

1 **Effects of fuel composition at varying air-fuel ratio on knock resistance**  
2 **during spark-ignition combustion**

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10 **Keywords:** air-fuel ratio, knock resistance, spark-ignition engines, primary refence  
11 fuel, gasoline fuel, RON, variable compression engine, paraffinic fuels, aromatic fuels.

12 **Abstract**

13 Knock resistance of liquid fuels for spark-ignition engines is determined using  
14 standardised tests (RON and MON), however, these do not involve consistent control of  
15 the air-fuel ratio ( $\lambda$ ). In contrast, modern engines have a highly controlled air-fuel  $\lambda$   
16 ratio, often operating in a very narrow range around stoichiometric in order to reduce  
17 pollutant emissions and achieve high thermal efficiencies. Hence, understanding the  
18 effect of  $\lambda$  on knock resistance of the fuel is imperative.

19 This paper investigates the influence of varying equivalence air-fuel ratio  $\lambda$  on the  
20 knock resistance of a range of fuels of equal RON values but differing chemical

21 composition. Binary component primary reference fuels and practical gasolines were  
22 tested with a Ricardo E6 variable compression engine operated at conditions similar to  
23 those used for RON tests. It was found that the knock resistance depended on the air-  
24 fuel ratio at which the engine was operated and the chemical composition of the test  
25 fuel. For all fuels, the knock resistance became insensitive to compression ratio at  
26 stoichiometric and very rich mixtures ( $\lambda=1$  and  $\lambda<0.88$ ). However, the knock resistance  
27 of highly paraffinic fuels was observed to be more sensitive to changes in  $\lambda$  than highly  
28 aromatic fuels.

## 29 **1 Introduction**

30 In spark-ignition engines, knock is an abnormal combustion event in which significant  
31 thermal energy can be undesirably released, potentially leading to severe engine  
32 mechanical damage or reducing engine operating life, performance, and efficiency  
33 (Hamilton and Cowart, 2008, Wang *et al.*, 2017). Knock is a major obstacle in the  
34 further improvement of SI engines. For example, the onset of knock limits the  
35 possibility of increasing the operating compression ratio, or applying turbocharging and  
36 downsizing strategies, that enhance thermal efficiency and power density of SI engines.  
37 Such strategies lead to an increase in pressure and temperature in the unburned mixture  
38 ahead of the flame front, accelerating the auto-ignition in the end-gas and causing  
39 knock (Ratcliff *et al.*, 2018).

40 In spark-ignition engines, knock occurs during flame propagation when the temperature  
41 of the unburned gases rises rapidly, exceeding the self-ignition temperature. Many  
42 mechanisms can therefore contribute to initiating knock. The overall temperature of the

43 in-cylinder contents changes in response to the piston movement as it compresses or  
44 expands the volume of the cylinder contents. The expansion of post-combustion gases  
45 compresses the unburned gases, elevating their temperatures. The unburned gas also  
46 receives radiation from the burned gas and the surrounding combustion chamber walls.  
47 If the local temperature of the unburned gases in the end-gas exceeds the mixture self-  
48 ignition temperature, knocking combustion due to end-gas auto-ignition occurs. Thus,  
49 gas pressure oscillations and an audible ringing sound caused by reflecting pressure  
50 waves driven by the sudden release of energy are observed (Syrimis and Assanis, 2003,  
51 Coetzer *et al.*, 2006, Zhen *et al.*, 2012).

52 Liquid fuels for spark-ignition engines are selected partly based on their ability to resist  
53 knock, which is usually expressed through their octane number (ON). The term ON was  
54 introduced in 1928 as a standardised measurement and specification of a fuel's  
55 resistance to auto-ignition (Stradling *et al.*, 2016; AlAbbad *et al.*, 2017; Kalghatgi and  
56 Stone, 2018). Fuels with a higher octane number can be more resistant to auto-ignition  
57 and thus are often used in higher compression ratio engines. The octane number concept  
58 has since been adopted globally for quantitative knock determination of liquid fuels for  
59 spark-ignition engines, utilising standardised equipment, a cooperative fuel research  
60 (CFR) engine, and the two following different sets of standardised operating conditions.  
61 The first set of operating conditions is used to evaluate the knock rating of fuels under  
62 mild operating conditions and provide a fuel parameter called the Research Octane  
63 Number (RON), while the second set is performed to rate a fuel under severe operating  
64 conditions and obtain the Motor Octane Number (MON). Table 1 shows the operating

65 conditions of RON and MON tests. Overall, fuels are rated for MON at a higher intake  
 66 mixture temperature and a higher engine operating speed than RON tests, representing  
 67 more severe engine operating conditions (Kolodziej and Wallner, 2017). Therefore,  
 68 during MON rating, the temperature of unburned gas is considerably higher at a given  
 69 pressure than that under RON conditions. Hence, practical fuels always have lower  
 70 MON values than RON, typically 10 values lower (Stradling *et al.*, 2016). The  
 71 difference between these two measured parameters (RON – MON) is known as fuel  
 72 sensitivity (S) (Kalghatgi and Stone, 2018).

73 Table 1: RON and MON engine operating conditions.

Parameter	RON	MON
Intake air temperature	52 °C	149 °C
Intake air pressure	Atmospheric	
Coolant temperature	100 °C	
Engine speed	600 rpm	900 rpm
Spark timing	13 °BTDC	14-26 °BTDC
Compression ratio	4 to 18	

74 In spark-ignition engines, the variation of equivalence air-fuel ratio ( $\lambda$ ) has a direct  
 75 influence on the occurrence and intensity of knock (Brock and Stanley, 2012). For a test  
 76 or sample fuel, RON and MON are determined in a CFR engine at a relative air-fuel  
 77 ratio of the mixture that produces the highest knock intensity. By using the falling level  
 78 technique, the carburettor fuel level can be changed from a high or rich mixture  
 79 condition to a low or lean mixture condition to reach the standardised knock intensity

80 (ASTM Int., 2019). However, this relative air-fuel ratio value cannot be quantified, and  
81 it is not known if the knock characteristics of test and sample fuels are compared at  
82 similar operating  $\lambda$ . In contrast, modern engines have a highly controlled  $\lambda$  ratio, often  
83 operating in a very narrow range around stoichiometric to reduce pollutant emissions  
84 and achieve high thermal efficiencies. Hence, understanding the effect of the  
85 equivalence air-fuel ratio  $\lambda$  on the knock resistance of a fuel is imperative, especially for  
86 future fuels that will potentially be derived from non-conventional sources, possessing a  
87 different chemical composition and a reduced variety of components.

88 The effects of the variation of equivalence air-fuel ratio  $\lambda$  and chemical composition of  
89 spark-ignition fuels have been investigated in several studies. Huber *et al.*, (2013)  
90 examined the standardised octane rating methods for RON and MON determination to  
91 develop a new engine-based test method that could better fit with modern engine  
92 technologies. It was observed that the variation of  $\lambda$  highly affected the ignition and  
93 combustion characteristics of the test fuels. Paraffinic primary reference fuels displayed  
94 different ignition and combustion characteristics than conventional gasoline. The  
95 variation of  $\lambda$  was also shown to have a significant impact on the resultant knock  
96 intensity of these test fuels. However, the effect was not explicitly investigated, and  
97 insufficient understanding about the behaviour of these fuels on a commercial gasoline  
98 engine operating at a fixed air-fuel ratio was obtained.

99 (Kolodziej and Wallner, 2017) investigated the effects of varying  $\lambda$  on the knocking  
100 characteristics and RON rating of a range of fuels all with a RON of 98 but with  
101 varying composition (achieved through different proportions of iso-paraffinic and

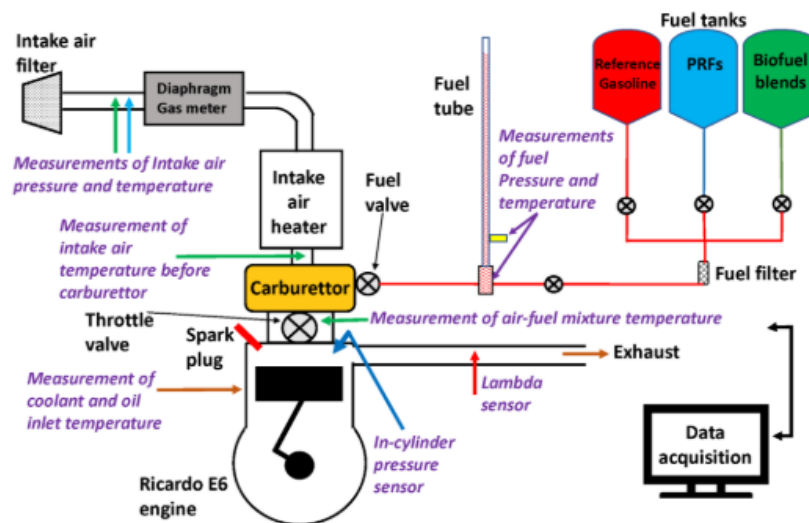
102 aromatic species and also ethanol). It was found that the knock tendency of PRFs  
103 reduced significantly as the mixture moved towards stoichiometric, displaying greater  
104 knock resistance in comparison to aromatic and ethanol blends of the same RON  
105 values. The results indicated that PRFs are more sensitive to changes in  $\lambda$  ratios than  
106 conventional gasoline fuels. A similar significant sensitivity in the combustion  
107 characteristics of PRF corresponding to a change in the rate of fuel supply was also  
108 reported by (Dec and Sjöberg, 2004) studying the effect of fuel chemistry on  
109 combustion phasing and ignition control of a single cylinder direct injection HCCI  
110 engine. In this study, three fuels; pure isooctane, gasoline and PRF 80, were examined.  
111 Slight changes in the autoignition characteristics of both isooctane and gasoline during  
112 the increase of fuel flow rate were observed compared to the PRF, attributable to the  
113 rise in the PRF cool-flame activity while increasing fuel mass flow rate.

