## 9-Chloroacridine

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#### Abstract

C}_{13} \mathrm{H}_{8} \mathrm{NCl}\), orthorhombic, $P 2_{1} 2_{12} 2_{1}, a=$ 6.850 (3), $b=11.662$ (5), $c=12.705$ (5) $\AA, D_{m}=$ $1.40(1), D_{c}=1.393 \mathrm{~g} \mathrm{~cm}^{-3}$ for $Z=4$. Equi-inclination Weissenberg intensities, scanned with an automatic densitometer. The structure, solved by direct methods, refined to $R \quad 0.0577$ and $R_{w} 0.0649$ for 1653 reflexions. The crystal structure is characterized by partial stacking of the chromophores; these are slightly non-planar.


Introduction. As part of a study of the effects of various ring substitutions on the molecular geometry and solidstate packing of the acridines (Achari \& Neidle, 1976) the crystal structure of the title compound was determined. Studies of several other acridines substituted only at the 9 -position have been reported: 9 -aminoacridine hydrochloride monohydrate (Talacki, Carrell \& Glusker, 1974), 9-chloromethylacridine hydrochloride (Zacharias \& Glusker, 1974), and 2-[3-(9-

Table 1. Final positional parameters for the nonhydrogen $\left(\times 10^{4}\right)$ and for the hydrogen atoms $\left(\times 10^{3}\right)$

For H , the atom numbering denotes the atom to which each H is bonded. The $X-H$ bond lengths, $r,(\AA)$ are given. Estimated standard deviations are in parentheses.

|  | $x$ | $y$ | $z$ | $r$ |
| :---: | :---: | :---: | :---: | :---: |
| C(1) | 6119 (5) | 4889 (3) | 2379 (3) |  |
| C(2) | 7303 (7) | 4785 (4) | 1505 (4) |  |
| C(3) | 6833 (7) | 3998 (4) | 703 (4) |  |
| C(4) | 5222 (6) | 3327 (4) | 760 (3) |  |
| C(5) | -596 (7) | 2108 (4) | 2478 (4) |  |
| C(6) | -1901 (7) | 2163 (4) | 3281 (4) |  |
| C(7) | -1610 (7) | 2920 (5) | 4138 (4) |  |
| C(8) | -3 (6) | 3605 (4) | 4180 (3) |  |
| C(9) | 3101 (5) | 4249 (3) | 3328 (3) |  |
| N(10) | 2340 (5) | 2733 (3) | 1643 (3) |  |
| C(11) | 3923 (6) | 3405 (3) | 1645 (3) |  |
| C(12) | 1414 (6) | 3576 (3) | 3355 (3) |  |
| C(13) | 4408 (5) | 4207 (3) | 2476 (3) |  |
| C(14) | 1105 (6) | 2824 (3) | 2478 (4) |  |
| Cl | 3585 (2) | 5172 (1) | 4376 (1) |  |
| H(1) | 618 (6) | 538 (4) | 301 (4) | 0.98 (4) |
| H(2) | 862 (11) | 518 (6) | 144 (7) | 1.02 (7) |
| H(3) | 749 (8) | 387 (4) | 0 (4) | 1.01 (5) |
| H(4) | 485 (8) | 281 (5) | 30 (4) | 0.88 (5) |
| H(5) | -71 (9) | 164 (5) | 192 (5) | 0.91 (5) |
| H(6) | -315 (9) | 169 (5) | 356 (5) | 1.08 (6) |
| H(7) | -250 (9) | 295 (5) | 477 (5) | 1.01 (6) |
| H(8) | 17 (9) | 402 (5) | 468 (6) | $0 \cdot 81$ (6) |

acridinylamino)propylaminolethanol (Glusker, Gallen \& Carrell, 1973).

9-Chloroacridine was crystallized from fluorobenzene as light-yellow plates elongated along a. Photographs revealed the crystals to be orthorhombic, with systematic absences $h 00: h=2 n+1,0 k 0: k=2 n+1$ and $00 l: l=2 n+1$, thus determining the space group

Table 2. Non-hydrogen-atom bond lengths ( $\AA$ ) and valence angles $\left(^{\circ}\right.$ )

Estimated standard deviations are in parentheses.

