

Non-destructive dose verification of two drugs within 3D printed polyprintlets

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Abstract

Three-dimensional printing (3DP) is a revolutionary technology in pharmaceuticals, enabling the personalisation of flexible-dose drug products and 3D printed polypills (polyprintlets). A major barrier to entry of this technology is the lack of non-destructive quality control methods capable of verifying the dosage of multiple drugs in polyprintlets at the point of dispensing. In the present study, 3D printed films and cylindrical polyprintlets were loaded with flexible, therapeutic dosages of two distinct drugs (amlodipine and lisinopril) across concentration ranges of 1-5% w/w and 2-10% w/w, respectively. The polyprintlets were non-destructively analysed for dose content using a portable near infrared (NIR) spectrometer and validated calibration models were developed using partial least squares (PLS) regression, which showed excellent linearity (R^2 Pred = 0.997, 0.991), accuracy (RMSEP= 0.24%, 0.24%) and specificity (LV1= 82.77%, 79.55%) for amlodipine and lisinopril, respectively. X-ray powder diffraction (XRPD) and thermogravimetric analysis (TGA) showed that sintering partially transformed the phase of both drugs from the crystalline to amorphous forms. For the first time, we report a non-destructive, RTR quality control of two separate active ingredients in a single 3D printed drug product using NIR spectroscopy, overcoming a major barrier to the integration of 3D printing into clinical practice.

1 **1. Introduction**

2 Hypertension is a silent killer, responsible for over 7.5 million deaths / year
3 (12.8% of all deaths) worldwide (WHO, 2019). Polypharmacy, which signifies
4 the concurrent use of multiple medications by one individual, is the current gold
5 standard for treating hypertension (Durden et al., 2013) (NICE, 2011). Due to
6 polypharmacy, medication adherence is a major challenge in the management
7 of hypertension, with over 65% of patients failing to adhere to their prescribed
8 regimens, and 50% ceasing their medication regime within one year of
9 prescription (Ruilope, 2011; Tibebe et al., 2017) (Abegaz et al., 2017; Corrêa
10 et al., 2016).

11
12 Critically, a long-term non-adherence to treatment has been associated with an
13 increased risk of cardiovascular events, including strokes, hospitalisations and
14 death (Abegaz et al., 2017; Herttua et al., 2013; Lee et al., 2017; Ong et al.,
15 2007). To resolves the issues arising from polypharmacy, the use of fixed drug
16 combinations (also known as ‘polypills’) have been explored. However,
17 although polypills have been shown to improve adherence, the dosage of each
18 drug is fixed making changes in dosage regimens inconvenient (Roy et al.,
19 2017). In order to overcome this, there is a need for a novel platform that
20 enables flexible dosing for polypills and three-dimensional printing (3DP) has
21 the potential to do so (Alomari et al., 2018; Trenfield et al., 2019a).

22
23 3DP has gained momentum in many industries such as the aeronautics,
24 robotics, electronics, manufacturing and food industries and, more recently,
25 within medicine and pharmaceuticals (Barnatt, 2013). Within the
26 pharmaceutical field, it is no longer a new idea to transition away from the
27 standard mass production of medicines of fixed strength towards creating
28 personalised dosage forms and dose combinations (Awad et al., 2018a; Awad
29 et al., 2018b; Goyanes et al., 2017). By creating medicines in a layer-by-layer
30 manner, this technology can produce printlets (3D printed tablets) that are
31 customised to a patient’s disease state, individual factors and therapeutic
32 needs (Florence and Lee, 2011; Goyanes et al., 2019b; Hamburg and Collins,
33 2010; Oblom et al., 2019; Trenfield et al., 2018a). Due to the ability for precise

34 material deposition, several studies have demonstrated the potential for 3DP to
35 create polypills (polyprintlets) containing more than one active pharmaceutical
36 ingredient (API) (Genina et al., 2017; Gioumouxouzis et al., 2018; Khaled et al.,
37 2015a, b; Robles-Martinez et al., 2019; Sadia et al., 2018b).

38

39 Whilst the evidence-based for 3DP of polyprintlets is increasing, the integration
40 of this technology into clinical practice has not yet been achieved (Alhnan et al.,
41 2016; Basit and Gaisford, 2018; Edinger et al., 2018). A major barrier
42 preventing 3DP uptake into pharmaceuticals is the absence of an at-line, non-
43 destructive quality control (QC) techniques to enable the real-time release of
44 3D printed medicines (Di Prima et al., 2016; Trenfield et al., 2018b; Trenfield et
45 al., 2019b). Analytical methods such as dose quantification using
46 chromatographic methods, as well as dissolution and disintegration testing, are
47 commonly used for QC of pharmaceuticals. However, these characterisation
48 methods are inherently destructive, which would be inconvenient for individually
49 fabricated printlets at the point-of-care (Awad et al., 2018a).

50

51 An alternative approach could involve the integration of real-time release (RTR)
52 testing, which is the ability to evaluate and ensure the quality of in-process
53 and/or final product based on process data (EMA, 2012). Vibrational
54 spectroscopic tools, such as near infrared (NIR) spectroscopy combined with
55 chemometrics, have previously been used as alternative QC tools within
56 pharmaceutical processes (Edinger et al., 2019; Trenfield et al., 2018b; Vakili
57 et al., 2017). NIR spectroscopy has been widely used for at-line analysis
58 because it has the capability to analyse and quantify drugs in a rapid, non-
59 destructive and user-friendly manner. Furthermore, it can be conveniently
60 integrated at the point of dispensing in the clinic due to its portability. The
61 potential for NIR spectroscopy as a non-destructive QC method was
62 demonstrated by our group previously whereby a point-and-shoot approach
63 was used to measure the drug content of paracetamol in printlets (Trenfield et
64 al., 2018b).

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66 To date, previous research had predominantly focused on quantifying single
67 active ingredients within individually fabricated dosage forms using NIR

68 spectroscopy (Trenfield et al., 2018b). However, one of the major benefits of
69 3DP is the ability to produce polyprintlets containing multiple APIs in the same
70 dosage form (Pereira et al., 2018; Xu et al., 2020). As such, for the first time,
71 we demonstrate the non-destructive QC of two distinct APIs (lisinopril and
72 amlodipine) at therapeutically-relevant dosages within 3D printed polyprintlets
73 using a portable, reflectance NIR spectrometer. The applicability of the model
74 to polyprintlets of different geometries (cylindrical and oral films) was evaluated,
75 and dosage forms were characterised using x-ray powder diffraction (XRPD)
76 and thermogravimetric analysis (TGA) to elucidate drug distribution and solid-
77 state characteristics.

