ACCEPTED MANUSCRIPT • OPEN ACCESS

Vibrationally resolved NO dissociative excitation cross sections by electron impact

To cite this article before publication: Vincenzo Laporta *et al* 2020 *Plasma Sources Sci. Technol.* in press <u>https://doi.org/10.1088/1361-6595/ab86d8</u>

Manuscript version: Accepted Manuscript

Accepted Manuscript is "the version of the article accepted for publication including all changes made as a result of the peer review process, and which may also include the addition to the article by IOP Publishing of a header, an article ID, a cover sheet and/or an 'Accepted Manuscript' watermark, but excluding any other editing, typesetting or other changes made by IOP Publishing and/or its licensors"

This Accepted Manuscript is © 2020 The Author(s). Published by IOP Publishing Ltd..

As the Version of Record of this article is going to be / has been published on a gold open access basis under a CC BY 3.0 licence, this Accepted Manuscript is available for reuse under a CC BY 3.0 licence immediately.

Everyone is permitted to use all or part of the original content in this article, provided that they adhere to all the terms of the licence https://creativecommons.org/licences/by/3.0

Although reasonable endeavours have been taken to obtain all necessary permissions from third parties to include their copyrighted content within this article, their full citation and copyright line may not be present in this Accepted Manuscript version. Before using any content from this article, please refer to the Version of Record on IOPscience once published for full citation and copyright details, as permissions may be required. All third party content is fully copyright protected and is not published on a gold open access basis under a CC BY licence, unless that is specifically stated in the figure caption in the Version of Record.

View the article online for updates and enhancements.

Vibrationally resolved NO dissociative excitation cross sections by electron impact

V. Laporta,^{1,*} J. Tennyson,² and I.F. Schneider^{3,4}

¹Istituto per la Scienza e Tecnologia dei Plasmi, CNR, Bari, Italy

²Department of Physics and Astronomy, University College London, London, UK

³Laboratoire Ondes et Milieux Complexes, CNRS-Université Le Havre Normandie, Le Havre, France

⁴Laboratoire Aimé Cotton, CNRS–Université Paris-Saclay, ENS Cachan, Orsay, France

A theoretical investigation of the dissociative excitation by electron impact on the NO molecule is presented, aiming to make up for the lack of data for this process in the literature. A full set of vibrationally-resolved cross sections and corresponding rate coefficients are calculated using the Local-Complex-Potential approach and five resonant states of NO⁻.

PACS numbers: xxxxx

Nitric oxide (NO) molecule is one of the minor components of terrestrial atmosphere. Generated in atmospheric plasma from chemical reactions of nitrogen with oxygen, NO and its radicals are very important in many industrial technologies [1-5] and play a key role in the combustion of fossil fuels [6, 7]. Of the nitrogen oxide compounds, the so-called NOx gasses, the NO molecule has the greater impact on environment and on pollution caused by human activities [8, 9].

In order to make kinetic plasma models involving nitric oxide, many sets of molecular data [10-12], spectroscopic properties [13-15] and reaction rate coefficients [16-19] are available in the literature but, in spite of its importance, none of them provide complete data on electron impact dissociation. More specifically, rate constants for electron-NO reactions, both theoretical [14, 20-22] and experimental [23-27], exist only for vibrational excitation and dissociative electron attachment processes at low-energy. A recent compilation of all the known electron collision cross sections is given by Song *et al.* [16].

To fill this gap, we present calculations – based on the formalism used previously [20] – of vibrational state resolved cross sections and the corresponding rate coefficients for dissociative excitation (DE) of NO by electron impact, *i.e.*:

$$e + \operatorname{NO}(X^{2}\Pi; v) \to \operatorname{NO}^{-} \to e + \operatorname{NO}(X^{2}\Pi; \epsilon_{c}) \to e + \operatorname{N}(^{4}S) + \operatorname{O}(^{3}P).$$

$$\tag{1}$$

We consider electron collision energies where numerous NO⁻ resonances exist, direct dissociation is negligible and the DE reaction is dominated by resonant processes [28, 29]. Our aim is to cover a large range of incident electron energies, so we take into account five resonance states of NO⁻: the three low-lying states of ${}^{3}\Sigma^{-}$, ${}^{1}\Sigma^{+}$ and ${}^{1}\Delta$ symmetries and two higher ones, with ${}^{3}\Pi$ and ${}^{1}\Pi$ symmetry, which lie close to the NO dissociation threshold. In the following, we number these resonances by r = 1, ..., 5, respectively. The vibrational excited states of which converge on the N(⁴S) + O({}^{3}P) dissociation limit of Eq. (1), due to their symmetry, have small oscillator strengths and short lifetimes [10, 30, 31] so their influence on the DE process can be neglected.

