

Compressive strength of bioactive sol-gel glass foam scaffolds

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Abstract

Larry Hench's 45S5 Bioglass[®] has been used in more than a million patients as a synthetic bone graft, in the form of a particulate. Bioglass is able to stimulate more bone regeneration than other bioactive ceramics. However, it is not commercially available as a porous scaffold with amorphous glass structure because the 45S5 composition crystallises during sintering. The sol-gel foaming process was developed in Hench's laboratory to overcome this problem. Here, we concisely review the work on scaffold development from Hench's group and report new data that show maximum compressive strengths in excess of 5 MPa can be achieved while maintaining the interconnected pore networks required for vascularised bone ingrowth. This was achieved through optimisation of sintering of the sol-gel foams. Sintering of the sol-gel foams was correlated to the network connectivity and silanol content of the sol-gel foams. Changes in strength after immersion in simulated body fluid were found to be small over the times investigated. Relating the dissolution results to *in vivo* studies indicates that the scaffolds degrade more rapidly *in vivo* than *in vitro*.

Introduction

Larry Hench's 45S5 Bioglass[®] was the first artificial material found to form a bond with bone^{1,2}, launching the field of bioactive materials³. It has the composition 46.1 mol% SiO₂, 24.4 mol% Na₂O, 26.9 mol% CaO and 2.6 mol% P₂O₅. *In vivo* studies showed bioactive glasses bond with bone more rapidly than other bioceramics⁴, such as synthetic hydroxyapatite and apatite/wollastonite glass-ceramic. To find out why Bioglass had this enhanced osteogenic potential, Hench formed a Tissue Engineering and Regenerative Medicine Centre with Professor Dame Julia Polak. Their *in vitro*

studies indicated their osteogenic properties are due to their dissolution products stimulating osteoprogenitor cells at the genetic level⁵⁻⁸. A common strategy for bone regeneration is the implantation of a synthetic bone graft to act as a temporary template (scaffold) for bone growth⁹. The role of the scaffold is to provide a platform on which bone cells can produce new bone matrix. The scaffold must bond with bone (be bioactive), biodegrade at a controlled rate, provide adequate mechanical support and have an interconnected pore network that allows vascularised bone ingrowth, which is hypothesised to be diameters in excess of 100 μm ⁹⁻¹¹. Synthetic hydroxyapatite (HA) is commercially available in porous granules and blocks with pore networks that mimic cancellous bone¹². The original Bioglass cannot be sintered into large scaffolds while maintaining its amorphous structure because the crystallisation onset temperature is too close to its glass transition temperature (T_g)¹³. T_g must be surpassed to enable viscous flow sintering, but crystallisation reduces control of dissolution and reduces bioactivity of the material¹³. This can be overcome, either by tailoring the glass composition to allow sintering^{13, 14}, or by using the bottom-up sol-gel approach. Using sol-gel, bioactive glass networks assemble at room temperature^{15, 16}. As melting is not required, sodium is not needed in sol-gel compositions, so typical compositions are ternary¹⁵, e.g. 58S (60 mol% SiO_2 , 36 mol% CaO , 4 mol% P_2O_5) or binary, e.g. 70S30C (70 mol% SiO_2 , 30 mol% CaO)¹⁷.¹⁸ Sol-gel glasses are inherently nanoporous, the pores form at the interstices between coalesced silica nanoparticles¹⁶ and their size (6-17 nm for 70S30C and 58S^{19, 20}) depends on the precursors used, the glass composition and pH^{15, 21}.

Sepulveda, Jones and Hench used the sol to gel transition, which is the formation of –Si-O-Si– network (polycondensation of Si-OH groups), to produce porous scaffolds that mimic the macropore architecture of cancellous bone²²⁻²⁴. They introduced a foaming

step (vigorous agitation of the sol with surfactant) into the conventional sol-gel process²². The gelation time was accelerated by adding hydrofluoric acid (HF) so that polycondensation occurred in a few minutes. The many variables in the process and their effect on structure property relationships were investigated^{23, 25-29}. The process was also up-scaled²⁷. By sintering at 800°C for 2 h, compressive strengths of 2.26 MPa were achieved with a modal interconnect diameter of 100 µm between spherical pores with diameters of 300-600 µm at 82% porosity²³. This is only just inside the range of compressive strength of the trabecular bone (2-12 MPa). Improved strength is needed. *In vitro* cell culture of human osteoblasts on the foams showed that the glass stimulated mineralised bone nodule formation without addition of growth factors or hormones³⁰⁻³². Osteoclasts also seem to remodel the glass surface in culture³³. *In vivo* studies confirmed the foams can regenerate bone defects and the scaffolds degrade over time³⁴. The reduction in compressive strength of these scaffolds as a function of dissolution time has not been reported in the literature. Sol-gel foaming was replicated by various groups³⁵⁻³⁹.

The aim here was to investigate whether the compressive strength of the foams could be improved without losing the interconnected pore network, and how immersion in simulated body fluid affects strength. Previous work found crystallisation of the 70S30C composition onset at 845 °C²³. To avoid the risk of crystallisation, sintering temperature could not be increased above 800 C therefore sintering time was investigated.

Materials and Methods

Scaffolds were produced by sol-gel foaming, with the 70S30C composition (70 mol% SiO₂, 30 mol% CaO) as previously described^{23, 27}. Sol was prepared by mixing the

following reagents (in order): deionised water, 2N nitric acid, tetraethyl orthosilicate (TEOS), and calcium nitrate tetrahydrate (all Sigma) at a molar ratio of water to TEOS (R ratio) of 12:1. 50 ml aliquots of sol were foamed by vigorous agitation (Kenwood electric hand mixer) with 0.5 ml Teepol (Thames Mead Ltd.) and 1.5 ml 5 vol% HF (a gelation catalyst). Teepol contains a low concentration mixture of anionic (15%) and nonionic surfactants (5%). As the gelation point neared, the foamed sol was poured into cylindrical polymethyl pentene moulds, which were then sealed. The samples were aged (60°C), dried (130°C) and thermally stabilised (600°C) then furnace cooled. Cylindrical foams were produced with a geometrically measured bulk density of $0.21 \pm 0.02 \text{ gcm}^{-3}$ and reheated to 600°C ($10^\circ\text{C min}^{-1}$), held at 600°C for 1 h and then heated (again at $10^\circ\text{C min}^{-1}$) to sintering temperature of 750°C, 775°C or 800°C and held for 2 h or 4 h before furnace cooling. They were then cut, using a scalpel, into blocks of 15 x 5 x 5 mm and 5 mm cubes.