114 (Montoya *et al.*, 2018) studied the effect of varying equivalence air-fuel ratio on knock  
115 tendency in two different engine configurations; a CFR unit and a converted Lister  
116 Petter TR2 Diesel engine (TR2) that operated as a spark-ignition engine, from lean to  
117 stoichiometric. Several fuel blends made of biogas, natural gas, propane, and hydrogen  
118 were tested. It was found that in the CFR engine, a lean mixture decreased the knocking  
119 tendency. Therefore, the engine could be operated at a higher critical engine  
120 compression ratio than a stoichiometric mixture. However, the opposite effect of  
121 varying equivalence air-fuel ratios was observed in the converted diesel engine,  
122 increasing the knock tendency with the supply of leaner mixtures. This was attributed to  
123 the increase in the mixture pressure at the end of engine compression stroke due to

124 increased intake charge density while introducing more air, thus leaner mixture, to the  
125 engine. At a fixed and super lean equivalence air-fuel ratio, (Naruke *et al.*, 2020)  
126 investigated the effect of fuels with different ignition characteristics on the knock  
127 propensity of a single cylinder spark-ignition engine operated at a fixed engine  
128 compression ratio of 15:1 and  $\lambda$  of 1.8. It was found that, at the occurrence of knock  
129 limit, the octane number and octane index of the fuels investigated did not correlate  
130 well with the crank angle position CA50.

131 While the variation of both equivalence air-fuel ratio  $\lambda$  and fuel composition has been  
132 observed to affect knock resistance significantly, there remains a limited systematic  
133 understanding of the combined influence of these parameters. Such insights are  
134 increasingly necessary with the uptake of alternative fuels and precise control of air-fuel  
135 stoichiometry. Therefore, this study investigates the effect of varying air-fuel ratio and  
136 fuel composition on knock resistance and knocking combustion characteristics, utilising  
137 practical gasoline fuels and PRFs of similar octane rating. All tests were conducted  
138 using a single cylinder E6 variable compression ratio spark-ignition engine, operated at  
139 standardised conditions similar to those used in RON measurements with consistent  
140 control of the air-fuel ratio ( $\lambda$ ) and knock limit.

141 **2 Experimental Setup**



142

143 Figure 1: Schematic diagram of the engine test rig and measurement systems.

144 Figure 1 shows a general layout of the experimental systems utilised in this study. A  
145 Ricardo E6 single cylinder variable compression engine (serial number of 98/67)  
146 configured for spark ignition combustion, was used for all experiments. The  
147 compression ratio of the test engine was manually varied between 4.5:1 and 9.1:1 while  
148 the engine was running by increasing or decreasing the relative position of the engine  
149 head to the crankshaft. The test engine was coupled to an electric dynamometer of  
150 swinging field direct current type. Table 2 summarises the test engine geometry  
151 specifications and valves timings.

152 The air-fuel mixture was prepared via the test engine carburettor (Solex 35 F.A.1) and  
153 ignited by a 14 mm spark plug with 0.7mm gap (NGK BPR6HS), situated at the side of  
154 the combustion chamber between the valves. To determine the air-fuel ratio  $\lambda$ , an  
155  $O_2/\lambda$  sensor (ECM AFRecorder 1200), fitted in an M18 hole approximately 120



156 mm downstream of the engine exhaust valve, was utilised. The O<sub>2</sub>/lambda sensor was  
157 calibrated for every test using the Hydrogen-to-Carbon (H/C) and Oxygen-to-Carbon  
158 (O/C) mass ratios of the test fuel. As the engine was operated at wide-open throttle for  
159 all tests, the ratio of the air-fuel mixture was varied and controlled during experiments  
160 using the test engine carburettor needle valve.

161 Several measuring transducers and sensors were installed for acquisition of pressure and  
162 temperature readings during tests as follows. Measurements of in-cylinder pressure  
163 were taken by a water-cooled piezo-electric pressure transducer (Kistler 6041B) in  
164 conjunction with a Kistler 5007 charge amplifier. The in-cylinder pressure  
165 measurements were referenced to the intake manifold pressure at a time at which the  
166 piston was at BDC and the inlet valve open. Temperature measurements for air, coolant  
167 water and lubricant oil were made by K-type thermocouples connected to thermocouple  
168 amplifiers of type Adafruit MAX31855 and placed at different positions. For  
169 monitoring and controlling the temperature of the air-fuel mixture just before delivery  
170 to the engine, a K-type thermocouple was also fitted after the carburettor and connected  
171 to a PID box so that the mixture intake temperature could be controlled. The relative  
172 humidity of the ambient air was measured using a capacitive humidity sensor type  
173 HPP805A031. All signals from the measuring instruments were acquired as analogue  
174 inputs by two separate PCs equipped by National Instruments multifunction I/O data  
175 acquisition cards with a high-speed sampling rate of 1.25 MS/s.

176 Table 2: Geometry specifications of the Ricardo E6 engine.

Number of cylinders	Single cylinder
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Compression ratio	Variable from 4.5:1 to 20:1
Cylinder Bore	76.2 mm
Cylinder Stroke	111.13 mm
Swept Volume	506.8 cm <sup>3</sup>
Number of Valves	1 inlet, 1 exhaust
Inlet Valve timing	IVO: 9 CAD BTDC    IVC: 35 CAD ABDC
Exhaust Valve timing	EVO: 42 CAD BBDC    EVC: 8 CAD ATDC
Cooling system	Water-cooled
Aspiration system	Natural

### 177      **3 Experimental Methodology**

#### 178      **3.1 Knock detection technique**

179      Knocking combustion cycles were detected directly using measurements of in-cylinder  
180      pressure. In order to distinguish between knocking and non-knocking cycles, the in-  
181      cylinder pressure signal was filtered by a bandpass filter according to the filter settings  
182      summarised in Table 3. The knock index MAPO, as described in Equation 1, was then  
183      obtained and compared to a pre-determined threshold of 0.5 bar, a value used in a  
184      previous study by (Kalghatgi, 2018). Therefore, the examined cycle was considered to  
185      be a knocking combustion cycle if it displayed an amplitude higher than the  
186      predetermined threshold within the filter window (10 CAD BTDC to 90 CAD ATDC).  
187      The analysis was applied for 50 consecutive combustion cycles. The percentage ratio  
188      between knocking combustion cycles and the total number of measured combustion  
189      cycles was calculated to find a term referred to as the knock frequency factor (KFRQ),  
190      see Equation 2. Hence, this factor was used as an index parameter in order to find the  
191      critical compression ratio of each test fuel at various air-fuel ratios.

192 
$$\text{MAPO (bar)} = \max [ [incylinder\ pressure] ]_{10\ CAD\ BTDC}^{90\ CAD\ ATDC}$$
 Equation 1

193 
$$\text{KFRQ (\%)} = \frac{\text{number of Knocking cycles}}{\text{total number of measured cycles}} \times 100$$
 Equation 2

194 Table 3: A summary of the bandpass filter settings for knock detection.

Filter type	Band-pass
Detection window	10 CAD BTDC to 90 CAD ATDC
Low Cut off frequency (Hz)	3600
High Cut off frequency (Hz)	18000
Sampling frequency (Hz)	36000
Threshold for knocking cycle detection (bar)	0.5

195 **3.2 Test procedure**

196 Table 4 summaries the test operating conditions utilised in this work. In general, all  
 197 experimental work was conducted at conditions similar to the International Standard  
 198 Test Method for Research Octane Number (RON) of Spark-Ignition Engine Fuel  
 199 (ASTM Int., 2019), excluding the coolant temperature; this was maintained at 70 °C  
 200 during motoring tests following the recommendation of the engine manufacturer as the  
 201 most suitable condition for engine operation.

202 Table 4: Test operating conditions.

Operating condition	Value
Inlet pressure	Atmospheric
Inlet air temperature	Based on the atmospheric pressure of the test day
Coolant in temperature	70 °C ± 1.5
Oil Temperature	55 °C ± 2
Engine speed and Load	600 rpm and Wide open throttle
Spark timing	13 CAD BTDC
Fuel/Air ratio	Varied between 1.00 and 0.82
Compression ratio	Variable

203

204 As shown in Table 4, the engine was always operated at wide open throttle, and fixed  
205 engine speed and spark timing of 600 rpm and 13 CAD BTDC, respectively. Each  
206 combustion test always started at engine CR of 6.91 and  $\lambda$  of 1. The engine CR was  
207 then increased gradually in order to find the CR, at  $\lambda=1$ , that resulted in engine  
208 knocking of 10% KFRQ. Next, exhaust lambda  $\lambda$  sweeps in the range from 0.98 to 0.84  
209 with increments of 0.025 were undertaken and the maximum engine CR at each  $\lambda$  point  
210 for a higher KFRQ of 30% was determined. The different values of KFRQs selected for  
211 t  $\lambda=1$  and richer mixtures, 10% and 30% respectively, were chosen as it was anticipated  
212 that a higher engine CR would be required to instigate knock at stoichiometric  
213 conditions and could result in knock events of significantly greater magnitude.

214 In order to investigate the experimental repeatability during combustion, the engine was  
215 fuelled by a reference gasoline fuel (Gasoline 91.3 RON) and operated in combustion  
216 mode. 13 repeated tests of this fuel, in combustion mode, were performed on different  
217 days throughout the experimental work period at the same set of operating conditions,  
218 Table 4, but at a constant engine CR of 6.91. Table 5 shows the variability of readings  
219 from the sensors and transduces in addition to the calculated standard deviation for each  
220 measurement.

221 Table 5: Variations of measurement readings and the calculated standard deviation for 13  
222 repeated tests of the reference test fuel Gasoline 91.3.

