diffusion studies, Fick's second law lays the foundation for diffusion modeling.  
96 Both numerical solutions (e.g., finite difference) and analytical solutions (e.g., Crank,  
97 1975) can be used to solve Fick's second law for the purposes of diffusion chronometry,  
98 but a lack of widely accepted protocols for diffusion modeling limits the consistency of  
99 different diffusion studies. Hence, a user-friendly program is extremely useful to  
100 standardize the usage of diffusion modeling, facilitate more diffusion studies, and  
101 enable us to better understand the duration, rate, and efficiency of geological events.

102 In addition to timescale estimates, diffusion chronometry requires robust  
103 propagation of uncertainties. The systematic uncertainties of experimentally  
104 determined diffusion coefficients can significantly affect the timescale uncertainties but  
105 are rarely considered in current diffusion studies. This impedes objective comparison  
106 of timescales derived from different elements. Another problem is that the algorithm of  
107 parameter uncertainties in the diffusion coefficients ~~are~~is rarely described in  
108 experimental studies, which introduces known but unquantified uncertainties. Available  
109 programs for diffusion modeling (e.g., Costa et al., 2008; Dunai, 2005; Faryad and  
110 Ježek, 2019; Girona and Costa, 2013; Jollands, 2020; Mutch et al., 2021; Robl et al.,  
111 2007; Smye et al., 2018) do not always estimate the aforementioned uncertainties.  
112 Therefore, a program that uses an explicit curve-fitting method to determine parameters  
113 in the diffusion coefficient and propagates their uncertainties into timescales will make  
114 diffusion studies more internally consistent and reproducible.

115 While all diffusion happens in three-dimensional space, diffusion problems can  
116 often be simplified into one-dimensional models, as long as the extent of diffusion is

117 short relative to the size of the crystal. For demonstration purposes, our program uses  
 118 analytical solutions to solve the one-dimensional model of single-element diffusion  
 119 with uncertainties being propagated using the Monte Carlo method. Our program can  
 120 also be adopted for 2D and 3D scenarios and multi-component systems (i.e., coupled  
 121 diffusion), which will be available in its future versions.

122

## 123 2. Methodology

124 Basic notations used in this study are defined in Table 1.

**Table 1.** Notations used in this study.

Notation	Unit	Description
$x$	m	position along a diffusion profile
$C$	as measured	composition along a diffusion profile
$C_1, C_2, C_3$	as measured	initial compositions of a diffusion profile as defined in Figure 1
erf, erfc		error function (erf) and complementary error function (erfc) used in analytical solutions of a diffusion profile (erfc = 1–erf)
$T$	K	initial temperature
$t$	s	diffusion time
$D_0$	m <sup>2</sup> /s	pre-exponential factor in the Arrhenius equation
$E_a$	J/mol	activation energy in the Arrhenius equation
R	J/(mol·K)	universal gas constant
$D$	m <sup>2</sup> /s	diffusion coefficient

125

### 126 2.1 Analytical solutions to one-dimensional diffusion

127 The goal of a diffusion problem in one dimension is to solve Fick's second law:

$$128 \quad \frac{\partial C}{\partial t} = \frac{\partial}{\partial x} \left( D \frac{\partial C}{\partial x} \right) \quad (1)$$

129 Ideally, this equation can be simplified as follows when  $D$  is independent of  $x$  and

130  $C$  (equation 2.2 in Crank, 1975):

131 
$$\frac{\partial C}{\partial t} = D \frac{\partial^2 C}{\partial x^2} \quad (2)$$

132 Analytical solutions to the above equation for diffusion in semi-infinite and infinite  
133 media are presented by Crank (1975) and adopted directly in Diffuser. For diffusion in  
134 minerals, a semi-infinite medium can be defined as a crystal in contact with an infinite  
135 long reservoir of a fixed composition while an infinite medium can be treated as  
136 chemical zoning in the interior of a crystal. For instance, in a homogeneous crystal  
137 where the boundary is at  $x = x_0$  with an initial condition of  $x = x_0$ ,  $C = C_1$  at the rim and  
138  $x > x_0$ ,  $C = C_2$  in the core (Figure 1A), if diffusion has not appreciably modified the initial  
139 core composition, the diffusion profile can be expressed as:

140 
$$C = (C_2 - C_1) \times \operatorname{erf}\left(\frac{x - x_0}{L}\right) + C_1 \quad (3)$$

141 where  $L$  is the characteristic diffusion length and defined as  $\sqrt{4Dt}$  when  $D$  is  
142 independent of time.

143 It should be emphasized that equation 3 is no longer strictly applicable if diffusion  
144 has erased the original core composition. This rule works for the profiles A–F in Figure  
145 1 of a semi-infinite medium. Other analytical solutions to diffusion profiles A–L in  
146 Figure 1 can be found in Crank (1975) and the Appendix. While analytical solutions for  
147 diffusion profiles A–H in Figure 1 have been compiled in a previous program (PACE),  
148 which focuses on deconvoluting the analytical beam effects (Jollands, 2020), Diffuser  
149 offers analytical solutions to four more forms of diffusion profiles (I–L in Figure 1) and  
150 associated uncertainty estimation of the  $D$  or  $t$  value. To make Diffuser more versatile,  
151 the main functions of PACE (Jollands, 2020) have been incorporated into our program.

152



153 **2.2 Fitting a model to the diffusion profile**

154 Measured profiles are fitted to the diffusion equations by nonlinear least squares  
155 regression (NLS) in MATLAB, which fits functions of the form

156 
$$C = f(x, \beta) + \varepsilon \quad (4)$$

157 where  $x$  and  $C$  are observed values,  $\beta$  is the fitted parameter(s),  $f$  is the fitted model  
158 and  $\varepsilon$  is the residual error of the model (i.e., the difference between the observed  $C$  and  
159 the predicted value). NLS finds the model parameter(s)  $\beta$ , which minimizes the residual  
160 sum of squares (RSS, equation 5) using an iterative optimization technique (e.g.,  
161 Levenberg-Marquardt algorithm used in this study; [Seber and Wild, 2003](#)). If the  
162 uncertainties of  $C$  are given ( $\sigma$ ), the associated weight of each point also can be  
163 considered as  $\frac{1}{\sigma^2}$ :

164 
$$\text{RSS} = \sum_1^n \frac{1}{\sigma_i^2} (C_i - f(x_i, \beta))^2 \quad (5)$$

165 where the subscript  $i$  denotes the  $i^{\text{th}}$  value and  $n$  is the total number of  $i$ .

166 For profiles A–L ([Figure 1](#)), it is straightforward to determine  $C_1$ ,  $C_2$  and  $C_3$  if a  
167 flat compositional platform exists. Although Diffuser can fit a model with  $C_1$ ,  $C_2$  and  
168  $C_3$  as free parameters, this should not be done in the absence of a flat compositional  
169 platform. Specifically, if the flat peak or trough is replaced by a bell shape after  
170 diffusion in profiles I–L of [Figure 1](#) (e.g.,  $t = t_2$ ), the composition associated with the  
171 initial flat peak or trough cannot be determined directly from the composition versus  
172 distance data. In such cases, instead of choosing an arbitrary value, Diffuser lets the  
173 initial composition vary during modeling, and then also considers its effect on the final  
174 timescales. Taking the profile I at  $t = t_2$  ([Figure 1](#)) as an example, because diffusion

175 decreases the height and broadens the peak simultaneously, assuming a higher initial  
176 composition plateau would result in a narrower initial bandwidth (Figure 2A). As  
177 expected, an optimal time value can always be obtained at an assumed flat peak value  
178  $C_0$  (Figure 2B). This means that there exist infinite solutions for the time when a flat  
179 compositional plateau is not recognizable for the profile I. Nevertheless, the modeled  
180 diffusion time converges with increasing  $C_0$  (Figure 2B), so it is still possible to obtain  
181 a time estimate according to a reasonable  $C_0$  or bandwidth, especially if ranges of  $C_0$   
182 can be estimated independently. Taking a diffusion profile of Rubin et al. (2017) as an  
183 example, despite  $C_0$  changing considerably, the goodness of fit ( $R^2$ ) of curve fitting  
184 (Figure 2B) shows a very weak response to the choice of  $C_0$ . As such, the true initial  
185 composition is unknown and using an assumed value will introduce additional  
186 uncertainties. Using variable  $C_0$ , the maximum timescale is nearly constant (~47 years)  
187 when  $C_0 > 300$  ppbw (Figure 2B; see the example section for details). To fully account  
188 for uncertainties introduced by an unknown initial composition, variable initial  
189 compositions are modeled in Diffuser. The same method is applicable to profiles J–L  
190 in Figure 1, when a flat compositional peak or trough cannot be recognized.

