

Direct Visualization of Soliton CO Overlayers on Supported Pd Nanoparticles

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Keywords: carbon monoxide, palladium, nanoparticles, scanning tunneling
microscopy

Abstract

The interaction of carbon monoxide (CO) molecules with the facets of noble metal nanoparticles forms the basis of many important catalytic reactions. Using scanning tunneling microscopy (STM), we have studied the adsorption of CO molecules on the (111) facets of Pd nanocrystals supported on a rutile TiO₂(110) substrate. We observed four compact CO overlayers with coverages ranging between 0.5 and 0.6 monolayer. Examination of the positions of the CO molecules in each of the unit cells reveals that one of the overlayers has a rhombic $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ -4CO structure. The other three form rectangular structures, namely $(7 \times \sqrt{3})$ rect-8CO, $c(5 \times \sqrt{3})$ rect-3CO and $c(9 \times \sqrt{3})$ rect-5CO. These are closely related via a soliton model previously proposed on the basis of infrared absorption spectroscopy and low energy electron diffraction. By imaging the CO molecules, we provide direct evidence for the soliton model.

Introduction

The adsorption of carbon monoxide (CO) on metal surfaces has been widely studied because of its relevance to many technological processes, including methanol synthesis,¹ the water-gas-shift reaction,² and CO oxidation in catalytic converters.³ CO molecules typically occupy a number of different adsorption sites and form a variety of ordered overlayers on metal surfaces. Under some conditions, they also cause the surface to restructure, which has important implications for heterogeneous catalysis.⁴

The adsorption behavior of CO on Pd(111) is well understood.⁵⁻¹⁴ At low CO coverages (θ_{CO}), CO molecules occupy threefold fcc-hollow (*H*) sites, forming a $(\sqrt{3} \times \sqrt{3})\text{R}30^\circ\text{-1CO}$ overlayer at 0.33 monolayer (where one monolayer (ML) equals one CO molecule per primitive Pd(111) unit cell and where 1CO indicates that there is one CO molecule in the primitive unit cell).¹² Above 0.33 ML, CO molecules also start to occupy twofold bridge (*B*) sites in addition to the *H* sites. At 0.5 ML, this leads to the formation of two different $c(4 \times 2)\text{-2CO}$ overlayers, one of which has fcc and hcp- hollow site occupation (*HH*), the other of which has bridge site occupation (*BB*).¹² As the CO coverage increases further to between 0.5 and 0.75 ML, CO molecules occupy a combination of different sites and form a series of $(n \times \sqrt{3})$, as well as other ordered structures.¹²⁻¹⁴ Finally, at 0.75 ML, CO molecules occupy atop (*T*-), fcc- and hcp- hollow sites, forming $TH\text{-}(2 \times 2)\text{-3CO}$.¹²

Nanoparticles exhibit properties distinct from their bulk counterparts.^{3,15,16} This, together with their technological relevance,¹⁷⁻¹⁹ has led to a great deal of research concerning the behavior of supported Pd nanoparticles.²⁰⁻²⁵ Recently, we have investigated the adsorption of CO on TiO₂(110)-supported Pd nanoparticles. The

Pd nanoparticles have atomically flat (111) top facets that makes them amenable to study by STM. While some of the CO overlayers observed are identical to their single crystal counterparts,^{12,25} some new CO structures were also observed²⁶ and attributed to the strain caused by the ‘carpet’ growth of Pd nanoparticles across the TiO₂(110) substrate steps.

Here we extend our work on CO adsorption on Pd nanoparticles to include compact CO phases with coverages of $\theta_{\text{CO}} = 0.5\text{-}0.6$ ML. Three of the observed overlayers, directly visualized for the first time, form closely-related rectangular structures. These are the $(7 \times \sqrt{3})_{\text{rect}}\text{-}8\text{CO}$, $c(5 \times \sqrt{3})_{\text{rect}}\text{-}3\text{CO}$, and $c(9 \times \sqrt{3})_{\text{rect}}\text{-}5\text{CO}$ structures, whilst the other overlayer has a rhombic $(\sqrt{7} \times \sqrt{7})\text{R}19.1^\circ\text{-}4\text{CO}$ unit cell.