NO is a stable, open-shell molecule with a ${}^{2}\Pi$ ground electronic state, and a very accurate theoretical treatment of the electron-NO scattering therefore needs to take into account the spin-dependence of the process. However, the spin-orbit coupling effects are only important at very low energies [23] and, consequently, in the following we will neglect them.

We start by briefly describing the theoretical model used to calculate the cross sections for the process (1), restricting ourselves to the major equations of the Local-Complex-Potential (LCP) model. For a comprehensive treatment of the resonant collisions, we refer to the seminal articles [28, 29, 32]. The LCP approach was used to calculate the low-energy vibrational excitation of NO [20] and CO [33] by electron impact and the DE of oxygen molecule [34, 35], which gave results in good agreement with experiment.

In the LCP model, the DE cross section for an NO molecule initially in vibrational level v by an incident electron of energy ϵ is given by [32]:

$$\sigma_{v}(\epsilon) = \sum_{r=1}^{5} \frac{2S_{r}+1}{(2S+1)2} \frac{g_{r}}{g^{2}} \frac{64 \pi^{5} m^{2}}{\hbar^{4}} \int_{\epsilon_{v}^{th}}^{\epsilon_{v}^{max}} d\epsilon_{c} \frac{k'}{k} \left| \langle \chi_{c} | \mathcal{V}_{r} | \xi_{v}^{r} \rangle \right|^{2} , \qquad (2)$$

where $2S_r + 1$ and 2S + 1 are the spin-multiplicities of the resonant anion state and of the neutral target state respectively, g_r and g represent the corresponding degeneracy factors, k(k') are the incoming (outgoing) electron

*Electronic address: vincenzo.laporta@istp.cnr.it

nergies; the coef	ficients c_r enter	ring in Eq. (4)	and the spin-s	statistical factor	$\text{rs } \frac{2S_r+1}{(2S+1)2} \frac{g_r}{g2}.$	
	$NO(X^2\Pi)$	$NO^{-}(^{3}\Sigma^{-})$	$NO^{-}(^{1}\Delta)$	$NO^{-}(^{1}\Sigma^{+})$	$NO^{-}(^{3}\Pi)$	$NO^{-}(^{1}\Pi)$
μ (a.u.)	13614.1					
$D_e \; (eV)$	6.610	5.161	5.90	5.60	1.85	0.729
R_e (a.u.)	2.175	2.393	2.38	2.37	3.18	3.14
0 (211)	1 / 8	1.20	1 18	1.20	0.75	1 0/10

0.775

0.60

(eV)

spin-stat

-0.015

0.81

TABLE I: Summary on the parameters used in the theoretical model: reduced mass μ ; Morse parameters for NO and NO⁻ potential energies; the coefficients c_r entering in Eq. (4) and the spin-statistical factors $\frac{2S_r+1}{(2S+1)2} \frac{g_r}{a^2}$.

momenta, *m* is the electron mass, χ_c stands for the continuum wave function of NO asymptotically converging to $N(^4S) + O(^3P)$, and ξ_v^r is the resonance nuclear-motion wave function which is obtained for each resonance as the solution of Schrödinger equation:

$$-\frac{\hbar^2}{2\mu}\frac{d^2}{dR^2} + V_r^-(R) - \frac{i}{2}\Gamma_r(R) - E\bigg]\xi_v^r(R) = -\mathcal{V}_r(R)\,\chi_v(R)\,, \qquad r = 1,\dots,5\,.$$
(3)