Percentage porosity was calculated using $\% \text{ porosity} = 1 - \rho_r$, where ρ_r is the relative density, where $\rho_r = \rho_f / \rho_s$, with ρ_f being the geometrical foam density and ρ_s the skeletal density⁴⁰, determined from helium pycnometry (Quantachrome). X-ray diffraction (XRD) patterns were obtained on crushed foams using a Philips PW 1700 series automated XRD spectrometer, using a step scanning method with $\text{CuK}\alpha$ radiation, at 40 kV and 40 mA, with a 0.040° 2θ step and a count rate of 30 s per step, from 2θ values of 10° to 50° . A LEO 1525 scanning electron microscope (SEM) equipped with a GEMINI field emission column was used to image the foamed scaffolds. Fragments of foamed scaffolds were mounted on sticky carbon tape and sputter coated with chromium to a maximum thickness of 15 nm before imaging. Images were collected on the in-lens secondary electron detector with an operating voltage of 5 kV and a working

distance of 5–8 mm. The interconnected pore size distributions were obtained by mercury intrusion porosimetry (Quantachrome PoreMaster 33).

Compression tests were carried out on foams 15 (h) x 5 x 5 mm using a Zwick with a parallel plate crosshead velocity of 0.5 mm min⁻¹ and a 500 N load cell. Five samples were used for each condition. Microcomputed tomography (μ CT) was performed on a bioactive glass foam 4 (h) x 7 x 7 mm sintered at 800°C for 2 h after it had just fractured under compression loading. Scans were performed using a commercial μ CT unit (Phoenix X-ray Systems and Services GmbH) at 8 μ m voxel resolution.

Foam cubes (5 mm dimensions) were immersed in 50 ml of (SBF)⁴¹ at 37°C and 175 rpm for 1, 4, 8, 24 and 168 h according to an agreed protocol for bioactive glass testing⁴². Since that the cube sized samples were not cut into constant shapes, the amount of the SBF needed was calculated to maintain a consistent mass to SBF volume ratio of 1.5 mg ml⁻¹⁴². Filtered SBF (using 1 μ m retention filter paper) was analysed by inductive coupled plasma spectroscopy (ICP, ARL instruments). The instrument detection limits for Si, Ca and P were 0.05, 0.10 and 0.20 μ g ml⁻¹ respectively. The scaffolds were dried in acetone and analysed by Fourier transform infrared spectroscopy (FTIR) to monitor the growth of the HCA layer. For collection of FTIR spectra, scaffolds were ground in KBr with a glass to KBr dilution ratio of 1:100. Spectra were collected using a Mattson Genesis II spectrometer, with a Pike Technologies EasiDiff diffuse reflectance accessory.

Results and Discussion

Figure 1a is a photograph of foam scaffolds post stabilisation, where the foam was dried and heated to 600°C (left) and a similar scaffold following the same thermal

stabilisation at 600°C and then further sintering at 800°C for 2 h. As 800°C was above T_g for the 70S30C glass composition ($T_{G \text{ ONSET}} = 717^\circ\text{C}^{23}$), shrinkage of the scaffold was observed due to viscous flow and further Si-O-Si bond formation through condensation. Pores that were visible to the naked eye were no longer visible after sintering. Figure 1b shows a fracture surface of a stabilised scaffold imaged under SEM. Interconnects between pores are visible (black holes). Other interconnects were present, but are out of the plane of the image. Figure 1c shows that the pores were highly interconnected in 3D (voxel size was 8 μm). In order to determine whether the pore networks are suitable for bone regeneration, interconnects must be in excess of 100 μm , to allow vascularised bone ingrowth¹⁰.

Figure 2 shows interconnect and pore size distributions for bioactive sol-gel foam scaffolds, stabilised at 600°C with initial ρ_f (foam density) of $0.22 \pm 0.02 \text{ g cm}^{-3}$ and porosity of 92% , then sintered to 800°C to give a ρ_f (foam density) of $0.41 \pm 0.02 \text{ g cm}^{-3}$ and porosity of 85%. Modal sizes were excess of 100 μm after stabilisation and sintering. After stabilisation at 600°C, the modal interconnect size was 153 μm by mercury porosimetry, decreasing 97 μm after sintering²⁴. Modal pore diameters from X-ray microtomography (μCT) were 743 μm after stabilization, decreasing to 561 μm after sintering²⁴. While the interconnect size is most important for tissue ingrowth, percentage porosity has the largest effect on compressive strength⁴⁰.

In this study, after thermal stabilisation at 600°C, the foam density was $0.21 \pm 0.02 \text{ g cm}^{-3}$, with a percentage porosity of 93%. Note the percentage porosity was high due to the presence of strut nanoporosity that is inherent to the sol-gel process²³. Figure 3 shows the interconnect diameter distributions for each condition tested. XRD patterns

showed amorphous halos for all conditions (data not shown), which was expected as sintering was below the $T_{c\text{ ONSET}}$ for the composition.

Figure 4 shows a 3D reconstruction of a foam scaffold immediately after fracture (Figure 4a) and Figure 4(b) is a cross-section through (a) showing several fractured pieces of the foam, segmented to different colours, always fracturing in the plane parallel to the load. The stress-strain curves that were obtained from the scaffolds were typical of brittle open cell foams^{23, 40}, with an elastic compression region, a point of fracture (maximum compressive strength) and then a region where there were continual fracture of pore layers and crushing/collapse of the pores. However the principle crack propagation was not in the direction predicted by Gibson and Ashby. They suggested cellular solids should fail through a crack propagating perpendicular to the load axis, due to sequential buckling/fracture of struts. Here, Figure 4a and Figure 4b show the cracks propagate along the loading axis, splitting the foam into several pieces. This could be due to the spherical nature of the pores and circular morphology of the interconnects, or, perhaps more likely, due to local stress concentrations causing crack initiation at the top of the scaffold in contact with the moving compression platen, then propagating catastrophically down the scaffold.

