BaBi₂O₆: a Promising *n*-Type Thermoelectric Oxide with the PbSb₂O₆ Crystal Structure — Supplementary Information

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Convergence

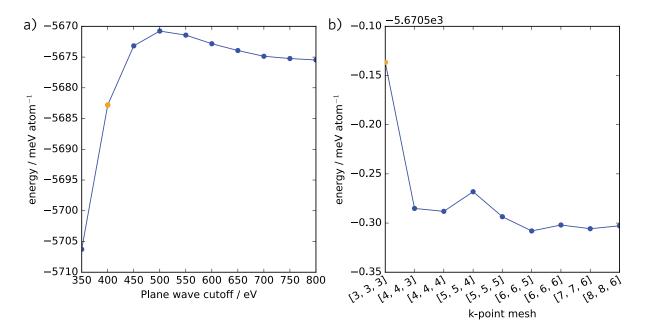


Figure S1: The total VASP energies of $BaBi_2O_6$ against (a) plane-wave energy cut-off and (b) k-point grid. The value converged to 10 meV atom⁻¹ is highlighted in orange.

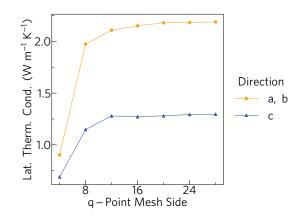


Figure S2: Convergence of the lattice thermal conductivity of $BaBi_2O_6$ against q-point mesh side broken down by lattice direction. Lattice thermal conductivities are converged to $0.01 \text{ W m}^{-1} \text{ K}^{-1}$ by the used $24 \times 24 \times 24$ q-point mesh.

BaBi₂O₆ Brillouin Zone

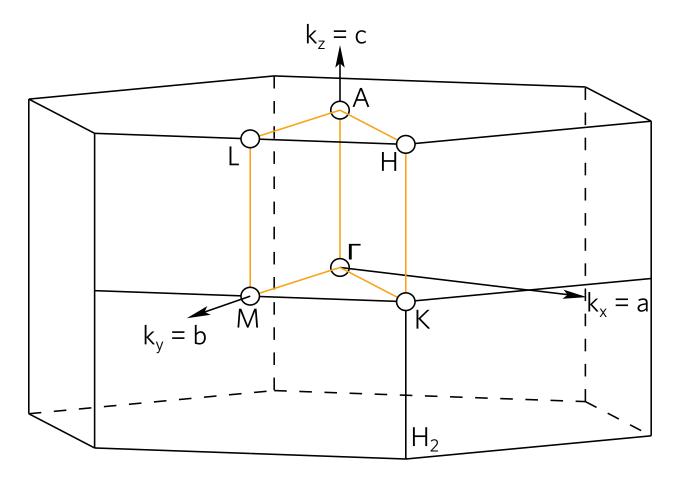


Figure S3: First Brillouin zone of $BaBi_2O_6$ (space group $P\bar{3}1m$ (162)), adapted from the Bilbao Crystallographic Server.¹ The Bradley–Cracknell paths² used in the electron and phonon dispersions (main text Figures 5(b) and 8(a)) are shown in orange.

Defects

Phase	$E_{\rm f}~({\rm eV})$
Ba_2Bi_3	-4.363
$\mathrm{Ba}_4\mathrm{Bi}_3$	-7.391
BaO	-5.069
BaO_2	-5.736
BiO_2	-2.856
$\mathrm{Bi}_{2}\mathrm{O}_{3}$	-5.801
$\mathrm{Bi}_4\mathrm{O}_7$	-11.71
$\mathrm{Ba}_{2}\mathrm{Bi}_{2}\mathrm{O}_{5}$	-17.26
$BaBi_2O_6$	-12.19

Table S1: The formation energies of $BaBi_2O_6$ and its competing phases using HSE06

Table S2: The chemical potential limits of the stable region of $BaBi_2O_6$ under O-poor conditions using HSE06, also illustrated in main text Figure 9

Competin	ng Phases	$\mid \mu_{Ba} \ (eV)$	μ_{Bi} (eV)
Elements	$\mathrm{Bi}_{2}\mathrm{O}_{3}$	-6.335	-2.927
$\mathrm{Bi}_{2}\mathrm{O}_{3}$	$\mathrm{Ba_2Bi_2O_5}$	-5.398	-2.404
$\mathrm{Ba_2Bi_2O_5}$	BaO_2	-5.290	-2.782
BaO_2	Elements	-5.736	-3.227