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80 **2. Materials and Methods**

81 Amlodipine (LKT Laboratories, Inc., US) (MW 408.879 g/mol, solubility at 25°C
82 75.3 mg/L) (Pubchem, 2003) and lisinopril dihydrate (Acros Organics, UK) (MW
83 441.525 g/mol, solubility at 25°C 216 mg/L) (DrugBank; Pubchem, 2005).
84 Polyethylene oxide (PEO) 100,000 (Sigma-Aldrich, UK) which has a molecular
85 weight of 100,000 g/mol and density of 1.13 g/mL (Pubchem, 2004) was used
86 as the thermoplastic polymer in the sintering process. Candurin[®] Gold Sheen
87 was purchased from Merck KGaA, Germany.

88

89 **2.1. 3D printed design**

90 Templates of the printlets were designed by using 123D Design Software
91 (Autodesk Inc, UK), a computer-aided design (CAD) software to create 3D
92 representations of the object. Oral square films (10 mm x 10 mm x 0.5 mm) and
93 standard cylindrical printlets (10 mm diameter x 3.6 mm height) were designed.
94 3D models were exported as a stereolithographic (.stl) file into 3D printer
95 Sintratec Central software Version 1.1.13.

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99 **2.2. Printing process**

100 PEO 100,000 was sieved by using a 180 μm orifice-size sieve to reduce its
101 particle size for printing. Twenty-five formulations were prepared for building
102 the calibration model for NIR spectroscopy across five different dosage
103 concentration ranges ($n = 5$, Table 1). The lowest and highest concentrations
104 of both drugs were selected to enable the provision of therapeutically-relevant
105 dosages, i.e. assuming a 200mg tablet is produced, amlodipine 1-5%w/w
106 covers a 2mg – 10mg dose range and lisinopril 2-10%w/w covers a 4mg – 20mg
107 dose range. For each formulation, 15g of a mixture of drugs and excipients
108 were blended using a pestle and mortar. 3% w/w of Candurin Gold Sheen was
109 added to all formulations to enhance absorption of the laser to allow printability
110 (Fina et al., 2017). The powder mixtures were transferred to a selective laser
111 sinter (SLS) printer (Sintratec Kit, AG, Brugg, Switzerland) for printing. The
112 chamber temperature (which indicates the temperature inside the printer body)
113 was maintained at 30 °C and the surface temperature (which refers to the
114 temperature of the powder bed surface in the build platform) was maintained at
115 40 °C. The laser scanning speed was set at 200 mm/s. The printing process
116 started with the activation of a 2.3 W blue diode laser (445 nm) to sinter the
117 powder within the build platform in a certain pattern based on the .stl file.
118 Powder in the reservoir platform (150 mm x 150 mm x 30 mm) of the printer
119 was moved by a sledge to a building platform (150 mm x 150 mm x 30 mm)
120 creating a flat and homogeneously distributed layer of powder. Then, the laser
121 would sinter on the powder particles together. This process was repeated layer-
122 by-layer until the object was completed. The dosage form was then removed
123 from the powder bed and excess powder was brushed off. Five oral films were
124 printed at the same time for each formulation. Three formulations (A2L4, A3L6
125 and A4L8) were chosen to be printed into cylindrical tablets for inclusion into
126 the building the calibration model ($n=3$).

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133 Table 1. Formulation composition for calibration model printing

Formulation code	Drug concentration (% w/w)		PEO (% w/w)	Candurin Gold Sheen (% w/w)
	Amlodipine	Lisinopril		
A1L2	1	2	94	3
A2L4	2	4	91	3
A3L6	3	6	88	3
A4L8	4	8	85	3
A5L10	5	10	82	3

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135 **2.3. Near infrared spectroscopy (NIR) data acquisition**

136 A portable benchtop Labspec 5000 NIR spectrometer (Analytical Spectral
 137 Devices, USA), equipped with three separate holographic diffraction gratings
 138 and three separate detectors; a 512-element silicon photo-diode array for
 139 wavelengths between 350–1000 nm, and two TE-cooled InGaAs for
 140 wavelengths between 1000–1800 nm and 1800–2500 nm was used to measure
 141 the NIR reflectance. An immobilised lab grade 1 m fibre optic cable (fibre core
 142 size 200 µm), which interfaced with the NIR equipment (BIF200- Vis-NIR,
 143 Ocean Optics Inc., FL, USA) was used to collect the spectra. A Spectralon 99%
 144 reflective standard (Labsphere, North Sutton, UK) was used for instrument
 145 calibration prior to spectra acquisition. UV–visible-NIR spectra were collected
 146 across the 350–2500 nm wavelength region (2150 data points) totalling 64
 147 scans, which were averaged. Each printlet was analysed at six different points
 148 to avoid potential sampling errors and to reduce the variability caused by
 149 different surface effects. All printlets were scanned three times on each side
 150 with the same format. The final spectrum (used to calculate amlodipine and
 151 lisinopril content) was the average of the spectra recorded at the six positions
 152 (6 averaged spectra/tablet). The data was processed by using Microsoft Excel
 153 and MATLAB software version R2017a (The MathWorks, CA, USA).

154

155 **2.4. Model development**

156 All five oral film concentrations (n=5) (amlodipine 1-5% w/w; lisinopril 2-10%
 157 w/w; Table 1) were selected for calibration model development. Two oral films
 158 from A2L4, A3L6 and A4L8 was used for internal validation. Multivariate data
 159 analysis was performed using MATLAB software version R2017a (The

160 MathWorks, CA, USA) with the PLS Toolbox version 8.6 (Eigenvector, CA, USA)
161 for data pre-processing and modelling. Partial least squares (PLS) regression
162 was performed on the datasets to build calibration models. The models were
163 internally cross-validated using Venetian blinds. Validation of the NIR
164 calibration model was performed according to guidance from the International
165 Conference on Harmonization (ICH) guidance Q2(R1)(ICH, 1994b)), European
166 Medicines Agency (EMA) (EMA, 2014b) and the Food and Drug Administration
167 (FDA) (FDA, 2015a), by assessing model specificity, linearity (expressed as
168 correlation coefficient, R^2) and accuracy (expressed as the root mean square
169 error of prediction; RMSEP). The calibration model developed covered a total
170 of 25 samples of oral films (with 19 samples being selected for calibration and
171 6 samples for internal validation) over an amlodipine concentration range of 1-
172 5% w/w and lisinopril concentration range of 2-10% w/w.

