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Occurrence, sources and seasonal variation of PM<sub>2.5</sub> carbonaceous aerosols in a water level fluctuation zone in the Three Gorges Reservoir, China

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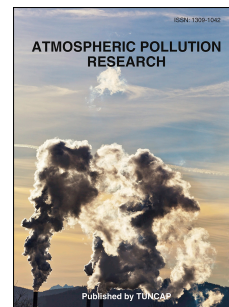
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**Xi Wang:** Writing-Original draft preparation.

**Fengwen Wang:** Supervision.

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**Fumo Yang:** Figures 1-5 integration.

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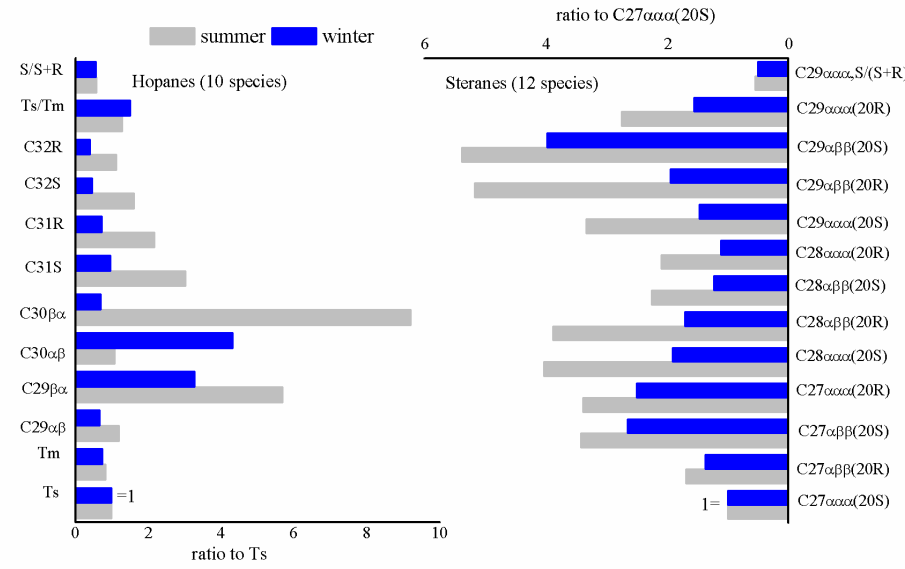
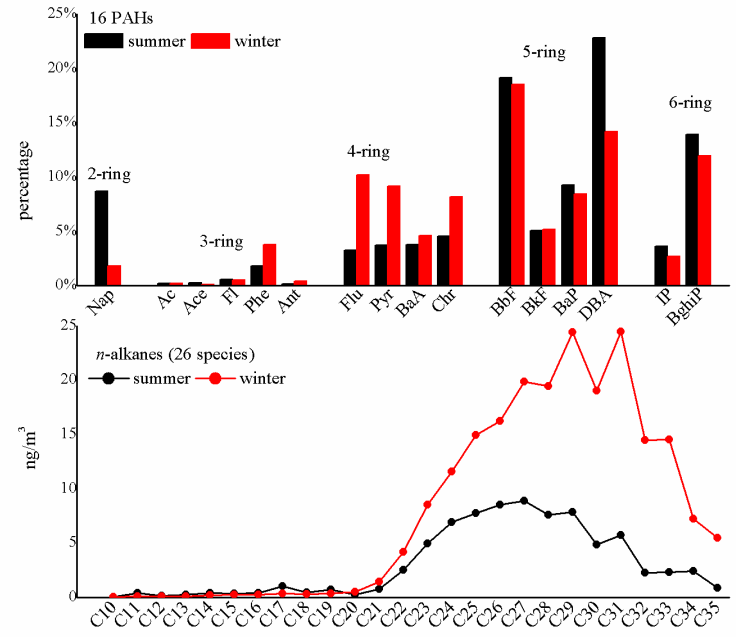
Winter: High water level



Summer: Low water level



water-level-fluctuation zone (WLFZ) in TGRR



1 **Occurrence, sources and seasonal variation of PM<sub>2.5</sub> carbonaceous aerosols in a**  
2 **water level fluctuation zone in the Three Gorges Reservoir, China**

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20 Abstract:

21 Periodic water impoundment and seasonal flood events result in a 30 m water  
22 level fluctuation zone (WLFZ) in the Three Gorges Reservoir Region (TGRR), China.

23 In order to assess the occurrence, sources and seasonal variation of airborne  
24 carbonaceous pollutants, a total of thirty-nine PM<sub>2.5</sub> samples, twenty-two from August  
25 2017 (summer: low water level ~135m) and seventeen from January 2018 (winter:  
26 high water level ~175m) were collected consecutively at a rural site in WLFZ in the  
27 TGRR. The results showed that OC, EC, char, soot, 16 PAHs and *n*-alkanes were  
28 higher in winter (mean: 9.17 μg/m<sup>3</sup>, 4.69 μg/m<sup>3</sup>, 4.25 μg/m<sup>3</sup>, 0.45 μg/m<sup>3</sup>, 19.8 ng/m<sup>3</sup>  
29 and 210 ng/m<sup>3</sup>, respectively) than in summer (mean: 6.55 μg/m<sup>3</sup>, 1.70 μg/m<sup>3</sup>, 1.37  
30 μg/m<sup>3</sup>, 0.32 μg/m<sup>3</sup>, 6.13 ng/m<sup>3</sup> and 79.1ng/m<sup>3</sup>, respectively). Compositional  
31 differences suggested air-soil/plant exchange as a source for 2-3-rings PAHs in  
32 summer and biomass burning for 4-5-rings in winter. Diagnostic ratios and PCA  
33 suggested more vehicular emission for PAHs in summer. Plant waxes contributed  
34 18.2 % and 26.2% of the *n*-alkanes in summer and winter, respectively, implying a  
35 relatively greater contribution from petroleum residues. The specific ratios and  
36 relative abundance of hopanes and steranes indicated mixed sources from both vehicle  
37 exhaust and petroleum residue; and petroleum residues contribute more steranes in  
38 summer compared with winter. As revealing basic information on the formation  
39 mechanism of carbonaceous aerosols, this study could contribute to a further  
40 understanding of their environmental geochemical behavior in the WLFZ of TGRR.

41 **Keywords:** carbonaceous pollutants; PM<sub>2.5</sub>; seasonal variations; water level  
42 fluctuation zone; Three Gorges Reservoir

### 43 **1 Introduction**

44 Following construction of the Three Gorges Dam (TGD), one of the biggest

45 hydroelectric dams in the world (2309 m long and 181 m high), a prominent reservoir  
46 known as the Three Gorges Reservoir Region (TGRR), was formed on a main  
47 tributary of the Yangtze River of China (<http://www.ctgpc.com.cn>). As one key  
48 function of TGD is to produce electricity, the TGRR has been allowed to maintain low  
49 water levels from May to September and higher water levels from October to April,  
50 respectively. A water-level-fluctuation zone (WLFZ) that experiences a novel  
51 hydrological regime exposed during May to September (summer: minimum to ~145m  
52 water depth) and submerged at October to April (winter: maximum to ~175m water  
53 depth) in the TGRR was therefore formed (Zhang et al., 2008). As the TGRR has a  
54 length of ~190 km and an area of ~350 km<sup>2</sup>, the WLFZ covers a large area and has  
55 given rise to environmental concerns since its complete impoundment in 2006. For  
56 example, Chen et al., (2009) compared the temporal-spatial CH<sub>4</sub> emissions among  
57 four vegetation stands in a typical drawdown area of WLFZ in 2008, finding  
58 important influence of water depth and dissolved organic carbon (DOC) on the spatial  
59 variations of emission flux and obvious seasonal variations with a maximal value in  
60 early July followed by a low and steady emission before winter flooding. Ye et al.,  
61 (2011) assessed soil heavy metal pollution status before and after submergence across  
62 12 sites in the WLFZ from 2008 to 2009, revealing that As and Cd were the primary  
63 pollutants before submergence; while afterwards, it was Hg, Cd, and Pb. Wang et al.,  
64 (2013) investigated the distribution characteristics of USEPA priority 16 PAHs in the  
65 water of WLFZ between late-April and early-June 2011, showing that PAHs increased  
66 from town or counties to big industrialized cities and therefore could be used to trace

67 the water pollution status of TGR. Floehr et al., (2015) conducted a sampling  
68 campaign on sediments and fish at several sites of the WLFZ from 2011 to 2013 to  
69 determine organic pollution levels and mutagenic potential. High PAH occurrence was  
70 detected at sediments of two hot spots, Chongqing (150-433 ng/g) and Kaixian  
71 (127-590 ng/g), conforming their key role as major pollutant sources and sites of  
72 eco-toxicological risk in this area. Gao et al., (2017) reported the spatiotemporal  
73 variability of organotins (OTs) in surface water under different water levels from 2012  
74 to 2013 along the TGR, highlighting the significant role of dynamic water level  
75 conditions on the occurrence and compositions of OTs. Sang et al., (2019) evaluated  
76 the effects of water impoundment and water-level manipulation on heavy metals,  
77 finding apparent bioaccumulation of Cu, Fe, Zn and Hg in fish and aquatic  
78 invertebrates in high water levels (winter) of the TGR. While these studies refer to  
79 the environmental behavior and ecological effects of major pollutants in the surface  
80 waters and sediments of the WLFZ, air-surface (water, soil and plant) exchange  
81 processes and fluxes of these pollutants could also be driven by the dynamic water  
82 level. However, to date, the variation in concentrations, compositions and sources of  
83 pollutants in the atmosphere between different seasons associated with water levels in  
84 the WLFZ have received little attention.

