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journal homepage: www.elsevier.com/locate/jqsrtMARVEL analysis of the measured high-resolution rovibrational spectra of H₂³²S

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ABSTRACT

44 325 measured and assigned transitions of H₂³²S, the parent isotopologue of the hydrogen sulfide molecule, are collated from 33 publications into a single database and reviewed critically. Based on this information, rotation-vibration energy levels are determined for the ground electronic state using the Measured Active Rotational-Vibrational Energy Levels (MARVEL) technique. The *ortho* and *para* principal components of the measured spectroscopic network of H₂³²S are considered separately. The verified set of 25 293 *ortho*- and 18 778 *para*- H₂³²S transitions determine 3969 *ortho* and 3467 *para* energy levels. The MARVEL results are compared with alternative data compilations, including a theoretical variational list.

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1. Introduction

Hydrogen sulfide, H₂S, is a volcanic gas present on earth and other geologically active planets and moons, such as Io [1], Venus [2], and, theoretically, hot super-earth “lava” exoplanets [3]. It has recently been detected above the clouds in the atmosphere of Uranus [4]. On earth, microbial respiration of seawater sulfate creates a hydrogen sulfide-rich ecosystem [5], suggesting that H₂S is a component for a potential microbial life sustaining atmosphere on Venus [2], and a potential biomarker for life on exoplanets [6]. H₂S influences many physiological processes [7], it is of importance in the treatment of respiratory diseases [8], and it is used as a measure of the quality of air near oil refineries [9]. The parent isotopologue, H₂³²S, was first detected in interstellar space in 1972 [10]. Many of the scientific and engineering applications mentioned re-

quire the detailed knowledge of the rovibrational energy levels of H₂³²S.

As part of this work, we present the largest compilation of published experimental rovibrational transition data for H₂³²S. The experimental database of H₂³²S transitions has been formatted and analysed using the MARVEL (Measured Active Rotational-Vibrational Energy Levels) spectroscopic network (SN) software [11–13]. This study builds on the data of the 2012W@DIS information system for the hydrogen sulfide molecule [14].

As to the structure of this paper, the next section provides the theory underlying the present study. Section 3 presents and discusses the experimental sources used, with results given in Section 4. Section 5 discusses these results and compares the empirical rovibrational energies presented in this work with corresponding levels previously determined both by experiment and theory. Finally, Section 6 provides our conclusions. All transition data and energy levels resulting from this work are included as supplementary data files.

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2. Theory

2.1. MARVEL

The MARVEL procedure [11–13] is based on the theory of spectroscopic networks [15,16], with the energy levels represented as nodes and the transitions between them as edges. The related MARVEL code can be used to critically evaluate and validate experimentally-determined transition wavenumbers and uncertainties collected from the literature, inverting the wavenumber information to obtain accurate empirical energy levels with an associated uncertainty. MARVEL has been successfully used to evaluate the energy levels for molecules such as $^{12}\text{C}_2$ [17], $^{48}\text{Tl}^{16}\text{O}$ [18], water vapour [19–23], H_3^+ [24], H_2D^+ and D_2H^+ [25], $^{12}\text{C}_2\text{H}_2$ [26], $^{14}\text{NH}_3$ [27,28], and $^{12}\text{C}_2\text{H}_2^{16}\text{O}$ [29]. MARVEL requires each measured transition to have an associated uncertainty and for each energy level considered to possess a unique set of quantum numbers.

2.2. Quantum number labelling

The seven quantum numbers that were used for labelling the upper and lower rovibrational states of H_2^{32}S are the same as those used in a previous MARVEL investigation to classify water [19]. These quantum numbers are summarised in Table 1. Normal-mode labelling is used for the vibrations, where ν_1 , ν_2 and ν_3 stand for the symmetric stretch, bend, and antisymmetric stretch vibrations, respectively. Standard asymmetric-top quantum numbers are used for the rotations, where J , K_a , and K_c are the three quantum numbers associated with rotational angular momentum, \mathbf{J} , and the two projections along the A and C axes. We also provide, as part of the label, the nuclear spin state (*ortho* or *para*), which is deduced by whether $(\nu_3 + K_a + K_c)$ is odd (*ortho*) or even (*para*) [26,30]. Hyperfine coupling associated with nuclear spins has been neglected.

When local-mode notation was used by an experimental source considered in this work, the quantum numbers were transformed to the normal-mode notation $(\nu_1\nu_2\nu_3)$ to describe a vibrational state. Note that some of the data sources considered in this study use J and $\tau = K_a - K_c$ instead of J , K_a , and K_c [31], where τ runs from $-J, -J+1, \dots, J-1, +J$, with energy increasing from $\tau = -J$. The $+/-$ parity of an asymmetric top molecule such as H_2^{32}S is defined by $(-1)^{K_c}$ [19]. H_2^{32}S belongs to the $\text{C}_{2v}(\text{M})$ molecular symmetry (MS) group [32], which contains irreducible representations A_1 , A_2 , B_1 , and B_2 , as given in Table 2.