	Measurement	Min	Max	Mean	$\pm$ SD
Temperatures [°C]	Coolant in	69.6	76.0	72.7	2.0
	Oil	52.5	57.3	54.8	1.3
	Air before carburettor*	47.0	48.5	47.8	0.8

	Exhaust gas	445.3	506.4	477.3	21.4
	Fuel	16.9	19.9	18.8	0.8
	Room	18.8	22.7	20.7	1.0
Relative Humidity	Relative humidity	24.6	37.2	30.9	4.7
Pressure [bar]	Atmospheric pressure	0.993	1.020	1.001	0.01
Flow rate [g/sec]	Mass flow rate of air	2.2	2.5	2.3	0.1
	Mass flow rate of fuel	0.1	0.2	0.1	0.004

223 \* For 3 days at a constant atmospheric pressure of 0.996 bar

### 224 **3.3 Fuels investigated**

225 Fuels of known RONs, chemical and physical properties but with different chemical  
 226 composition were selected for the investigation. The test fuels were divided into three  
 227 sets. Firstly, two commercial highly aromatic gasoline fuels of RONs of 97 and 91.3,  
 228 respectively, were obtained from Haltermann Carless UK LTD. Secondly, an  
 229 oxygenated gasoline blend, A 5 % (V/V) ethanol and reference gasoline with RON of  
 230 90.9 was blended and delivered by BP Formulated Products Technology, UK. Lastly,  
 231 Primary Reference fuels (PRFs) were prepared in-house by blending n-Heptane into  
 232 iso-Octane on a volumetric basis corresponding to the required RON, varying the  
 233 percentage of n-Heptane (3%, 10% and 15%) to produce PRFs of RON 97, 90 and 85,  
 234 respectively. Both n-Heptane and iso-Octane with GC purity of 99% and 99.8%,  
 235 respectively, were purchased from MERCK CHEMICALS.

236 Table 6 summarises the physical and chemical properties of the test fuels and also of the  
 237 molecules utilised in preparing the test blends: Ethanol, n-Heptane and iso-Octane. In  
 238 Table 6, it can be seen that the RON values of the fuels investigated fall in the range of  
 239 97 and 85. Two fuels; Gasoline 97 and PRF 97 possessed equivalent RON, while the  
 240 PRF 85 had the lowest RON value of 85. The other fuels G90.09+5%(V/V) Ethanol,

241 Gasoline 91.3, and PRF90 were distributed between the upper and lower borders of the  
 242 RON range with small differences in their RON values; 93.7, 91.3 and 90, respectively.

243 Table 6: Physical and chemical properties of the test fuels.

Fuels	Density <sup>a</sup> g/cm <sup>3</sup>	Heat of vaporization <sup>b</sup> kJ/mol	H/C <sup>c</sup>	O/C <sup>c</sup>	Stoichio- metric AFR <sup>c</sup>	Calorific value <sup>d</sup> MJ/kg	RON <sup>e</sup>
PRF 85	0.693	---	2.255	0.0	15.14	---	85
PRF 90	0.694	---	2.254	0.0	15.14	---	90
PRF 97	0.695	---	2.251	0.0	15.13	---	97
Gasoline 90.9*	0.744	---	1.864	0.0	14.56	42.95	90.9
Gasoline 91.3	0.731	---	1.822	0.002	14.52	43.44	91.3
Gasoline 90.9 + 5% (V/V) Ethanol	0.746	---	1.901	0.016	14.3	---	93.7
Gasoline 97	0.750	---	1.708	0.008	14.19	42.78	97
iso-Octane*	0.692	35.14	2.250	0.0	15.13	44.31	100
n-heptane*	0.680	36.57	2.286	0.0	15.18	44.57	0
Ethanol*	0.789	42.32	3.000	0.5	9.00	26.8	108.6

a = Data provided by the supplier for gasoline fuels, measured for the PRF blends at T= 20 C, and taken from U.S. National Library of Medicine for iso-Octane, n-Heptane and Ethanol (2022).

b = Data taken from U.S. National Library of Medicine, at 25 °C (2022).

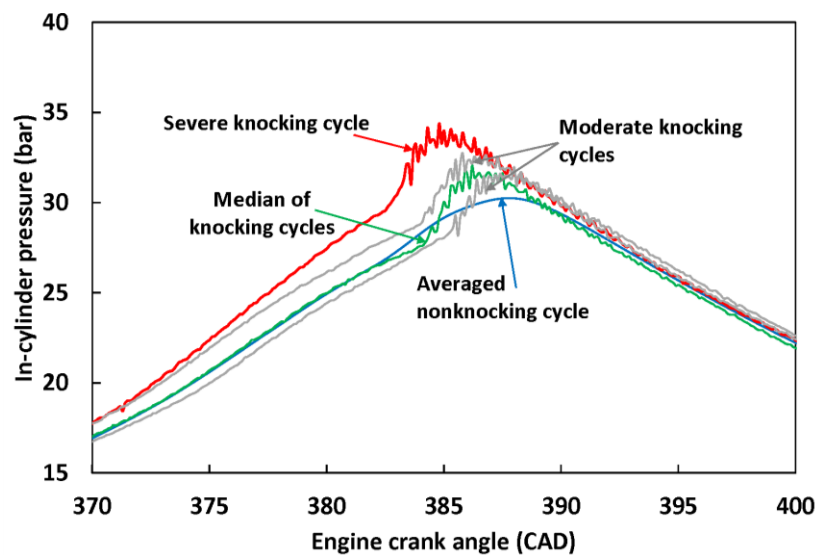
c = Data provided by the supplier for gasoline fuels, calculated for the PRF blends, iso-Octane, n-Heptane and Ethanol.

d = Net calorific values as provided by the supplier for gasoline fuels and taken from NIST Chemistry WebBook for the other fuels (2022).

e = Data provided by the supplier for gasoline fuels, calculated for the PRF blends and taken from (Vallinayagam et al., 2017) for Ethanol.

\* Those molecules were not tested on their own, however, the data provided for reference as they were utilised as blending components with gasoline fuels.

### 244 3.4 Selection of a representative knocking combustion cycle



245

246 Figure 2: In-cylinder pressure for a typical knocking test of Gasoline 91.3 tested at  
247 stoichiometric  $\lambda$ , and engine CR and speed of 7.38 and 600 rpm, respectively.

248 Figure 2 shows an example of a knocking test conducted during the investigation of the  
249 reference gasoline 91.3 test fuel. The engine was operated at the steady-state conditions  
250 of 600 rpm, WOT and stoichiometric air-fuel ratio, with the engine compression ratio  
251 gradually increased from 6.91 to find the knock limit of the fuel at the required KFRQ.

252 As can be seen in Figure 2, the engine started to knock as the engine CR increased to  
 253 7.38, producing a KFRQ of 8% (four random knocking cycles out of 50 consecutive  
 254 combustion cycles).

255 It can be seen, in Figure 2, that all non-knocking cycles were isolated, grouped,  
 256 averaged and plotted, shown in blue, while the four random knocking cycles were  
 257 plotted individually and evaluated for knock characteristics. The knocking combustion  
 258 cycles show differences in knock induced pressure oscillations, peaks and time of  
 259 occurrence relative to one another (Figure 2). Therefore, to find a representative  
 260 knocking combustion cycle for the test fuel that could be used to describe the knocking  
 261 combustion parameters of a given fuel, the median cycle of the knocking cycles was  
 262 selected. In the case the total number of knocking combustion cycles was even, the  
 263 selected representative cycle is that which has the amplitude immediately after the  
 264 median, see Table 7. The amplitude of bandpass filtered in-cylinder pressure at the first  
 265 knock-point and time of occurrence in crank angle degree were found as representative  
 266 knocking characteristics of the selected knocking combustion cycle for a test fuel.

267 Table 7: A summary of knocking combustion cycles analysis of Gasoline 91.3 tested at  
 268 stoichiometric  $\lambda$ , and engine CR and speed of 7.38 and 600 rpm, respectively.

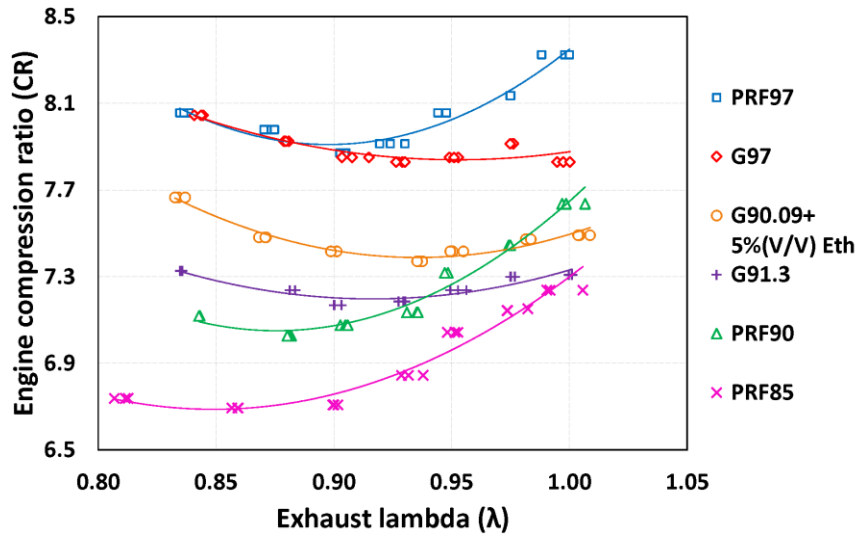
Knocking combustion cycles number	#7	<b>#19</b>	#41	#48	Median	Values of the Selected cycle
Amplitude of in-cylinder pressure at first knock-point (bar)	0.545	<b>0.556</b>	0.524	0.56	0.551	0.556



Time of first knock-point (CAD)	385.5	<b>386.2</b>	385.5	383.7	385.5	386.2
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269 **4 Results and Discussion**

270 **4.1 Knock resistance**



271

272 Figure 3: Operating limit of engine CR of the PRFs and gasoline fuels at varying  $\lambda$  between

273

1.00 and 0.8.