85 With a length of over 600 km and an area of 1,084 km<sup>2</sup>, the TGR extends from  
86 Jiangjin District of Chongqing Municipality to Yichang City of Hubei province (Chen  
87 et al., 2013). Approximately 80% area of the TGR is under the jurisdiction of  
88 Chongqing Municipality. Chongqing, the largest city by population (over 30 million

89 in 2018), is one of the four municipalities and national financial and economic centers  
90 in southwest China. Due to rapid industrialization and urbanization, the PM<sub>2.5</sub>  
91 pollution in Chongqing has deteriorated in recent decades, especially with respect to  
92 the occurrence of atmospheric carbonaceous pollutants (e.g., Organic Carbon: OC and  
93 Elemental Carbon: EC) (Chow et al., 2006; Yang et al., 2011; Chen et al., 2014;  
94 Zhang et al., 2015). Influenced by the mountainous topography and unique  
95 meteorological conditions (low wind speeds, high frequency of temperature  
96 inversions) of the TGRR, these carbonaceous pollutants are readily trapped in the  
97 WLFZ following transport from Chongqing municipality and have a detrimental  
98 effect on its ecosystems (Wang et al., 2016b). Studying the variation of carbonaceous  
99 pollutants in PM<sub>2.5</sub> in different seasons could provide useful information on these  
100 pollutants and their formation mechanisms. However, to the best of our knowledge,  
101 there has been no report on the details of carbonaceous species in the atmosphere of  
102 WLFZ in TGRR. In this study, thirty-nine PM<sub>2.5</sub> samples, twenty-two from August  
103 2017 (summer: low water level ~135m) and seventeen in January 2018 (winter: high  
104 water level ~175m), were collected consecutively at a rural site in a typical WLFZ of  
105 the TGRR. These samples were analyzed for carbonaceous pollutants, including OC,  
106 EC, char-EC (char), soot-EC (soot), 16 USEPA priority polycyclic aromatic  
107 hydrocarbons (16 PAHs), *n*-alkanes (26 species, C<sub>10</sub>–C<sub>35</sub>), as well as relative  
108 abundances of hopanes (10 species, C<sub>29</sub>–C<sub>32</sub>) and steranes (12 species, C<sub>27</sub>–C<sub>29</sub>). The  
109 purpose of this study was to assess the status of carbonaceous aerosol pollution in  
110 PM<sub>2.5</sub> driven by the dynamic water level and seasonal factors. This provides important



111 information on the formation mechanism of carbonaceous aerosols and could also  
112 contribute to a further understanding of their environmental geochemical behavior in  
113 the WLFZ of TGRR.

## 114 **2 Materials and Methods**

### 115 **2.1 Sampling sites and sample collection**

116 The sampling site (31°9'6" N, 108°33'52" E) is located at Pengxi River wetland  
117 nature reserve in TGRR (Figure 1). With low residential density and almost no  
118 industrial activity nearby, this nature reserve is about 30 km away from the center of  
119 Kaizhou district of Chongqing. The TGR water level in the reserve can reach ~175 m  
120 in January and drop to ~145 m in August, forming an area of 36 km<sup>2</sup> and a 30-m  
121 vertical drawdown zone in the wetlands. These features make this zone an ideal site to  
122 compare the characteristics of carbonaceous pollutants in the atmosphere between  
123 different seasons associated with water levels in TGRR.

124 The sampling apparatus was placed on the roof of a three-story building, 8 m  
125 away from a roof fence and 200 m above sea level. With an area of approximately 500  
126 m<sup>2</sup>, this sampling location is a 'super' scientific monitoring station affiliated to  
127 Chongqing Forestry Bureau and has been used previously by researchers to observe  
128 atmospheric deposition of polycyclic aromatic compounds (Tian et al., 2017) and  
129 inorganic nitrogen (NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup>) (Wang et al., 2018). In this study, PM<sub>2.5</sub> samples  
130 were collected on quartz filters (20×25 cm<sup>2</sup>, T2600, Pall Corporation, Port  
131 Washington, NY, USA) using a PM<sub>2.5</sub> sampler (ASM-1, Guangzhou Mingye Huanbao  
132 Technology Company, Guangzhou, China) at a flow rate of 300 L min<sup>-1</sup>. Each sample

133 was based on 23.5-hours collection, starting at 09:00 and ending at 08:30 the  
134 following day. Two sampling campaigns, 22 Jul-13 Aug 2017 (summer, n=22, average  
135 temperature: 33°C) and 15 Jan-31 Jan 2018 (winter, n=17, average temperature: 8°C),  
136 were conducted to represent low and high water level, respectively. For each sampling  
137 period, we obtained two parallel operational sample blanks, which were collected by  
138 placing filters into the sampler without pumping air. Prior to sampling, the quartz  
139 filters were wrapped in aluminum foil and baked in a muffle furnace for 4 h at 450°C.  
140 The filters were then sealed in valve bags and stored in a desiccator. Following  
141 exposure, all samples including the operational blanks were stored at -20°C until  
142 further analysis. The filters were each divided into two halves. One was used for PAH  
143 and *n*-alkane analysis while the other half was used for EC and OC.

## 144 **2.2 Sample analysis for OC and EC**

145 A Thermal/Optical Carbon Analyzer (Desert Research Institute: DRI, Model  
146 2015) was used to analyze OC and EC by IMPROVE thermal/optical reflectance  
147 (TOR) as suggested by Chow et al., (2007). For each sample, four OC fractions (OC1,  
148 OC2, OC3, and OC4), three EC fractions (EC1, EC2, and EC3) and a pyrolyzed  
149 carbon fraction (OP) were analyzed by taking areas of 0.544 cm<sup>2</sup> from the sampling  
150 filter. The OC fractions (1-4) were analyzed in a helium atmosphere at 140 °C, 280 °C,  
151 480°C and 580 °C respectively. The EC fractions (1-3) were analyzed in a 2%  
152 oxygen/98% helium atmosphere at 580 °C, 740 °C and 840 °C, respectively. OP was  
153 used to monitor the charring of OC. OC is the sum of the four OC fractions plus OP.  
154 EC is the sum of the three EC fractions minus OP. Based on the identification of two

155 EC fractions provided by Han et al., (2009), char could be calculated as EC1 minus  
156 OP and soot as the sum of EC2 and EC3.

### 157 **2.3 Sample analysis for PAHs, *n*-alkanes**

158 In order to extract PAHs and *n*-alkanes, dichloromethane (DCM) was used for  
159 Soxhlet extraction lasting 48 hours. Before extraction, a known PAHs standard sample  
160 consisting of deuterated naphthalene (Nap-d<sub>8</sub>, *m/z* 136) (200 ng), deuterated  
161 acenaphthene (Ace-d<sub>10</sub>, *m/z* 164) (200 ng), deuterated phenanthrene (Phe-d<sub>10</sub>, *m/z* 188)  
162 (200 ng), deuterated chrysene (Chr-d<sub>12</sub>, *m/z* 240) (200 ng) and deuterated perylene  
163 (Per<sub>12</sub>, *m/z* 264) (200 ng) was added to DCM to monitor the efficiency of extraction.  
164 After extraction, the mixed DCM was rotary evaporated to about 5 mL at 40°C and 50  
165 rpm/min in a vacuum rotary evaporator. The extracts were transferred into 22 mL  
166 glass bottles and evaporated down to about 2 mL by N<sub>2</sub> with a purity of 99%. The  
167 concentrates were then purified on chromatography columns (8 mm in diameter, 20  
168 cm in length), which, from bottom to top, were filled with 3 cm deactivated Al<sub>2</sub>O<sub>3</sub>, 3  
169 cm SiO<sub>2</sub> and 1 cm Na<sub>2</sub>SO<sub>4</sub>. Subsequently, the columns were eluted three times with  
170 20 mL dichloromethane/hexane (1:1, v: v). Before PAH analysis, the eluate was  
171 further concentrated to about 500 µL under gentle N<sub>2</sub> stream.