2.3. Selection rules

The rigorous selection rules governing rotation-vibration transitions for a molecule of the $\text{C}_{2v}(\text{M})$ MS group are given by:

$$\Delta J = 0, \pm 1, \quad (1)$$

$$J' + J'' \neq 0, \quad (2)$$

$$-\leftrightarrow + \quad (3)$$

Table 1

Quantum numbers used to label the upper and lower energy states of H_2^{32}S .

Label	Description
ν_1	S-H symmetric stretch ($\sim 2614.4 \text{ cm}^{-1}$)
ν_2	Symmetric bending mode ($\sim 1182.6 \text{ cm}^{-1}$)
ν_3	S-H antisymmetric stretch ($\sim 2628.5 \text{ cm}^{-1}$)
J	Rotational angular momentum
K_a, K_c	Projections of rotational angular momentum
<i>ortho/para</i>	Nuclear spin state (see text)

Table 2

Symmetry of the rovibrational states of H_2^{32}S .

Symmetry	A_1	A_2	B_1	B_2
Parity	+	-	-	+
Nuclear spin state	<i>para</i>	<i>para</i>	<i>ortho</i>	<i>ortho</i>

The *ortho* states of H_2^{32}S have the nuclear spin statistical weight $g_{\text{ns}} = 3$, while for the *para* states $g_{\text{ns}} = 1$, thus, *ortho* transitions have three times the intensity of *para* transitions. This is sometimes referred to as intensity alternation. It is assumed that *ortho* and *para* states do not interconvert. Such transitions are very weakly allowed [33] but have yet to be observed for H_2^{32}S .

3. Experimental sources

A large number of experimentally determined transition wavenumbers can be found in the literature for the main isotopologue of hydrogen sulfide, H_2^{32}S . We have attempted to conduct a rigorous and comprehensive search for all useable spectroscopic data. Fortunately, much of the data up to 2012 was previously collated as part of the 2012 W@DIS information system for hydrogen sulfide [14], in which some of the authors of this paper were involved. These data were converted to MARVEL format for this work and analysed alongside data from newly collected sources. This requires the transition wavenumber (in cm^{-1}) and the associated uncertainty, along with quantum number assignments for both the upper and lower energy states, and a unique reference label for each transition. This reference indicates the data source the transition originates from. The data source tag is based on the notation employed by an IUPAC Task Group on water spectroscopy [19]. An extract of the input file in the required format is given in Table 3; the full file can be found in the supplementary data for this publication.

33 sources of experimental data were used in the final data set. The data from more recent papers are generally provided in digital format, but some of the older papers had to be processed through digitalisation software, or even manually entered in the worst cases. After digitalisation the data were converted to MARVEL format, as described above.

Table 4 gives a summary of all the data sources used in this work, along with the energy range, number of transitions (as-

Table 3

Extract from the MARVEL input file for the *ortho* transitions for H_2^{32}S . The full file is supplied as part of the supplementary information to this paper. All energy term values and uncertainties are in units of cm^{-1} . The meaning of the upper and lower state assignments can be found in Table 1.

Transition	Uncertainty	Upper state assignment	Lower state assignment	Reference
33.12631	0.00015	0 0 0 5 2 3 <i>ortho</i>	0 0 0 5 1 4 <i>ortho</i>	94YaKL_1
33.12631	0.00018	0 0 0 3 0 3 <i>ortho</i>	0 0 0 2 1 2 <i>ortho</i>	94YaKL_2
34.00529	0.00010	0 0 0 8 7 2 <i>ortho</i>	0 0 0 8 6 3 <i>ortho</i>	94YaKL_5
34.15779	0.00010	0 0 0 11 9 2 <i>ortho</i>	0 0 0 11 8 3 <i>ortho</i>	94YaKL_7
34.21980	0.00010	0 0 0 6 4 3 <i>ortho</i>	0 0 0 6 3 4 <i>ortho</i>	94YaKL_8
34.24062	0.00012	0 0 0 4 2 3 <i>ortho</i>	0 0 0 4 1 4 <i>ortho</i>	94YaKL_9
35.73512	0.00010	0 0 0 8 6 3 <i>ortho</i>	0 0 0 8 5 4 <i>ortho</i>	94YaKL_13

Table 4

Data sources used in this study with frequency range, numbers of transitions (A/V for assigned/verified), and comments, which are detailed in Section 3.1. As far as we are aware all experiments were conducted at room temperature.