191 For profiles K–L in Figure 1, it is noticed that when  $C_3$  is close to the flat peak (or  
192 trough) composition, the diffusion profile will resemble Figure 1G–H after sufficient  
193 time of diffusion (e.g.,  $t_3$  in Figure 1K–L). It should be emphasized that care should be  
194 taken to distinguish these two cases in real studies. For instance, profiles of an element  
195 with a very low diffusion rate could potentially be used to identify the initial boundary  
196 shape.

197

### 198 2.3 Time-dependent diffusion coefficient

199 All the above models apply the parameter  $L$  for curve fitting, which equals  $\sqrt{4Dt}$   
200 when  $D$  is independent of time. Here, we consider the case of a non-constant  $D$  over  
201 time, if, for example, temperature ( $\tau$ ) changes with time ( $t$ ), then

$$202 \quad \tau(t) = F(T, t) \quad (6)$$

203 where  $T$  denotes the initial temperature which is a constant estimated independently  
204 by other methods, e.g., geothermometers, and  $F$  represents a linear, exponential, or  
205 parabolic function.

206 Then, Diffuser employs an isobaric Arrhenius equation for the diffusion coefficient:

$$207 \quad D(t) = D_0 e^{\frac{-E_a}{R\tau(t)}} \quad (7)$$

208 where  $D_0$  and  $E_a$  are determined experimentally.

209 Therefore,  $Dt$  becomes an integral ( $\xi$ ):

$$210 \quad \xi(t) = \int_0^t D(t) dt \quad (8)$$

211 and Fick's second law for equation 2 becomes

$$212 \quad \frac{\partial C}{\partial \xi} = \frac{\partial^2 C}{\partial x^2} \quad (9)$$

213 Equation 9 can be solved analytically by replacing  $Dt$  with  $\xi$  (Crank, 1975). Models  
214 can then be fit to data to obtain  $L$  and  $\xi$  values as:

$$215 \quad L = \sqrt{4\xi} \quad (10)$$

216 The  $t$  value can be obtained by solving integral equation 8 directly in MATLAB  
217 (e.g., using the `fzero` function, a combination of bisection, secant, and inverse quadratic  
218 interpolation methods). Furthermore, Diffuser outputs the parameter  $\xi$  for users to

219 explore further in case of a more complex time-dependent diffusion coefficient (e.g.,  
 220 an abrupt temperature change caused by magma recharge).

221

## 222 **2.4 Uncertainties of the modeled diffusion time**

223 According to previous studies, the uncertainty budget of the modeled diffusion time  
 224 is dominated by the temperature and parameters that control the diffusion coefficient  
 225 (Costa et al., 2008; Costa and Morgan, 2010). Diffuser propagates the uncertainties of  
 226 1) the parameter  $L$  calculated by curve fitting, 2) the initial temperature  $T$  estimated by  
 227 other methods such as geothermometers, 3) and, importantly, the experimentally  
 228 determined diffusion parameters ( $D_0$  and  $E_a$ ) themselves, into the uncertainty of the  
 229 diffusion time. In the case of isothermal diffusion, the relative time uncertainty can be  
 230 calculated directly as follows, assuming that there is negligible or no covariance  
 231 between  $D$  and  $L$ .

$$232 \quad \left(\frac{\sigma_t}{t}\right)^2 = \left(\frac{\sigma_D}{D}\right)^2 + \left(\frac{2\sigma_L}{L}\right)^2 \quad (11)$$

233 where  $\sigma_t$ ,  $\sigma_D$  and  $\sigma_L$  represent absolute uncertainties of  $t$ ,  $D$ , and  $L$  respectively.  $\sigma_D$   
 234 is calculated as:

$$235 \quad \sigma_D = \sqrt{J'\Sigma J} \quad (12)$$

236 where  $\Sigma$  is the covariance matrix,  $J$  is the Jacobian matrix and  $J'$  is its transpose:

$$237 \quad \Sigma = \begin{bmatrix} (\sigma_{\ln[D_0]})^2 & \text{Cov}(\ln[D_0], E_a) & 0 \\ \text{Cov}(\ln[D_0], E_a) & (\sigma_{E_a})^2 & 0 \\ 0 & 0 & (\sigma_T)^2 \end{bmatrix} \quad (13)$$

$$238 \quad J = D \begin{bmatrix} 1 \\ -\frac{1}{RT} \\ \frac{E_a}{RT^2} \end{bmatrix} \quad (14)$$

239 where  $\sigma_{\ln[D_0]}$ ,  $\sigma_{E_a}$  and  $\sigma_T$  represent absolute uncertainties of  $\ln[D_0]$ ,  $E_a$ , and  $T$   
240 respectively.  $\text{Cov}(\ln[D_0], E_a)$  is the covariance between  $\ln[D_0]$  and  $E_a$ .  $\sigma_{\ln[D_0]}$ ,  $\sigma_{E_a}$  and  
241  $\text{Cov}(\ln[D_0], E_a)$  are calculated in Diffuser by refitting the original experimental data ( $D$   
242 at different  $T$ ) by linear least-squares regression (e.g., [Montgomery and Runger, 2018](#)).

243 If the temperature varies with time (e.g., during cooling), a Monte Carlo method is  
244 introduced to estimate the time uncertainty. First, a data set is generated consisting of  
245  $L_i$  and  $T_i$  values, which follows a Gaussian distribution with uncertainties of  $\sigma_L$  and  $\sigma_T$ ,  
246 respectively. Second, the diffusion time  $t_i$  is calculated using equations 6–10 for each  
247 paired  $L_i$  and  $T_i$ . Finally, we calculate the average  $t$  value and its uncertainty from all  $t_i$   
248 results. Because  $t_i$  values are log-normally distributed based on the Lilliefors test ([Figure](#)  
249 [3](#); [Lilliefors, 1967](#)), we calculate the average  $\ln[t]$  value and its uncertainty and then  
250 convert them into  $t$ , yielding a geometric mean value with upper and lower boundaries  
251 at a 95% confidence level. The numerical stability of the geometric mean and  
252 confidence interval increases with the number of iterations. Taking a diffusion profile  
253 of [Rubin et al. \(2017\)](#) as an example, repeating Monte Carlo modeling 15 times using  
254 100, 500, 1000, 2000, 3000, 4000, and 5000 trials, respectively, shows a convergence  
255 of  $t$  and its uncertainty (range <1 year) when the number of trials is larger than 2000  
256 ([Supplementary Figure S1](#)). In some cases, the distribution of  $\ln[t]$  values does not  
257 follow a normal distribution, and more trials are needed to obtain robust uncertainty  
258 estimates.

259

## 260 **3. Program design**

261 Diffuser was written in MATLAB and comes with an intuitive graphical user  
262 interface (GUI) that can either be used offline (on Windows, Mac or Linux); or online  
263 (in any HTML-5 compatible web browser) at <http://www.geoapp.cn>. The latter is our  
264 suggested method for accessing the program regardless of the computer platform being  
265 used. A brief introduction of how to use this program interactively is given below. A  
266 more detailed manual also is provided along with the program.

267

### 268 **3.1 Data input**

269 The measured diffusion profile can be fed into the software through a delimited  
270 text (e.g., txt or csv format), spreadsheet file (Microsoft® Excel), or clipboard on the  
271 data input panel (Figure 4A). The distance and composition data should be in two  
272 columns. If the compositional uncertainties ( $\sigma$ ) are assigned in another column, Diffuser  
273 asks the user to input the column names of  $x$ ,  $C$ , and  $\sigma$ . Otherwise, all compositional  
274 data are treated with equal weights. After data import, the data is directly visualized by  
275 means of a pre-formatted plot (Figure 4A).