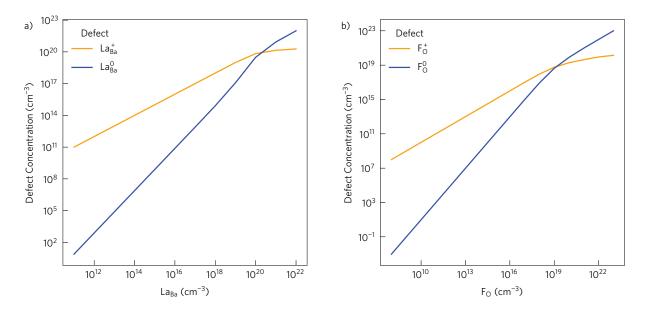


Figure S4: Concentrations of the different charge states for (a) La_{Ba} and (b) F_{O} in $BaBi_2O_6$. The x-axis value shows the total dopant amount that would be needed to get the carrier concentration given by the orange line. At high carrier concentrations, these values differ because a significant number of dopant atoms form neutral defects (blue line).

Oxygen Interstitial

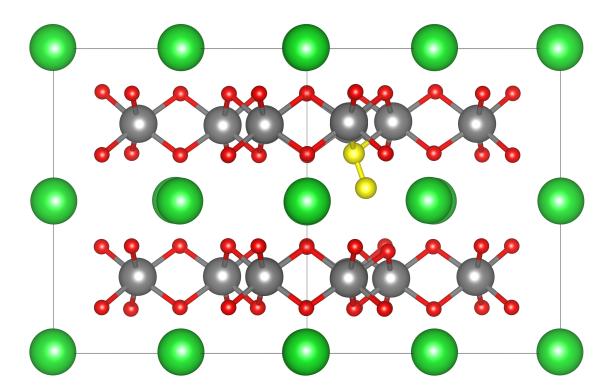


Figure S5: Relaxed O interstitial, which forms a peroxide-like species. Ba is green, Bi grey and O red, except for the peroxide-like defect in yellow.

AMSET inputs

Table S3: The k-point meshes used in the density of states (DoS), the interpolated DoS used in AMSET, density functional perturbation theory (DFPT) and optics calculations for $BaBi_2O_6$. The layers are in the *ab* plane. The high-frequency dielectric constant was calculated from the optics calculation, the ionic dielectric constant, elastic constant, piezoelectric constant (of 0) and polar optical phonon frequency were calculated from DFPT. The static dielectric constant is the sum of the high-frequency and ionic dielectric constants.

Calculation	DoS	Interpolated DoS	DFPT	Optics
k-point mesh	$12 \times 12 \times 12$	$91\times91\times75$	$14 \times 14 \times 12$	$10\times10\times9$

High-frequency dielectric constant $(\epsilon_0) =$	6.90 0 0	0 6.90 0	0 0 6.2	25		
Static dielectric constant $(\epsilon_0) =$	$\begin{bmatrix} 17.26 \\ 0 \\ 0 \end{bmatrix}$	5 0 17.1 0	18	0 0 9.79		
Elastic constant (GPa) =	173 11 82	11 174 82	82 82 80	0 0 0	0 0 0	79 -79 0
	0 0	0 0	0 0	82 -79	-79 180	0 0
	79	-79	0	0	0	108

Polar optical phonon frequency (THz) = 8.70

Mobility

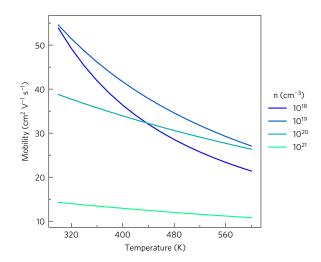


Figure S6: Electron mobility in the in-plane direction against temperature at various doping concentrations.

References

- Aroyo, M. I.; Orobengoa, D.; de la Flor, G.; Tasci, E. S.; Perez-Mato, J. M.; Wondratschek, H. Brillouin-Zone Database on the Bilbao Crystallographic Server. Acta Crystallogr. A 2014, 70, 126.
- (2) Bradley, C.; Cracknell, A. The Mathematical Theory of Symmetry in Solids: Representation Theory for Point Groups and Space Groups; Oxford University Press, 2009.