Tag	Ref.	Range (cm ⁻¹)	A/V	Comments
72HeCoLu	[34]	1.17–25.55	37/35	(4a)
95BeYaWiPo	[35]	4.39–85.41	112/84	(4b)
68CuKeGa	[36]	5.63–14.15	6/6	
71Huiszoon	[37]	5.63–7.23	2/2	
14CaPu	[38]	7.23–53.17	70/70	
85BuFeMeSh	[39]	10.02–20.9	6/6	
94YaKl	[40]	33.13–259.76	366/366	
13CaPu	[41]	33.97–37.45	4/4	
13AzYuTeMa	[42]	45.25–359.79	1158/1139	(4c)
83FlCaJo	[43]	50.77–307.51	426/387	(4d)
18UIBeGr	[44]	729.78–1735.41	2267/2267	
82LaEdGiBo	[45]	1003.46–1495.28	397/396	(4e)
83Strow	[46]	1082.03–1257.07	123/123	
96UIMaKoAl	[47]	1178.05–1359.78	41/41	
98BrCrCrNa	[48]	2141.30–4249.85	7473/7473	
18Horneman	[49]	2180.35–4220.46	4460/4460	
84LeFlCaJo	[50]	2180.36–2945.81	2113/2111	(4f)
81GiEd	[51]	2192.48–2823.11	715/704	(4g)
96UIOnKoAl	[52]	3614.70–3887.66	106/106	
05ULiBeGr	[53]	4000.59–6653.79	2347/2347	
97BrCrCrNa	[54]	4500.88–5595.01	5221/5219	(4h)
18Liu	[55]	4514.79–5555.58	3337/3335	(4i)
04BrNaPoSi_c	[56]	5688.27–6676.71	3178/3178	
04BrNaPoSi_a	[57]	7169.19–7898.97	2878/2876	(4j)
04ULiBeGr_b	[58]	7226.81–7994.09	1855/1855	
04ULiBeGr_a	[59]	8405.91–8905.28	589/589	
04BrNaPoSi_b	[60]	8412.73–8906.11	1179/1175	(4k)
03DiNaHuZh	[61]	9541.01–10000.71	1736/1728	(4l)
01NaCa_a	[62]	10787.33–11297.99	1105/1097	(4m)
94GrRaStDe	[63]	11948.91–12246.28	227/145	(4n)
97VaBiCaFl	[64]	12324.55–12670.68	399/387	(4o)
99CaFl	[65]	13060.51–13357.14	219/206	(4p)
01NaCa_b	[66]	16186.25–16436.57	173/154	(4q)
Total		1.17–16436.57	44 325/44 071	

signed (A) and verified (V)), and comments, which are detailed in Section 3.1. Table 5 lists those data sources which were considered but not used, with comments on the reasons for their exclusion from the analysis. The reference tag given in these tables matches those used in the unique labels in the MARVEL input files, given in the supplementary data and illustrated in Table 3.

As transitions have never been observed between *ortho* and *para* states of H₂³²S, they form two separate principal components (PCs) of the experimental spectroscopic network. All input and out-

put files supplied in the supplementary data to this work are split into either *ortho* or *para*.

3.1. Comments on the experimental sources of Table 4

(4a) 72HeCoLu [34] contains 2 lines which have been cited as taken from other experimental sources, for which the original data are already in our dataset. These duplicates were removed.

(4b) 95BeYaWiPo [35] contains 28 lines which have been cited as from other experimental sources, for which the original data are already in our dataset. These duplicates were removed.

(4c) 13AzYuTeMa [42] gives data for rotational lines recorded at room temperature. The equipment used provides spectra with very high sensitivity and this enabled transitions between highly rotationally excited states to be recorded, even at room temperature. These high rotational excitations may not be so accurately assigned as those of lower energy, as there are no data from other sources in the same region to confirm these measurements. The supplementary data from the original paper does not contain the original experimental transitions; this was provided by the corresponding author. There were 19 lines which could not be validated against other data and so were removed from our dataset (see Section 3.2).

(4d) 83FlCaJo [43] contains 39 lines which have been cited as from other sources which are already present in our dataset. The duplicates were removed.

(4e) 82LaEdGiBo [45] contains one line which could not be validated and so has been removed from our dataset.

(4f) 84LeFlCaJo [50] contains 2 lines which could not be validated against other more recent data and were removed from our dataset. All data from this source were recalibrated; the calibration factor employed was 0.999 999 746 below 2450 cm⁻¹ and 1.000 000 0714 above 2450 cm⁻¹.