173

174 **2.5. Determination of drug content**

175 Three individual oral film printlets of each formulation were placed in separate
176 volumetric flasks with deionised water and methanol (50:50). Samples of the
177 solution were then filtered using a 0.45 μm membrane filter (Millipore Ltd.,
178 Ireland) into the HPLC vials. HPLC analysis was performed using UV-HPLC
179 equipped with an Eclipse Plus C18 column (150 x 4.6 mm, 5 μm particle size)
180 (Agilent, UK) at a temperature of 40°C. Analyses were carried at a detection
181 wavelength of 215 nm, a flow rate of 1 mL/min, an injection volume of 100 μL
182 and a run time of 18 mins. The mobile phase consisted of a gradient of solvent
183 A (HPLC water adjusted to pH 3 with phosphoric acid) and solvent B
184 (acetonitrile). The method entailed the following: 1) solvent A and solvent B
185 were set at 83:17 at the start time; 2) then adjusted to 80:20 at the 6th minute;
186 3) then adjusted to 10:90 at the 15th minute 4) adjusted to 83:17 at the 18th
187 minute. Elution times for amlodipine and lisinopril were 1.9 mins and 11.1 mins,
188 respectively.

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191 **2.6. X-ray powder diffraction**

192 A Rigaku MiniFlex 600 (Rigaku, USA) with a Cu $K\alpha$ X-ray source ($\lambda = 1.5418$
193 Å) and accompanying software Miniflex Guidance version 1.2.01 were used to

194 record x-ray powder diffraction (XRPD) patterns of printlets (ground to fine
195 powder), formulation blends and pure amlodipine, lisinopril, PEO 100,000 and
196 Candurin Gold. The intensity and voltage applied were 15 mA and 40 kV
197 respectively. The angular range of data acquisition was 3–40° 2 θ , with a step
198 size of 0.02° at a speed of 2° min⁻¹.
199

200 **2.7. Thermal analysis**

201 Thermogravimetric analysis (TGA) was also used for characterisation. All the
202 samples were heated at 10 °C min until 400 °C in open aluminium pans using
203 Discovery TGA (TA instruments, Waters, LLC, USA). The purge gas used was
204 nitrogen gas with a flow rate of 25 mL/ min. Data were collected and analysed
205 by using TA Instruments Trios software and percentage mass loss and onset
206 temperature were calculated. The results from thermal analysis were plotted
207 using OriginPro Software (OriginPro 2017 (64 bit) SR2 b9.4.2.380).

208

209 **2.8. Characterisation of the printlets**

210 **2.8.1. Determination of printlet weight variability**

211 All cylindrical printlets were weighed by using a weighing balance (Sartorius AG
212 CPA225D, Germany). Printlets were measured in triplicate, and the mean and
213 standard deviation for each printlet was calculated.

214

215 **2.8.2. Determination of printlet strength**

216 A traditional tablet hardness tester TBH 200 (Erweka GmbH, Heusenstamm,
217 Germany) was used to determine the crushing strength of three cylindrical
218 printlets of each drug combination. The mean and standard deviation for each
219 printlet was calculated.

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223 **2.8.3. Determination of printlet friability**

224 Three tablets of each concentration were weighed and placed into the drum of
225 a Friability Tester Erweka type TAR 10 (Erweka GmbH, Heusenstamm,

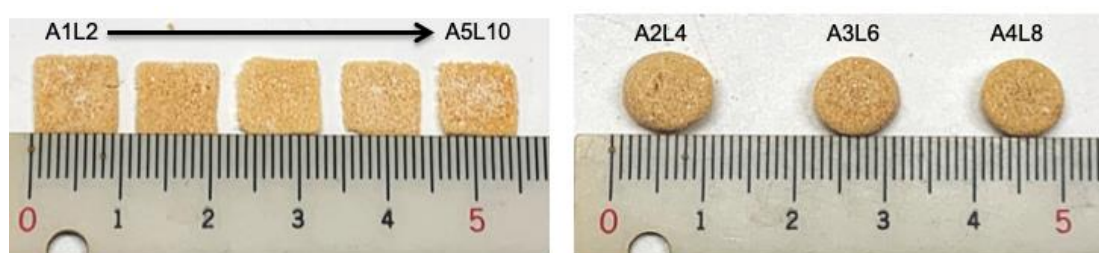
226 Germany). The drum was rotated at 25 rpm for 100 rounds and the samples
227 were reweighed. The friability of these samples was analysed in terms of weight
228 loss and it was expressed as percentage of original sample weight.

229

230 3. Results and Discussion

231 For the first time, it was possible to use a low temperature SLS printing process
232 to manufacture 3D printed oral square films and cylindrical printlets containing
233 two drugs (amlodipine and lisinopril) at therapeutically relevant concentrations
234 (up to 5% w/w and 10% w/w, respectively; Figure 1). Previously studies have
235 proven the feasibility of using SLS 3DP technology in the pharmaceutical field
236 by successfully manufacturing immediate and modified release tablets (Barakh
237 Ali et al., 2019; Fina et al., 2017; Fina et al., 2018a), as well as fast
238 disintegrating oro-dispersible tablets (Fina et al., 2018b). Awad *et al.* has also
239 shown the capability of SLS 3D printing to produce pellets (miniprintlets)
240 containing more than one drug (Awad et al., 2019). However, the majority of
241 these studies required the use of elevated temperatures (80 – 135 °C) to enable
242 effective sintering. Favourably, due to low T_g of the polymer (PEO 100,000, T_g
243 of -67 °C), it was possible to manufacture the dosage forms at a low
244 temperature (40 °C), which could be highly beneficial for thermally-labile drugs
245 that are unsuitable for higher temperature 3D printing processes (Goyanes et
246 al., 2015a).

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249 Figure 1. Printlets of two different geometries; square oral film (left) and cylindrical shape tablet
250 (right). Drug content increases from left to right. The scale is in cm.

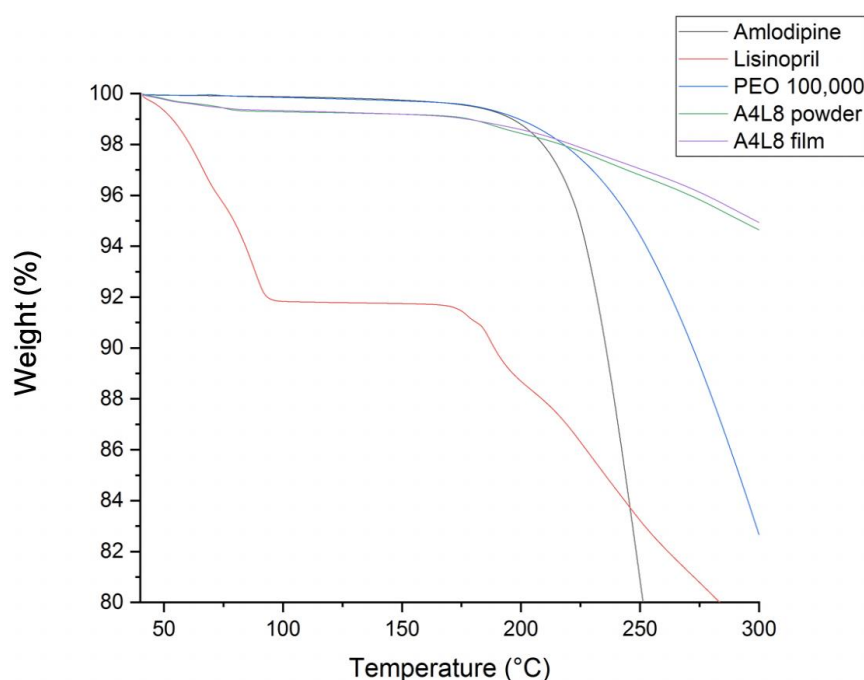
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253 3.1. Polyprintlet characterisation