276

### 277 **3.2 Diffusion modeling**

278 After importing the diffusion data, the user can choose the relevant solution to  
279 Fick's second law graphically, on the diffusion modeling panel ('Diffusion profile' in  
280 Figure 4B) by comparing the shape of the measured data profile with the diffusion types  
281 presented in Figure 1. The unit of position  $x$  should be set and initial compositions ( $C_1$ ,

282  $C_2$ , and  $C_3$ ) can be fixed or determined by fitting. Specifically, if the user does not define  
283 an initial flat peak or trough for profiles I–L in [Figure 1](#), more parameters are required,  
284 including minimum and maximum values and a step length of the assumed flat peak or  
285 trough ( $C_0$ ). Then, the user can start a diffusion model to obtain a modeled  $Dt$  value.

286

### 287 **3.3 Time and diffusion coefficient calculation**

288 To model timescales in natural samples, the diffusion coefficient, initial  
289 temperature with its uncertainty, trials for the Monte Carlo calculation, and cooling path  
290 with its constant coefficient should be set ([Figure 4C](#)). Additionally, diffusion  
291 coefficient data collected from the literature are provided in an embedded Excel file  
292 (see below). The program reads these values automatically and users can access them  
293 via a drop-list on the diffusion coefficient panel ([Figure 4C](#)) once the program interface  
294 is opened. At the time of release, Diffuser contains a library of diffusion coefficients for  
295 the following systems: olivine (Ca, Al, P, REE, Ti, H, Li, Be), garnet (Hf, REE), quartz  
296 (Ti, Al, H), zircon (REE, Ti, Al, Li), orthopyroxene (REE, Ti, Cr, H), clinopyroxene (Ti,  
297 REE, H), and feldspar (Sr, Ba, REE, H). These have been compiled in Diffuser by  
298 refitting original experimental data in previous studies. Thus, the algorithm of  
299 parameter uncertainties in Diffuser is internally consistent. Diffusion coefficients from  
300 other systems can be added [intoto](#) Diffuser by modifying the relevant template file.  
301 Requests to add other diffusion coefficients to future versions of Diffuser can be sent  
302 to the first author.

303 To calculate diffusion coefficients in isothermal experimental studies, the user can

304 enter the experimental duration and get an estimated  $D$  with its uncertainty (Figure 4D).  
305 Then, the parameters of  $\ln[D_0]$  and  $E_a$  in the diffusion coefficient and their covariance  
306 can be calculated by ordinary linear least-squares regression of experimental data ( $D$  at  
307 different  $T$ ).

308

### 309 **3.4 Data output**

310 There are two basic types of output in our program after each modeling process:  
311 first, a figure showing the curve-fitting result, which can be saved as vector graphics  
312 (e.g., Figure 5A); and second, a text or spreadsheet file recording the calculation result  
313 (including information of trials,  $R^2$ ,  $C_1$ ,  $C_2$ ,  $C_3$ ,  $x_0$ ,  $Dt$ ,  $D$ ,  $t$ , and  $t$  uncertainty). The figure  
314 compares the measured diffusion profiles directly with modeled zoning profiles (e.g.,  
315 Figure 5A). If the temperature is assigned an uncertainty, a marginal plot shows  
316 distributions of the temperature and diffusion timescale and the trade-off between these  
317 two parameters (e.g., Figure 5B). The uncertainty budget of the diffusion time estimate  
318 is broken down into a sequence of histograms that show the contributions of curve  
319 fitting, temperature, and experimental parameters in the diffusion coefficient (e.g.,  
320 Figure 5C).

321 Especially, during calculation for profiles I–L in Figure 1, if there is no flat peak or  
322 trough defined by the user, then an additional figure is created to show the process of  
323 estimating timescales with an assumed composition range of the initial flat peak/trough  
324 (e.g., Figure 2B). Furthermore, if the temperature decreases with time (non-isothermal  
325 systems), Monte Carlo results of timescales are displayed on histograms to show



326 whether they are log-normally distributed based on [the](#) Lilliefors test (e.g., [Figure 3](#)).  
327 The Monte Carlo results and histograms can be saved so that users can evaluate the  
328 calculation process.

329

## 330 **4. Examples**

331 We will demonstrate the functionality, convenience and efficiency of Diffuser with  
332 three examples.

### 333 **4.1 Ti diffusion in quartz**

334 Quartz is a common mineral formed in magmatic, metamorphic, and hydrothermal  
335 processes. It often has a large crystal size ( $\gg 100 \mu\text{m}$ ) and displays clear trace element  
336 zoning (e.g., Ti and Al), and thus has been extensively used for diffusion modeling (e.g.,  
337 [Ackerson et al., 2018](#); [Cernuschi et al., 2018](#); [Li et al., 2022](#); [Spear et al., 2012](#)). For  
338 example, a titanium diffusion profile in quartz from the Valles caldera, U.S.A. ([Boro et](#)  
339 [al., 2021](#)) yielded a timescale of  $180 \pm 15$  years between magma recharge and volcanic  
340 eruption, using the Ti-in-quartz diffusivity of [Cherniak et al. \(2007\)](#) and an assumed  
341 temperature of  $750 \pm 20$  °C. Diffuser gives a timescale of  $195_{-131}^{+399}$  years which is  
342 identical within uncertainty to the time determined by [Boro et al. \(2021\)](#) using a finite  
343 difference method ([Figure 5A–B](#)). However, the uncertainty is an order of magnitude  
344 larger than that previously estimated. It also can be seen in Diffuser that the main  
345 contribution to the time uncertainty is curve fitting ( $\sim 72\%$ , [Figure 5C](#)). In addition, a  
346 better way to define the initial diffusion interface position ( $x_0$ ) is by measuring as many  
347 elements as possible and studying transects of slow and fast diffusing elements. In

348 reality, this is not always possible. Instead of adopting an interface position visually  
349 from the diffusion profile, Diffuser determines it directly by curve fitting when an  
350 element with a lower diffusivity is unavailable, which is at least convenient and more  
351 reproducible.

352 It should be noted that a recent experimental study of [Jollands et al. \(2020\)](#) yielded  
353 a much lower Ti-in-quartz diffusivity than that proposed by [Cherniak et al. \(2007\)](#).  
354 Using this new Ti-in-quartz diffusivity will give timescales about three orders of  
355 magnitude longer ( $224_{-185}^{+1064}$  kyr). This has promoted discussions about addressing the  
356 discrepancy of time-scales derived from the two Ti-in-quartz diffusivities ([Audétat et](#)  
357 [al., 2021](#); [Boro et al., 2021](#); [Gualda and Pamukçu, 2020](#)), which is beyond the scope of  
358 this study and not discussed further.

359

## 360 **4.2 Ca diffusion in olivine**

361 Olivine is a common phenocryst in basalt and has been widely used to constrain  
362 timescales between magma intrusion and eruption, which provides important  
363 information for volcano monitoring and forecasting. Furthermore, multiple elements  
364 including rare earth elements, Ti, P, Cr, Ca, Ni, Fe–Mg, Li, Be, and H in a single olivine  
365 crystal enable verifying diffusion and crystal-growth zoning models (see reviews in  
366 [Costa et al, 2020](#) and [Costa, 2021](#)). Although diffusion of major elements in olivine  
367 (e.g., Fe–Mg) shows multi-component effects that generally cannot be modeled using  
368 simple analytical solutions of the diffusion equation, trace elements in olivine that have  
369 composition-independent diffusivities can be modeled by Diffuser. For example, a Ca

370 diffusion profile in olivine from the IODP Hole U1309D (Ferrando et al., 2020) records  
371 a timescale of  $150 \pm 40$  years for melt-rock interaction, using the Ca-in-olivine  
372 diffusivity of Coogan et al. (2005), an initial temperature of  $1230 \pm 20$  °C and linear  
373 cooling rate of 0.004 °C/year. Diffuser recovers a timescale of  $135_{-53}^{+86}$  years with a  
374 fixed initial CaO composition of 0.065 wt.% at the rim and 0.105 wt.% in the core  
375 (Figure 6A–B). This result is identical within uncertainty to the time determined by  
376 Ferrando et al. (2020) using a 3D finite difference method. It also can be seen in  
377 Diffuser that the main contribution to the time uncertainty is curve fitting (~80%, Figure  
378 6C). Uncertainties of experimentally determined diffusion coefficients are systematic  
379 and should be considered when comparing diffusion studies using different elements,  
380 or using the same element but different diffusion coefficients.