Table 1 shows the structural and mechanical characterisation results and time to HCA formation in SBF for the scaffolds in this study. Sintering at 750 °C for 4 h did not increase the foam density of the scaffolds. The modal interconnect diameter was 104 µm and compressive strength was 2.3 MPa, which was similar to the maximum strength obtained in previous studies, where scaffolds sintered for 2 h at 800°C²³. Sintering for 2 h at 800°C, in this study, increased the foam density to $0.45 \pm 0.02 \text{ g cm}^{-3}$, but the modal interconnect diameter was larger, at 116 µm. The increase in density was due to

the increase in bridging oxygen bonds, decrease in non-bridging bonds (reduction in Si-OH groups) due to the polycondensation process and viscous flow. Previous data support this, with nitrogen sorption data showing modal nanopore diameter to decrease from 17 nm to 12 nm²³ and calculations from solid state proton MAS NMR studies indicating that the ratio of Si:OH in the glass increased from 2.0 to 8.7⁴³ as sintering temperature increased from 600°C to 800°C (2 h dwells).

Compressive strength, of scaffolds sintered at 800°C for 2 h, was higher than previously reported even though the modal interconnected pore diameter and percentage porosity were higher. Modal interconnected pore diameter was previously 98 µm, compared to 116 µm, and percentage porosity was 82%, compared to 84% here. Previous studies used foams that were 27 mm in diameter and 9 mm in height. Here samples were taller than they were wide (15 mm (h) x 5 x 5 mm). The higher compressive strength measured in this study could be due to the difference in the sample sizes used. In the previous study samples with volume >5000 mm³ were used while in this study it was ~375 mm³, this is more than 13 times less, which means there would be much lower number of flaws inherent to the material within the smaller samples used in this study when compared to the previous study.

Sintering for 4 h at 800°C reduced the interconnect size from 116 µm to 31 µm, but the compressive strength remained similar. There was therefore no benefit to sinter for longer for bone regeneration applications as the interconnect size became too small. Sintering for longer time seemed to cause reduction in interconnect size in general because sintering at 775°C for 4 h produced a smaller interconnect diameter than sintering at 800°C for 2 h. The extra time holding above T_g seems to cause excess viscous flow. Sintering should therefore not exceed 2 h for sol-gel foams.

Apatite formation in SBF

Apatite formation in SBF is an approximate indication of whether the scaffolds will form HCA *in vivo*^{44, 45}. FTIR spectra can contain P-O bending bands at ~571 and 602 cm^{-1} that represent crystalline orthophosphate, which, for bioactive glasses in SBF, is likely to be HCA. In FTIR spectra of scaffolds that were immersed in SBF for 1 to 8 h, dual P-O bending bands were not present on any scaffolds, indicating there was no HCA layer present. However, after 4 h, a broad band was observed in spectra taken from all scaffolds, between 620 and 550 cm^{-1} , which is often considered to signify amorphous calcium phosphate deposition (data not shown). Figure 5 shows FTIR spectra for the sol-gel foam scaffolds following immersion in SBF for 24 h (Figure 5a) and 1 week (Figure 5b). After 24 h, P-O bending bands were found at wavenumbers 563 and 605 cm^{-1} in spectra from scaffolds sintered at 750°C and 775 °C for 4 h. After 1 week, P-O bending bands were present in spectra from all samples.

Compressive strength after immersion in SBF

Compressive strength was measured after scaffolds were immersed in SBF. The sintering protocol of 800°C for 2 h was used for the scaffolds. Samples used in this study had a slightly higher foam density (0.6 g cm^{-3} compared to the 0.45 g cm^{-3}) used in the earlier data. This meant the scaffolds had lower percentage porosity (78% compared to 84%) and higher compressive strength (6.0 MPa compared to 4.6 MPa). After 2 h immersion in SBF, the foam density and percentage porosity changed little and while the mean compressive strength decreased to 5.3 MPa, the change was within the uncertainty of the experiment. Little change was observed up until 72 h immersion in SBF. However after 1 week, which was the time at which HCA formation was observed, porosity increased to 86% and compressive strength decreased to 2.1 MPa. This is still within the compressive strength of

cancellous bone. This implies that once dissolution has occurred to the point of HCA formation, porosity increases and mechanical strength begins to decrease. This would imply bone ingrowth should begin before 1 week *in vivo*, to prevent the risk of fracture. *In vivo* studies on 70S30C foam scaffolds showed excellent vascularized bone ingrowth after 11 weeks³⁴, in a rat tibia model, but only when the foams were preconditioned in culture media for 3 days *in vitro*. This was attributed to the scaffolds that were not preconditioned releasing a burst of Ca²⁺ that remained trapped inside the pores, causing a local pH increase inside the pores.

Figure 6a shows FTIR spectra of the scaffolds used in the SBF study. It shows that HCA formation started at 24 h and 3 days immersion in SBF. SEM images show that a precipitate formed on the the surface of the scaffolds after 3 days (Figure 6b) and matured into typical HCA-like morphology after 7 days (Figure 6c).

Conclusions

Bioactive sol-gel foams developed in Hench's laboratory were the first amorphous bioactive glass scaffolds with interconnected pore networks that mimic the morphology of cancellous bone with interconnects suitable for vascularised bone ingrowth. The 70S30C composition with modal interconnect diameters of 116 μm can be made to have a compressive strength of 4.6 MPa. Immersion in SBF for three days has little effect on mechanical properties or percentage porosity but after 1 week, compressive strength can be decreased to less than half of the dry value.