254 Initially, TGA was performed to evaluate whether the drugs would be stable at
255 the temperatures required during the sintering process (Figure 2). The results

256 showed that lisinopril decomposition occurred gradually in three steps. A weight
257 loss of ~8% was observed up to 100°C, attributed to the loss of water due to
258 lisinopril being in the dihydrate form. A constant weight was maintained
259 between 100-175°C indicated that no alteration occurred in the dehydrated
260 lisinopril crystal during this stage. Beyond 175 °C, the lisinopril crystal melted
261 ($T_m = 178-179^\circ\text{C}$) and degraded, which is similar to findings reported in the
262 literature (Hinojosa-Torres et al., 2008). TGA data of the other components
263 (amlodipine and PEO 100,000) and the formulation blends predicted that all the
264 components would remain stable and no degradation of the drugs and
265 excipients was likely to occur at the printing temperatures (40 °C). HPLC
266 analysis was also used to confirm stability of amlodipine and lisinopril post-
267 printing, with the HPLC trace showing only evidence of the main APIs peaks
268 after sintering.



269
270 Figure 2. Thermogravimetric analysis of amlodipine, lisinopril, PEO 100,00, A4L8 formulation
271 blend and A4L8 film.

272
273 XRPD analysis of the drug, polymers, formulation blends and printlets were
274 performed to determine the physical state of the drugs and the degree of
275 incorporation within the polymers (Figure 3). Characteristic patterns from the
276 XRPD focused between 3 to 16° 2θ showed that the lisinopril drug peak present

277 at about $7.5^\circ 2\theta$ was also present in the powder blend. However, in the films,
 278 the peak became broader and showed a significant reduction in peak height.
 279 This indicates that lisinopril had been converted partially into the amorphous
 280 phase. Characteristic amlodipine peaks were present at 10° and $12^\circ 2\theta$ in the
 281 powder formulation but not the printed film indicating that either there was a
 282 complete conversion to the amorphous state or, alternatively, the remaining
 283 crystalline content was below the sensitivity of the XRPD method. Consistent
 284 drug and polymer peak shifts of $\sim +1^\circ 2\theta$ was apparent in the printed
 285 formulation, which was attributed to the stress-strain influence or the change
 286 in height presentation of a printed disc versus the raw powder. Such peak shifts
 287 have been observed in previous studies (Robles-Martinez et al., 2019).

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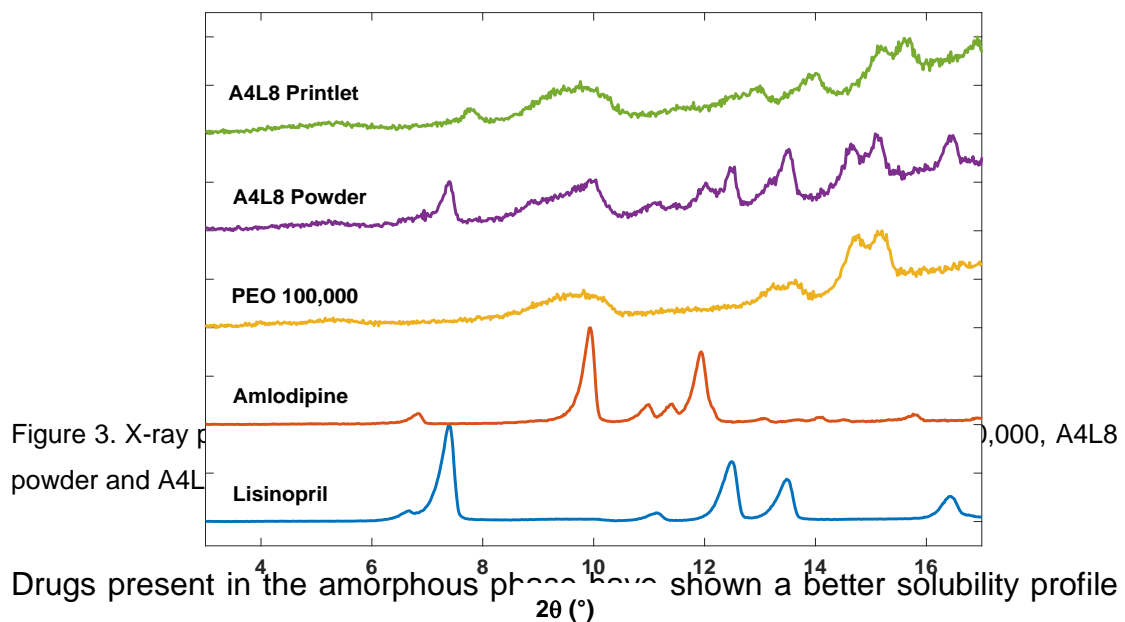
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302 Drugs present in the amorphous phase have shown a better solubility profile

303 compared with crystalline drugs (Babu and Rangia, 2011). Therefore, there are

304 a number of advantages of delivering drugs in an amorphous form especially

305 for drugs under BCS Class II or IV that has low solubility to enhance its

306 dissolution and bioavailability (Capretto et al., 2017; Martinez et al., 2014).

307 Several papers have shown the feasibility of using 3D printing technologies to

308 formulate drugs in amorphous or semi-amorphous states to achieve enhanced

309 drug release profiles (Goyanes et al., 2019a; Kollamaram et al., 2018; Sadia et

310 al., 2016). During the SLS process used here, the application of the laser may

311 lead to complete drug melting to enable formation of the non-crystalline
 312 matrices (Trenfield et al., 2018b). However, this process will vary based on the
 313 drugs, excipients and printing parameters used, such as laser scanning speed
 314 and chamber temperature. It is worth mentioning however, that despite the
 315 benefits of formulating drugs in the amorphous phase, there could also be a
 316 risk of conversion back to its crystalline state and hence in the future
 317 accelerated stability studies are required to determine shelf life of the 3D printed
 318 drug products.