274 Figure 3 shows the operating limit of the engine compression ratio (CR) for the fuels

275 investigated at different air-fuel ratios (exhaust lambda  $\lambda$ ). The operating limit of the

276 engine CRs at  $\lambda = 1$  and the other richer  $\lambda$  values were found at KFRQ of 10% and 30%,

277 respectively (Section 3.2). In Figure 3, in general, it can be seen that the engine CR

278 operating limit increases as the RON of fuels increases. Hence, fuels with high RON

279 such as Gasoline RON97 (G97) and PRF97, which are of higher knock resistance, were

280 able to be tested at a higher engine CR prior to observation of the same frequency of

281 knock as compared to other fuels investigated. This correlation between the ability of  
282 fuels to resist knock (the octane number) and the operating limit of engine CR was  
283 expected and agrees with the concept of the standard test method for research octane  
284 number of spark-ignition engine fuels (ASTM Int., 2019).

285 It can also be seen, in Figure 3, that the operating limit of engine CR was affected by  
286 changing the exhaust lambda  $\lambda$ . In general, for all fuels, engine CR decreases as the  
287 mixture becomes richer to a certain limit and subsequently increases as the mixture  
288 becomes richer still in order to maintain the same level of knocking frequency. Similar  
289 observations have been made in a CFR engine by other researchers (Brock and Stanley,  
290 2012; Wang *et al.*, 2017; Montoya *et al.*, 2018; Hoth *et al.*, 2019). It has been suggested  
291 that the influence of varying exhaust lambda on the engine CR can be explained as  
292 follows: initially, increasing fuel supply to the engine results in a richer mixture, which  
293 increases the burning rate and promotes knock (Chen and Raine, 2015). Furthermore, as  
294 a mixture becomes more richer,  $\lambda < 0.97$ , at a constant throttle position, it means there  
295 is more fuel available to be burned and release energy, leading to an increase in  
296 combustion and in-cylinder wall temperatures due to increased heat transfer. This effect  
297 on the in-cylinder charge temperature persists from cycle to cycle; as the in-cylinder  
298 wall temperature increases, more heat is transferred to the in-cylinder charge during  
299 intake and compression strokes. Therefore, reaction rates during the end-gas  
300 autoignition process are increased through elevated temperatures, decreasing the  
301 ignition delay time (Gauthier *et al.*, 2004) and hence, autoignition in the end-gas occurs  
302 more easily (Bolt and Henein, 1969; Dec and Sjöberg, 2004). As a result, the operating

303 limit of engine CR must be lowered so as to offset the additional in-cylinder  
304 temperature gained by mixture enrichment, in order to reduce the severity of the  
305 resultant knock. However, Figure 3 also shows that beyond a certain level of mixture  
306 richness, varying between fuels in the range  $\lambda = 0.85$  to  $0.95$ , a further increase in fuel  
307 mass increases engine CR. This can be attributed to insufficient availability of oxygen  
308 to fully oxidise the fuel, with combustion therefore becoming oxygen limited, leading to  
309 a decrease in the temperature after compression as well as the burning rate,  
310 consequently increasing the ignition delay of the end gases (Machrafi *et al.*, 2007;  
311 Zheng *et al.*, 2019). Accordingly, an increase in engine CR operating limit is required  
312 so that the same knocking frequency is maintained.

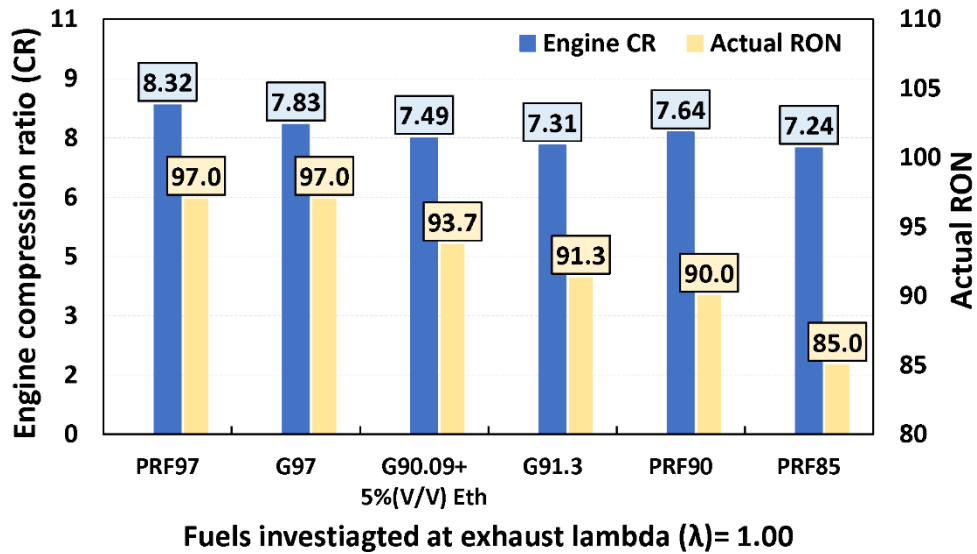
313 From Figure 3, it can be seen that the effect of exhaust lambda  $\lambda$  on the operating limit  
314 of engine CR for PRFs is different as compared to gasoline fuels. Overall, it appears  
315 that the PRFs are more sensitive to changes in the mixture strength, especially in the  
316 range between  $\lambda=1$  and  $\lambda = 0.90$ . This observation is even more pronounced when  
317 comparing two fuels of equivalent octane rating, for example, PRF97 and Gasoline 97.  
318 At  $\lambda = 1$ , the PRF97 displayed a significantly lower knock propensity as it was able to  
319 resist knock until a higher compression ratio than gasoline 97 by 0.5 unit. However, this  
320 divergence in engine CR operating limit between the two fuels decreases as the mixture  
321 becomes richer, until both fuels show a similar engine CR operating limit at around  $\lambda =$   
322  $0.9$ . Similar observations can be made for the other PRFs (90 and 85). At  $\lambda=1$ , the  
323 PRF90 showed a higher knock limit than the fuels Gasoline 91.3 and G90.09+5%(V/V)  
324 Ethanol, which have higher RON by 1.3 and 3.7 units, respectively. Similarly, the

325 PRF85, at  $\lambda=1$ , was able to exhibit a knock resistance similar to Gasoline 91.3, which  
326 has 6.3 units of RON higher than it. Therefore, the results suggest that at stoichiometric  
327 and slightly rich conditions, PRFs possess higher knock resistance than expected  
328 relative to gasoline fuels with comparable or higher RON.

329 The dissimilar effect of varying exhaust lambda  $\lambda$  on the knock tendency of the gasoline  
330 fuels and the PRFs is a point of interest and can be discussed by referring to the effect  
331 of the fuel composition. The PRFs investigated are highly iso-paraffinic fuels as they  
332 consist of high concentrations of iso-octane blended with n-heptane. However, gasoline  
333 fuels contain about 30 %(m/m) aromatic compounds along with varying proportions of  
334 paraffins, iso-paraffins, olefins, naphthenes molecules, giving gasoline fuels an octane  
335 sensitivity of about 10 in comparison to 0 for PRFs (Sluder *et al.*, 2016). A similar  
336 experimental observation of the dissimilar knock tendencies of iso-paraffinic and  
337 aromatic fuels was reported by Hoth *et al.*, (2019).

338 According to Iqbal *et al.*, (2011), the presence of olefins and aromatics in fuels such as  
339 gasoline changes significantly the chemical kinetics of the fuel oxidation process  
340 compared to paraffinic fuels; in contrast to PRFs olefins and aromatics fuels do not  
341 exhibit NTC. It was observed experimentally by Iqbal *et al.*, (2011) that PRFs  
342 experience an increase in cool flame activity during first stage ignition as the mixture  
343 becomes richer due to an increase in energy release, leading to initiation of a second  
344 stage ignition. Therefore, the overall ignition delay was decreased much more than  
345 olefins and aromatics fuels (Iqbal *et al.*, 2011). A similar observation was reported by  
346 Dec and Sjöberg (2004) suggesting that fuels with cool-flame chemistry such as PRFs

347 require immediate compensation in engine operating conditions, such as lowering the  
 348 engine wall temperature, as the mixture becomes richer; otherwise, the control of  
 349 combustion will be difficult. These previous observations explain the necessarily sharp  
 350 decrease in the operating limit of engine CR for PRFs apparent in Figure 3 as lambda  
 351 becomes increasingly rich.

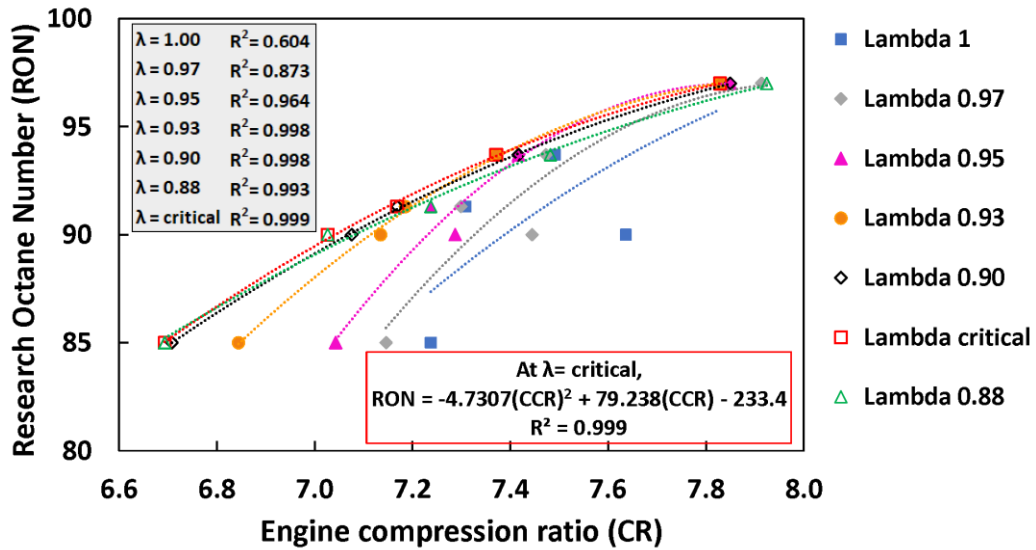


352

353 Figure 4: Operating limit of engine CR and RON values of the PRFs and gasoline fuels  
 354 investigated at  $\lambda=1$ .