Experimental

The experiments were carried out using an *Omicron GmbH* low temperature STM described in detail elsewhere.²⁵ TiO₂(110) samples (*PiKem*) were cleaned by cycles of Ar⁺ ion sputtering and vacuum annealing to 1000 K. Sample cleanliness and long-range order were checked using X-ray photoelectron spectroscopy (XPS) and low energy electron diffraction (LEED), respectively. Pd was deposited onto the as-prepared TiO₂(110) surface with the sample held at ~ 720 K. The Pd evaporator comprises a Pd wire (*Advent*, 99.95 %) wrapped by a W filament that is resistively heated. Depositing Pd in this way leads to the formation of well-defined Pd nanocrystals on the TiO₂(110) surface that have (111) top facets.²⁵ These nanocrystals have measured average diameters of 26.3 ± 4.8 nm and heights of 2.1 ± 0.4 nm. Due to this relatively large size, we expect their top facets to have properties similar to that

of the Pd(111) single crystal surface. CO (*SIP Analytical Ltd.*, 99.3 %) was dosed *in-situ* onto the sample at 124 K via a directional doser that is placed ~100 mm away from the STM stage. Based on the dosing geometry, the actual CO dose is estimated to be 100 times the nominal dosage as monitored with the ion gauge in the analysis chamber. It is this estimated CO dosage that is reported here.

STM measurements were made in the constant current mode using electrochemically-etched W tips that were conditioned by outgassing at ~500 K and by high voltage pulses in STM. All STM images were obtained at 124 K.

Results and Discussions

Figure 1a shows a STM image recorded from the (111) top facet of a Pd nanocrystal that has an average diameter (d) of 33 nm and a height (h) of 2.3 nm. Before imaging, the sample was exposed to 11 L of CO (1 L = 1.33×10^{-6} mbar.s). This gives rise to the ordered CO overlayer that can be observed in Fig. 1a. The overlayer is characterized by bright spots that are organized in alternating rows of W and V shapes. Some of these W and V shapes are marked in Fig. 1a. The structure can be described with a rectangular unit cell containing eight protrusions. The dimensions of the unit cell are $(1949 \pm 20) \times (476 \pm 20)$ pm², consistent with a $(7 \times \sqrt{3})a^2$ overlayer, where $a = 275$ pm is the lattice spacing on the (111) face of Pd. On this basis, we propose that this overlayer corresponds to a $(7 \times \sqrt{3})\text{rect-}8\text{CO}$ structure.