172 The targeted PAHs were 16 USEPA priority PAHs. The following are the number  
173 of rings, names and acronyms of these 16 PAHs. 2-ring: naphthalene (Nap); 3-ring:  
174 acenaphthylene (Ac), acenaphthene (Ace), fluorene (Fl), phenanthrene (Phe),  
175 anthracene (Ant); 4-ring: fluoranthene (Flu), pyrene (Pyr), benzo[a]anthracene (BaA),  
176 chrysene (Chr); 5-ring: benzo[b]fluoranthene (BbF), benzo[k]fluoranthene (BkF),

177 benzo[a]pyrene (BaP), dibenzo[a,h]anthracene (DBA); 6-ring: indeno[1,2,3-cd]pyrene  
178 (IP), benzo[ghi]perylene (BghiP). The molecular formulae of the 26 targeted  
179 *n*-alkanes are C<sub>10</sub>H<sub>22</sub>, C<sub>11</sub>H<sub>24</sub>, C<sub>12</sub>H<sub>26</sub>, C<sub>13</sub>H<sub>28</sub>, C<sub>14</sub>H<sub>30</sub>, C<sub>15</sub>H<sub>32</sub>, C<sub>16</sub>H<sub>34</sub>, C<sub>17</sub>H<sub>36</sub>, C<sub>18</sub>H<sub>38</sub>,  
180 C<sub>19</sub>H<sub>40</sub>, C<sub>20</sub>H<sub>42</sub>, C<sub>21</sub>H<sub>44</sub>, C<sub>22</sub>H<sub>46</sub>, C<sub>23</sub>H<sub>48</sub>, C<sub>24</sub>H<sub>50</sub>, C<sub>25</sub>H<sub>52</sub>, C<sub>26</sub>H<sub>54</sub>, C<sub>27</sub>H<sub>56</sub>, C<sub>28</sub>H<sub>58</sub>,  
181 C<sub>29</sub>H<sub>60</sub>, C<sub>30</sub>H<sub>62</sub>, C<sub>31</sub>H<sub>64</sub>, C<sub>32</sub>H<sub>66</sub>, C<sub>33</sub>H<sub>68</sub>, C<sub>34</sub>H<sub>70</sub>, C<sub>35</sub>H<sub>72</sub>.

182 For GC-MS analysis, hexamethylbenzene (HMB) was quantitatively (200ng)  
183 added to the samples as an internal standard. The GC-MSD (Agilent GC 6890 N  
184 coupled with 5975C MSD) was equipped with DB5-MS column (30 m × 0.25 mm ×  
185 0.25 μm) and highly purified helium (carrier gas, 99% in purity). The GC-MS  
186 operation is as follows: The temperature is set initially at 60°C for 2 min, further  
187 ramped to 290°C at 3 °C/min and held for 20 min. The samples were injected with the  
188 split-less mode at 290°C. The post-run time was 5 min at 310 °C.

189 Authentic standards of the 16 PAHs (Nap, Ac, Ace, Fl, Phe, Ant, Flu, Pyr, BaA,  
190 Chr, BbF, BkF, BaP, DBA, IP, BghiP) and 14 even numbered *n*-alkanes (C<sub>10</sub>, C<sub>12</sub>, C<sub>14</sub>,  
191 C<sub>16</sub>, C<sub>18</sub>, C<sub>20</sub>, C<sub>22</sub>, C<sub>24</sub>, C<sub>26</sub>, C<sub>28</sub>, C<sub>30</sub>, C<sub>32</sub>, C<sub>34</sub> and C<sub>36</sub>) were used to determine the  
192 concentration of PAHs and *n*-alkanes respectively. PAHs were quantified by the ions  
193 of *m/z* 128 for Nap, 152 for Ac, 153 for Ace, 166 for Fl, 178 for Phe and Ant, 202 for  
194 Pyr and Flu, 228 for BaA and Chr, 252 for BbF, BkF and BaP, 276 for IP and BghiP,  
195 278 for DBA. *N*-alkanes were quantified by the ions at *m/z* 57 and 71. For the odd  
196 alkanes, we use the close even alkanes to calculate the abundance based on the  
197 retention times and mass spectra (Wang et al., 2017). The target component  
198 identification was based on the charge-mass ratio and retention times of the

199 chromatographic peaks with the standards.

### 200 **2.3 Sample analysis for hopanes and steranes**

201 The pretreatment process and GC-MS analysis of hopanes and steranes was the  
202 same as those for PAHs and *n*-alkanes. The 10 targeted hopanes and 12 steranes were  
203 detected by separating the characteristic ions: *m/z* 191 for hopanes, *m/z* 217 for 5 $\alpha$ -,  
204 14 $\alpha$ - and 17 $\alpha$ -steranes and *m/z* 218 for 5 $\alpha$ -, 14 $\beta$ - and 17 $\beta$ -steranes. Without authentic  
205 standard samples (i.e., reference material) of the targeted hopanes and steranes, we  
206 still can present their relative abundances rather than concentration data. The  
207 abbreviation and forms of the 10 targeted hopanes and 12 steranes are shown in Table  
208 S1 in Supporting Information.

### 209 **2.4 Quality Assurance / Quality Control (QA/QC)**

210 The Thermal/Optical Carbon Analyzer was calibrated with known quantities of  
211 sucrose every day. One replicate analysis was performed for every 10 samples. The  
212 difference determined from replicate analyses was 3.0% for OC and EC, 4.5% for TC  
213 (total carbon) and 8.6% for char and soot. Results were corrected relative to the  
214 average blank concentrations.

215 The purity of organic solvent used in this study (DCM, Hexane) was 95% HPLC  
216 grade. Prior to pretreatment, all containers were soaked and washed with hot  
217 potassium dichromate and sulfuric acid immersion. These were then rinsed with  
218 de-ionized water (18.2 M $\Omega$  Milli-Q), wrapped in aluminum foil and baked at 450°C  
219 for 4 hours in a muffle furnace. Before use, all containers were rinsed three times with  
220 DCM.

221 For PAHs, the average surrogate recoveries were  $63\% \pm 8\%$  for Nap-d<sub>8</sub>,  $73\% \pm 5\%$   
222 for Ace-d<sub>10</sub>,  $91\% \pm 7\%$  for Phe-d<sub>10</sub>,  $91\% \pm 4\%$  for Chr-d<sub>12</sub> and  $94\% \pm 5\%$  for Per-d<sub>12</sub>,  
223 respectively. The reported PAHs concentrations were recovery corrected in this study.  
224 The QA/QC for *n*-alkanes was based on the recovery rate of Phe-d<sub>10</sub> and Chr-d<sub>12</sub>  
225 according to Guo et al., (2009). Nominal detection limits for individual PAHs ranged  
226 from 8 to 80 pg/m<sup>3</sup>; and from 10 to 100 ng/m<sup>3</sup> for the 20 *n*-alkanes. Procedural blanks  
227 and standard-spiked blanks were analyzed and the results are within QA/QC  
228 requirements. Sample results were corrected by subtracting the average blank.

### 229 **3 Results and discussion**

#### 230 **3.1 Concentrations of OC, EC, char, soot, PAHs and *n*-alkanes**

231 Table S2 in Supporting Information presents the eight fractions of OC, EC, 16  
232 PAHs and 26 *n*-alkanes concentrations in all PM<sub>2.5</sub> samples. The average  
233 concentrations and some indices of carbonaceous pollutants in winter and summer are  
234 presented in Table 1. The average concentrations of OC ( $9.17 \pm 5.03 \mu\text{g}/\text{m}^3$ ) in winter  
235 were higher than those of summer ( $6.55 \pm 2.27 \mu\text{g}/\text{m}^3$ ). This difference was more  
236 apparent for EC,  $4.69 \pm 3.40 \mu\text{g}/\text{m}^3$  in winter and  $1.70 \pm 0.71 \mu\text{g}/\text{m}^3$  in summer,  
237 respectively. Similar results were observed for both char and soot:  $4.25 \pm 3.09 \mu\text{g}/\text{m}^3$   
238 for char and  $0.45 \pm 0.47 \mu\text{g}/\text{m}^3$  for soot in winter and  $1.37 \pm 0.5 \mu\text{g}/\text{m}^3$  for char and  
239  $0.32 \pm 0.21$  for soot in summer, respectively. Figure 2 shows the individual  
240 concentrations of OC, EC, char and soot in order to provide a better direct comparison.  
241 It could be seen that apart from soot, the differences between OC, EC and char are  
242 very similar and distinct during the sampling period. As regard PAHs and *n*-alkanes,

243 the average concentrations of PAHs was  $19.8 \pm 9.18 \text{ ng/m}^3$  in winter and  $6.13 \pm 1.27$   
244  $\text{ng/m}^3$  in summer; for *n*-alkanes, they were  $210 \pm 147 \text{ ng/m}^3$  and  $79.1 \pm 17.7 \text{ ng/m}^3$ ,  
245 respectively. These large differences indicate that there may be specific sources of  
246 high carbonaceous pollutants between summer and winter.