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References

1. L. L. Hench, R. J. Splinter, W. C. Allen, T. K. Greenlee. Bonding mechanisms at the interface of ceramic prosthetic materials, *Journal of Biomedical Materials Research Biomedical Materials Symposium* **2**(1) 117-141 (1971).
2. L. L. Hench, H. A. Paschall. Direct chemical bonding of bioactive glass-ceramic materials and bone, *J Biomed Mater Res Symp* **4** 25-42 (1973).
3. L. L. Hench. Opening paper 2015- Some comments on Bioglass: Four Eras of Discovery and Development, *Biomedical Glasses* **1**(1) 1-11 (2015).
4. H. Oonishi, L. L. Hench, J. Wilson, F. Sugihara, E. Tsuji et al. Quantitative comparison of bone growth behavior in granules of Bioglass (R), A-W glass-ceramic, and hydroxyapatite, *Journal of Biomedical Materials Research* **51**(1) 37-46 (2000).
5. I. D. Xynos, A. J. Edgar, L. D. K. Buttery, L. L. Hench, J. M. Polak. Gene-expression profiling of human osteoblasts following treatment with the ionic products of Bioglass (R) 45S5 dissolution, *Journal of Biomedical Materials Research* **55**(2) 151-157 (2001).
6. L. L. Hench, J. M. Polak. Third-generation biomedical materials, *Science* **295**(5557) 1014-1017 (2002).
7. I. D. Xynos, A. J. Edgar, L. D. K. Buttery, L. L. Hench, J. M. Polak. Ionic products of bioactive glass dissolution increase proliferation of human osteoblasts and induce insulin-like growth factor II mRNA expression and protein synthesis, *Biochemical and Biophysical Research Communications* **276**(2) 461-465 (2000).
8. I. D. Xynos, M. V. J. Hukkanen, J. J. Batten, L. D. Buttery, L. L. Hench et al. Bioglass 45S5 stimulates osteoblast turnover and enhances bone formation in vitro: Implications and applications for bone tissue engineering, *Calcified Tissue International* **67**(4) 321-329 (2000).
9. J. R. Jones. Review of bioactive glass: From Hench to hybrids, *Acta Biomaterialia* **9**(1) 4457-4486 (2013).
10. S. F. Hulbert, S. J. Morrison, J. J. Klawitte. Tissue Reaction to Three Ceramics of Porous and Non-Porous Structures, *Journal of Biomedical Materials Research* **6**(5) 347-374 (1972).
11. C. E. Holy, M. S. Shoichet, J. E. Davies. Engineering three-dimensional bone tissue in vitro using biodegradable scaffolds: Investigating initial cell-seeding density and culture period, *Journal of Biomedical Materials Research* **51**(3) 376-382 (2000).
12. R. Z. LeGeros. Properties of osteoconductive biomaterials: Calcium phosphates, *Clinical Orthopaedics and Related Research* (395) 81-98 (2002).

13. Z. Y. Wu, R. G. Hill, S. Yue, D. Nightingale, P. D. Lee et al. Melt-derived bioactive glass scaffolds produced by a gel-cast foaming technique, *Acta Biomaterialia* **7**(4) 1807-1816 (2011).
14. Q. Fu, E. Saiz, A. P. Tomsia. Bioinspired strong and highly porous glass scaffolds, *Advanced Functional Materials* **21**(6) 1058-1063 (2011).
15. R. Li, A. E. Clark, L. L. Hench. An investigation of bioactive glass powders by sol-gel processing, *Journal of Applied Biomaterials* **2**(4) 231-239 (1991).
16. S. Lin, C. Ionescu, K. J. Pike, M. E. Smith, J. R. Jones. Nanostructure evolution and calcium distribution in sol-gel derived bioactive glass, *Journal of Materials Chemistry* **19**(9) 1276-1282 (2009).
17. P. Saravanapavan, J. R. Jones, R. S. Pryce, L. L. Hench. Bioactivity of gel-glass powders in the CaO-SiO₂ system: A comparison with ternary (CaO-P₂O₅-SiO₂) and quaternary glasses (SiO₂-CaO-P₂O₅-Na₂O), *Journal of Biomedical Materials Research Part A* **66A**(1) 110-119 (2003).
18. J. Brinker, G. W. Scherer. Sol-gel science : the physics and chemistry of sol-gel processing. Boston: Academic Press; 1990.
19. P. Sepulveda, J. R. Jones, L. L. Hench. Characterization of melt-derived 45S5 and sol-gel-derived 58S bioactive glasses, *Journal of Biomedical Materials Research* **58**(6) 734-740 (2001).
20. P. Saravanapavan, L. L. Hench. Mesoporous calcium silicate glasses. II. Textural characterisation, *Journal of Non-Crystalline Solids* **318**(1-2) 14-26 (2003).
21. E. M. Valliant, C. A. Turdean-Ionescu, J. V. Hanna, M. E. Smith, J. R. Jones. Role of pH and temperature on silica network formation and calcium incorporation into sol-gel derived bioactive glasses, *Journal of Materials Chemistry* **22**(4) 1613-1619 (2012).
22. P. Sepulveda, J. R. Jones, L. L. Hench. Bioactive sol-gel foams for tissue repair, *Journal of Biomedical Materials Research* **59**(2) 340-348 (2002).
23. J. R. Jones, L. M. Ehrenfried, L. L. Hench. Optimising bioactive glass scaffolds for bone tissue engineering, *Biomaterials* **27** 964-973 (2006).
24. J. R. Jones, G. Poologasundarampillai, R. C. Atwood, D. Bernard, P. D. Lee. Non-destructive quantitative 3D analysis for the optimisation of tissue scaffolds, *Biomaterials* **28** 1404-1413 (2007).
25. J. R. Jones, L. L. Hench. Effect of surfactant concentration and composition on the structure and properties of sol-gel-derived bioactive glass foam scaffolds for tissue engineering, *Journal of Materials Science* **38**(18) 3783-3790 (2003).
26. J. R. Jones, L. L. Hench. Factors affecting the structure and properties of bioactive foam scaffolds for tissue engineering, *Journal of Biomedical Materials Research Part B-Applied Biomaterials* **68B**(1) 36-44 (2004).
27. J. R. Jones, S. Ahir, L. L. Hench. Large-scale production of 3D bioactive glass macroporous scaffolds for tissue engineering, *Journal of Sol-Gel Science and Technology* **29**(3) 179-188 (2004).
28. V. FitzGerald, K. O. Drake, J. R. Jones, M. E. Smith, V. Honkimaki et al. In situ high-energy X-ray diffraction study of a bioactive calcium silicate foam immersed in simulated body fluid, *Journal of Synchrotron Radiation* **14** 492-499 (2007).
29. V. FitzGerald, R. A. Martin, J. R. Jones, D. Qiu, K. M. Wetherall et al. Bioactive glass sol-gel foam scaffolds: Evolution of nanoporosity during processing and in situ monitoring of apatite layer formation using small- and