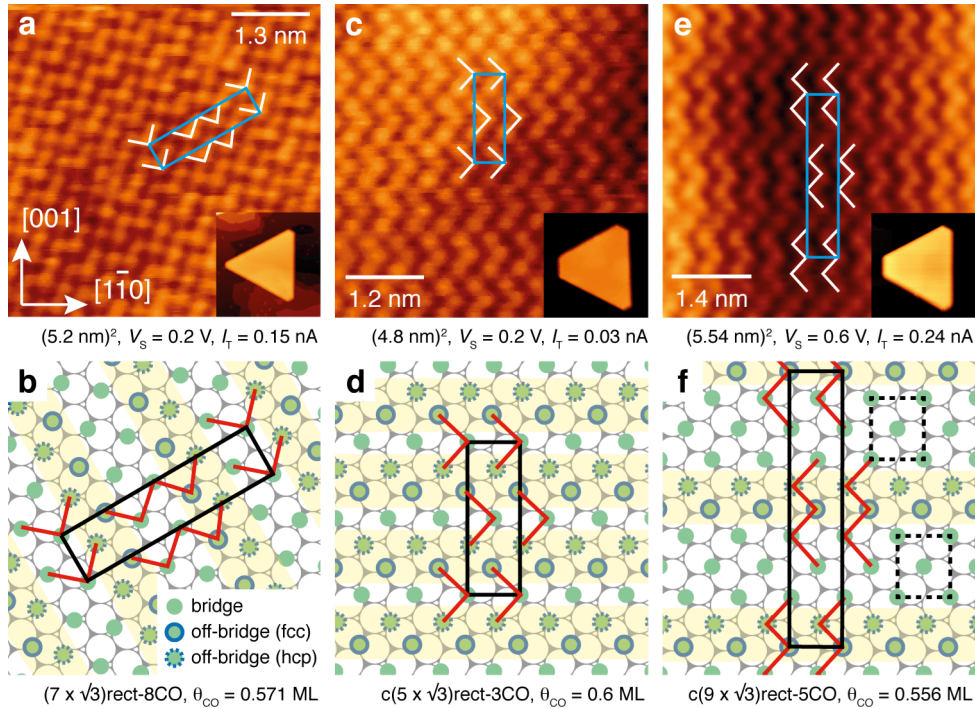


Figure 1. (a,c,e) STM images recorded from the (111) top facets of three different Pd nanocrystals (insets) supported on the $\text{TiO}_2(110)$ surface. The nanocrystals have measured average diameters of 26.3 ± 4.8 nm and heights of 2.1 ± 0.4 nm. The (111) top-facets were covered with different amounts of CO (0.5-0.6 ML), leading to formation of three different CO overlayers, all of which have $(n \times \sqrt{3})$ rect unit cells: (a) $(7 \times \sqrt{3})$ rect -8CO, characterized by alternating rows of ‘V’ and ‘W’ shapes (marked by white lines); (c) $c(5 \times \sqrt{3})$ rect -3CO, characterized by rows of ‘V’ shapes (marked by white lines); (e) $c(9 \times \sqrt{3})$ rect -5CO, characterized by rows of ‘W’ shapes (marked by white lines). All STM images were recorded at 124 K. (b,d,f) Corresponding proposed models for the $(n \times \sqrt{3})$ rect overlayers of CO shown in (a,c,e), respectively. Green circles mark the CO molecules that occupy regular bridge sites. Green circles enclosed with solid green lines mark the CO molecules that occupy off-bridge sites offset towards neighboring fcc hollow sites while those enclosed with dashed lines mark the CO molecules offset towards the neighboring hcp hollow sites. Domain walls, which are comprised of off-bridge CO molecules and separate the antiphase domains of $(2 \times \sqrt{3})$ rect -2CO are shaded yellow. Black solid lines mark the $(n \times \sqrt{3})$ rect unit cells. Red lines in (b), (d) and (f) indicate the alternating ‘V’ and ‘W’, ‘V’, and ‘W’ arrangements of CO molecules observed in (a), (c) and (e) respectively. In (f), a $(2 \times \sqrt{3})$ rect -2CO unit cell is marked with a dashed line.

Using infrared absorption spectroscopy (IRAS) and LEED, Tüshaus et al. investigated compact CO structures on the Pd(111) single crystal surface,¹³ and proposed a so-called *soliton* model to account for their experimental results. At the base CO coverage (θ_{CO}) of 0.5 ML, all CO molecules occupy bridge sites, forming a $(2 \times \sqrt{3})_{\text{rect}}\text{-2CO}$ overlayer. As θ_{CO} increases, additional CO is accommodated in domain walls. While one might expect the CO molecules inside the domain walls to occupy bridge sites, they are actually offset from the bridge sites in order to minimize intermolecular repulsion, leading to off-bridge occupation. Across the domain walls, anti-phase domains of the existing $(2 \times \sqrt{3})_{\text{rect}}$ structures are formed.

Based on Tüshaus et al.'s interpretation,¹³ we propose a similar model to account for the overlayer observed in Fig. 1a. Our model is shown in Fig. 1b and has a unit cell comprising eight CO molecules: four occupy regular bridge sites (green filled circles), two occupy off-bridge sites offset towards the neighboring fcc-hollow sites (green circles enclosed with solid lines), and the remaining two occupy off-bridge sites offset towards the neighboring hcp-hollow sites (green circles enclosed with dashed lines). Together, the CO molecules at the off-bridge sites form domain walls (shaded in yellow in Fig. 1b) that separate the antiphase domains of the existing bridge-bonded $(2 \times \sqrt{3})_{\text{rect}}\text{-2CO}$. This arrangement of CO molecules leads to the structure of $(7 \times \sqrt{3})_{\text{rect}}\text{-8CO}$ ($\theta_{\text{CO}} = 0.571$ ML). The W and V shapes shown in the model are reproduced to scale and superimposed onto the image in Fig. 1a, confirming the good fit with the proposed model.