247 Table 2 provides a comparison of the carbonaceous pollutants, including OC, EC,  
248 char, soot, PAHs and *n*-alkanes in  $\text{PM}_{2.5}$  measured in this study with other studies  
249 worldwide. The OC and EC concentrations in  $\text{PM}_{2.5}$  at WLFZ ( $7.86$  and  $3.20 \text{ }\mu\text{g/m}^3$ ,  
250 respectively) were lower than a rural site at Shaanxi, China ( $38.1$  and  $4.91$ ,  
251 respectively) (Zhu et al., 2012), comparable to background/rural sites at Dongguan,  
252 Lin'an and Haikou (Feng et al., 2015; Liu et al., 2017; Wang et al., 2015d), but higher  
253 than a remote site at Yulong mountain, China ( $1.84$  and  $0.55 \text{ }\mu\text{g/m}^3$ , respectively)  
254 (Zhang et al., 2018) and a rural site at Simcoe, Canada ( $1.30$  and  $0.64 \text{ }\mu\text{g/m}^3$ ,  
255 respectively) (Jeong et al., 2013). With regard to char and soot, the concentrations of  
256  $\text{PM}_{2.5}$  at WLFZ ( $2.81$  and  $0.39 \text{ }\mu\text{g/m}^3$ , respectively) were lower than a rural site in  
257 Shaanxi, China ( $4.05$  and  $0.86 \text{ }\mu\text{g/m}^3$ , respectively) (Zhu et al., 2012), comparable to a  
258 rural site at Suthep Mountain in Chiangmai, Indochina ( $2.97$  and  $0.36 \text{ }\mu\text{g/m}^3$ ,  
259 respectively) (Chuang et al., 2013), but higher than at Qinghai Lake in China ( $0.16$   
260 and  $0.22 \text{ }\mu\text{g/m}^3$ , respectively) (Li et al., 2013). For PAHs and *n*-alkanes, the total  
261 concentrations of these two compounds in  $\text{PM}_{2.5}$  at WLFZ ( $12.1$  and  $145 \text{ ng/m}^3$ ,  
262 respectively) were much higher than at Yulong mountain ( $0.97$  and  $6.53 \text{ ng/m}^3$ ,  
263 respectively) (Zhang et al., 2018), Qinghai Lake ( $0.69$  and  $6.47 \text{ ng/m}^3$ , respectively)  
264 (Li et al., 2013), Singapore ( $0.76$  and  $25.5 \text{ ng/m}^3$ , respectively) (Zhang et al., 2017)

265 and Qatar (0.56 and 8.53 ng/m<sup>3</sup>, respectively) (Javed et al., 2019). However, at  
266 Lin'an, China, concentrations of PAHs were 26.9 ng/m<sup>3</sup> while *n*-alkanes  
267 concentrations were 62.2 ng/m<sup>3</sup> (Feng et al., 2015). These results demonstrate that the  
268 concentrations of *n*-alkanes in PM<sub>2.5</sub> at WLFZ of TGRR are up to ten times higher  
269 than those elsewhere (Table 2). It may be inferred that these *n*-alkanes may be derived  
270 from specific sources, such as epicuticular waxes, associated with higher plants in the  
271 TGRR.

### 272 3.2 SOC estimation and sources from OC/EC and char/soot ratios

273 EC is essentially a primary pollutant while OC is resulted from both primary  
274 emitted sources (primary organic carbon: POC) and secondary transformation from  
275 gaseous precursors (secondary organic carbon: SOC). When the OC/EC ratios exceed  
276 2.0, the relationship of OC and EC can be used to estimate the SOC as (Turpin et al.,  
277 1995):

$$278 \quad \text{SOC} = \text{OC} - \text{POC}$$

$$279 \quad \text{POC} = \text{EC} \times (\text{OC}/\text{EC})_{\text{min}}$$

280 where  $(\text{OC}/\text{EC})_{\text{min}}$  is the the minimum ratio of OC/EC. This equation has been  
281 used by researchers to identify source categories and estimate levels of secondary  
282 organic carbon (SOC) (Chow et al., 1996; Castro et al., 1999; Cao et al., 2005; Feng  
283 et al., 2009). However, the ratio of char to soot (char/soot) may also be a valuable  
284 indicator to imply source variations since there are distinct changes between different  
285 emission sources (Han et al., 2010; Wang et al., 2019). Table 1 summarize the SOC  
286 concentrations,  $2.47 \pm 1.32 \mu\text{g}/\text{m}^3$  in winter and  $3.11 \pm 2.00 \mu\text{g}/\text{m}^3$  in summer, and



287 their relative contribution to OC (SOC/OC in %),  $33.0 \pm 22.3$  in winter and  $44.6 \pm$   
288  $22.2$  in summer, respectively. Higher SOC concentrations (mean  $3.11 \mu\text{g}/\text{m}^3$ ) and  
289 SOC/OC (mean 44.6%) were observed in summer compared with winter,  $2.47 \mu\text{g}/\text{m}^3$   
290 for SOC and 33.0% for SOC/EC, respectively. In summer, the air temperature could  
291 be as high as  $40 \text{ }^\circ\text{C}$  with a high frequency of sunny days, generating more intense of  
292 solar radiation. Under these meteorological conditions, there could be much higher  
293 photochemical activity resulting in enhanced formation of SOC (Castro et al., 1999).  
294 In winter, the temperature fluctuated around  $10 \text{ }^\circ\text{C}$  with a high frequency of rain,  
295 providing suitable meteorological conditions to preserve POC (Safai et al., 2014). As  
296 a consequence, carbonaceous pollutants could accumulate in the water and sediments  
297 of WLFZ in winter; while in summer, the conditions of high temperature, intensive  
298 solar radiation and oxygen availability in the exposed surface sediments could result  
299 in the re-emission of these carbonaceous pollutants to the atmosphere. Therefore, the  
300 difference between SOC and SOC/OC in  $\text{PM}_{2.5}$  of WLFZ at TGRR between summer  
301 and winter could be due to the distinct seasonal meteorological conditions enhanced  
302 by air-water, air-sediment and air-plant exchange in this region.

303 Table 1 also presents the mean OC/EC and char/soot in  $\text{PM}_{2.5}$  between winter  
304 and summer. The OC/EC was  $2.47 \pm 1.11$  in winter and  $4.24 \pm 1.67$  in summer,  
305 respectively. For char/soot, these were  $14.3 \pm 15.6$  in winter and  $4.69 \pm 2.12$  in  
306 summer, respectively, displaying a much more pronounced variation than OC/EC.  
307 According to Schauer et al., (2002) and Wang et al., (2015a), higher OC/EC (e.g.,  $>4$ )  
308 are associated with wood/coal burning or SOC; while Saarikoski et al., (2008)

309 suggested that lower OC/EC (e.g., <4) could characterize the dominance of fossil-fuel  
310 burning sources, such as vehicle emissions. Therefore, the OC in PM<sub>2.5</sub> in summer  
311 could be from wood/coal burning and SOC; while EC was solely from wood/coal  
312 burning. Fossil-fuel combustion could be the main contributor to these two  
313 carbonaceous pollutants in winter. As regards char/soot, ratios of 0.60 for motor  
314 vehicle exhaust, 1.31 for coal combustion and 22.6 for biomass burning have been  
315 suggested by Chow et al., (2004); while ratios of >3 for biomass burning/coal  
316 combustion and <1.0 for motor vehicle exhausts were found by Han et al., (2010).  
317 The higher char/soot ratio in winter indicated dominance of biomass burning and  
318 could be associated with the burning of agricultural straw or firewood. In rural China,  
319 these materials are often burned in stoves for cooking and indoor heating in winter.  
320 The lower char/soot in summer indicates a major influence from fossil-fuel  
321 combustion, and could be explained by vehicle emissions as a result of intensive  
322 agricultural cultivation on the large exposed area of the WLFZ. Source identification  
323 from char/soot could be more reliable than that from OC/EC as it is not influenced by  
324 secondary organic formation processes and therefore will maintain the source  
325 fingerprints of the fuel type.

### 326 **3.3 Compositions of PAHs and *n*-alkanes**

327 Figure 3 shows the relative percentages of 16 PAHs and concentrations of 26  
328 *n*-alkanes in PM<sub>2.5</sub> between winter and summer. Distinct differences in the  
329 percentages of Nap (2-ring), Phe, Ant (3-ring) and Flu, Pyr, BaA, Chr (4-ring) in total  
330 PAHs between these two seasons could be observed. In winter, Phe, Ant, Flu, Pyr,

331 BaA and Chr contributed 36.4% to the 16 PAHs; while in summer it was only 17.4%.  
332 BbF, BkF, BaA, DBA, IP, and BghiP (5~6-ring) dominated, and contributed 61.0% in  
333 winter and 74.0% in summer, respectively. Nap, Ac and Ace accounted for 2.21% in  
334 winter and 9.23% in summer, respectively. It has been reported that 4-ring PAHs are  
335 derived from both from coal combustion and biomass burning while 5-6-ring PAHs  
336 could originate from diesel and gasoline vehicle emissions (Harrison et al., 1996).  
337 Natural mineral dusts are also found to contain Nap, Ace, Fl, and Phe (Moon et al.,  
338 2008). The higher contribution from Nap, Ac and Ace in summer compared with  
339 winter indicates a possible “volatilization” from exposed surfaces (e.g., soils and  
340 plants) in WLFZ. The volatilization here was achieved by re-emission of aged PAHs  
341 from contaminated sediments that received high levels of pollutants during  
342 impoundment. Nap, Ac and Ace are emitted in the gaseous phase and then perhaps  
343 adsorbed onto PM<sub>2.5</sub> in the atmosphere (Wang et al., 2014). Further study that focuses  
344 on the air-soil/plant exchange of individual PAHs under different seasons associated  
345 with water levels are therefore needed and suggested. The higher contribution from  
346 Phe, Ant, Flu, Pyr, BaA and Chr in winter indicates organic pollutants emitted as a  
347 result of coal combustion and biomass burning such as agricultural straw or firewood.