1.08

0.50

5.70

0.023

6.80

0.013

where μ is the NO reduced mass, V_r^- and Γ_r represent the potential energies and the autoionization widths respectively for the five resonant NO⁻ states included in the calculation, χ_v is the wave function of the initial vibrational state of NO with energy ϵ_v , and $E = \epsilon + \epsilon_v$ is the total energy of the system. Following Ref. [32], the widths Γ_r were expressed as:

$$\Gamma_r = c_r \left(V_r^- - V_0 \right)^{l_r + \frac{1}{2}} H \left(V_r^- - V_0 \right), \qquad r = 1, \dots, 5,$$
(4)

where V_0 is the NO potential energy, H is a Heaviside step-function and l_r is the angular momentum of the lowest contributing partial wave associated with the incident electron. Experimental results show that the dominant contribution come from the *p*-wave [22] so l_r was set to one. In order to reproduce the positions and widths of the peaks in the low-energy region of the experimental cross sections, the constants c_r in Eq. (4) were introduced as phenomenological external parameters as reported in the paper [20]. Their values are given in Table I. In Eqs. (2) and (3), \mathcal{V}_r is the discrete-to-continuum coupling given by [21]:

$$\mathcal{V}_{r}^{2} = f_{r}^{2l_{r}+1} \frac{\hbar}{2\pi} \frac{\Gamma_{r}}{\sqrt{2m\left(V_{r}^{-} - V_{0}\right)}}, \qquad r = 1, \dots, 5,$$
(5)

where f_r is the so-called penetration factor introduced to reproduce the correct dependence of the cross sections on the energy near the excitation threshold, namely,

$$f_r^2(\epsilon) = \begin{cases} \epsilon / (V_r^- - V_0) & \text{if } \epsilon < (V_r^- - V_0) \\ 1 & \text{otherwise} \end{cases}, \qquad r = 1, \dots, 5.$$
(6)

The sum in the cross section formula (2) runs over all the five resonance states of the NO⁻ anion involved in the dissociation process. The integral extends to the continuum part of the NO potential from the dissociation threshold energy ϵ_v^{th} corresponding to the vibrational level v up to $\epsilon_v^{max} = \epsilon_v^{th} + 10$ eV. The spin-statistical factors are listed in Table I.

In the model shown above, the potentials $V_0(R)$ and $V_r^-(R)$ are expressed as a standard Morse function $U(R) = D_e \left[1 - e^{-\alpha(R-R_e)}\right]^2 + W$ whose parameters, for the NO molecule and for the three low-lying resonances ${}^3\Sigma^-$, ${}^1\Sigma^+$ and ${}^1\Delta$, were determined by a fit procedure explained in [20]. In order to take into account the recent results presented in [12], the asymptotes for the singlet states ${}^1\Sigma^+$ and ${}^1\Delta$ have been shifted to the correct threshold $N^-({}^3P) + O({}^3P)$. We have checked, and the results in [20] are not affected by these changes. Analogously, the ${}^3\Pi$ and ${}^1\Pi$ symmetry parameters were obtained by a fit to the data presented in [11] and the *ab-initio* R-Matrix results in the Ref. [12]. All Morse parameters are summarized in Table I. The NO ground state potential energy curve, as well as those for the five NO⁻ resonances and their corresponding autoionization widths are reported in Fig. 1. Table II contains the list of the vibrational levels supported by the NO molecule.

Figures 2 and 3 summarize the results of the present letter. The cross sections were computed up to 15 eV, at which point they drop off and become negligible, yo cover temperatures up to 50000 K for the reaction rates, relevant for the applications mentioned above.





FIG. 1: (Plot on the left) Potential energy curves for the ground electronic state of NO molecule (solid line) and for the five NO⁻ resonances (broken lines) included in the calculations; (Plot on the right) The corresponding widths of the resonances.

TABLE II: Energies of the vibrational levels of the electronic ground state of the NO molecule. $D_0 = 6.490 \text{ eV}$

v	$\epsilon_v \; (eV)$	v	$\epsilon_v (\mathrm{eV})$	v	$\epsilon_v ({ m eV})$
0	0.000	18	3.581	36	5.745
1	0.236	19	3.739	37	5.823
2	0.468	20	3.892	38	5.897
3	0.695	21	4.040	39	5.967
4	0.918	22	4.185	40	6.033
5	1.137	23	4.324	41	6.094
6	1.351	24	4.460	42	6.150
7	1.561	25	4.591	43	6.203
8	1.767	26	4.718	44)	6.251
9	1.968	27	4.840	45	6.294
10	2.164	28	4.958	46	6.333
11	2.357	29	5.072	47	6.368
12	2.545	30	5.181	48	6.399
13	2.729	31	5.286	49	6.425
14	2.908	32	5.386	50	6.446
15	3.083	33	5.483	51	6.464
16	3.253	34	5.574	52	6.477
17	3.419	35	5.662	53	6.485