319

320 Printlet hardness was evaluated to determine the ease of handling of the
 321 developed formulations. For all the cylindrical printlets, the hardness exceeded
 322 the maximum value that the equipment could measure because the printlets
 323 did not break but they were physically deformed (Table 2). Friability of all the
 324 formulations of cylindrical printlets were less than 1%, complying with the British
 325 Pharmacopoeia (BP) requirements for uncoated tablets, making them suitable
 326 for handling and packing (BP, 2018). Favourably, percentage recoveries of both
 327 amlodipine and lisinopril were determined using HPLC, and were all found to
 328 be between the 85-115% limits that have been set by BP for content uniformity
 329 testing (Table 2). All dosage forms were found to pass weight variation tests
 330 according to the BP (<7.5% variation).

331

332 Table 2. Physical properties and recovery of the cylindrical printlets

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335 **3.2. Quantitative analysis using NIR spectroscopy**

Formulation	Weight (mg) \pm SD	Crushing strength (N) \pm SD	Friability (%) \pm SD	Amlodipine recovery (%)	Lisinopril recovery (%)
A2L4	170.5 \pm 1.08	>483.7 \pm 0.58	0.23 \pm 0.05	103.9 \pm 4.7	96.9 \pm 0.7
A3L6	168.6 \pm 5.71	>484.0 \pm 0.00	0.57 \pm 0.48	99.1 \pm 7.4	99.9 \pm 0.3
A4L8	163.6 \pm 3.65	>483.7 \pm 0.58	0.93 \pm 0.90	103.1 \pm 2.4	98.7 \pm 7.6

336 In order to facilitate the integration of 3D printing for the production of
337 antihypertensive polyprintlets at the point-of-care, a non-destructive method is
338 required to enable at-line quality control and batch release. Previously, we have
339 proven the feasibility of using process analytical technologies (PAT) to quantify
340 a single model drug (paracetamol) in SLS printlets (Trenfield et al., 2018b).
341 Here, we have investigated the use of a portable NIR spectrometer to quantify
342 both therapeutically-relevant dosages of amlodipine and lisinopril in 3D printed
343 oral films and cylindrical tablets. Initially, the pure drugs (amlodipine and
344 lisinopril) and pure PEO 100,000 were scanned to identify unique peaks of
345 interest for calibration model development (Figure 4). For amlodipine, the
346 wavelengths selected ranged between 1450-1600 nm and 2000-2100 nm,
347 whereas for lisinopril the wavelength selected was between 1600-1730 nm.
348 Lisinopril also displayed a high absorbance at ~1920nm, however this peak is
349 attributed to the presence of water due to the drug being in the dihydrate form;
350 as such, this peak was excluded for model development. The feasibility of using
351 the selected absorbance peaks was evaluated by scanning formulation blends
352 of increasing drug concentrations (amlodipine: 1-4% w/w and lisinopril 2-8%
353 w/w) (Figure 5). The NIR absorbance was found to increase upon increasing
354 concentrations of both amlodipine (Figures 5A and B) and lisinopril (Figure 5C)
355 formulations, indicating their suitability for calibration model development.

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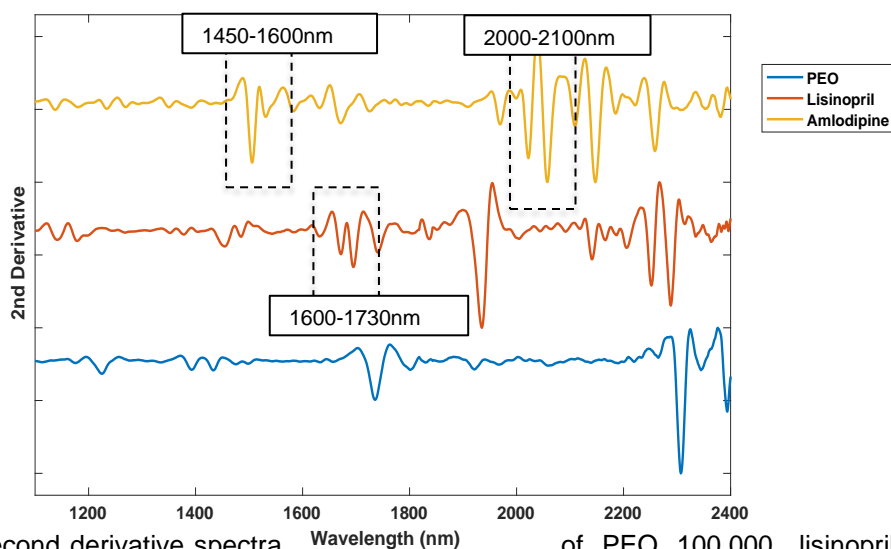
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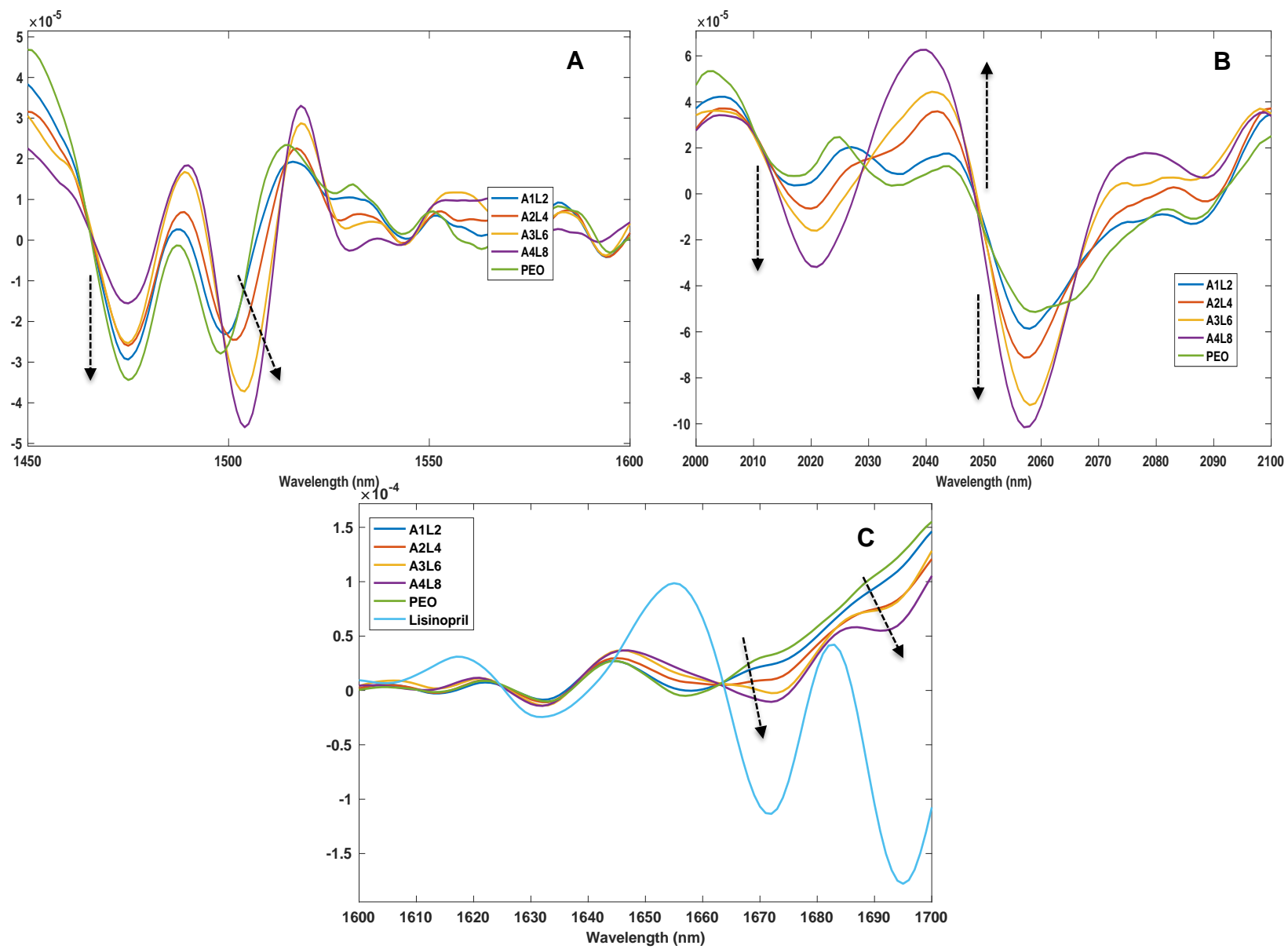
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Figure 4. Second derivative spectra of PEO 100,000, lisinopril and amlodipine. Spectral features of interest were 1450-1600 nm and 2000-2100 nm for amlodipine, and 1600-1730 nm for lisinopril.