348 The compositions of 26 *n*-alkanes (C<sub>10</sub>-C<sub>35</sub>) were similar between these two  
349 seasons. However, differences in their concentrations, especially for C<sub>25</sub>-C<sub>33</sub>, were  
350 clearly observed. In order to distinguish the biogenic and anthropogenic origins of  
351 *n*-alkanes, the carbon preference index (CPI) has been used to assess source  
352 information (Simoneit, 1986). The CPI in this study was calculated as the sum of the

353 concentrations of the odd carbon-number (e.g., C<sub>11</sub>, C<sub>13</sub>, C<sub>15</sub>...) divided by the even  
354 carbon-number (e.g., C<sub>10</sub>, C<sub>12</sub>, C<sub>14</sub>...), i.e.:  $CPI = \frac{\sum(C_{11}-C_{35})}{\sum(C_{10}-C_{34})}$ . The CPI  
355 was similar for the two seasons, averaging  $1.24 \pm 0.17$  in winter and  $1.27 \pm 0.15$  in  
356 summer, respectively (Table 1). The CPI values around 1.00 were also observed at  
357 TSP in a small town of Linzhi on the southeastern Tibetan Plateau, China (Chen et al.,  
358 2014) and PM<sub>2.5</sub> of a remote island in East China Sea (Wang et al., 2015b). A CPI of  
359 1.2 in PM<sub>2.5</sub> of December, 2011 in urban Shanghai, China was also presented by Wang  
360 et al., (2016a). According to Simoneit, (1986), a CPI equivalent to 1.0 suggests a  
361 dominance of sources from anthropogenic activity, such as gasoline  
362 emissions/petroleum residues). Therefore, anthropogenic sources substantially  
363 contributed to the *n*-alkanes in PM<sub>2.5</sub> of WLFZ in TGRR.

#### 364 **3.4 Sources of PAHs and *n*-alkanes**

365 The molecular diagnostic ratios (MDRs), such as Phe/(Phe+Ant), Flu/(Flu+Pyr),  
366 BaA/(BaA+Chr) and IP/(IP+BghiP), have been widely used to identify PAH sources  
367 (Guo et al., 2009; Wang et al., 2014; Wang et al., 2015b). However, the use of these  
368 ratios was based on a number of assumptions (Ravindra et al., 2008). First, each  
369 emitting sources always release individual PAHs at the same ratios. Second, PAHs of  
370 the same molecular weight have similar environmental fates, i.e., the concentration  
371 ratio of PAH A against PAH B always remains constant during their entire  
372 environmental lifespan. These assumptions make MDRs be useful tools especially on  
373 the determination or speciation of the “fresh” emission sources around mixed  
374 combustion sources such as urban atmosphere. Recently, Harrison, et al., (2018)

375 revealed the mechanism of distinct reactivity of PAHs during atmospheric transport  
376 and found OH and NO<sub>3</sub> radicals responsible for the gas phase oxidation of low  
377 molecular weight PAHs (fluoranthene and pyrene). Therefore, use of diagnostic ratios  
378 at remote locations could be misleading since there would be aged process associated  
379 with re-emission from re-suspended soil or plant surface. In this study, we combined  
380 MDRs with principal component analysis (PCA) to apportion the PAHs sources as to  
381 improve the ability of discrimination.

382 Figure 4 compares the Phe/(Phe+Ant), Flu/(Flu+Pyr), BaA/(BaA+Chr) and  
383 IP/(IP+BghiP) between summer and winter. Phe/(Phe+Ant) was 0.89 in winter and  
384 0.92 in summer, indicating a dominant source from coal combustion/biomass burning.  
385 However, Flu/(Flu+Pyr) was 0.53 in winter and 0.48 in summer, respectively. This  
386 discrepancy suggests that, in winter, grass, wood, coal combustion dominated; while  
387 in summer fossil fuel combustion contributed the most. BaA/(BaA+Chr) was 0.36 in  
388 winter and 0.45 in summer, both characterizing vehicular emission. The values of  
389 IP/(IP+BghiP) in winter and summer were 0.54 and 0.62, respectively, indicating an  
390 origin from coal, wood and grass combustion.

391 PCA using SPSS 16.0 (SPSS Inc., Chicago, IL, USA) was performed on datasets  
392 containing eight MDRs PAHs (Phe, Ant, Flu, Pyr, BaA, Chr, IP, and BghiP). Table 3  
393 presents two extracted PCs, PC1 and PC2, for both winter and summer. PCs explained  
394 95.8% (53.2 and 42.6%, respectively) of the total variance for winter and 89.0% (62.1  
395 and 26.9%, respectively) for summer. PC1 in winter had high loadings of Phe, Ant,  
396 Flu and Pyr, whereas PC1 in summer was dominated by Flu, Pyr, BaA, IP and BghiP.

397 Therefore, PC1 in winter was associated with coal combustion/biomass burning; PC1  
398 in summer was more associated with vehicular emissions in addition with coal  
399 combustion/biomass burning (Harrison, et al., 1996; Rajput, et al., 2011; Wang et al.,  
400 2014). PC2 in winter had high loadings of IP and BghiP, whereas PC1 in summer was  
401 dominated by Phe and Ant. Hereby, PC2 in winter was characterized as vehicular  
402 emissions and PC2 in summer was assigned as crude oil leakage or refined petroleum  
403 release (Zakaria et al., 2002). The sources apportioned using these MDRs therefore  
404 complement each other, generating a more comprehensive understanding of the  
405 sources of PM<sub>2.5</sub>-bound PAHs in WLFZ in TGR.

406 Simoneit et al., (1991) suggested that the sources of *n*-alkanes could be mainly  
407 divided into plant waxes (natural) and petroleum residues (anthropogenic). The  
408 *n*-alkane with the highest concentration, namely C<sub>max</sub>, was C<sub>26</sub>-alkanes or C<sub>27</sub>-alkanes  
409 in winter and C<sub>29</sub>-alkanes or C<sub>31</sub>-alkanes in summer, characterizing a source from  
410 epicuticular waxes of higher plants. Table 1 lists the relative contribution of plant  
411 waxes which was calculated as: Wax C<sub>n</sub> = [C<sub>n</sub> - (C<sub>n+1</sub> + C<sub>n-1</sub>)/2]/C<sub>n</sub>. The contribution  
412 from plant wax was 26.2 ± 13.6% in winter and 18.2 ± 8.90% in summer, respectively.  
413 In order to adapt to the periodic flooding conditions, the artificial cultured trees (e.g.,  
414 *Aeschynomene indica*, *Cyperus*) could lose their leaves or these could be abraded,  
415 possibly emitting more epicuticular waxes to the atmosphere. In summer, there could  
416 be petroleum leakage due to fuel handling and/or refueling of the vehicles that  
417 undertake intensive agricultural cultivation on the exposed areas. The slight difference  
418 in percentage of plant waxes between the two seasons indicates that anthropogenic

419 activity contributes most of the total 26 PM<sub>2.5</sub>-bound *n*-alkanes in WLFZ of TGRR.  
420 Actually, there have been some studies that also estimate the contribution from plant  
421 wax to total *n*-alkanes at other places. For example, as to the TSP samples that were  
422 collected from Lulang on the southeastern Tibetan Plateau, China, the percent of plant  
423 wax to total *n*-alkanes was higher in winter than that in summer, which was 27.9%  
424 and 18.5%, respectively (Chen et al., 2014). While in PM<sub>2.5</sub> of East China Sea, the  
425 percent of plant wax to total *n*-alkanes in winter was only 3.30% and in summer it was  
426 9.00% (Wang et al., 2015a). With a CPI value of 1.2, a ratio of 8.20% from plant wax  
427 to total *n*-alkanes was observed in PM<sub>2.5</sub> of winter in urban Shanghai (Wang et al.,  
428 2016). These differences between these studies suggest spatial-temporal variations  
429 and uncertainties associated with the sample number and sampling apparatus and  
430 further research is required.

### 431 **3.5 Relative abundances of hopanes, steranes, and their source implications**

432 The ratios of Ts/Tm,  $\alpha\beta\text{C}_{31} \text{ S}/(\text{S}+\text{R})$ , and  $\alpha\alpha\alpha\text{C}_{29} \text{ S}/(\text{S}+\text{R})$  are common  
433 parameters used to identify anthropogenic sources (Oros et al., 2000; Alves et al.,  
434 2008). As summarized in Table 1, Ts/Tm was  $1.52 \pm 0.36$  in winter and  $1.33 \pm 0.48$  in  
435 summer, respectively.  $\alpha\beta\text{C}_{31} \text{ S}/\text{S}+\text{R}$  was  $0.57 \pm 0.02$  in winter and  $0.57 \pm 0.07$  in  
436 summer. With regard to  $\alpha\alpha\alpha\text{C}_{29} \text{ S}/(\text{S}+\text{R})$ , it was  $0.49 \pm 0.08$  in winter and  $0.55 \pm 0.06$   
437 in summer. According to Feng et al., (2005), a ratio of Ts/Tm < 1.0 indicates the  
438 influence of less thermally mature fuels (e.g., biomass and coal). Ts/Tm was >1.0 in  
439 both seasons, implying an impact from vehicle exhaust. The ratios of  $\alpha\beta\text{C}_{31}\text{S}/(\text{S} + \text{R})$   
440 both averaged 0.57, indicating a dominant source from traffic emissions (Fraser et al.,

441 1998). The  $\alpha\alpha\alpha\text{C}_{29} \text{ S}/(\text{S}+\text{R})$  was  $0.49 \pm 0.08$  in winter and  $0.55 \pm 0.06$  in summer.  
442 The relatively high  $\alpha\alpha\alpha\text{C}_{29} \text{ S}/(\text{S}+\text{R})$  suggests more mature petroleum residues in the  
443 atmosphere in summer (Zaghden et al., 2007), consistent with the suggestion of  
444 agricultural activity at this time.