We observed two other rectangular CO overlayers formed on the (111) top-facets of the Pd nanocrystals. One of the overlayers is characterized by rows of V

shapes (Fig. 1c) and has a unit cell containing six bright spots. The unit cell has dimensions of $(1430 \pm 20) \times (480 \pm 20) \text{ pm}^2$ and is consistent with a $(5 \times \sqrt{3})a^2$ overlayer. The other overlayer is characterized by rows of W shapes (Fig. 1e) and has a unit cell containing ten bright spots. This unit cell has dimensions of $2.44 \times 0.47 \text{ nm}^2$, consistent with a $(9 \times \sqrt{3})a^2$ overlayer.

As before, we construct models for each of these overlayers based on the *soliton* model.¹³ As shown in Fig 1d, the model proposed for the $(5 \times \sqrt{3})\text{rect}$ overlayer has a unit cell containing six CO molecules: two occupy bridge sites, while the other four occupy off-bridge sites. This leads to a $c(5 \times \sqrt{3})\text{rect}-3\text{CO}$ structure ($\theta_{\text{CO}} = 0.6 \text{ ML}$). Note that such a model has already been proposed for identical overlayers on Pd(111) single crystal surfaces.¹³ The model proposed for the $(9 \times \sqrt{3})a^2$ overlayer is shown in Fig. 1f and has a unit cell comprising ten CO molecules: six occupy the regular bridge sites while the other four occupy off-bridge sites, leading to a $c(9 \times \sqrt{3})\text{rect}-5\text{CO}$ structure ($\theta_{\text{CO}} = 0.556 \text{ ML}$). The V and W shapes shown in the models of Fig 1c,e are reproduced to scale and superimposed onto the images in Fig. 1c,e thereby confirming the good fit to the models. As with the model in Fig. 1b, the CO molecules at the off-bridge sites are shaded yellow in Fig. 1d,f and form domain walls that separate antiphase domains.

Figure 2 show schematic models for the $(n \times \sqrt{3})\text{rect}$ CO structures that have coverages between $3/5$ and $7/13 \text{ ML}$. Fig. 2a shows the $(5 \times \sqrt{3})\text{rect}$ structure which comprises anti-phase domain walls (highlighted in yellow) that alternate with domains that consist of single rows of CO molecules in bridge sites (marked with

green circles). Each domain wall contains two rows of CO molecules. To minimize intermolecular repulsion,¹³ these are displaced away from each other towards the neighboring threefold sites, leading to off-bridge occupation. For the $(5 \times \sqrt{3})_{\text{rect}}$ structure shown in Fig. 2a, molecules in one of the CO rows (marked with green circles enclosed with blue solid lines) are offset to the bottom-left, i.e. towards the neighboring fcc-hollow sites, whilst the other row contains CO molecules (marked with green circles with blue dashed lines) offset in the opposite direction towards the top-right, i.e. towards the neighboring hcp-hollow sites. The displacement directions of the CO molecules in the domain walls are indicated with arrows in Fig. 2a(left). We will refer to domain walls with this arrangement of CO molecules as type *A*.