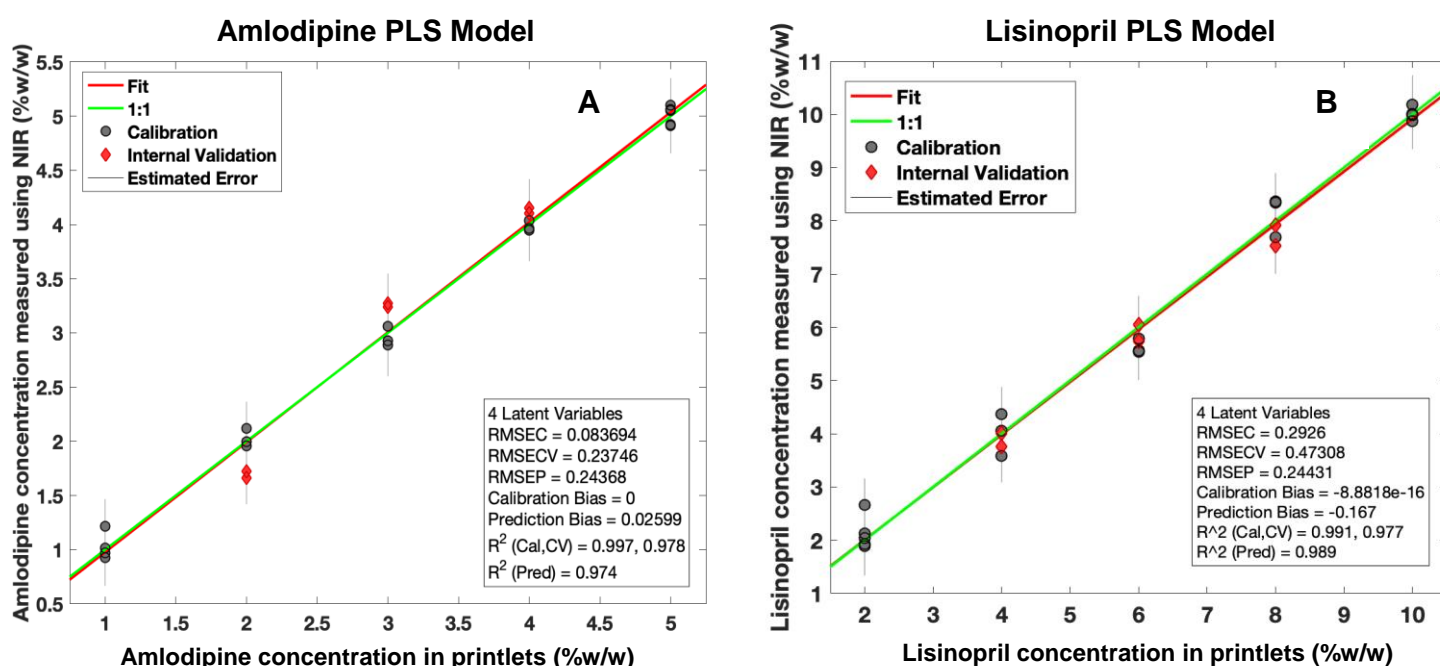
370 Data pre-treatment is essential to eliminate or minimise variability unrelated to
371 the property of interest and to minimise physical effects prior to multivariate
372 calibration, to ensure the development of an effective model (Huang et al.,
373 2010). Pre-treatment improves the accuracy of quantification by enhancing
374 spectral information and reducing baseline drift (Chalus et al., 2005). Evaluation
375 of a variety of pre-processing methods was performed to create a reliable
376 multivariate calibration model (data not shown). In this study, for amlodipine,
377 the model selected has 4 latent variables (LVs), covers between 1450-1600 nm
378 and 2000-2100 nm wavelength range with a second derivative (Savitzky and
379 Golay method: filter width of 21 with a second polynomial (Savitzky and Golay,
380 1964)), followed by multiplicative scatter correction (MSC) and mean centering
381 pre-processing techniques.



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Figure 5. Change in second derivative NIR absorbances with the concentration of amlodipine in A) 1450-1600 nm, B) 2000-2100 nm and lisinopril in C) 1600-1700 nm.