- wide-angle X-ray scattering, *Journal of Biomedical Materials Research Part A* **91A**(1) 76-83 (2009).
30. ISO 23317 Implants for surgery–Hydroxyapatite–Part 1: Ceramic hydroxyapatite 2000.
 31. J. R. Jones, O. Tsigkou, E. E. Coates, M. M. Stevens, J. M. Polak et al. Extracellular matrix formation and mineralization of on a phosphate-free porous bioactive glass scaffold using primary human osteoblast (HOB) cells, *Biomaterials* **28** 1653-1663 (2007).
 32. P. Valerio, M. M. Pereira, A. M. Goes, M. F. Leite. Effects of extracellular calcium concentration on the glutamate release by bioactive glass (BG60S) preincubated osteoblasts, *Biomedical Materials* **4**(4) 7 (2009).
 33. S. Midha, W. van den Bergh, T. B. Kim, P. D. Lee, J. R. Jones et al. Bioactive glass foam scaffolds are remodelled by osteoclasts and support the formation of mineralized matrix and vascular networks in vitro, *Advanced Healthcare Materials* **2**(3) 490-499 (2013).
 34. S. Midha, T. B. Kim, W. van den Bergh, P. D. Lee, J. R. Jones et al. Preconditioned 70S30C bioactive glass foams promote osteogenesis in vivo, *Acta Biomaterialia* **9**(11) 9169-9182 (2013).
 35. S. Wang, M. M. Falk, A. Rashad, M. M. Saad, A. C. Marques et al. Evaluation of 3D nano-macro porous bioactive glass scaffold for hard tissue engineering, *Journal of Materials Science-Materials in Medicine* **22**(5) 1195-1203 (2011).
 36. R. M. Almeida, A. Gama, Y. Vueva. Bioactive sol-gel scaffolds with dual porosity for tissue engineering, *Journal of Sol-Gel Science and Technology* **57**(3) 336-342 (2011).
 37. A. C. Marques, R. M. Almeida, A. Thiema, S. Wang, M. M. Falk et al. Sol-gel-derived glass scaffold with high pore interconnectivity and enhanced bioactivity, *Journal of Materials Research* **24**(12) 3495-3502 (2009).
 38. P. Valerio, M. H. R. Guimaraes, M. M. Pereira, M. F. Leite, A. M. Goes. Primary osteoblast cell response to sol-gel derived bioactive glass foams, *Journal of Materials Science-Materials in Medicine* **16**(9) 851-856 (2005).
 39. M. D. Coelho, M. M. Pereira. Sol-gel synthesis of bioactive glass scaffolds for tissue engineering: Effect of surfactant type and concentration, *Journal of Biomedical Materials Research Part B-Applied Biomaterials* **75B**(2) 451-456 (2005).
 40. L. J. Gibson, M. F. Ashby. Cellular Solids Structure and Properties. Oxford: Pergamon Press; 1988.
 41. T. Kokubo, H. Kushitani, S. Sakka, T. Kitsugi, T. Yamamuro. Solution able to reproduce in vivo surface-structure in bioactive glass-ceramic A-W3, *Journal of Biomedical Materials Research* **24**(6) 721-734 (1990).
 42. A. L. B. Macon, T. B. Kim, E. M. Valliant, K. Goetschius, R. K. Brow et al. A unified in vitro evaluation for apatite-forming ability of bioactive glasses and their variants, *Journal of Materials Science-Materials in Medicine* **26**(2) (2015).
 43. J. R. Jones, T. F. Kemp, M. E. Smith. Effect of OH content on the bioactivity of sol-gel derived glass foam scaffolds, *Key Engineering Materials* **309-311** 1031-1034 (2006).
 44. T. Kokubo, H. Takadama. How useful is SBF in predicting in vivo bone bioactivity?, *Biomaterials* **27**(15) 2907-2915 (2006).
 45. M. Bohner, J. Lemaitre. Can bioactivity be tested in vitro with SBF solution?, *Biomaterials* **30**(12) 2175-2179 (2009).