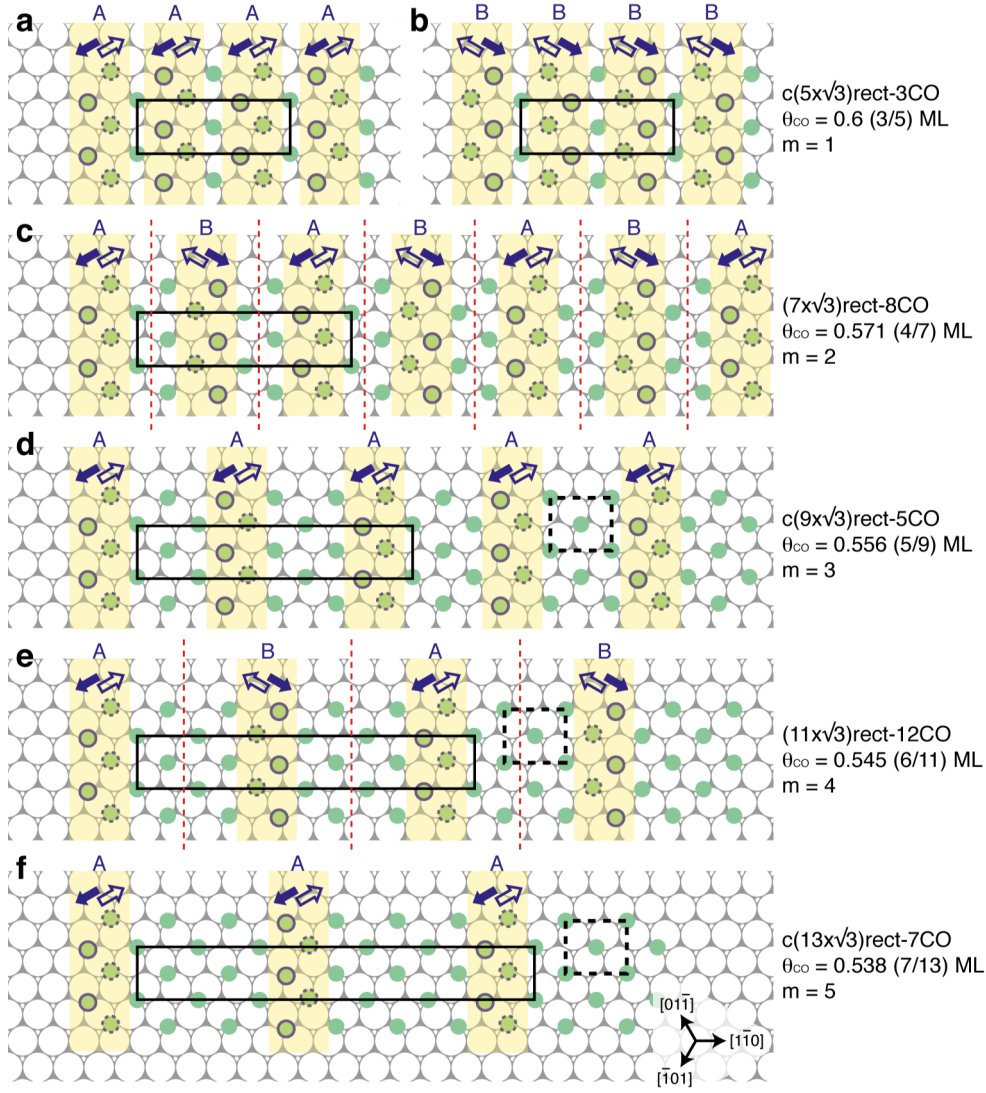


Figure 2. Schematic illustrations of the $(n \times \sqrt{3})\text{rect}$ structures on Pd(111) with CO coverages between 0.6 and 0.538 ML. Green circles mark the CO molecules that occupy bridge site in the $(2 \times \sqrt{3})\text{rect}$ domains. Yellow shading marks the anti-phase domain walls. Inside the domain walls, green circles enclosed with blue solid lines mark the off-bridge CO molecules that are offset towards the neighboring fcc-hollow sites, while those enclosed with blue dashed lines mark the off-bridge CO molecules that are offset towards the neighboring hcp-hollow sites. Filled (open) arrows indicate the displacement directions of the off-bridge CO molecules that are offset towards the neighboring fcc- (hcp-) hollow sites, respectively. The difference in their displacement directions leads to two different types of domain walls that we denote ‘A’ and ‘B’. Black solid lines indicate the $(n \times \sqrt{3})\text{rect}$ unit cells whereas black dashed lines mark some of the $(2 \times \sqrt{3})\text{rect}$ unit cells. The CO coverage (θ_{CO}) and the number of CO rows, m , in each of the $(2 \times \sqrt{3})\text{rect}$ domains are indicated. In (c) and (e), red dashed lines mark glide reflection planes formed between CO patches with domain walls of different types.