406 For lisinopril, the model selected had 4 LVs, and covers a wavelength range
 407 between 1600-1730 nm with a second derivative (Savitzky and Golay method:
 408 filter width of 15 with a second polynomial (Savitzky and Golay, 1964)), followed
 409 by standard normal variant (SNV) and mean centering pre-processing
 410 techniques. These models were selected due to having a high linearity ($R^2 =$
 411 0.997 for amlodipine; 0.991 for lisinopril) and high accuracy (RMSEP = 0.24%
 412 for amlodipine; 0.24% for lisinopril) (Figures 6A and B). These values confirmed
 413 that the NIR test results were proportional to the amlodipine and lisinopril
 414 concentrations in the stated range. There are several parameters including
 415 model linearity, specificity and accuracy that are recommended by ICH (ICH,
 416 1994a), EMA (EMA, 2014a) and FDA (FDA, 2015b) guidelines that the
 417 developed models need to satisfy to be validated.
 418

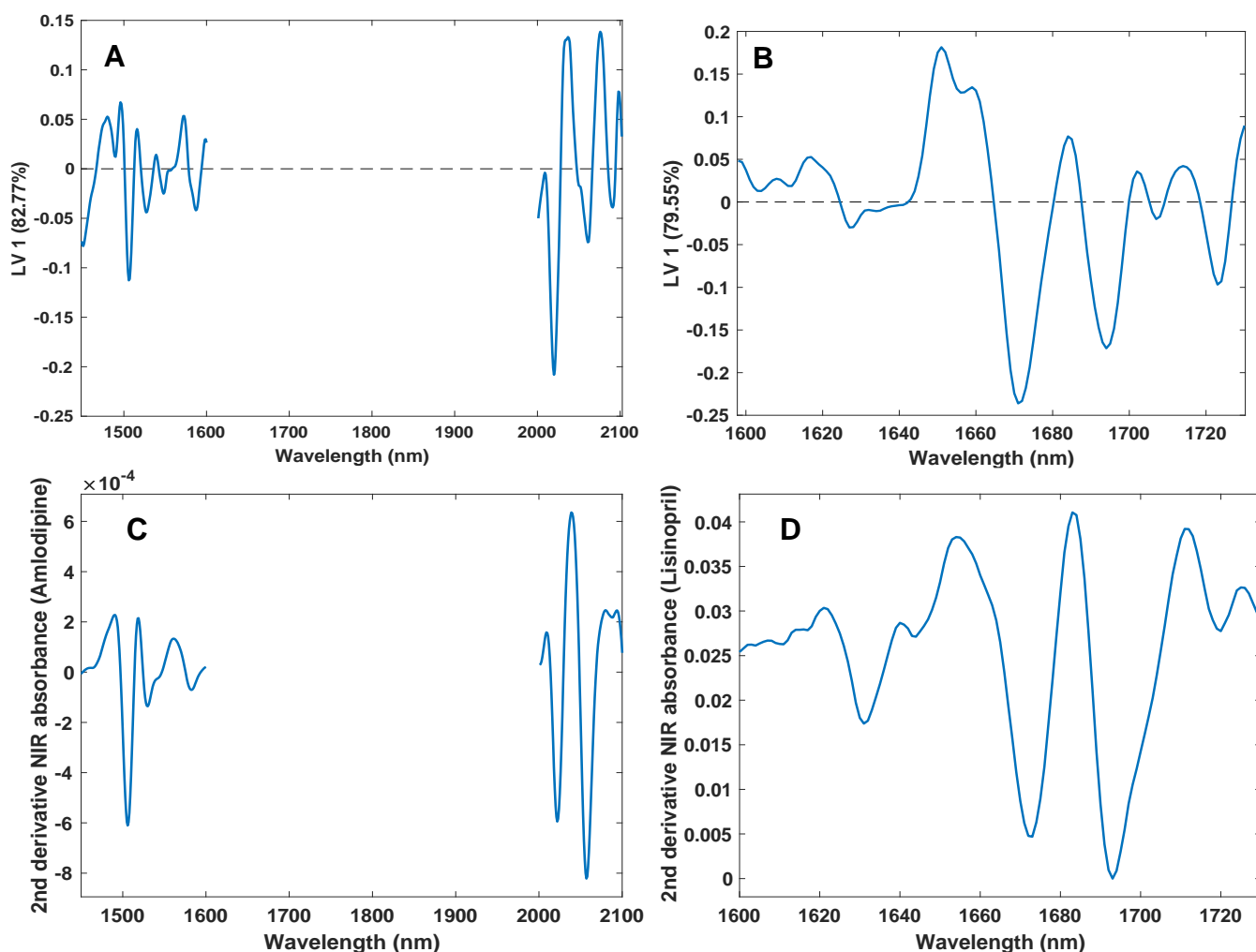


431 Figure 6. PLS calibration models of NIR predicted drug content (%w/w) of oral film printlets for
 432 A) Amlodipine and B) Lisinopril. Grey points are calibration (19 points from 5 concentrations);
 433 Red points are internal validation (6 points from 3 concentrations).

434
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440 3.2.1 Specificity

441 Specificity is defined as the ability to identify definitely the analyte (amlodipine
442 and lisinopril) from the other excipients (Candurin Gold Sheen and PEO
443 100,000) (Patel et al., 2012). This was evaluated by comparing the loadings
444 spectra of the 1st latent variable (LV 1) to the pure material reference spectra,
445 which accounted for variation of 82.77% and 79.55% for amlodipine and
446 lisinopril, respectively (Figure 7). The LV1 spectrum was found to model well-
447 known amlodipine spectral features at 1450-1600 nm and 2000-2100 nm
448 (demonstrated in Figures 7A and C) and well-known lisinopril spectral features
449 at 1600-1730 nm (highlighted in Figures 7B and D). The LV1 spectra of both
450 amlodipine and lisinopril models were not found to be modelling common
451 spectral features of PEO 100,0 00 (Figure 4).



452

453 Figure 7. NIR absorbance spectra of A) Amlodipine PLS model LV1 loading spectra, B)

454 Lisinopril PLS model LV1 loading spectra, C) Amlodipine pure and D) Lisinopril pure

455 **3.2.2 Accuracy**

456 The accuracy of a calibration model can be defined as the closeness in the
 457 agreement between the actual and the predicted NIR values (ICH, 1994a). As
 458 such, oral film printlets from 3 concentration levels (A2L4, A3L6 and A4L8) were
 459 scanned and the model was evaluated for prediction accuracy. An excellent
 460 predictive performance was observed with both drugs as the RMSEP for
 461 amlodipine was 0.24% and RMSEP was 0.24% for lisinopril (Figures 6A and B).
 462 Table 3 shows the difference between the HPLC and NIR predicted amlodipine
 463 and lisinopril concentrations. Paired t-test results showed that there were no
 464 significant differences between HPLC and NIR predictions as $p > 0.05$ across
 465 all three concentrations. This confirmed that NIR is a suitable quantification
 466 method for standard printlets. NIR prediction showed a higher SD compared to
 467 HPLC, which may be due to the minute differences in the surface effects of the
 468 printlets (Trenfield et al., 2018b). Generally, the model maintained a good
 469 predictive performance due to a majority of the data variation being attributed
 470 to the changes in the drugs' concentration (LV1 for amlodipine = 82.77%; LV1
 471 for lisinopril = 79.55%).