Figures

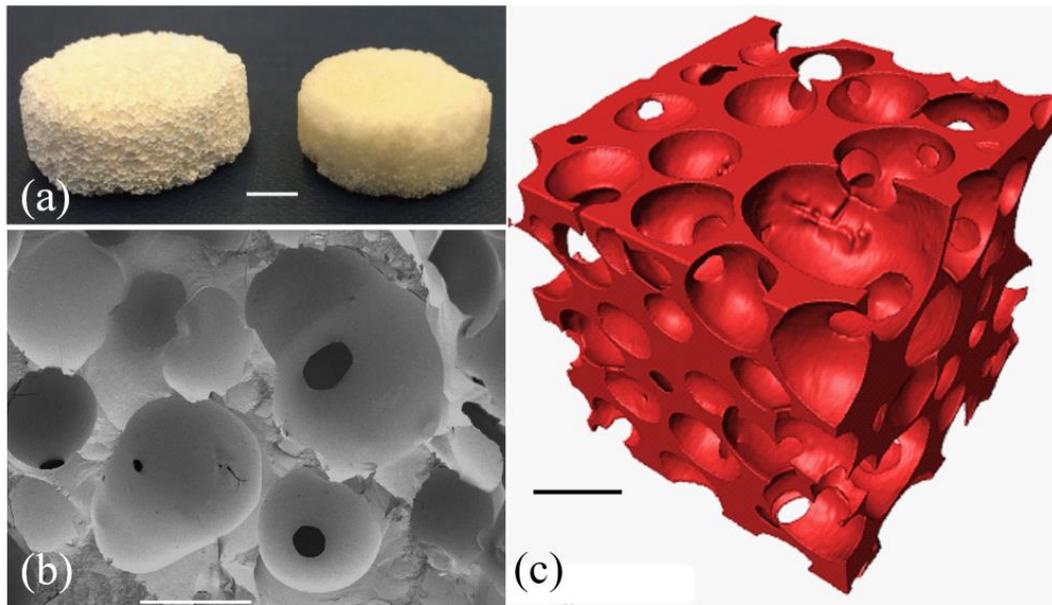


Figure 1. Bioactive glass sol-gel foam scaffolds (a) photographs after stabilisation at 600°C (left) and after sintering at 800°C, scale bar = 5 mm; (b) SEM image of a fracture surface, scale bar = 500 μm ; (c) X-ray microtomography 3D image, scale bar = 500 μm .

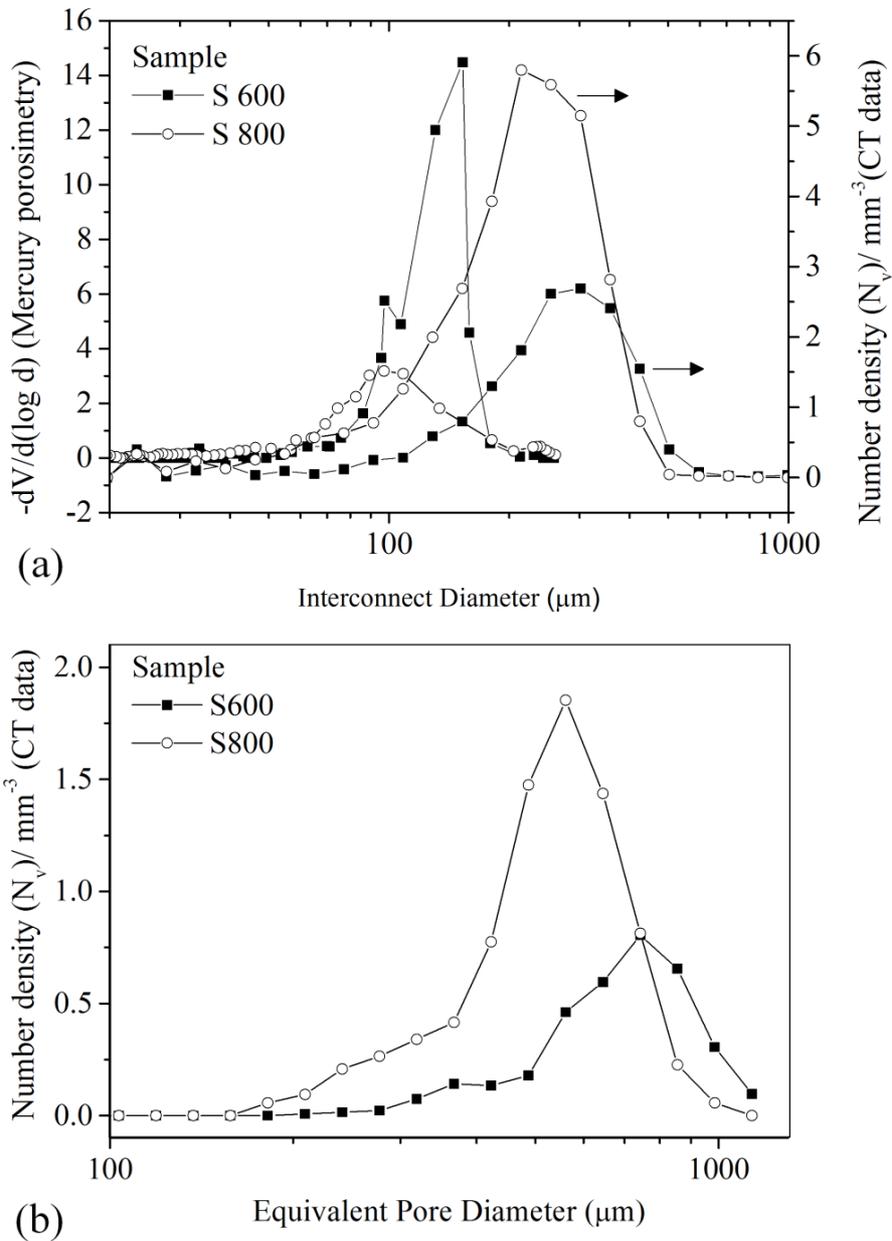


Figure 2. Pore size distributions for bioactive sol-gel foam scaffolds, stabilised at 600°C with initial ρ_f (foam density) of 0.22 g cm⁻³ and porosity of 92%, then sintered to 800°C to give a ρ_f (foam density) of 0.41 g cm⁻³ and porosity of 85%: (a) Interconnect size distributions from mercury porosimetry and X-ray microtomography (data replotted from Jones *et al.*²⁴); (b) Pore size distributions from X-ray microtomography (data replotted from Jones *et al.*²⁴).