445 The relative abundance diagrams of 10 hopanes and 12 steranes between summer  
446 and winter are shown in Figure 5. These abundances were calculated based on  
447  $18\alpha(\text{H})$ -22,29,30-trisnorneohopane (Ts: where  $\text{Ts} = 1$ ) and  $\text{C}_{27}\text{-}5\alpha(\text{H})$ ,  $14\alpha(\text{H})$ ,  
448  $17\alpha(\text{H})$ -steranes ( $\text{C}_{27}\alpha\alpha\alpha(20\text{S})$ : where  $\text{C}_{27}\alpha\alpha\alpha(20\text{S}) = 1$ ). It could be seen that the  
449 dominant hopanes in winter were  $\text{C}_{30}\alpha\beta$ , followed by  $\text{C}_{29}\beta\alpha$ ; while the dominant  
450 hopanes in summer were  $\text{C}_{30}\beta\alpha$  and  $\text{C}_{29}\beta\alpha$ . Non-catalytic converter equipped  
451 gasoline-powered vehicles could emit particle-bound  $\text{C}_{30}\alpha\beta$  (Schauer et al., 2002).  
452 The dominant species of steranes during the two seasons were  $\text{C}_{29}\alpha\beta\beta(20\text{S})$  and  
453  $\text{C}_{29}\alpha\beta\beta(20\text{R})$ . These two species were found to be abundant in the particulate phase  
454 of non-catalytic converter equipped gasoline-powered vehicle tailpipes (Schauer et al.,  
455 2002). Similar patterns, namely  $\text{C}_{29}$  being more abundant than  $\text{C}_{28}$  and  $\text{C}_{27}$  steranes,  
456 was also found in surface sediments of the Bohai Sea, China, where it was attributed  
457 to crude oil leakage (Hu et al., 2009).

458 The ratios of  $\text{Ts}/\text{Tm}$ ,  $\alpha\beta\text{C}_{31} \text{ S}/(\text{S}+\text{R})$  (Table 1) and the relative abundance of  
459 hopanes (Figure 5) indicate mixed sources from vehicle exhaust and mature petroleum  
460 residues to the  $\text{PM}_{2.5}$ -bound hopanes in WLFZ of TGRR. The ratios of  $\alpha\alpha\alpha\text{C}_{29}$   
461  $\text{S}/(\text{S}+\text{R})$  (Table 1) and the relative abundance of steranes (Figure 5) also indicate  
462 vehicle exhaust and petroleum residues are the major sources of steranes and



463 petroleum residues contribute more of steranes in summer compared with winter.

#### 464 **4 Conclusions**

465 This study provides the first datasets on the concentrations, sources and seasonal  
466 variations of carbonaceous pollutants in PM<sub>2.5</sub> of two distinct water levels, ~175 m in  
467 winter and 145 m in summer, at WLFZ of TGR. All the carbonaceous pollutants had  
468 higher concentrations in winter with respect to summer. The seasonal SOC and  
469 SOC/OC ratio between summer and winter differs mainly as a result of distinct  
470 air-water, air-soil and air-plant exchange in this region. The different compositions of  
471 16 PAHs in the two sampling periods implies that the dominant sources were distinct,  
472 with air-soil/plant exchange contributing 2~3-ring PAHs in summer and biomass  
473 burning associated with agricultural straw or firewood contributing 4-5-ring PAHs in  
474 winter. The results obtained by MDRs and source apportionment using PCA  
475 complemented each other well, highlighting a prominent contribution from traffic  
476 emissions to PM<sub>2.5</sub>-bound PAHs in summer. The contribution of plant waxes to  
477 *n*-alkanes in winter ( $26.2 \pm 13.6\%$ ) was higher than that in summer ( $18.2 \pm 8.90\%$ ),  
478 possibly due to the loss or abrasion of leaves from trees (e.g., *Aeschynomene indica*,  
479 *Cyperus*) under periodic flooding conditions. Petroleum residues are the dominant  
480 source of steranes in summer, likely due to agricultural vehicles, while vehicle  
481 exhaust was the most common source of hopanes and steranes in both seasons.

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- 704

705 **Table captions**

706 Table 1 Concentrations and some indices of carbonaceous pollutants in PM<sub>2.5</sub> between  
 707 summer and winter in WLFZ (n=22 and 17 in summer and winter, respectively,  
 708 average ± stdev).

709 Table 2 Comparisons of the carbonaceous pollutants in PM<sub>2.5</sub> measured this study  
 710 with other places worldwide (OC, EC, Char, Soot in μg/m<sup>3</sup>, PAHs and *n*-alkanes in ng  
 711 /m<sup>3</sup>).

712 Table 3 Correlation factor loading matrix of selected eight PAHs in PM<sub>2.5</sub> (" - "  
 713 indicate values of the factor loading below zero).

714 **Figure captions**

715 Figure 1 Sampling site at WLFZ of TGRR.

716 Figure 2 Comparisons of OC, EC, char and soot in PM<sub>2.5</sub> between summer and winter  
 717 associated with water levels.

718 Figure 3 The compositions of PAHs and *n*-alkanes in PM<sub>2.5</sub> between summer and  
 719 winter.

720 Figure 4 The diagnostic ratios of Phe/(Phe+Ant), Flu/(Flu+Pyr), BaA/(BaA+Chr) and  
 721 IP/(IP+BghiP) of PM<sub>2.5</sub> between summer and winter.

722 Figure 5 Abundance distribution diagrams of relative hopane and sterane of PM<sub>2.5</sub>  
 723 between summer and winter.

724 (Ts:18α(H)-22,29,30-trisnorneohopane;Tm:17α(H)-22,29,30-trisnorhopane;C<sub>29</sub>αβ:17α(H),21

725 β(H)-norhopane;C<sub>29</sub>βα:17β(H),21α(H)-norhopane;C<sub>30</sub>αβ:17α(H),21β(H)-hopane;C<sub>30</sub>βα:17β(

726 H),21α(H)-hopane;C<sub>31</sub>S:22S-17α(H),21β(H)-homohopane;C<sub>31</sub>R:22R-17α(H),21β(H)-homoho

727 **pane;C<sub>32</sub>S:22S-17 $\alpha$ (H),21 $\beta$ (H)-bishomohopane;C<sub>32</sub>R:22R-17 $\alpha$ (H),21 $\beta$ (H)-bishomohopane)**  
728 **(C27 $\alpha\alpha\alpha$ (20S), C27 $\alpha\beta\beta$ (20R), C27 $\alpha\beta\beta$ (20S), C27 $\alpha\alpha\alpha$ (20R), C28 $\alpha\alpha\alpha$ (20S), C28 $\alpha\beta\beta$ (20R),**  
729 **C28 $\alpha\beta\beta$ (20S), C28 $\alpha\alpha\alpha$ (20R), C29 $\alpha\alpha\alpha$ (20S), C29 $\alpha\beta\beta$ (20R), C29 $\alpha\beta\beta$ (20S), C29 $\alpha\alpha\alpha$ (20R), and**  
730  **$\alpha\alpha\alpha$  = 5 $\alpha$ (H), 14 $\alpha$ (H), 17 $\alpha$ (H)-steranes;  $\alpha\beta\beta$  = 5 $\alpha$ (H), 14 $\beta$ (H), 17 $\beta$ (H)-steranes, R and S =**  
731 **C-20 R and S configuration, respectively.)**  
732

Table 1 Concentrations and some indices of carbonaceous pollutants in PM<sub>2.5</sub> between summer and winter in WLFZ (n=22 and 17 in summer and winter, respectively, average  $\pm$  stdev).