472

473

474 Table 3. Results of dose predicted from oral film printlets test set using the NIR model vs the
 475 reference HPLC method

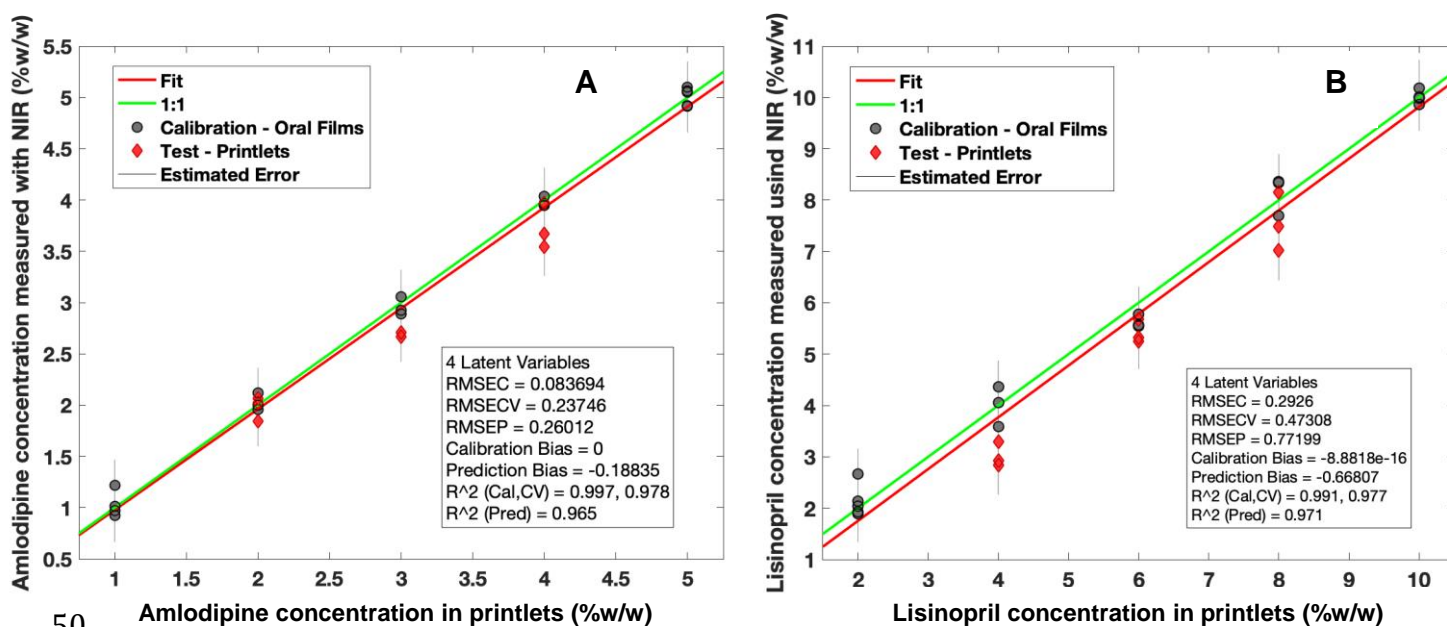
Formulation	Test Validation					
	Amlodipine (% w/w)			Lisinopril (% w/w)		
	HPLC	NIR	P value	HPLC	NIR	P value
A2L4	2.08±0.001	1.99±0.09	0.23	2.90±0.0001	3.03±0.27	0.54
A3L6	2.97±0.002	2.62±0.12	0.10	5.99±0.0001	5.72±0.25	0.26
A4L8	4.13±0.000	3.73±0.31	0.15	7.90 ±0.006	7.46±0.46	0.41

476

477 One of the main benefits of 3D printing for personalised medicine is the ability
 478 to tailor solid dosage form shape and size, depending on the patient preference
 479 or therapeutic needs (Trenfield et al., 2018a). Several studies have already
 480 shown that changing printlet geometry can alter the dose and drug release
 481 characteristics (Goyanes et al., 2015b; Martinez et al., 2018; Sadia et al.,

482 2018a). It is also well known that NIR absorbance can be affected by surface
 483 effects (Jamrógiewicz, 2012; Saeed et al., 2009) and, as such, it was important
 484 to evaluate the performance of the developed PLS model when scanning
 485 printlets of different geometries.

486 To determine this, cylindrical shaped printlets (amlodipine 2-4% w/w and
 487 lisinopril 4-8% w/w) with the same formulation compositions as the oral films
 488 were 3D printed and scanned using NIR at six different points on the dosage
 489 form. Figures 8A and 8B show that the model predicted the concentrations well,
 490 as the points fitted on the calibration curve. For amlodipine prediction, the
 491 change in printlet geometry from a thin film to cylindrical tablet caused a slightly
 492 higher error compared to oral films (RMSEP values of 0.26% and 0.24%,
 493 respectively). A similar occurrence observed with the prediction of lisinopril,
 494 with RMSEP values of 0.77% and 0.24% for cylindrical tablets and oral films,
 495 respectively. This phenomenon is likely due to the complex rounded surface
 496 structure of this shape compared with the flat films. However, overall the model
 497 continued to be fit-for-purpose for use with differently shaped tablets of the
 498 same composition.



509 Figure 8. Application of developed PLS models of NIR predicted drug content of A) amlodipine
 510 and B) lisinopril. Grey points are calibration (based on oral films); red points are a test set of
 511 cylindrical printlets.

512 Currently, PLS regression is widely used for a full quantitative characterisation
513 as it gives the highly accurate predictions (Ravn et al., 2008; Roggo et al., 2007).
514 The non-destructive, at-line QC method demonstrated here clearly shows a lot
515 of advantages as it is highly user-friendly and provides rapid dose prediction
516 with the scanning time for each tablet is only roughly 10 seconds. Since the
517 FDA and EMA guidance also recognise the use of PLS regression as a
518 quantitative tool, the developed model in the present study is suitable to be
519 used for quality control purposes in the clinic (EMA, 2014a; FDA, 2015b). The
520 validation of the developed PLS models have proven the feasibility of the use
521 of NIR spectroscopy to replace conventional destructive dose verification
522 methods (such as HPLC and UV spectroscopy).

523

524

525 **4. Conclusion**

526 For the first time, we report the at-line dose verification of two separate drugs
527 (amlodipine and lisinopril) within 3D printed antihypertensive polyprintlets.
528 Calibration models were developed across therapeutically relevant dosages of
529 two drugs (amlodipine: 1-5% w/w, and lisinopril 2-10% w/w) and were
530 applicable to polyprintlets of different geometries (oral films and cylindrical
531 tablets). The developed models demonstrated excellent linearity (R^2 pred =
532 0.997, 0.991), accuracy (RMSEP = 0.24%, 0.24%) and specificity (LV1 =
533 82.77%, 79.55%) for both amlodipine and lisinopril respectively, and were
534 validated according to current international standards. This manuscript
535 provides a novel method for the dual quantification of two drugs, facilitating the
536 integration of 3D printing into clinical practice.

537

538

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546 **6. References**

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