355 Figure 4 shows a comparison of the operating limit of the engine CR obtained for the  
 356 known-RON PRF and gasoline fuels at stoichiometric  $\lambda$ . It is known, by definition, that  
 357 there is a slightly non-linear increase of the operating limit of engine CR with  
 358 increasing RON of a given fuel (ASTM Int., 2019). However, it can be seen in Figure 4  
 359 that, as was discussed previously in the context of Figure 3 due to the chemical  
 360 composition of the fuels investigated, the iso-paraffinic fuels, such as PRF 97 and  
 361 PRF90, can be operated at higher engine CRs relative to fuels with similar RON but of

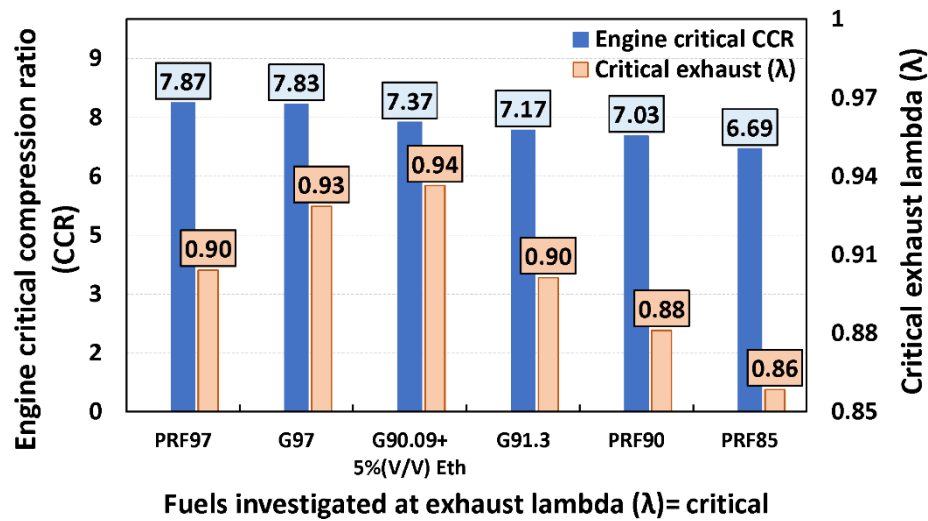
362 composition including aromatic compounds, such as Gasoline 97, G90.9+5Eth and  
 363 Gasoline 91.3. It can be seen that PRF97 operated at a higher engine CR than Gasoline  
 364 97 by 0.49 unit. Also, while the engine CR operating limit decreased with decrease in  
 365 RON from Gasoline 97, to G90.9+5Eth and Gasoline 91.3, PRF90 showed a higher  
 366 engine CR than the aforementioned fuels despite a lower RON value.



367  
 368 Figure 5: Relationship between the operating limit of engine CR and RON value of the PRFs  
 369 and gasoline fuels at different exhaust  $\lambda$ .

370 Figure 5 shows a comparison of the relationship between the operating limit of the  
 371 engine CR and RON values for the PRF and gasoline fuels at different exhaust lambda  
 372  $\lambda$  between 1.00 and 0.88. It can be seen that the trend obtained at stoichiometric  $\lambda$   
 373 shows no correlation and does not reflect the theoretical relationship between engine  
 374 CR and RON of the fuels. Although the trend becomes more representative as the fuels  
 375 were tested at richer mixtures, exhaust  $\lambda \approx 0.9$ , each fuel has a particular value of  
 376 exhaust  $\lambda$ , as was observed in Figure 3, which causes its minimum operating limit of

377 engine CR (CCR). These values of exhaust  $\lambda$  for all fuels investigated were selected and  
 378 subsequently referred to as the critical  $\lambda$  and engine CR (CRR) values. This  
 379 combination of critical  $\lambda$  and critical engine compression ratio CCR were found to  
 380 provide the best correlation,  $R^2=0.999$ , between the RON of a given fuel and its knock-  
 381 limit engine compression ratio CRR despite the difference in the chemical composition.



382

383 Figure 6: A comparison of the critical operating limit of engine CR and value of critical  $\lambda$  of the  
 384 PRFs and gasoline fuels investigated.

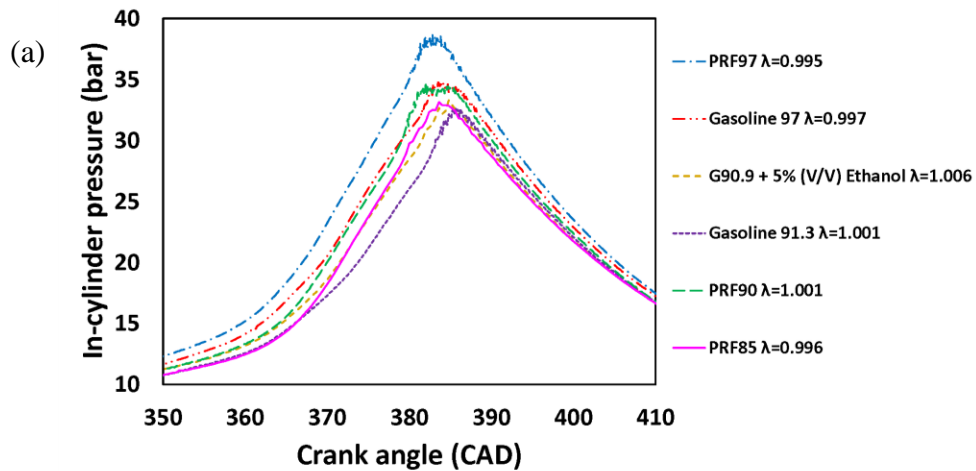
385 Figure 6 shows the critical operating limits of engine CR (CCR) of the PRFs and  
 386 gasoline fuels while operating the critical  $\lambda$  values. In contrast to Figure 4, which made  
 387 a similar comparison but at  $\lambda = 1$ , it can be seen that the order of the fuels in terms of  
 388 RON agrees with the determined operating limit of engine CR (CCR), in spite of the  
 389 varying fuel composition. For example, the difference between the engine CR for  
 390 PRF97 and Gasoline 97 is reduced significantly from 0.49 to 0.04 units of CR. In

391 addition, PRF90 displays as expected a lower engine CCR compared to  
392 G90.09+5%(V/V) Ethanol and Gasoline 91.3.

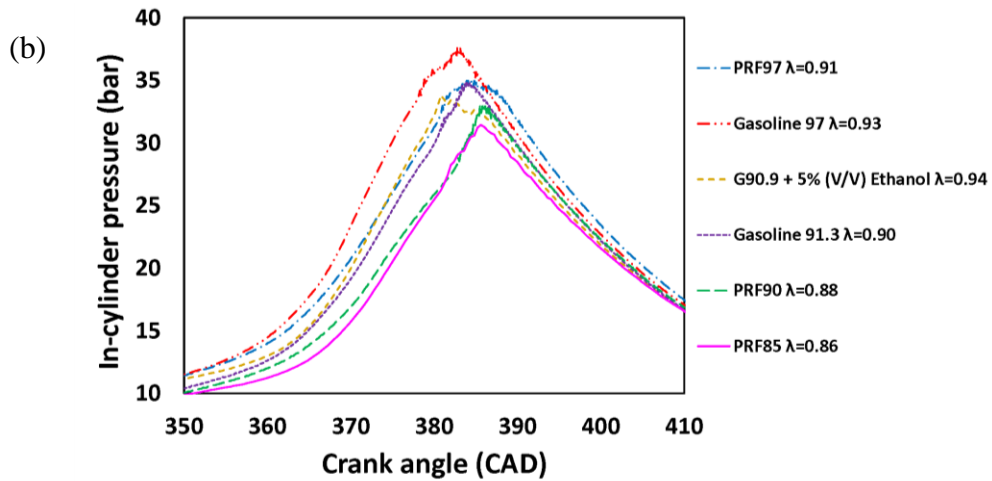
393 With regards to the effect of exhaust lambda  $\lambda$ , it can be seen that most of the fuels  
394 investigated have a critical  $\lambda$  at very rich mixtures in the range between 0.90 and 0.86,  
395 excluding gasoline 97 and the gasoline with ethanol blend (G90.09+5%(V/V) Ethanol)  
396 which they have a critical  $\lambda$  at less rich mixtures of 0.93 and 0.94, respectively. These  
397 findings agree with the values available in the literature, suggesting that blends  
398 including ethanol have a critical  $\lambda$  around 0.93, while iso-paraffinic fuels, PRFs, tend to  
399 have a higher knock intensity in the range from 0.90 to 0.88 (Kolodziej, 2017; Foong *et*  
400 *al.*, 2017). It is interesting to see that when considering the PRFs, there is a trend of  
401 decreasing critical lambda with decreasing RON. This can be attributed to the increase  
402 in the level of n-heptane, which is known to be significantly more reactive at low  
403 temperature conditions. The occurrence of knock is less sensitive to oxygen availability  
404 as the oxygen that is available is more likely to be consumed by heptane present.  
405 Therefore, the increasing temperature with increasing richness becomes the dominant  
406 factor.

#### 407 **4.2 Knocking combustion characteristics**





408



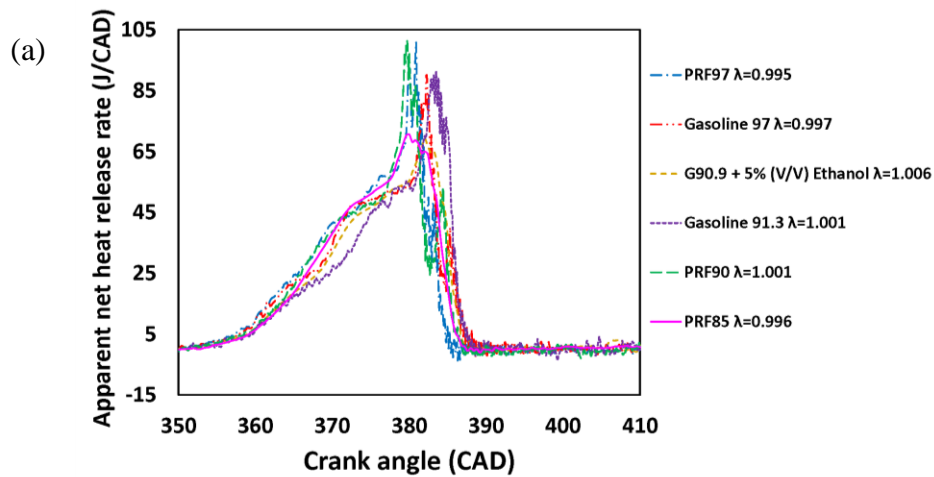
409

410 Figure 7: In-cylinder pressure of the representative knocking combustion cycles of the PRFs  
 411 and gasoline fuels tested at operating limit of engine CR at (a)  $\lambda=1$  and (b)  $\lambda=\text{critical}$ .