Sampling period	Seasons (water level)	OC	EC	SOC	Char	Soot	OC/EC	Char/Soot	SOC/OC (%)
		$\mu\text{g}/\text{m}^3$							
2017/07/22-08/25	summer (145 m)	6.55 $\pm$ 2.27	1.70 $\pm$ 0.71	3.11 $\pm$ 2.00	1.37 $\pm$ 0.57	0.32 $\pm$ 0.21	4.24 $\pm$ 1.67	4.69 $\pm$ 2.12	44.6 $\pm$ 22.2
2018/01/15-01/31	winter (175 m)	9.17 $\pm$ 5.03	4.69 $\pm$ 3.40	2.47 $\pm$ 1.32	4.25 $\pm$ 3.09	0.45 $\pm$ 0.47	2.47 $\pm$ 1.11	14.3 $\pm$ 15.6	33.0 $\pm$ 22.3
		16 PAHs	<i>n</i> -alkanes (C <sub>10</sub> -C <sub>35</sub> , 26 species)			hopanes		steranes	
		ng/m <sup>3</sup>	yield (ng/m <sup>3</sup> )	CPI	C <sub>max</sub>	waxed (%)	Ts/Tm	$\alpha\beta\text{C}31$ S/S+R	$\alpha\alpha\alpha\text{C}29$ S/S+R
2017/07/22-08/25	summer (145 m)	6.13 $\pm$ 1.27	79.1 $\pm$ 17.7	1.27 $\pm$ 0.15	C <sub>29</sub> /C <sub>31</sub>	18.2 $\pm$ 8.90	1.33 $\pm$ 0.48	0.57 $\pm$ 0.07	0.55 $\pm$ 0.06
2018/01/15-01/31	winter (175 m)	19.8 $\pm$ 9.18	210 $\pm$ 147	1.24 $\pm$ 0.17	C <sub>26</sub> /C <sub>27</sub>	26.2 $\pm$ 13.6	1.52 $\pm$ 0.36	0.57 $\pm$ 0.02	0.49 $\pm$ 0.08

Table 2 Comparisons of the carbonaceous pollutants in PM<sub>2.5</sub> measured this study with other places worldwide (OC, EC, Char, Soot in  $\mu\text{g}/\text{m}^3$ , PAHs and *n*-alkanes in  $\text{ng}/\text{m}^3$ ).

Location	Type of site	Time	Sample number	OC	EC	Char	Soot	$\Sigma\text{PAHs}$	$\Sigma n\text{-alkanes}$	Reference
WLFZ in TGRR	rural	Jul., 2017-Jan., 2018	n=39	7.86	3.20	2.81	0.39	12.1	145	This study
Sanya, China	suburban	Jan., 2012-Jul., 2013	n=42	3.30	1.10	0.95	0.15	6.85	14.8	Wang et al. (2015c)
Dongguan, China	rural	Feb.,2010-Dec., 2012	n=156	10.4	2.68	-	-	8.97	44.7	Wang et al. (2015d)
Shaanxi, China	rural	Nov., 2007-Dec., 2008	n=53	38.1	4.91	4.05	0.86	-	-	Zhu et al. (2012)
Lin'an, China	background	Apr., 2008-Jan., 2009	n=110	10.3	1.54	-	-	26.9	62.2	Feng et al. (2015)
Haikou, China	background	Jan., 2015-Sep., 2015	-	5.70	2.40	2.10	0.30	-	-	Liu et al. (2017)
Mt. Yulong, China	remote	Mar., 2012-Apr., 2012	n=30	1.84	0.55	-	-	0.97	6.53	Zhang et al. (2018)
Qinghai Lake, China	remote	Jul., 2010-Aug., 2010	n=56	1.58	0.37	0.16	0.22	0.69	6.47	Li et al. (2013)
Singapore	background	Nov., 2015-Feb., 2016	n=24	3.49	1.06	-	-	0.76	25.5	Zhang et al. (2017)
Simcoe, Canada	rural	Feb., 2005-Nov., 2007	n=143	1.30	0.64	0.07	0.57	-	-	Jeong et al. (2013)

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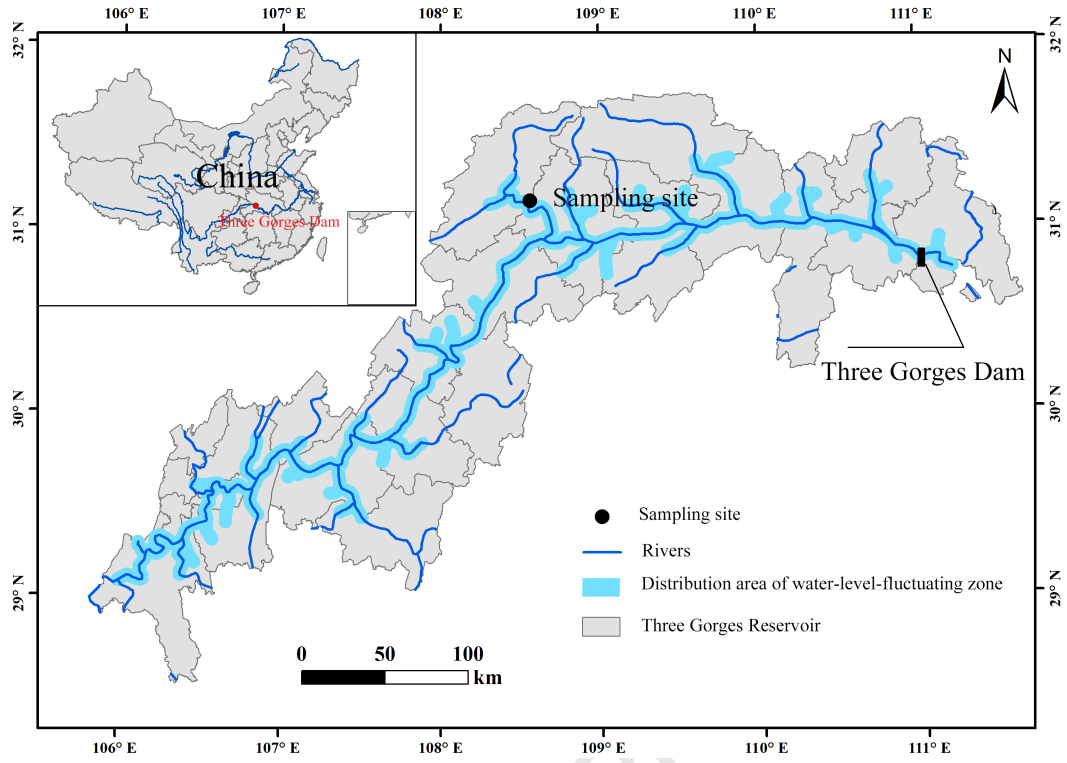
West Midlands, UK	rural	May., 2007-Apr., 2008	n=60	2.50	1.10	-	-	1.49	23.4	Harrison et al. (2010)
Chiangmai, Indochina	rural	Mar., 2010-Apr., 2010	n=15	18.6	3.33	2.97	0.36	-	-	Chuang et al. (2013)
Doha, Qatar	suburban	May. 2015-Dec., 2015.	n=105	1.78	2.61	-	-	0.56	8.53	Javed et al. (2019)
Jeju Island, South Korea	coastal	Aug., 2007-Sep., 2008	n=41	4.00	1.70	1.30	0.40	-	-	Lim et al. (2012)

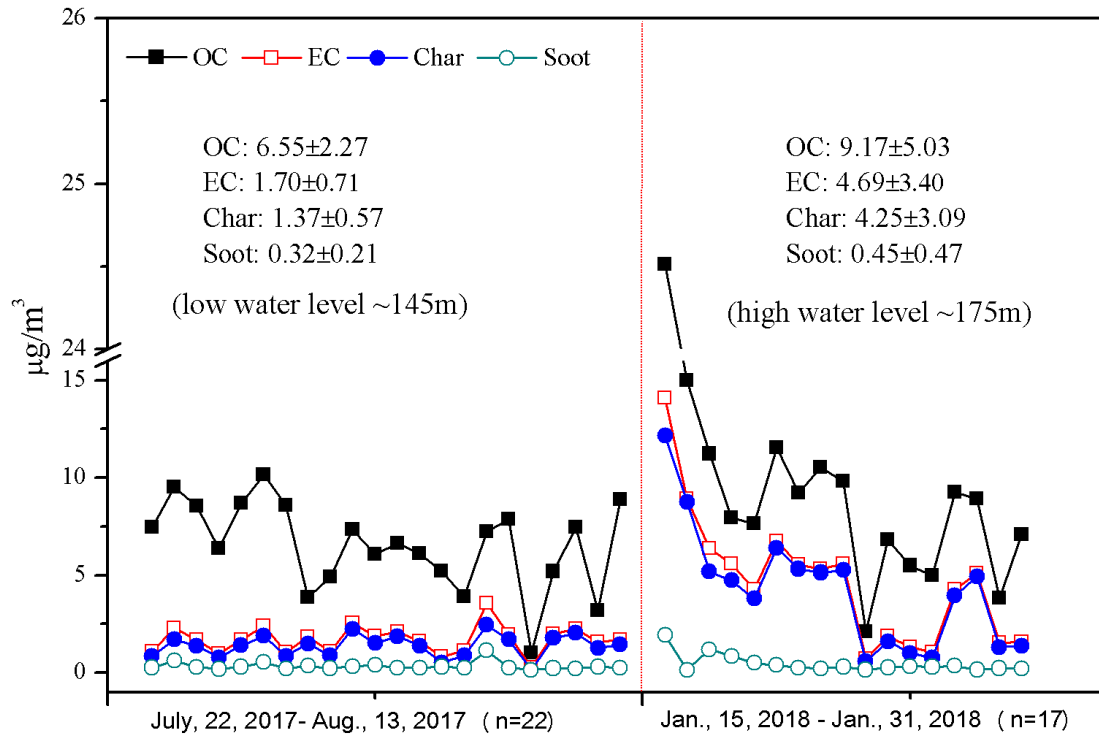
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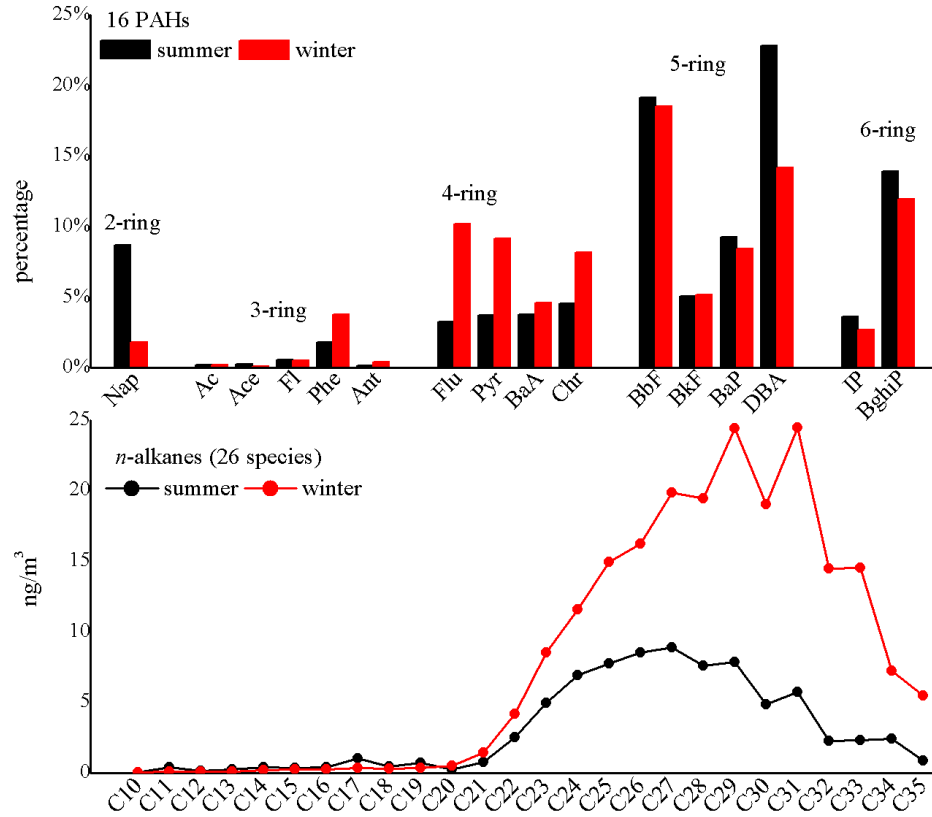