381

### 382 4.3 Li diffusion in zircon

383 Zircon is an invaluable mineral for understanding igneous, metamorphic, and  
384 sedimentary environments, as highlighted by the wide utilization of its U–Pb ages, trace  
385 elements, and Hf–O–Si–Zr isotopes. Trace element diffusion in zircon that recovers  
386 magmatic residence time is making it more versatile (e.g., Bloch et al., 2022; Cherniak,  
387 2021; Cherniak and Watson, 2010; Cherniak et al., 2007; Trail et al., 2016). For example,  
388 a Li diffusion profile in zircon from the Taupo Volcanic Zone, New Zealand (Rubin et  
389 al., 2017) records a timescale of 22 years between magma injection and eruption, using  
390 the Li-in-zircon diffusivity of Trail et al. (2016) and a fixed temperature of  $700 \pm 20$  °C.  
391 The modeling result of Diffuser ( $23_{-13}^{+30}$  years) is identical to the finite difference

392 method of Rubin et al. (2017) when initial Li compositions are fixed at 0 ppbw outside  
393 the band and 120 ppbw in the band (Figure 7A–B). It also can be seen that the major  
394 contribution of the time uncertainty comes from the temperature (~60%, Figure 7C).  
395 However, because Li diffusion has decreased the initial peak which is now unknown,  
396 Rubin et al. (2017) assumed the maximum measured Li composition (~120 ppbw) to  
397 represent the initial composition plateau. Such an assumption is plausible but very hard  
398 to test. To evaluate the effect of varying initial composition on final timescales, assumed  
399 initial Li ~~composition~~compositions ranging from 120 to 600 ppbw are modeled in  
400 Diffuser. The results show that obtained timescales range from  $23_{-13}^{+30}$  years to  $47_{-26}^{+60}$   
401 years and have become stable when the maximum Li composition goes beyond ~300  
402 ppbw (Figure 2B). Therefore, it is more reasonable to propagate these uncertainties and  
403 use a time range (e.g., 10–107 years) rather than a single value to represent the diffusion  
404 time. A more precise time estimate can be obtained if a reasonable plateau value or  
405 bandwidth can be constrained. For example, we can consider an upper limit of Li  
406 composition in natural zircon for the plateau or use another element with a very low  
407 diffusion coefficient (e.g., Hf, Th, and U; Cherniak et al., 1997) to determine the initial  
408 bandwidth.

409 As emphasized by Trail et al. (2016), when Li zoning in zircon is utilized for  
410 diffusion chronometry, it is important to evaluate whether Li compositions correlate  
411 with the REE and P, which may have charge balance effects related to zircon growth.  
412 The potential for such coupled diffusion to modify derived timescales should be  
413 evaluated carefully in future diffusion studies.

414

## 415 **5. Conclusion**

416 Diffuser is a program designed for diffusion chronometry. In particular, it is coded  
417 with robust uncertainty propagation of curve fitting, temperature, and experimentally  
418 determined diffusion coefficients. The code offers an intuitive and user-friendly GUI to  
419 control the data import, diffusion modeling, time calculation, and diffusivity calculation.  
420 The program conducts an automatic curve fit and provides fitting parameters with  
421 associated uncertainties. Data output includes figures showing curve-fitting results and  
422 text or spreadsheet files recording modeling results which can be saved as vector  
423 graphics and used for further processing, respectively. Accelerated by improvements in  
424 microanalysis techniques, diffusion profiles at microscales now are readily obtained.  
425 Together with the easy-to-handle program to generate diffusion models of multiple  
426 elements and multiple minerals, studies on diffusion timescales are expected to become  
427 more common for a more comprehensive understanding on the duration and rates of  
428 geological processes.

429

## 430 **Authorship statement**

431 Y. Li designed the project with inputs from P. Vermeesch, M. Jollands and X.H. Li; L.G. Wu and Y.  
432 Li developed the methodology and software, and wrote the initial draft. All authors contributed to  
433 the revisions, led by Y. Li.

434

## 435 **Declaration of competing interest**

436 The authors declare that they have no known competing financial interests nor personal relationships

437 that could have appeared to influence the work reported in this paper.  
438

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446 reviewers and editor Pierre Lanari for very constructive comments that significantly improved the  
447 quality and presentation of this paper and the program.  
448

## 449 **Computer code availability**

450 The source code and user manual of Diffuser can be downloaded from  
451 <https://github.com/liguangwu/diffuser.git> and a web version of Diffuser is available at  
452 <http://www.geoapp.cn/>. Comments, bug reports, and requests for extra diffusion coefficients to be  
453 added to the database can be sent to the authors via [wuliguang@mail.iggcas.ac.cn](mailto:wuliguang@mail.iggcas.ac.cn) (L.G. Wu) and  
454 [geolij@outlook.com](mailto:geolij@outlook.com) (Y. Li).  
455

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614 Geochemistry*, Boston.

615 **List of Figures**

616 **Figure 1.** **A–D**, half diffusion profiles from rim to core in a crystal; **E–F**, full diffusion  
617 profiles from rim to rim in a crystal. **G–H**, diffusion profiles for chemical zoning with  
618 an initial sharp boundary in a crystal; **I–L**, diffusion profiles for chemical zoning with  
619 an initially confined band in a crystal. The grey dashed line shows the initial boundary  
620 and composition conditions at time  $t_0$  ( $t_0 = 0$ ) and the blue solid line shows the diffusion  
621 profile at time  $t_1$ ,  $t_2$ , and  $t_3$  ( $t_0 < t_1 < t_2 < t_3$ ). The right panel shows composition distributions  
622 within minerals before ( $t = t_0$ ) and after diffusion ( $t = t_1$ ) which corresponds to profiles  
623 A–L, respectively. The black arrow shows the direction of the diffusive flux.

624

625 **Figure 2.** **A**, a schematic diagram showing that different values of the initial flat peak  
626 and bandwidth can have a same diffusion profile. **B**, modeled diffusion time with an  
627 assumed composition range of the initial flat peak using an example data from [Rubin](#)  
628 [et al. \(2017\)](#). Uncertainties of the curve fitting, temperature and experimentally  
629 determined diffusion coefficient are propagated into the uncertainty of timescales.

630

631 **Figure 3.** Distribution of  $\ln[t]$  and timescales calculated by the Monte Carlo method  
632 (with an artificial cooling rate of 200 °C/Ma and temperature of  $700 \pm 20$  °C) using an  
633 example data from [Rubin et al. \(2017\)](#). Trials are 5000 in total.

634

635 **Figure 4.** A screenshot of Diffuser interface. **A**, data import panel. **B**, diffusion  
636 modeling panel. **C**, time calculation panel. **D**, diffusivity calculation panel.

637

638 **Figure 5.** An example of Ti-in-quartz diffusion modeling in Diffuser using raw data  
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640 showing distributions of the temperature and diffusion timescale and the trade-off  
641 between these two parameters. The superscript a and b denote timescales calculated by  
642 the finite difference method from [Boro et al. \(2021\)](#) and by Diffuser, respectively. **C**,  
643 histogram showing the uncertainty budget of modeled timescales.

644

645 **Figure 6.** An example of Ca-in-olivine diffusion modeling in Diffuser using raw data  
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648 between these two parameters. The superscript a and b denote timescales calculated by  
649 the finite difference method from [Ferrando et al. \(2020\)](#) and by Diffuser, respectively.  
650 **C**, histogram showing the uncertainty budget of the modeled timescale.