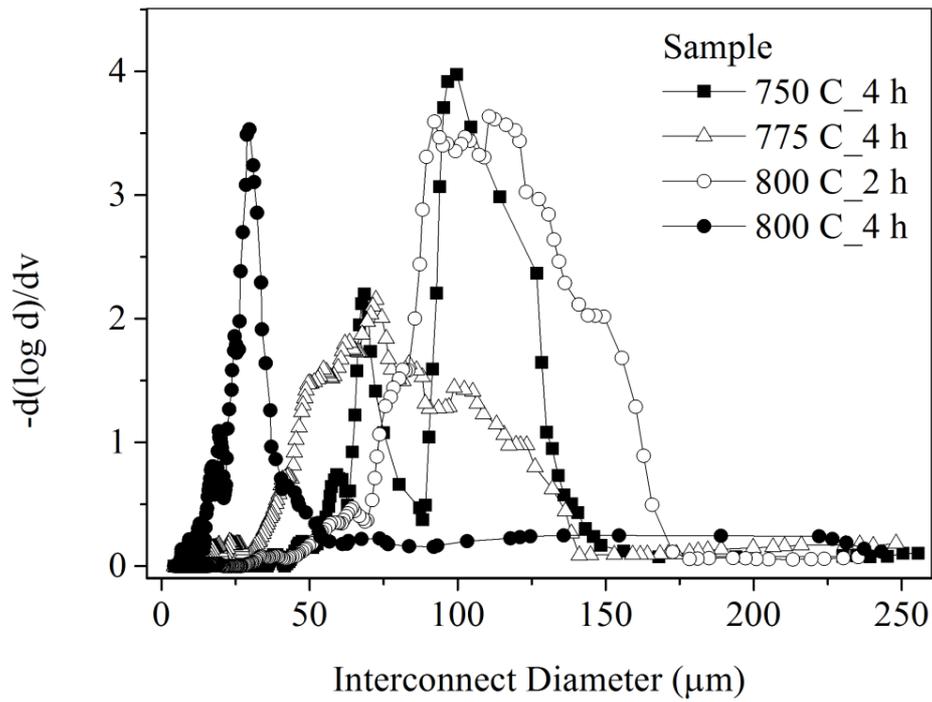


Figure 3. Interconnect size distributions for bioactive sol-gel foam scaffolds, stabilised at 600 °C with initial ρ_f (foam density) of 0.21 g cm⁻³ and porosity of 93%, then sintered to different temperatures for 2 h or 4 h.

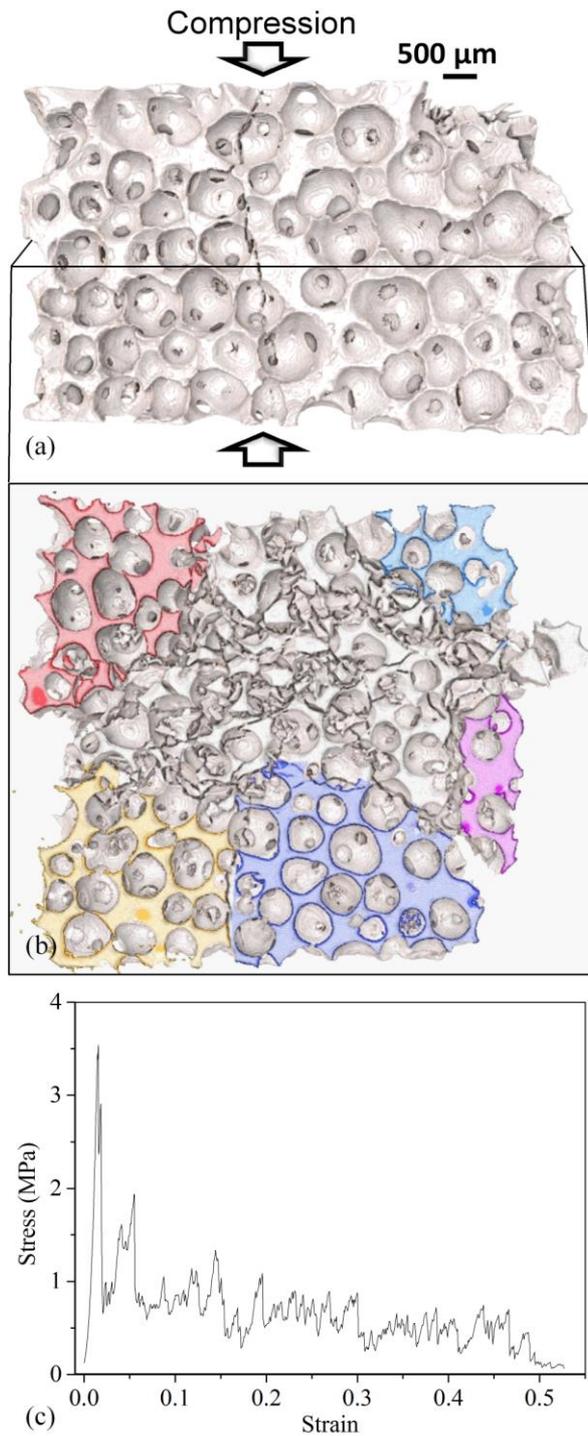


Figure 4. (a) X-ray microtomography (μ CT) image of a crack in a bioactive glass foam scaffold immediately after fracture; (b) A μ CT image of a cross-section through (a) with several individual fractured pieces of the foam coloured differently; (c) Representative compression stress-strain graph for a sol-gel foam scaffold. This example was a scaffold heated to 750°C for 4 h.