| $\mathrm{C}(1)-\mathrm{C}(2) \quad 1$ | 1.381 (6) | $\mathrm{C}(8)-\mathrm{C}(12)$ | 1.429 (5) |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}(1)-\mathrm{C}(13) \quad 1$ | 1.421 (5) | $\mathrm{C}(9)-\mathrm{C}(12)$ | 1.397 (5) |
| $\mathrm{C}(2)-\mathrm{C}(3) \quad 1$ | 1.409 (7) | $\mathrm{C}(9)-\mathrm{C}(13)$ | 1.407 (5) |
| $\mathrm{C}(3)-\mathrm{C}(4) 1$ | 1.355 (6) | $\mathrm{C}(9)-\mathrm{Cl}$ | 1.744 (4) |
| $\mathrm{C}(4)-\mathrm{C}(11) \quad 1$ | 1.437 (5) | $\mathrm{N}(10)-\mathrm{C}(11)$ | 1.338 (5) |
| $\mathrm{C}(5)-\mathrm{C}(6) \quad 1$ | 1.358 (6) | $\mathrm{N}(10)-\mathrm{C}(14)$ | 1.362 (5) |
| $\mathrm{C}(5)-\mathrm{C}(14) \quad 1$ | 1.433 (5) | $\mathrm{C}(11)-\mathrm{C}(13)$ | 1.449 (5) |
| $\mathrm{C}(6)-\mathrm{C}(7)$ | 1.416 (7) | $\mathrm{C}(12)-\mathrm{C}(14)$ | 1.434 (5) |
| $\mathrm{C}(7)-\mathrm{C}(8) \quad 1$ | 1.361 (6) |  |  |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(13)$ | 120.3 (4) | $\mathrm{C}(4)-\mathrm{C}(11)-\mathrm{N}(10)$ | 117.6 (3) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 120.3 (4) | $\mathrm{C}(4)-\mathrm{C}(11)-\mathrm{C}(13)$ | 118.0 (3) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 121.6 (4) | $\mathrm{N}(10)-\mathrm{C}(11)-\mathrm{C}(13)$ | 124.5 (3) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(11)$ | $120 \cdot 6$ (4) | $\mathrm{C}(8)-\mathrm{C}(12)-\mathrm{C}(14)$ | 118.9 (3) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(14)$ | 120.5 (3) | $\mathrm{C}(9)-\mathrm{C}(12)-\mathrm{C}(14)$ | 116.6 (3) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | 120.9 (4) | $\mathrm{C}(8)-\mathrm{C}(12)-\mathrm{C}(9)$ | 124.5 (3) |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | 120.7 (4) | $\mathrm{C}(1)-\mathrm{C}(13)-\mathrm{C}(9)$ | 124.9 (3) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(12)$ | 120.5 (4) | $\mathrm{C}(1)-\mathrm{C}(13)-\mathrm{C}(11)$ | 119.2 (3) |
| $\mathrm{Cl}-\mathrm{C}(9)-\mathrm{C}(12)$ | 119.0 (3) | $\mathrm{C}(9)-\mathrm{C}(13)-\mathrm{C}(11)$ | 116.0 (3) |
| $\mathrm{Cl}-\mathrm{C}(9)-\mathrm{C}(13)$ | 119.3 (3) | $\mathrm{C}(5)-\mathrm{C}(14)-\mathrm{C}(10)$ | 117.4 (3) |
| $\mathrm{C}(12)-\mathrm{C}(9)-\mathrm{C}(13)$ | 121.7 (3) | $\mathrm{C}(10)-\mathrm{C}(14)-\mathrm{C}(12)$ | 124.1 (3) |
| $\mathrm{C}(11)-\mathrm{N}(10)-\mathrm{C}(14)$ | 4) 117.1 (3) | $\mathrm{C}(5)-\mathrm{C}(14)-\mathrm{C}(12)$ | 118.5 (3) |



Fig. 1. The numbering scheme.
uniquely to be $P 22_{1} 2_{1}$. Cell dimensions were obtained from least-squares refinement of $2 \theta$ values measured from calibrated Weissenberg photographs. Intensities were collected by the equi-inclination Weissenberg method with multi-film packs. Layers $0 \mathrm{kl}-5 \mathrm{kl}$ and $\mathrm{h} 0 \mathrm{l}-$ $h 10 l$ were photographed with Ni -filtered Cu Ka radiation. The films were scanned by the Science Research Council microdensitometer service with an Optronics P- 1000 Photoscan. 1653 unique observed reflexions were obtained after interlayer scaling and merging of equivalent reflexions; the average agreement $R_{s}$ between multiple measurements of a reflexion was $0 \cdot 102\left(R_{s}=\Sigma\left|\bar{I}-I_{i}\right| \Sigma I_{i}\right.$, where $\hat{I}$ is the mean of multiple measurements $I_{i}$ ).

The structure was solved by direct methods, and refined by full-matrix least squares; the H atoms were located in a difference synthesis. The refinement converged with anisotropic temperature factors for the nonhydrogen atoms, and isotropic for H. $R$ was 0.0577 and $R_{w}$ was 0.0649 . Weights were of the form $w=$ $1 /\left(\sigma^{2}|F|+0.001 F^{2}\right)$. Table 1 lists the final positional parameters.*

Discussion. Fig. 1 shows the numbering scheme. Table 2 gives the bond lengths and angles. The crystal structure does not utilize any potential molecular

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Fig. 2. The $c$ axis projection of the crystal structure.


Fig. 3. The stacking of the chromophores in the crystal structure, viewed perpendicular to their planes.
symmetry [contrast 3,6-diaminoacridine, proflavine, which has crystallographically imposed mirror symmetry (Achari \& Neidle, 1976)]. Within the limits of error, the molecule has mirror symmetry along the $\mathrm{C}(9)-\mathrm{N}(10)$ plane; the differences between $\mathrm{C}(11)-\mathrm{C}(13)$ and $\mathrm{C}(12)-\mathrm{C}(14)$, and between $\mathrm{N}(10)-\mathrm{C}(11)$ and $\mathrm{N}(10)-\mathrm{C}(14)$ are not significant. A comparison of 9 -chloroacridine with related structures shows that the electron-withdrawing Cl substituent has induced a significant shortening of $\mathrm{C}(9)-\mathrm{C}(12)$ and $\mathrm{C}(9)-\mathrm{C}(13)$, compared with their lengths in 9 -aminoacridine (Talacki, Carrell \& Glusker, 1974), and 2-[3-(9-acridinylamino)propylaminolethanol (Glusker, Gallen \& Carrell, 1973). Their average length here is 1.404 $\AA$, as against 1.436 and $1.434 \AA$, respectively. As has been noted (Jones \& Neidle, 1975), many acridines are significantly non-planar, the extent of the deviation increasing with the amount of side-chain substitution. In 9 -chloroacridine, the two outer rings are mutually inclined at $1.8^{\circ}$.

The crystal structure is shown in Fig. 2. There is relatively little chromophore stacking: Fig. 3 shows the extent of this $3.51 \AA$ interaction, which contrasts with the extensively stacked structures of many more complex acridines (Glusker, Carrell, Berman \& Gallen, 1975).

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[^0]:    * Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 32799 ( 12 pp .). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CH1 1NZ, England.