651

652 **Figure 7.** An example of Li-in-zircon diffusion modeling in Diffuser using raw data  
653 from [Rubin et al. \(2017\)](#). **A**, curve fitting result of the diffusion profile. **B**, marginal plot  
654 showing distributions of the temperature and diffusion timescale and the trade-off  
655 between these two parameters. The superscript a and b denote timescales calculated by  
656 the finite difference method from [Rubin et al. \(2017\)](#) and by Diffuser, respectively. **C**,  
657 histogram showing the uncertainty budget of the modeled timescale.

658

659

## 660 Appendix

661 Analytical solutions to diffusion in semi-infinite and infinite media are listed below.

### 662 1 Diffusion in a semi-infinite medium

663 For a rim-to-core profile with an initial condition of  $x = x_0$ ,  $C = C_1$  and  $x > x_0$ ,  $C = C_2$

664 (Figure 1A):

$$665 \quad C = (C_2 - C_1) \times \operatorname{erf}\left(\frac{x - x_0}{L}\right) + C_1 \quad (\text{A1})$$

666 For a rim-to-core profile with an initial condition of  $x < x_0$ ,  $C = C_2$  and  $x = x_0$ ,  $C = C_1$

667 (Figure 1B):

$$668 \quad C = (C_2 - C_1) \times \operatorname{erf}\left(\frac{-x + x_0}{L}\right) + C_1 \quad (\text{A2})$$

669 For a rim-to-core profile with an initial condition of  $x = x_0$ ,  $C = C_2$  and  $x > x_0$ ,  $C = C_1$

670 (Figure 1C; analogous to equation 2.45 in Crank, 1975):

$$671 \quad C = (C_2 - C_1) \times \operatorname{erfc}\left(\frac{x - x_0}{L}\right) + C_1 \quad (\text{A3})$$

672 For a rim-to-core profile with an initial condition of  $x < x_0$ ,  $C = C_1$  and  $x = x_0$ ,  $C = C_2$

673 (Figure 1D):

$$674 \quad C = (C_2 - C_1) \times \operatorname{erfc}\left(\frac{-x + x_0}{L}\right) + C_1 \quad (\text{A4})$$

675 For a rim-to-rim profile with an initial condition of  $(x_0 - W) < x < (x_0 + W)$ ,  $C = C_2$  and  $x \leq$

676  $(x_0 - W)$  or  $x \geq (x_0 + W)$ ,  $C = C_1$  (Figure 1E):

$$677 \quad C = (C_2 - C_1) \times \left[ \operatorname{erf}\left(\frac{W + x - x_0}{L}\right) + \operatorname{erf}\left(\frac{W - x + x_0}{L}\right) \right] + 2C_1 - C_2 \quad (\text{A5})$$

678 where  $W$  is the rim-to-core length.

679 For a rim-to-rim profile with an initial condition of  $(x_0 - W) < x < (x_0 + W)$ ,  $C = C_1$  and  $x \leq$

680  $(x_0 - W)$  or  $x \geq (x_0 + W)$ ,  $C = C_2$  (Figure 1F):

681 
$$C = (C_2 - C_1) \times \left[ \operatorname{erfc}\left(\frac{W+x-x_0}{L}\right) + \operatorname{erfc}\left(\frac{W-x+x_0}{L}\right) \right] + C_1 \quad (\text{A6})$$

682 **2 Diffusion in an infinite medium**

683 For an initial sharp boundary when the position of the initial diffusion interface is fixed  
 684 at  $x = x_0$  (Figure 1G), if an initial high composition ( $C_2$ ) lies on the right side,

685 
$$C = \frac{C_2 - C_1}{2} \times [1 + \operatorname{erf}\left(\frac{x-x_0}{L}\right)] + C_1 \quad (\text{A7})$$

686 If an initial high composition ( $C_2$ ) lies on the left side (Figure 1H; analogous to equation  
 687 2.14 in Crank, 1975),

688 
$$C = \frac{C_2 - C_1}{2} \times \operatorname{erfc}\left(\frac{x-x_0}{L}\right) + C_1 \quad (\text{A8})$$

689 For an initially confined region of  $(x_0-h) < x < (x_0+h)$ , if the diffusion is symmetric with  
 690 an initial high composition ( $C_2$ ) in the region (Figure 1I; analogous to equation 2.15 in  
 691 Crank, 1975),

692 
$$C = \frac{C_2 - C_1}{2} \times \left[ \operatorname{erf}\left(\frac{h-x+x_0}{L}\right) + \operatorname{erf}\left(\frac{h+x-x_0}{L}\right) \right] + C_1 \quad (\text{A9})$$

693 where  $h$  is the half bandwidth.

694 If the diffusion profile is symmetric with an initial low composition ( $C_1$ ) in the region  
 695 (Figure 1J),

696 
$$C = \frac{C_2 - C_1}{2} \times \left[ \operatorname{erfc}\left(\frac{h-x+x_0}{L}\right) + \operatorname{erfc}\left(\frac{h+x-x_0}{L}\right) \right] + C_1 \quad (\text{A10})$$

697 If the diffusion profile is asymmetric with an initial high composition ( $C_2$ ) in the region  
 698 (Figure 1K),

699 
$$C = \frac{C_2 - C_3}{2} \times \operatorname{erf}\left(\frac{h-x+x_0}{L}\right) + \frac{C_2 - C_1}{2} \times \operatorname{erf}\left(\frac{h+x-x_0}{L}\right) + \frac{C_3}{2} + \frac{C_1}{2} \quad (\text{A11})$$

700 If the diffusion profile is asymmetric with an initial low composition ( $C_1$ ) in the region  
 701 (Figure 1L),

$$702 \quad C = \frac{c_3 - c_1}{2} \times \operatorname{erfc}\left(\frac{h-x+x_0}{L}\right) + \frac{c_2 - c_1}{2} \times \operatorname{erfc}\left(\frac{h+x-x_0}{L}\right) + C_1 \quad (\text{A12})$$