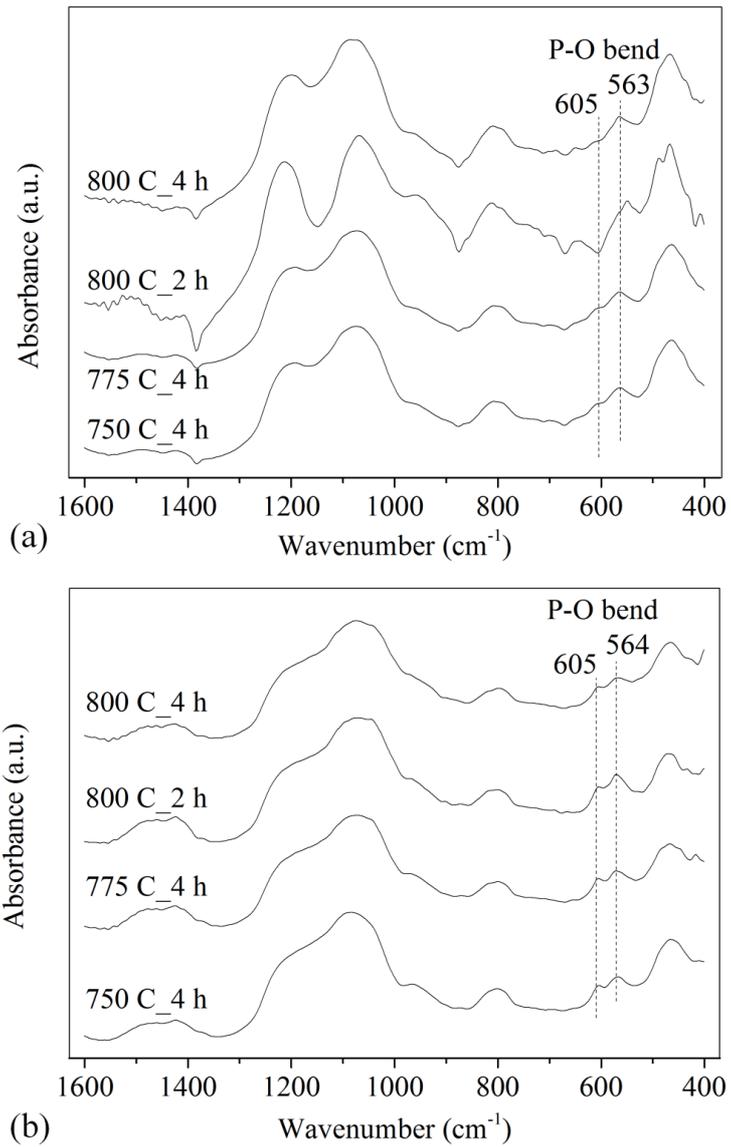


Figure 5. FTIR spectra from sol-gel foam 70S30C scaffolds, for different sintering temperatures and times, following immersion in SBF for: (a) 24 h; (b) 1 week.

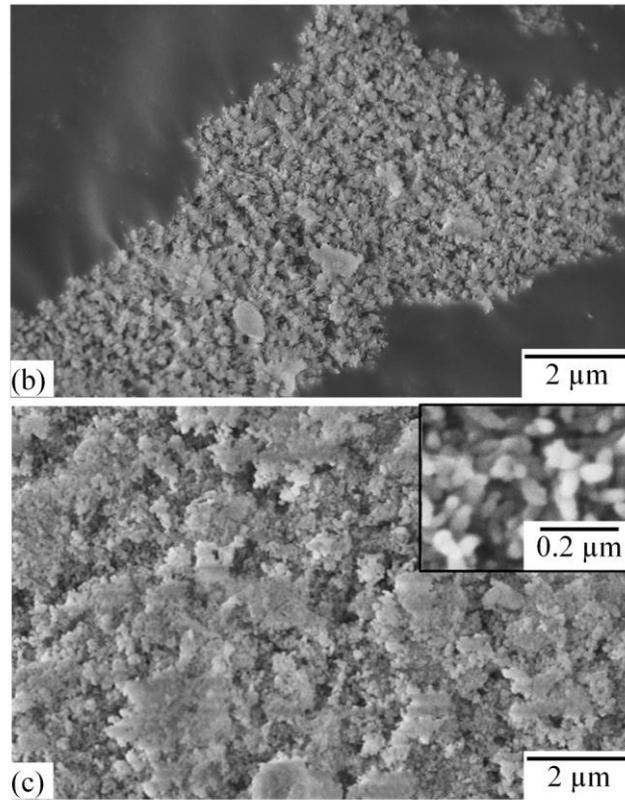
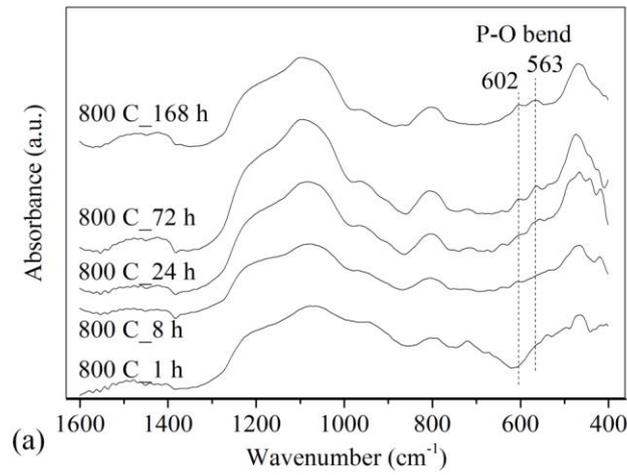


Figure 6. (a) FTIR spectra from sol-gel foam 70S30C scaffolds, sintered at 800°C for 2 hours, following immersion in SBF for 1, 8, 24, 72 and 168 h; SEM images of the sol-gel foam 70S30C scaffolds sintered at 800°C for 2 h after soaking in SBF for (b) 3 days and (c) 7 days. Inset of (c) shows high magnification SEM images of the needle like HCA crystals formed on the surface of the scaffolds.

Table 1. Characterisation summary for 70S30C bioactive glass sol-gel foams: T_s

= sintering temperature, t_s = sintering hold time ρ_f = foam density, D =

interconnected pore diameter, σ_{max} = maximum compressive strength; t_{HCA} =

time of HCA layer formation (from FTIR).

T_s [°C]	t_s [h]	ρ_f [g cm ⁻³]	Modal D [μm]	σ_{max} [MPa]	t_{HCA} [h]
750	4	0.22	104	2.3 ± 0.9	24
775	4	0.42	71	4.48 ± 1.0	168
800	2	0.45	116	4.58 ± 0.9	168
800	4	0.48	31	4.56 ± 1.0	168

Table 2. Compressive strength of 70S30C bioactive glass sol-gel foams sintered at

800 °C for 2 h, following immersion in SBF: ρ_f = foam density; %P = percentage

porosity; σ_{max} = maximum compressive strength.

Time in SBF [h]	ρ_f [g cm ⁻³]	%P	σ_{max} [MPa]
0	0.60	78	6.0 ± 0.8
2	0.58	79	5.2 ± 0.6
24	0.57	79	4.4 ± 0.4
72	0.56	80	5.0 ± 0.5
168	0.38	86	2.1 ± 0.1