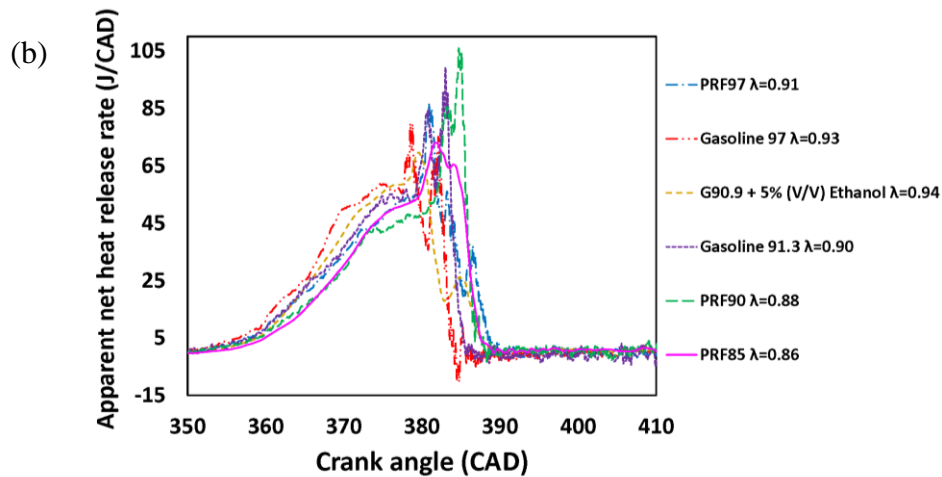
412 Figures 7 (a) and (b), show the in-cylinder pressure for the representative knocking  
 413 combustion cycles of the PRF and gasoline fuels tested at the operating limit of engine  
 414 CRs at  $\lambda=1$  and  $\lambda=\text{critical}$ , respectively. It can be seen, in Figure 7 (a), that at  
 415 stoichiometric  $\lambda=1$  the iso-paraffinic fuels (PRFs) displayed higher levels of in-cylinder  
 416 pressure relative to the aromatic fuels (gasoline fuels) although all fuels operated at  
 417 engine CRs that produced the same knocking frequency. This means that, for example,

418 PRF97 reached the knocking frequency threshold at a peak in-cylinder pressure higher  
419 than Gasoline 97 by about 10%. The results also show that PRF90 and PRF85 exhibited  
420 a peak in-cylinder pressure equivalent to aromatic fuels of approximately seven units  
421 higher octane number. These observations were expected as the PRFs, at  $\lambda=1$ , operated  
422 at a much higher engine CR in comparison to gasoline fuels, as was discussed in the  
423 context of Figure 4.

424 Comparing the fuels investigated at their critical conditions of exhaust lambda  $\lambda$  and  
425 engine CR shows a contrary observation to that at  $\lambda=1$ . It can be seen in Figure 7 (b)  
426 that the iso-paraffinic fuels (PRFs), for example PRF97 and PRF90, tend to reach the  
427 pre-determined knocking threshold at an in-cylinder pressure much lower than the  
428 aromatic fuels Gasoline 97 and Gasoline 91.3, respectively. It is worth noting that, at  
429 the critical test conditions, the addition of ethanol to gasoline fuel causes the knock to  
430 occur at lower in-cylinder pressure compared to Gasoline 91.3, even though it was  
431 operated at a higher engine CR. This could be attributed to the faster burning velocity  
432 gained by the addition of ethanol, advancing the peak of in-cylinder pressure closer to  
433 the engine TDC combustion occurring at a smaller volume (Turner *et al.*, 2011; Jiang *et*  
434 *al.*, 2017).



435

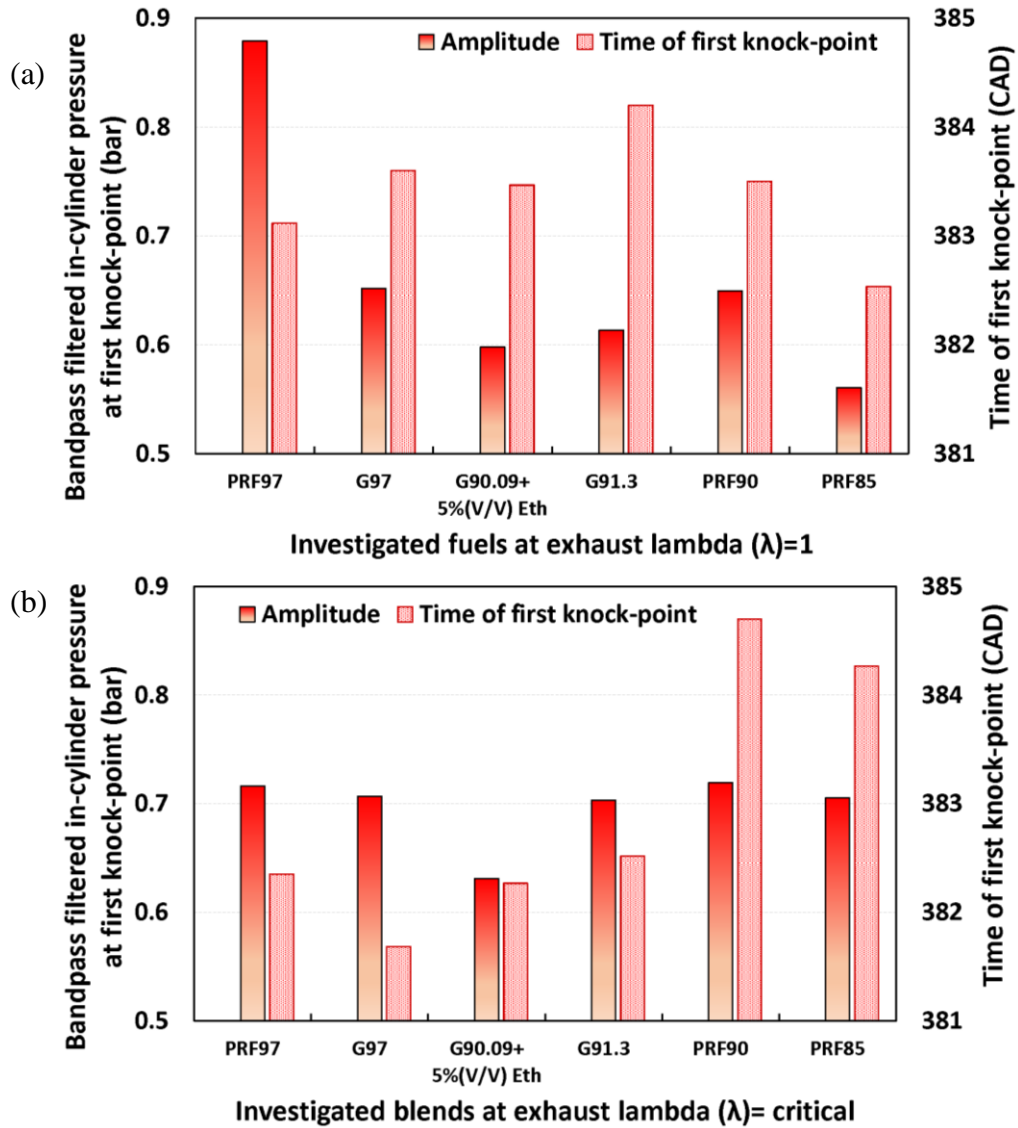


436

437 Figure 8: Apparent net heat release rate of the representative knocking combustion cycles of the  
 438 PRFs and gasoline fuels tested at operating limit of engine CR at (a)  $\lambda=1$  and (b)  $\lambda$ =critical.

439 The apparent net heat release rate of the representative knocking combustion cycles of  
 440 the fuels investigated at the operating limit of engine CRs at  $\lambda=1$  and  $\lambda$ =critical are  
 441 shown in Figures 8 (a) and (b), respectively. At  $\lambda=1$ , in general, all fuels experienced  
 442 multiple peaks of heat release due to knocking combustion between 377 and 386 CAD,  
 443 with the iso-paraffinic fuels (PRFs) advanced relative to the aromatic fuels. This is also  
 444 apparent in the in-cylinder pressure traces, Figures 7 (a) and (b), and can be attributed to

445 testing of the PRFs at a higher engine CR (Figure 3). The results also show that the  
446 PRFs produced higher cumulative energy release rate by approximately 10% than the  
447 gasoline fuels, due to the higher calorific value of the former. However, at the critical  $\lambda$   
448 operation, it can be seen, in Figure 8 (b), that the aromatic fuels tend to knock earlier  
449 than the iso-paraffinic fuels, as the initial peaks of heat release (present due to knock in  
450 the end-gas zone) occur approximately 1.5 CAD, earlier.



451

452

453 Figure 9: Amplitude of bandpass filtered in-cylinder pressure and angle at first knock-point of  
 454 the representative knocking combustion cycles of the PRFs and gasoline fuels tested at  
 455 operating limit of engine CR at (a)  $\lambda=1$  and (b)  $\lambda=\text{critical}$ .