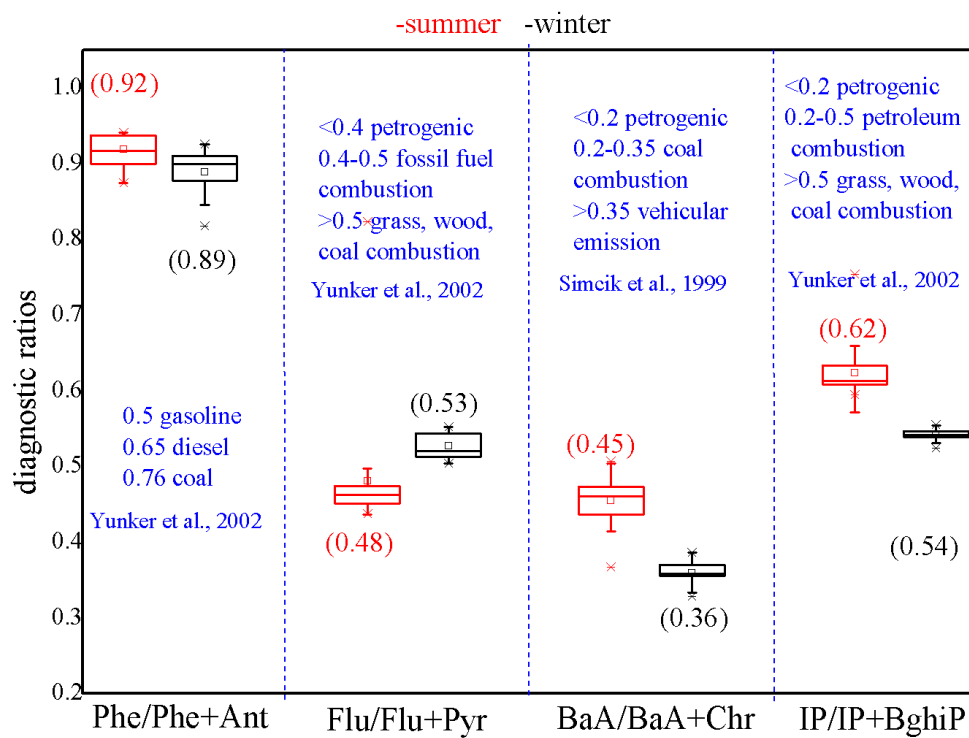
Table 3 Correlation factor loading matrix of selected eight PAHs in PM<sub>2.5</sub> (" - " indicate values of the factor loading below zero).

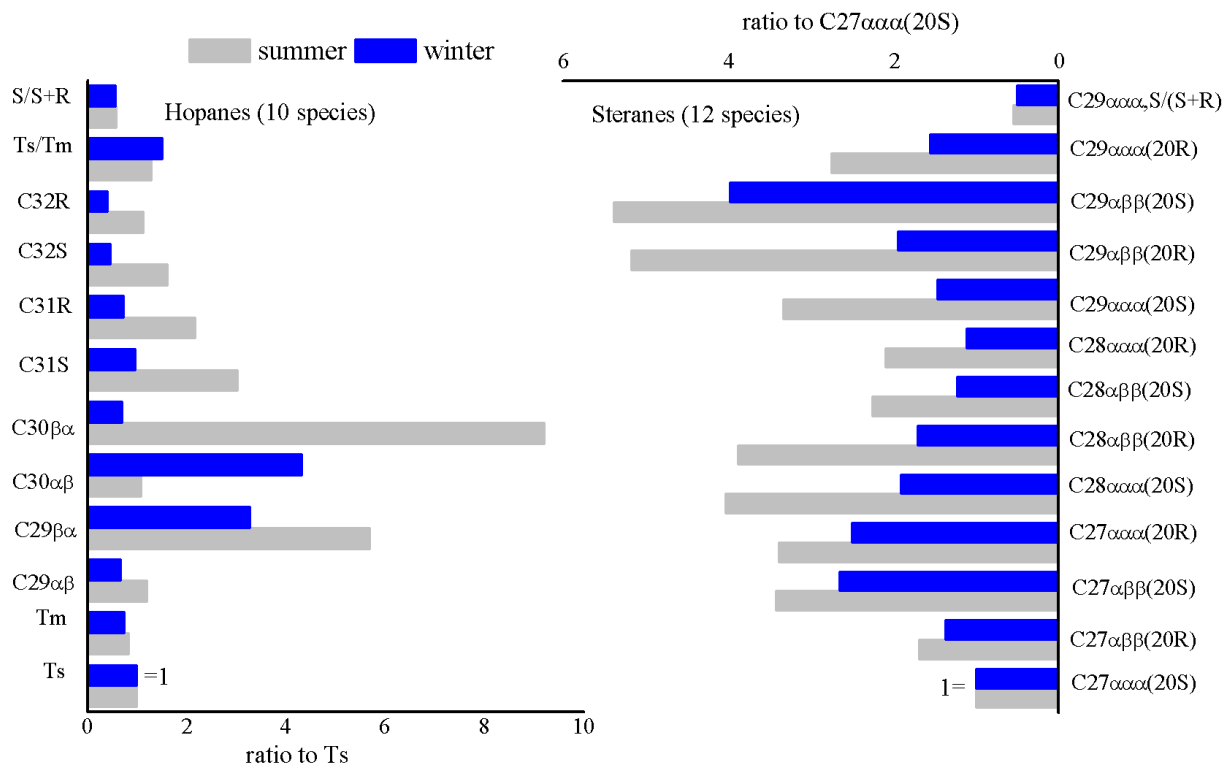
PM <sub>2.5</sub>		winter		summer	
8 species		PC1	PC2	PC1	PC2
PAHs	Phe	0.979	0.120	-	0.721
	Ant	0.791	0.442	0.007	0.821
	Flu	0.911	0.386	0.927	0.064
	Pyr	0.849	0.508	0.958	0.001
	BaA	0.668	0.720	0.939	-
	Chr	0.722	0.668	0.741	-
	IP	0.288	0.952	0.961	-
	BghiP	0.263	0.960	0.908	-
explained variance %		53.2	42.6	62.1	26.9











1. Char/soot revealed dominance of biomass burning in winter and a major influence from fossil-fuel combustion in summer.
2. Air-soil/plant exchange contributed 2~3-ring PAHs in summer and biomass burning contributed 4-5-ring in winter.
3. Vehicle exhaust was the most common source of hopanes and steranes both in summer and winter.

Journal Pre-proof

Xi Wang, Ting Feng, Siyuan Zhang and Peili Lu performed the research; Fumo Yang, Jiaxin Liu and Zhigang Guo analyzed data; Fengwen Wang, Li Liu and Neil L. Rose wrote the paper. All the co-authors substantially contributed to commenting and revising it. All authors read and approved the final manuscript. The authors have declared no conflict of interest.

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