Figure 2 contains the results of cross sections for the process of dissociation in (1) for three specific vibrational levels of NO molecule. Partial contributions coming from the five resonances as well as the sum are shown. Some features can be noticed: (i) Basically, as expected, for all cases, the major contribution to the total cross section comes from the ³II and ¹II resonances due to its closeness to the dissociation threshold, whereas the ¹ Σ^+ and ¹ Δ resonances, in general, make a minor contribution, in particular for low and middle vibrational levels (v = 0 and v = 10). (ii) As a consequence of the Franck-Condon overlap, the ³II and ¹II contributions to the cross section for v = 0 extends up to 11.5 eV, with a maximum around 9 eV. (iii) Beyond 11.5 eV, the asymptotic behavior is driven by the ³ Σ^- resonance. (iv) Sinve high vibrational levels (v = 40) approach the NO dissociation limit, the contributions from ³ Σ^- and, in particular, from the ¹ Σ^+ and ¹ Δ states, become comparable to those from the ³II and ¹II resonances at threshold.

Finally, Fig. 3 reports the full set of DE cross section results resolved over the vibrational ladder. By assuming a Maxwellian distribution for the electrons, the corresponding rate coefficient K_v is given, as a function of the electron temperature T_e , by:

$$K_v(T_e) = \left(\frac{1}{m\pi}\right)^{1/2} \left(\frac{2}{k_B T_e}\right)^{3/2} \int \epsilon \,\sigma_v(\epsilon) \, e^{-\epsilon/k_B T_e} \, d\epsilon \,, \tag{7}$$

where k_B is the Maxwell-Boltzmann constant.





FIG. 2: Contributions coming from the five resonant states (broken lines, same colors as in Fig. 1) to the total dissociative excitation cross section (solid line) for v = 0 (plot on the left), v = 10 (plot on the middle) and v = 40 (plot on the right).



FIG. 3: Vibrational state-resolved cross sections and the corresponding rate coefficients for NO dissociative excitation by electron impact.

In conclusion, vibrational state-resolved cross sections for dissociation of nitric oxide by electron-impact are computed for the first time using a phenomenological Local-Complex-Potential approach. Among the five resonances we considered in the calculations, the ${}^{3}\Pi$ symmetry is the one which makes the largest contribution. The full set of data obtained in the present work is available as supplementary material to this letter and in the LXCat database [36].

Acknowledgements

VL and IFS wish to thank the 'Institut de recherche Energie Propulsion & Environnement' (I-EPE, CNRS-Normandie Université, France), la Region Normandie, FEDER, and LabEx-EMC³ via the projects EMoPlaF and CO_2 -VIRIDIS for the financial support. VL thanks the 'Lab. Ondes et Milieux Complexes' (CNRS-Université du Havre, France), where this work was started, for the kind hospitality.

- [1] L. Campbell and M. Brunger, International Reviews in Physical Chemistry 35, 297 (2016).
- [2] A. Bultel and J. Annaloro, Plasma Sources Science and Technology 22, 025008 (2013).
- [3] L. Campbell and M. J. Brunger, Plasma Sources Science and Technology 22, 013002 (2012).
- [4] O. Motapon, M. Fifirig, A. Florescu, F. O. W. Tamo, O. Crumeyrolle, G. Varin-Bréant, A. Bultel, P. Vervisch, J. Tennyson, and I. F. Schneider, Plasma Sources Sci. Technol. 15, 23 (2006).
- [5] C. A. Barth, K. D. Mankoff, S. M. Bailey, and S. C. Solomon, Journal of Geophysical Research: Space Physics 108 (2003).
- [6] R. I. Slavchov, M. Salamanca, D. Russo, I. Salama, S. Mosbach, S. M. Clarke, M. Kraft, A. A. Lapkin, and S. V. Filip, Fuel 267, 117218 (2020).
- [7] G. A. Lavoie, J. B. Heywood, and J. C. Keck, Combustion Science and Technology 1, 313 (1970).
- [8] L. B. Kreuzer and C. K. N. Patel, Science **173**, 45 (1971).
- [9] C. W. Spicer, Atmospheric Environment (1967) 11, 1089 (1977).
- [10] R. de Vivie and S. D. Peyerimhoff, The Journal of Chemical Physics 89, 3028 (1988).
- [11] R. Locht and J. Momigny, International Journal of Mass Spectrometry and Ion Physics 4, 379 (1970).