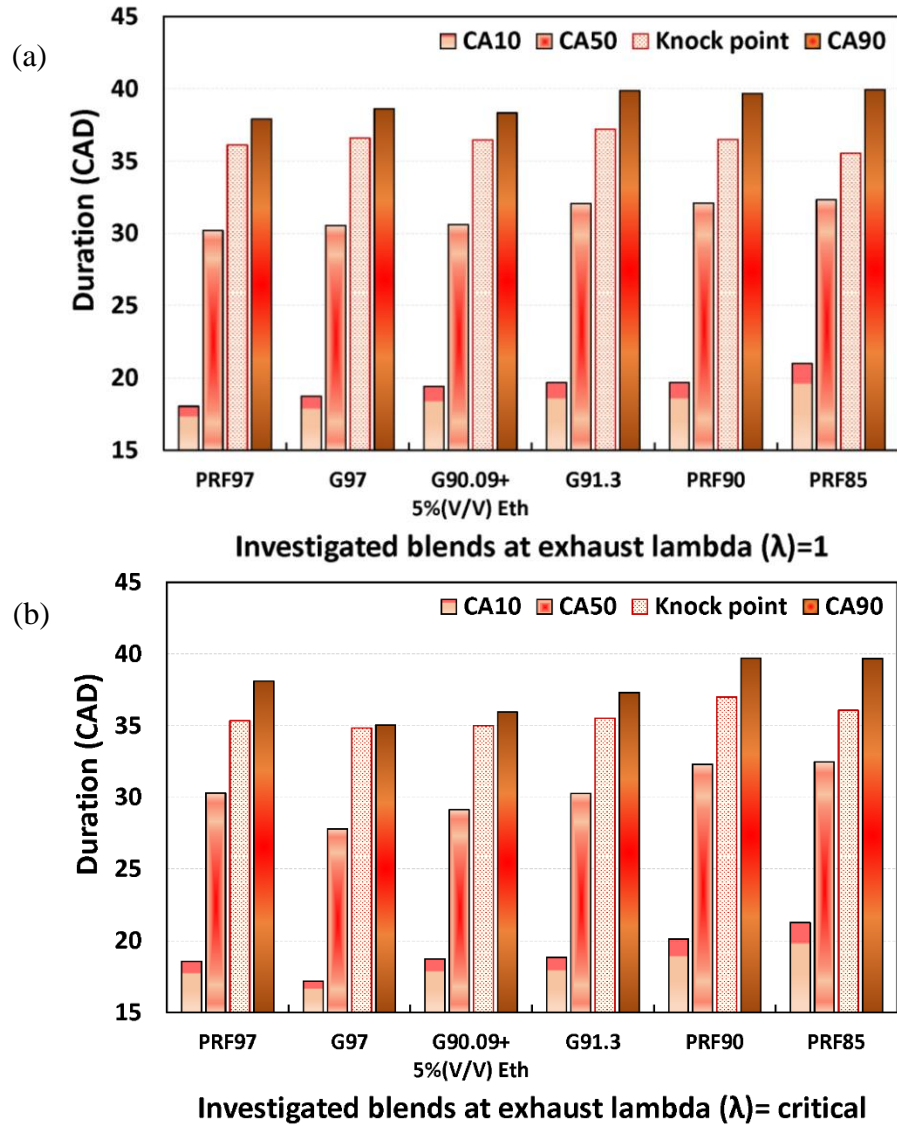
456 Detection of knock during the experiments was undertaken by application of a bandpass  
 457 filter to the in-cylinder pressure data (as described in Section 3.1). Figures 9 (a) and (b)  
 458 show the increase in amplitude of the bandpass filtered in-cylinder pressure and the  
 459 time of occurrence for the first knock-point of the representative knocking combustion

460 cycles of the PRFs and gasoline fuels tested at the operating limit of engine CRs at  $\lambda=1$   
461 and  $\lambda=\text{critical}$ , respectively. In general, increases in the amplitude of the in-cylinder  
462 pressure of the representative knocking cycles are shown to be lower at  $\lambda=1$  than at  
463  $\lambda=\text{critical}$ , except PRF97 which required the highest engine CR in order to reach the  
464 pre-determined knock threshold (Figure 3). This characteristic is also apparent in Figure  
465 (7 a), where fluctuations are apparent at the peak in-cylinder pressure of PRF97.  
466 However, at  $\lambda=1$ , it can be seen that the increase in amplitude for the iso-paraffinic fuels  
467 decreases significantly by 26% and 36% as the n-heptane content increases from 3% to  
468 15%, respectively. On the other hand, the Figure 9 shows a slightly higher increase in  
469 amplitude in the case of Gasoline 97 relative to Gasoline 91.3, while the addition of  
470 ethanol (G90.09 + 5% (V/V) Eth) reduces it compared to Gasoline 97 and Gasoline  
471 91.3.

472 Additionally, Figures 9 (a) and (b) show that the time of the first knock-point is delayed  
473 for most of the fuels at  $\lambda=1$  compared to  $\lambda=\text{critical}$ , except PRF90 and PRF85. These  
474 observations can likely be attributed to the increase in the burning velocity of the fuels  
475 at the richer  $\lambda=\text{critical}$  conditions relative to  $\lambda=1$ , which increases the rate of  
476 combustion and reduces its duration (Figure 8). As a result, at the critical lambda  
477 operating condition, which is rich for all fuels (Figure 6), the amplitude of in-cylinder  
478 pressure shows little variation despite the differences fuel composition, (with the  
479 exception of the ethanol/gasoline blend). The addition of ethanol to fuels was found to  
480 reduce peak-to-peak in-cylinder pressure fluctuations which may be the reason why the  
481 amplitude of in-cylinder pressure is lower than the other fuels. These observations are

482 in agreement with the study of (Kolodziej, 2017). With regards to the delayed time of  
483 the amplitude of in-cylinder pressure for PRF90 and PRF85, these fuels were seen in  
484 (Figure 7 b) to reach a peak in-cylinder pressure later compared to the other fuels  
485 investigated as they were operated at lower engine CR corresponding to their RON  
486 values.

487 Figures 10 (a) and (b) show the burn duration of the combustion characteristics CA10,  
488 CA50, CA90, in addition to the interval to the first knock-point relative to the spark  
489 timing for the knock representative cycle of the fuels investigated at the operating limit  
490 of engine CRs at  $\lambda=1$  and  $\lambda=\text{critical}$ , respectively. It can be seen that at  $\lambda=1$ , the burn  
491 duration CA10 increases gradually by less than a CAD as the RON of the aromatic fuel  
492 decreases. At the same time, a more significant increase, (by about 1.5 CAD), is  
493 apparent for the iso-paraffinic fuels as the RON decreases from 97 to 90 and then to 85,  
494 with PRF85 exhibiting the longest CA10 burn duration of 21 CAD.



495

496

497 Figure 10: Durations of CA10, CA50 and CA90 of the representative knocking combustion  
 498 cycles of the PRFs and gasoline fuels tested at operating limit of engine CR  
 499 at (a)  $\lambda=1$  and (b)  $\lambda$ =critical.

500 As for the CA50 and CA90 durations, fuels with high RONs such as PRF97, Gasoline  
 501 97 and G90.09+5%(V/V) Ethanol displayed comparably shorter durations of about 2  
 502 CAD less in comparison to fuels with lower RONs such as Gasoline 91.3, PRF90 and  
 503 PRF85. At constant RON, it can be seen that PRF 97 exhibited slightly lower burn

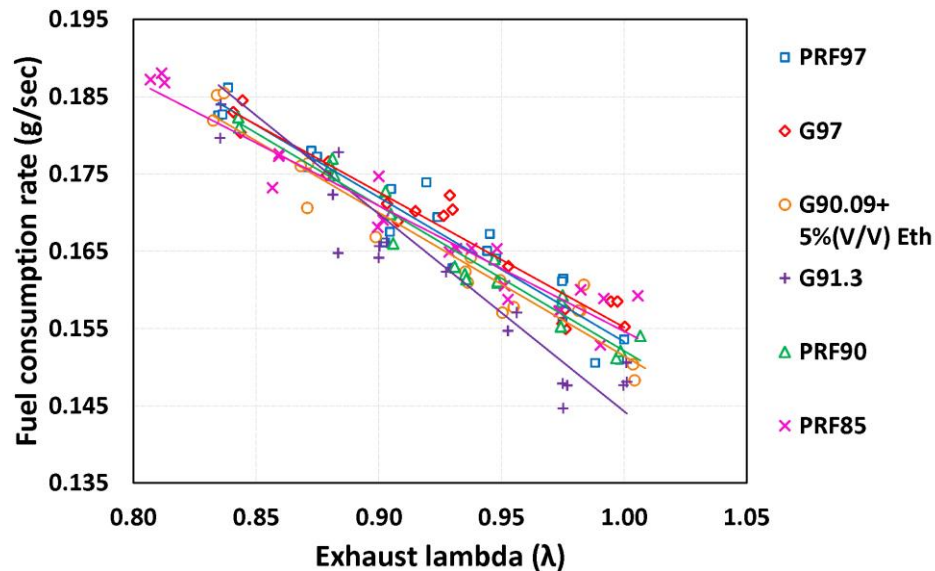


504 durations of CA50 and CA90 as it was operated at a higher compression ratio than  
505 Gasoline 97. It can be observed that, at a constant  $\lambda=1$ , as the RON of the fuels  
506 decreases, iso-paraffinic fuels, especially PRF97 and PRF90, experience a more  
507 considerable delay in flame initiation, combustion rate and overall combustion  
508 durations than aromatic fuels. However, at a comparable RON value, iso-paraffinic  
509 fuels show a slightly shorter duration as they require operation at a higher engine CR  
510 than aromatic fuels to achieve the same knock frequency and intensity (Figure 3). The  
511 addition of ethanol to gasoline shows a slight decrease in the CA10 duration, while a  
512 significant reduction (by about 1.5 CAD) can be seen in the combustion phasing CA50  
513 and CA90 compared to non-oxygenated gasoline (Gasoline 91.3). With regards the  
514 interval to the first knock-point, it can be observed that the iso-paraffinic fuels  
515 displayed shorter durations relative to the aromatic fuels, in agreement with Figure 8 in  
516 which the PRFs were observed to exhibit peaks of ANHRR earlier than the aromatic  
517 fuels.