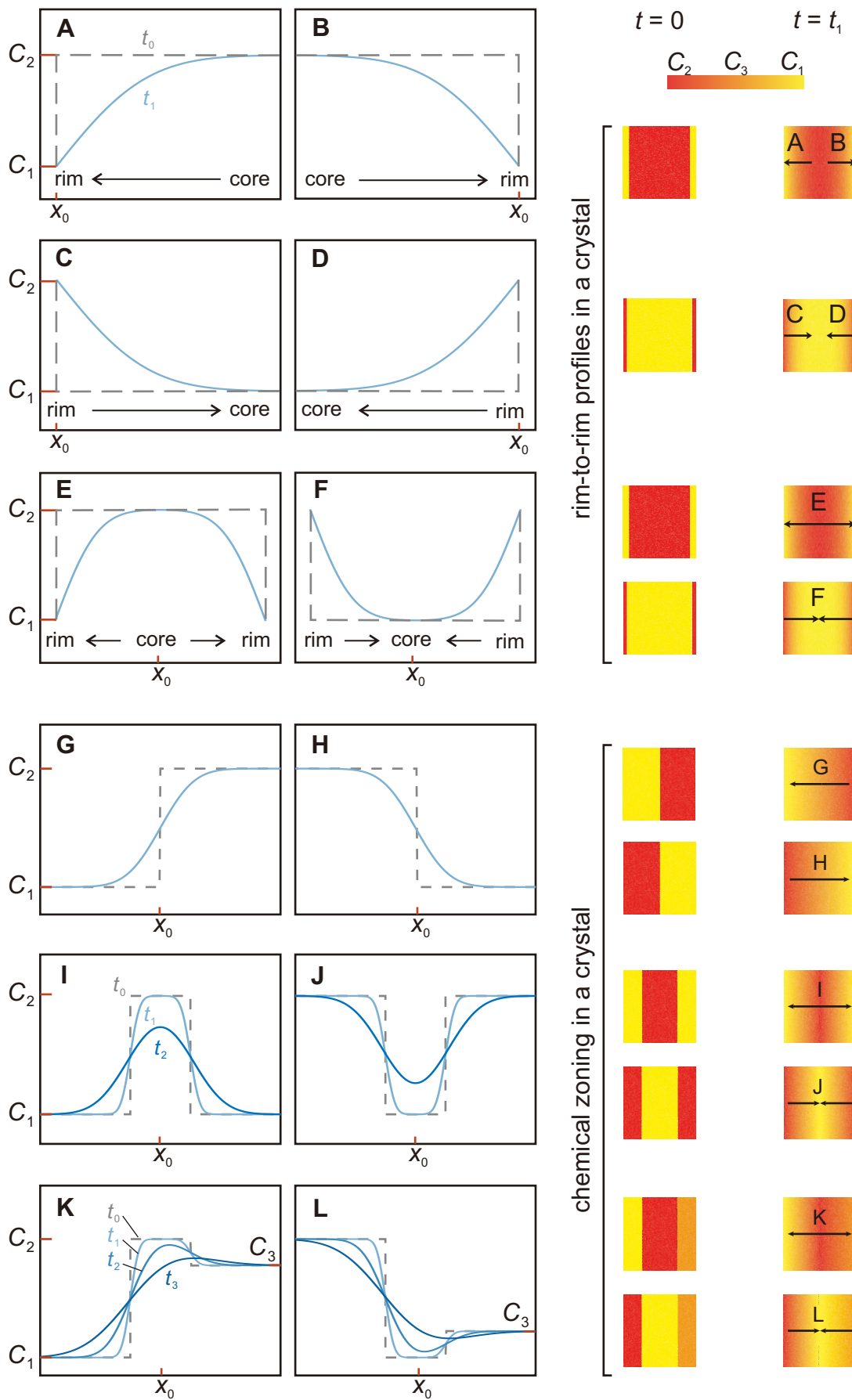


Figure 1



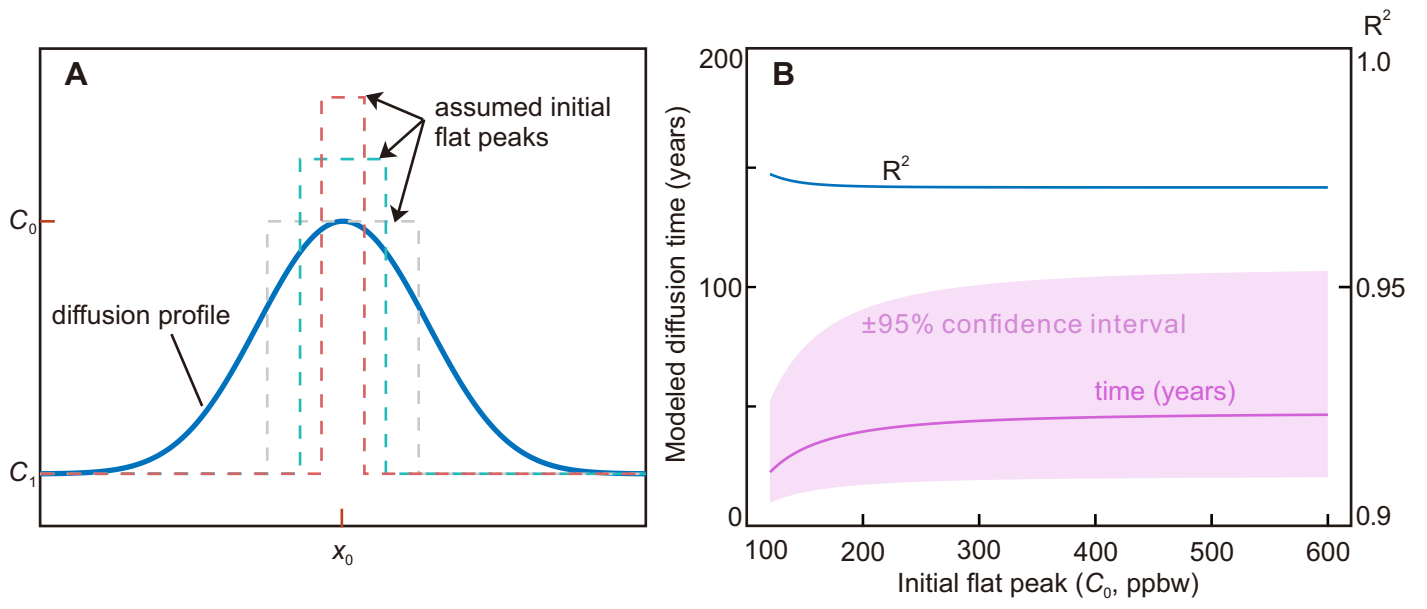


Figure 2

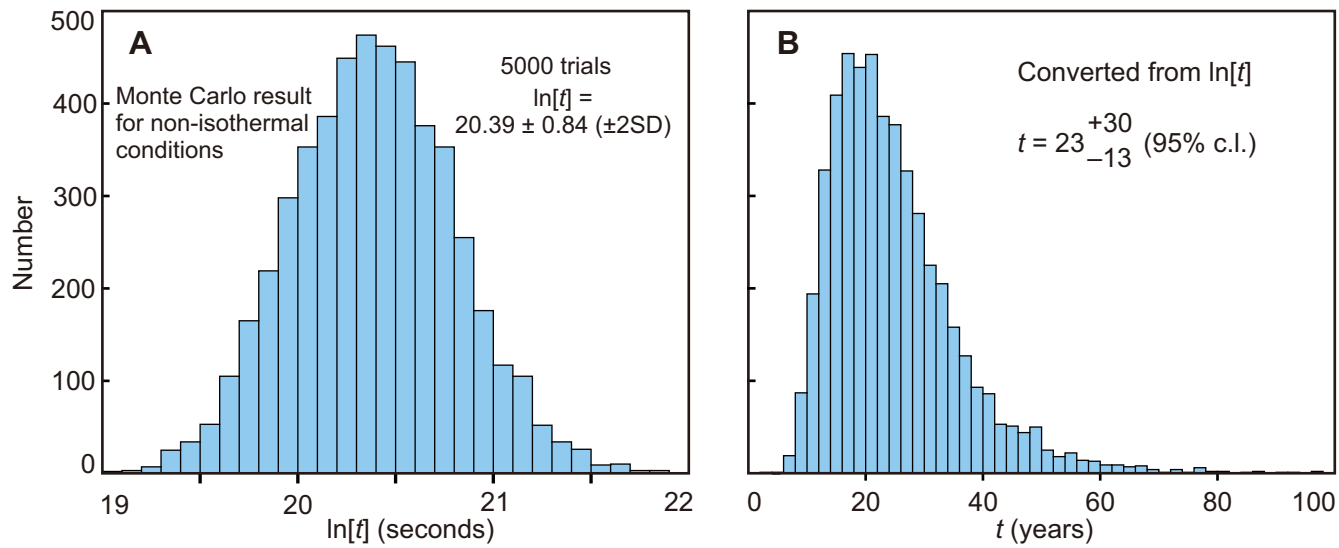
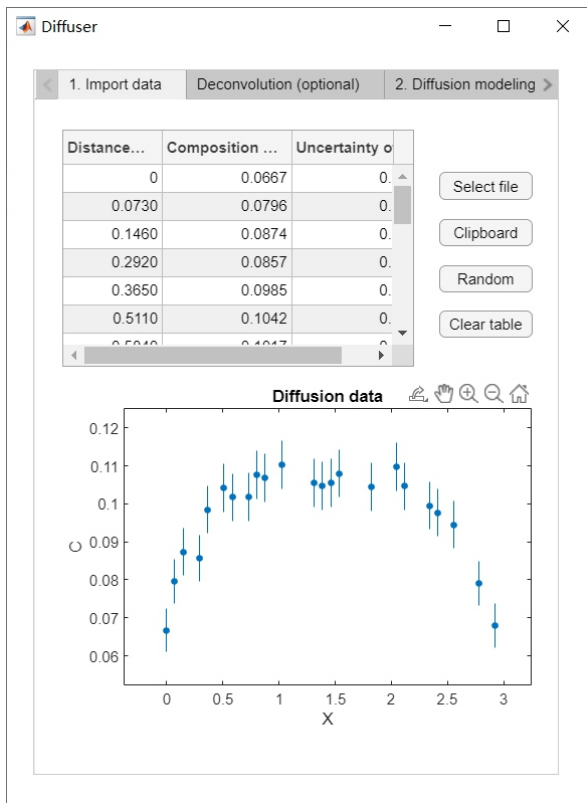
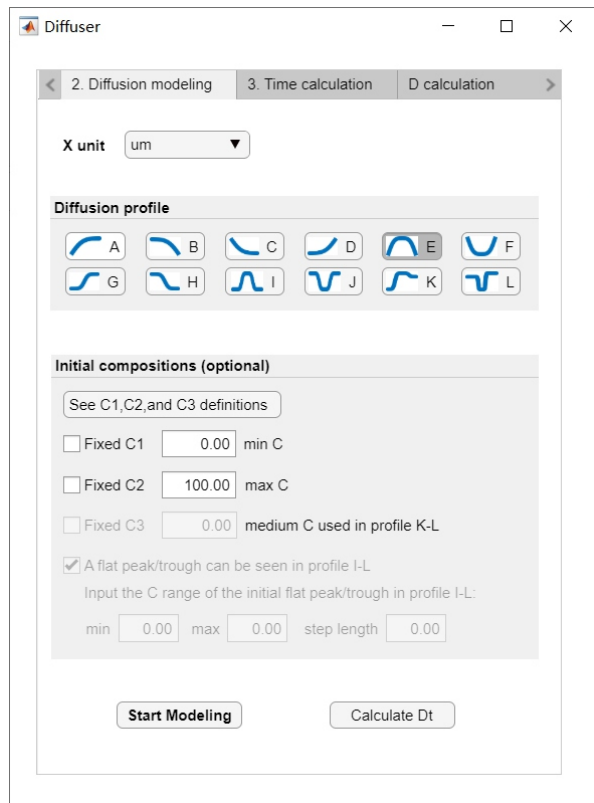