Figure 2b shows another $(5 \times \sqrt{3})_{\text{rect}}$ structure that has a different arrangement of CO molecules inside its domain walls that we denote as type *B*. It consists of two rows of off-bridge CO, one of which contains CO molecules offset to the top-left, i.e. towards the neighboring hcp-hollow sites, the other containing CO molecules offset to the bottom-right, i.e. towards the neighboring fcc-hollow sites. The two models in Fig. 2a,b cannot be distinguished from the STM images (Fig. 1).

Figure 2c shows a model of the $(7 \times \sqrt{3})_{\text{rect}}$ structure. It contains $(2 \times \sqrt{3})_{\text{rect}}$ domains that are separated alternately by type *A*, then type *B* domain walls. In Fig. 2d-f, as we continue to increase the value of *n* in the $(n \times \sqrt{3})_{\text{rect}}$ structures, it becomes apparent that there is a pattern to the type of domain wall observed. When there are an odd number of CO rows in the antiphase domains, the domain walls are all of either type *A* or type *B*, whereas with an even number of CO rows, the domain walls alternate between *A* and *B*. As indicated by the red dashed lines in Fig. 2c, e, the CO arrangement in the latter case also gives rise to glide-reflection symmetry between CO patches containing domain walls of different types.

If we denote the number of rows in the $(2 \times \sqrt{3})_{\text{rect}}$ domains as an integer variable, *m*, we can also express a number of other relationships. The length of the soliton $(n \times \sqrt{3})_{\text{rect}}$ unit cell along $\{\bar{1}01\}$, is given by *na*, where $n = 2m + 3$ and the number of CO molecules present in the $(n \times \sqrt{3})_{\text{rect}}$ unit cell is $(2m+4)$ which gives a CO coverage of $(m+2)/(2m+3)$ ML.

Exposing the $c(9 \times \sqrt{3})_{\text{rect}}$ -5CO overlayer (Fig. 1c) to an extra 0.33 L CO leads to a structural conversion into another compact CO structure. As shown in the

STM image of Fig. 3a, the newly-formed structure is characterized by a rhombic arrangement of CO and has around a quarter of the CO molecules appearing brighter than the remainder. The white rhombus in Fig. 3a is drawn with the vertices over the brightest spots and measurements reveal the side lengths to be 718 ± 10 pm, very close to the value of $\sqrt{7}a$. Furthermore, one of its sides is aligned at an angle of 19.1° from the vertical direction (marked by the blue guideline). As such, we propose a $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ unit cell for this overlayer, although we note that the ordering is not perfect. For instance, the green rhombus only has bright spots at two of its vertices. A $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ structure has been reported on several single crystal metal surfaces: Pd(111),¹² Rh(111),²⁷ and Ni(111).²⁸ On Ni(111), the $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ -4CO unit cell has one CO occupying an atop site and the remaining three CO molecules in bridge sites.²⁸ On this basis, we propose our $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ structure in Fig. 3a to have the same occupancy, as illustrated by the model in Fig. 3c. To further confirm our assignment of the structure, we constructed a Pd(111) lattice at the center of the STM image (Fig. 3b) based on the assumption that the brightest spots are CO molecules at atop sites. Examination of the positions of the STM spots with respect to the Pd(111) grid reveals that all other CO molecules in the rhombus (darker STM spots) occupy bridge sites, hence confirming a $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ -4CO overlayer with atop and bridge site occupation.

We note that the $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ -4CO overlayer has the same coverage as the $(7 \times \sqrt{3})\text{rect}$ -8CO overlayer. In the $(7 \times \sqrt{3})\text{rect}$ -8CO overlayer, half of the CO molecules occupy regular bridge sites while the other half occupy off-bridge sites. In

contrast, the $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ -4CO structure has 3/4 of CO molecules occupying regular bridge sites with the remaining 1/4 of CO molecules occupying atop sites. Presumably, the decrease in stability in having two out of every four CO molecules in off-bridge sites is similar to that incurred by accommodating one out of every four CO molecules in atop sites.