518 Figure (10 b) shows the same combustion parameters where the fuels were tested at  
519 their critical conditions of engine CR and  $\lambda$  ratio. Overall, as was seen at the constant  
520  $\lambda=1$ , an increase in the burn duration CA10 with decreasing fuel RON is apparent.  
521 However, this increase is also visible in the other burn durations, CA50 and CA90. It is  
522 interesting to note that the duration of CA10 of the iso-paraffinic fuels become slightly  
523 longer than that found at  $\lambda=1$ , while no significant changes in the durations of CA50  
524 and CA90 can be seen. However, the aromatic fuels experience, at the critical  
525 conditions relative to  $\lambda =1$ , considerable reduction by about 3 CAD in all durations.

526 Therefore, it can be summarised that iso-paraffinic fuels have more prolonged  
 527 combustion initiation and development durations than the aromatic fuels at critical  
 528 knock operating conditions of engine CR and air-fuel ratios. This explains the late  
 529 occurrence of ANHRR peaks for the iso-paraffinic fuels, shown in Figure (8 b),  
 530 compared to the aromatic fuels, which tend to have faster heat release and earlier  
 531 knocking peaks. The results of knock-point durations show that iso-paraffinic fuels tend  
 532 to knock at an angle closer to CA50. In contrast, the aromatic fuels resist knocking for a  
 533 more extended period, with a knock-point much closer to the end of the combustion  
 534 phase.

### 535 4.3 Fuel flow characteristics



536  
 537 Figure 11: Fuel consumption rate of the investigated PRFs and gasoline fuels tested at operating  
 538 limit of engine CR and varying exhaust  $\lambda$ .

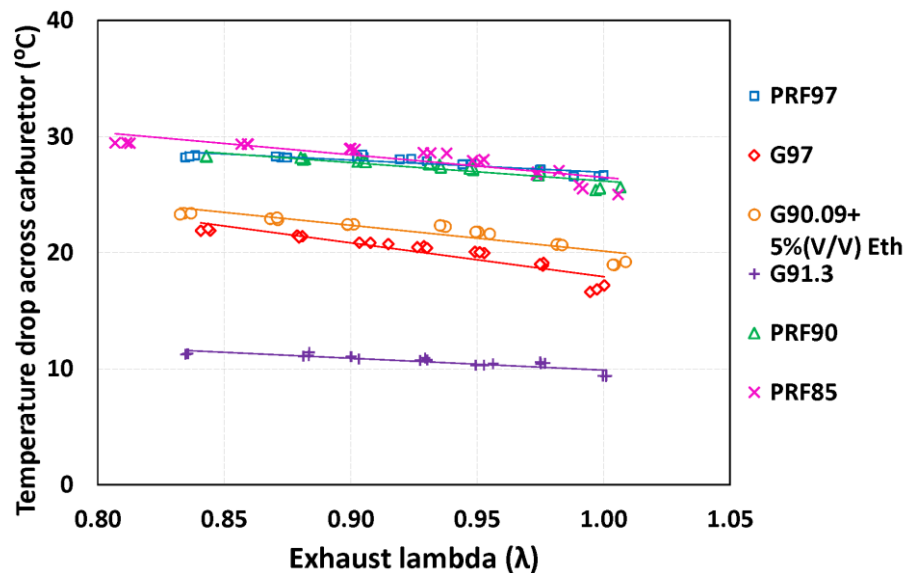
539 Figure 11 shows the fuel consumption rates, over the operating range of  $\lambda$ , for the PRFs  
 540 and gasoline fuels. Overall, all fuels exhibit comparable consumption rates with a linear

541 increase of about 25% when increasing the mixture stoichiometry from ( $\lambda=1$   
 542 to  $\lambda= 0.82$ ). However, fuels with higher heating values and lower stoichiometric AFR,  
 543 such as Gasoline 91.3 and G90.09+5%(V/V) Ethanol, tend to display slightly lower fuel  
 544 consumption rates relative to the other fuels. Of the PRFs, PRF85, which has the  
 545 highest content of *n*-heptane, exhibits a relatively higher fuel flow rate compared to  
 546 PRF97 and PRF90. Table 8 shows the indicated specific fuel consumption rates of the  
 547 fuels investigated at exhaust  $\lambda=1$ .

548 Table 8: Indicated specific fuel consumption rates of the fuels investigated at exhaust  $\lambda=1$ .

Fuels	PRF97	G97	G90.09 + 5%(V/V) Eth	G91.3	PRF90	PRF85
ISFC (g/kWh)	254.8	271.0	277.3	273.8	267.8	289.0

549



550

551 Figure 12: Temperature drop across the carburettor during supply of the investigated PRFs and  
 552 gasoline fuels tested at operating limit of engine CR and varying exhaust  $\lambda$ .

553 Figure 12 shows the temperature drop across the carburettor during the supply of the  
554 fuels investigated over the entire operating range of exhaust  $\lambda$ . Overall, it can be seen  
555 that the reduction in temperature increases with the increase of fuel supply to the  
556 engine, as is expected given that there is more fuel to be mixed with the hot air causing  
557 a further reduction in the charge temperature. The heat of vaporisation (HoV) of each  
558 fuel plays a significant role here, therefore, it is expected there is a different reduction in  
559 temperature among the fuels investigated. For PRFs, it can be seen in Figure 11, an  
560 increase in fuel flow rate with increasing *n*-heptane content. It is known that relative to  
561 iso-octane, *n*-heptane has a slightly higher (HoV) by about 1.4 kJ/mol. As a result, a  
562 slight increase in the temperature drop can be seen as the *n*-heptane content in the PRFs  
563 increases. With regard the gasoline fuels, it is interesting to see that the addition of 5%  
564 (V/V) of ethanol, which has a higher HoV than iso-octane, increased the temperature  
565 drop by about 50% compared to Gasoline 91.3. It can also be seen that Gasoline 97  
566 displayed a significant reduction in the temperature drop relative to Gasoline 91.3, but  
567 of slightly lower magnitude than that exhibited by G90.09+5%(V/V) Ethanol. The  
568 cooling effect of fuels is an essential factor for suppressing engine knock. As for  
569 ethanol, it is quantitatively described as an equivalent octane number (Wang, Zeraati-  
570 Rezaei, *et al.*, 2017).

## 571 **5 Conclusions**

572 In this study, the effect of varying air-fuel ratio and fuel composition on knock  
573 resistance during spark-ignition combustion were experimentally investigated. In  
574 addition, knocking combustion characteristics and cooling effect of the test fuels were

575 evaluated and compared. It was found that the determination of the operating limit of  
576 engine CR, at the targeted knocking frequency, depended significantly on the air-fuel  
577 ratio and the chemical composition of a test fuel. By investigating these effects, the  
578 following conclusions can be drawn:

- 579 • A richer mixture exhibited knocking combustion at a lower engine CR than a  
580 leaner mixture, due to the decrease in ignition delay caused by the increase in  
581 end- gas reaction rates at the elevated in-cylinder temperature present when  
582 enriching the mixture. However, introducing too much fuel increased the engine  
583 CR required to maintain the same level of knocking frequency, attributable to an  
584 increase in the ignition delay with decreasing in-cylinder temperature and  
585 burning rate as a result of insufficient oxygen availability to fully oxidise the  
586 fuel.
- 587 • Highly aromatic fuels were less sensitive to changes in exhaust lambda  $\lambda$  than  
588 highly iso-paraffinic fuels and, exhibited less variation in the operating limit of  
589 engine CR at varying exhaust lambda  $\lambda$ . However, at the same operating  
590 condition of exhaust lambda  $\lambda$ , especially at stoichiometric and slightly rich  $\lambda$ ,  
591 iso-paraffinic fuels displayed greater knock resistance than highly aromatic fuels  
592 of the same RON values, requiring operation at a higher engine CR in order to  
593 exhibit the same level of knocking frequency.
- 594 • The effect of fuel composition on the determination of the operating limit of  
595 engine CR, at the targeted knocking frequency, indicated the importance of  
596 comparing test fuels with different chemical composition at their critical exhaust

597 lambda  $\lambda$  rather than at fixed exhaust lambda  $\lambda$ . Thus, a great degree of  
598 correlation remained between the operating limit of the engine CR and known  
599 RON values despite differences in fuel composition.

600 Regarding the effect of chemical composition of the test fuels on knocking combustion  
601 characteristics, it was found that:

- 602 • For all fuels, the in-cylinder pressure fluctuations at the critical  $\lambda$  were stronger  
603 than at other  $\lambda$  values, attributed to the effect of multiple auto-ignition  
604 occurrences that increased the in-cylinder pressure fluctuations. At these  
605 conditions, the iso- paraffinic fuels produce higher peak-to-peak in-cylinder  
606 pressure fluctuations and peak of heat release rates with longer combustion  
607 initiation and development durations, and a propensity to knock at an angle  
608 much closer to the mid-point of combustion (CA50) and earlier compared to the  
609 aromatic fuels.
- 610 • The addition of ethanol to gasoline increased the rate of combustion and  
611 decreased peak-to-peak in-cylinder variation compared to PRFs and other  
612 gasoline fuels, attributable to the presence of an alcohol functional group.

613 The effect of chemical composition of the test fuels on the fuel consumption rate and  
614 cooling effect, as a temperature drop across the carburettor, while varying the exhaust  
615 lambda  $\lambda$ , can be summarised as follows:

- 616 • In general, all fuels have a comparable rate of fuel consumption. However, a  
617 more significant reduction in temperatures across the carburettor was observed

618 with the PRFs and (G90.09+5%(V/V) Ethanol) fuel, due to their high heat of  
619 vaporisation (HoV). It is interesting to note that the addition of 5% (V/V)  
620 ethanol to gasoline 90.9 reduced the charge temperature by 50% compared to  
621 Gasoline 91.3.

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