- [12] D. Nandi, V. S. Prabhudesai, B. M. Nestmann, and E. Krishnakumar, Phys. Chem. Chem. Phys. 13, 1542 (2011).
- [13] A. Wong, S. N. Yurchenko, P. Bernath, H. S. P. Müller, S. McConkey, and J. Tennyson, Monthly Notices of the Royal Astronomical Society 470, 882 (2017).
 - [14] D. Teillet-Billy and F. Fiquet-Fayard, Journal of Physics B: Atomic and Molecular Physics 10, L111 (1977).
 - [15] R. J. Hargreaves, I. E. Gordon, L. S. Rothman, S. A. Tashkun, V. I. Perevalov, A. A. Lukashevskaya, S. N. Yurchenko, J. Tennyson, and H. S. Müller, Journal of Quantitative Spectroscopy and Radiative Transfer 232, 35 (2019).
 - [16] M.-Y. Song, J.-S. Yoon, H. Cho, G. P. Karwasz, V. Kokoouline, Y. Nakamura, and J. Tennyson, Journal of Physical and Chemical Reference Data 48, 043104 (2019).
 - [17] Y. Itikawa, Journal of Physical and Chemical Reference Data 45, 033106 (2016).
- [18] A. Zecca, G. P. Karwasz, R. S. Brusa, and T. Wróblewski, International Journal of Mass Spectrometry 223-224, 205 (2003).
- [19] M. J. Brunger and S. J. Buckman, Physics Reports **357**, 215 (2002).
- [20] V. Laporta, R. Celiberto, and J. M. Wadehra, Plasma Sources Science and Technology 21, 055018 (2012).
- [21] C. S. Trevisan, K. Houfek, Z. Zhang, A. E. Orel, C. W. McCurdy, and T. N. Rescigno, Phys. Rev. A 71, 052714 (2005).
- [22] Z. Zhang, W. Vanroose, C. W. McCurdy, A. E. Orel, and T. N. Rescigno, Phys. Rev. A 69, 062711 (2004).
- [23] M. Allan, J. Phys. B: At. Mol. Opt. Phys. **38**, 603 (2005).
- [24] M. Jelisavcic, R. Panajotovic, and S. J. Buckman, Phys. Rev. Lett. 90, 203201 (2003).
- [25] L. Josic, T. Wróblewski, Z. L. Petrovic, J. Mechlinska-Drewko, and G. P. Karwasz, Chemical Physics Letters 350, 318 (2001).
- [26] E. Krishnakumar and S. K. Srivastava, Journal of Physics B: Atomic, Molecular and Optical Physics 21, L607 (1988).
- [27] R. J. Van Brunt and L. J. Kieffer, Phys. Rev. A 10, 1633 (1974).
- [28] J. N. Bardsley and F. Mandl, Reports on Progress in Physics **31**, 471 (1968).
- [29] W. Domcke, Physics Reports **208**, 97 (1991).
- [30] J. Brzozowski, N. Elander, and P. Erman, Physica Scripta 9, 99 (1974).
- [31] R. Camilloni, E. Fainelli, G. Petrocelli, and G. Stefani, Journal of Physics B: Atomic and Molecular Physics 20, 1839 (1987).
- [32] L. Dubé and A. Herzenberg, Phys. Rev. A **20**, 194 (1979).
- [33] V. Laporta, C. M. Cassidy, J. Tennyson, and R. Celiberto, Plasma Sources Science and Technology 21, 045005 (2012).
- [34] V. Laporta, R. Celiberto, and J. Tennyson, Plasma Sources Science and Technology 22, 025001 (2013).
- [35] V. Laporta, R. Celiberto, and J. Tennyson, Phys. Rev. A 91, 012701 (2015).
 - [36] L. C. Pitchford, L. L. Alves, K. Bartschat, S. F. Biagi, M.-C. Bordage, I. Bray, C. E. Brion, M. J. Brunger, L. Campbell, A. Chachereau, et al., Plasma Processes and Polymers 14, 1600098 (2017).