Figure 3

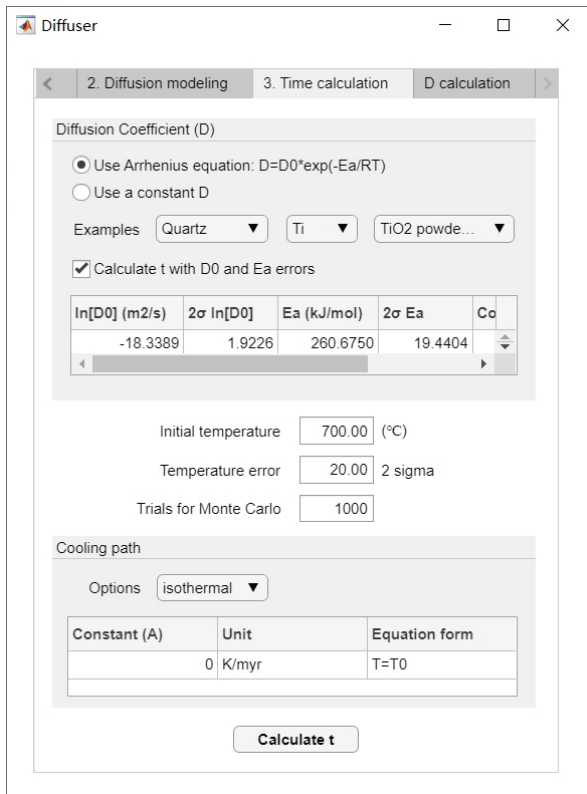
## A. data import



## B. diffusion modeling



## C. time calculation



## D. diffusivity calculation

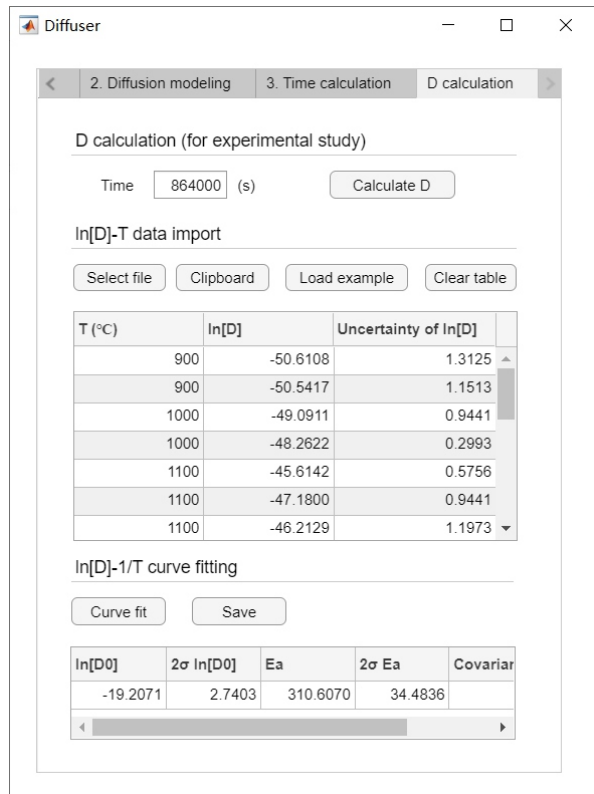


Figure 4

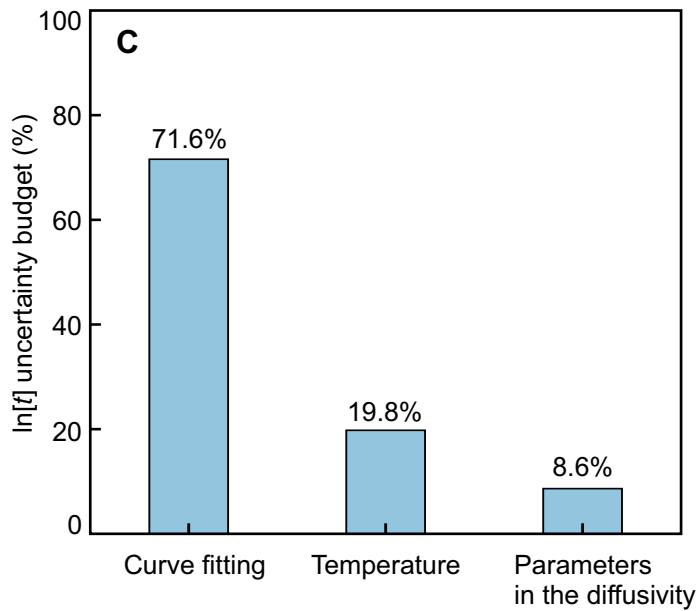
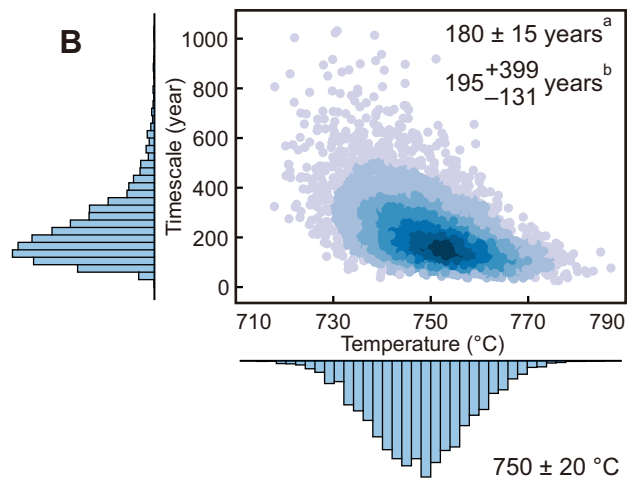
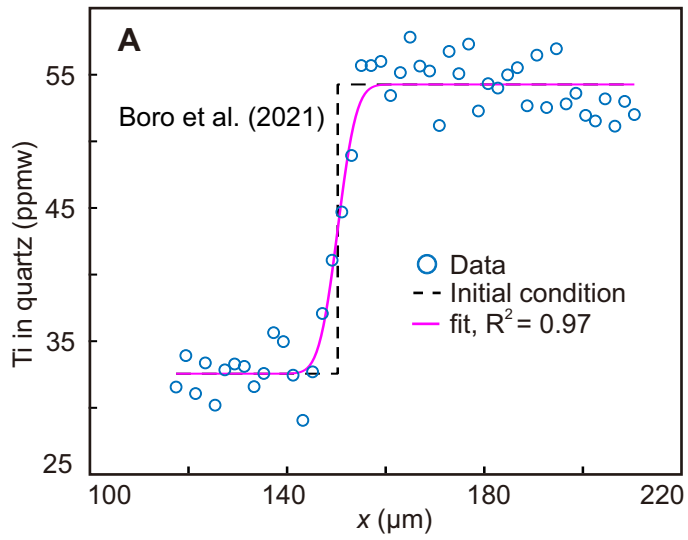