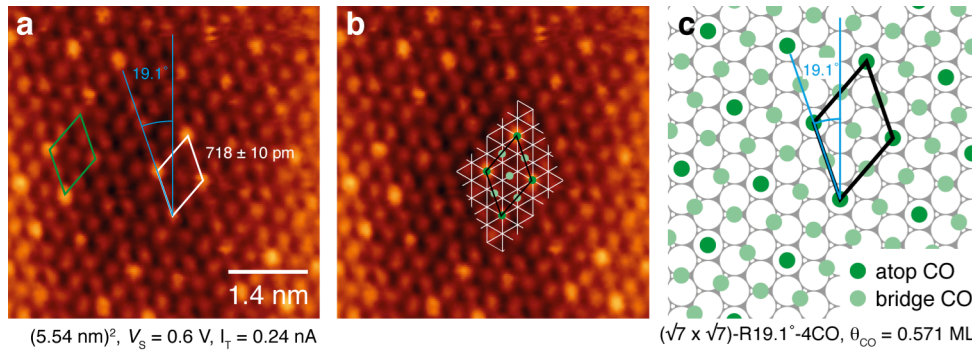


Figure 3. (a) STM image of $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ -4CO structure following an additional exposure of 0.33 L CO to the same island as shown in the inset of Fig. 1c. A rhombic structure is marked with white lines and is formed by four brighter spots at each of the vertices. Green lines mark a rhombus structure that has only two brighter spots at its vertices. The rhombic unit cell has a side-length of 718 ± 10 nm, very close to the value of $\sqrt{7}a$ (where $a = 275$ pm is the interatomic distance on a Pd(111) surface). The light blue guideline is aligned to one side of the rhombic unit cell and is exactly 19.1° from the vertical, as would be expected from the ideal $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ structure. The STM image was recorded at 124 K. (b) As (a). On the basis that the brightest spots correspond to CO molecules at atop sites, a perfect hexagonal Pd(111) grid is superimposed over part of the image, revealing a $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ unit cell with atop and bridge site occupation. (c) Proposed model for the $(\sqrt{7} \times \sqrt{7})R19.1^\circ$ -4CO unit cell. Dark and light green circles mark the CO molecules that occupy atop and bridge sites, respectively.

Summary

In conclusion, using STM we have observed four compact overlayers of CO that form on the (111) top-facets of $\text{TiO}_2(110)$ -supported Pd nanocrystals. The CO coverages of these overlayers range between 0.5 and 0.6 ML. Three overlayers were observed with rectangular unit cells: $(7 \times \sqrt{3})\text{rect-8CO}$, $c(5 \times \sqrt{3})\text{rect-3CO}$, and $c(9 \times \sqrt{3})\text{rect-5CO}$. The models proposed for these phases are all related to the *soliton* model suggested by Tüshaus et al.¹³ CO molecules occupy off-bridge sites in domain walls that separate CO in bridge sites that form anti-phase domains. The

other phase observed has a rhombic unit cell, $(\sqrt{7} \times \sqrt{7})R19.1^\circ-4\text{CO}$, with CO molecules occupying atop and bridge sites.

Acknowledgements

This work was supported by the European Cooperation in Science and Technology Action CM1104, the European Research Council Advanced Grant ENERGYSURF (GT), the Royal Society (UK), and Alexander von Humboldt Stiftung (Germany).

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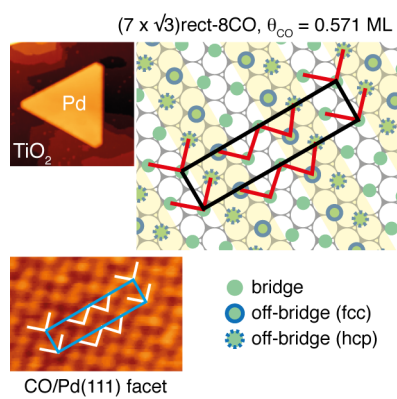
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