Figure 5

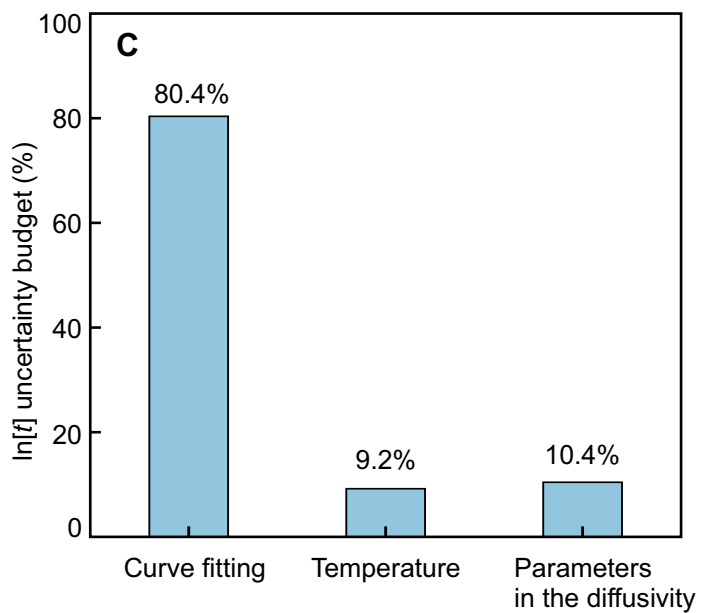
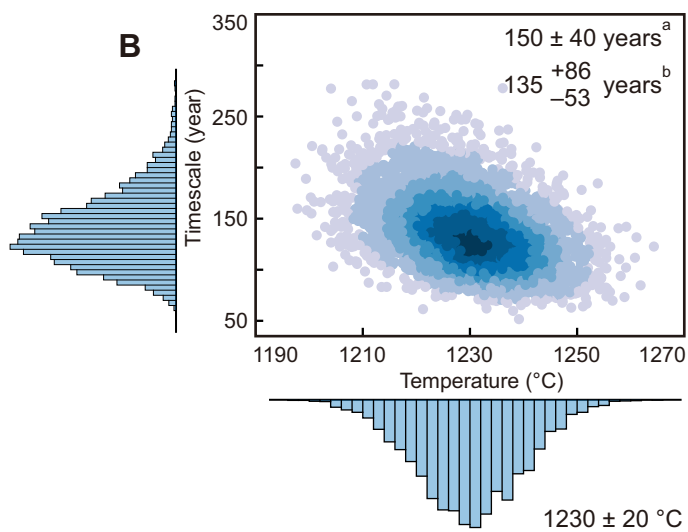
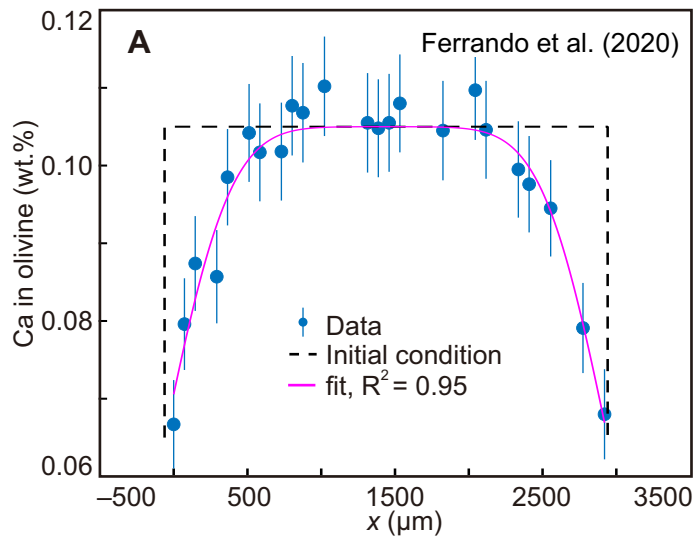


Figure 6

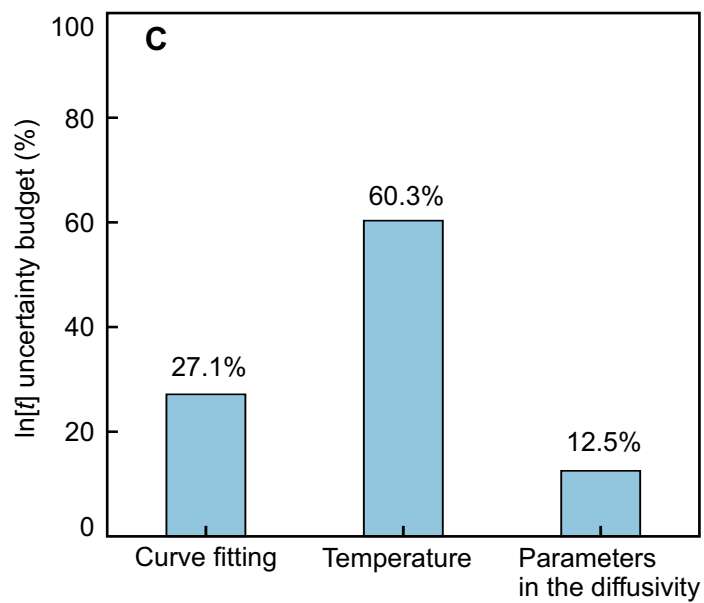
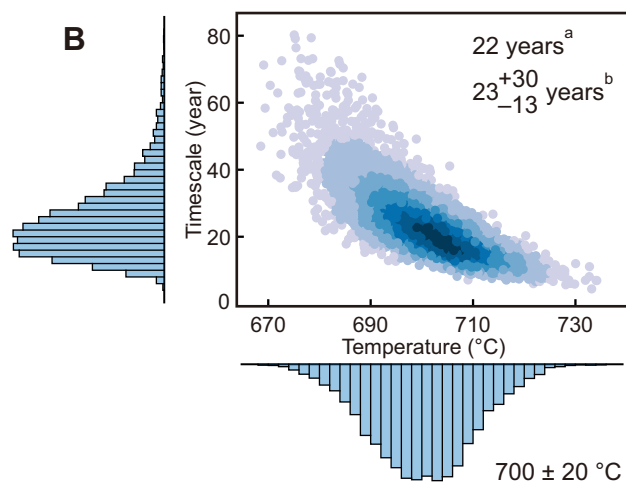
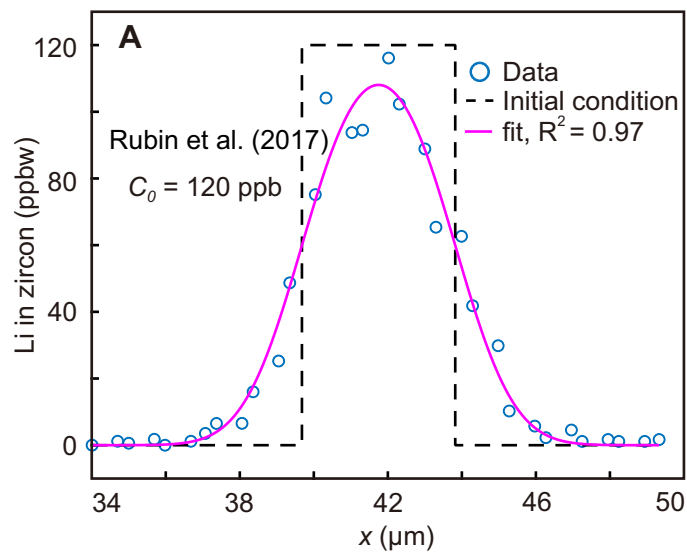


Figure 7