(4g) 81GiEd [51] contains 11 lines which could not be validated against more recent data and thus were removed from our dataset.

(4h) 97BrCrCrNa [54] contains 2 lines with no corresponding energy in the AYT2 linelist (see Section 5) and so were removed from our dataset.

(4i) 18Liu [55] contains 2 lines with no corresponding energy in the AYT2 linelist and so were removed from our dataset.

(4j) 04BrNaPoSi_a [57] contains 2 lines with no corresponding energy in the AYT2 linelist and so were removed from our dataset.

(4k) 04BrNaPoSi_b [60] contains 4 lines which could not be validated and were removed from our dataset.

(4l) 03DiNaHuZh [61] contains one blended line which has been removed from our dataset. A further 7 lines were found to include

Table 5

Data sources considered but not used in this work.

Tag	Ref.	Comments
02CoRoTy	[67]	Data taken from other sources
94WaKuSu	[68]	Data taken from other sources
96SuMeKr	[69]	Data taken from other sources
97SuMeKr	[70]	Data taken from other sources
97Sumpf	[71]	Data taken from other sources
98PiPoCoDe	[72]	Data from http://spec.jpl.nasa.gov . Data does not appear to be experimental.
73HeDeKi	[73]	Data taken from other sources, [34,36,37]
02KiSuKrTi	[74]	Data appears to be taken from the 1996 edition of HITRAN [75]
06Polovtseva	N/A	Thesis, no citation. Referenced in [14]
87LeFlCaAr	[76]	Energy levels
13Azzam	[77]	All results are given in 13AzYuTeMa [42]
94KoJe	[78]	Energy levels
95FlGrRa	[79]	Energy levels
85LaEdGiBo	[80]	Calculated values, compared to experimental values from 83FlCaJo [43]
94ByNaSmSi	[81]	Energy Levels
01TyTaSc	[82]	Vibrational energy Levels
69MiLeHa	[83]	Could not be validated with the other sources to a reasonable accuracy (see Section 3.2)
56AIPI	[31]	Could not be validated with the other sources to a reasonable accuracy (see Section 3.2)
69SnEd	[84]	Could not be validated with the other sources to a reasonable accuracy (see Section 3.2)

a level with no corresponding energy in the AYT2 linelist and so were removed from our dataset.

(4m) 01NaCa_a [62] contains 8 transitions which were found to have no corresponding levels in the AYT2 linelist and so were removed from our dataset.

(4n) 94GrRaStDe [63] tentatively assigned the vibrational band in their data as $(\nu_1, \nu_2, \nu_3) = (2, 2, 2)$; however, in 95FlGrRaSt [79], a paper published later with the same authors, the revised assignment is $(\nu_1, \nu_2, \nu_3) = (3, 0, 2)$. We adopt the latter vibrational label in our dataset. No other data sources probe this vibrational band. The frequency is only given to 3 decimal places, so the uncertainty was altered to match. The original dataset has 82 transitions either labelled with an asterisk to indicate lines from the H_2^{34}S isotopologue, or labelled with a dagger to indicate lines which have been perturbed due to those from H_2^{34}S [63]. We have commented these out by adding a minus sign to the wavenumber and added the label *_pt* to the end.

(4o) 97VaBiCaFl [64] contains 4 lines which could not be validated and so were removed from our dataset. A further 8 lines were found to contain a level with no corresponding energy in the AYT2 linelist and so were removed from our dataset.

(4p) The assignment for 99CaFl_185_na_ct from 99CaFl [65] leads to different spin states (*ortho/para*) for the upper and lower state, which is forbidden. We have removed this transition from our dataset. 5 other lines could not be validated and so were removed from our dataset. 7 lines were found to have no variational counterparts in the AYT2 linelist and so were removed.

(4q) 01NaCa_b [66] contains one line which could not be validated and so was removed from our dataset. A further 18 lines were found to contain levels with no corresponding energies in the AYT2 linelist and so were removed from our dataset.

3.2. General comments

All transitions which were considered but not processed in the final dataset have a minus sign in front of the transition wavenumber (indicating that MARVEL will ignore them) in the input files provided as supplementary material and are labelled with a comment at the end of the reference (see the comments of Section 3.1). We used a cut-off of 0.035 cm^{-1} as the largest acceptable uncertainty and removed any transitions with an uncertainty greater than this, with a note in Section 3.1 to indicate how many transitions from a particular data source could not be validated to within this accuracy.

The publications 97BrCrCrNa [54], 04BrNaPoSi_c [56], 04BrNaPoSi_a [57], and 04BrNaPoSi_b [60] include just a short description of the assignment and modelling of the Fourier Transform (FT) spectra between 4500 and 8900 cm^{-1} . Detailed information on the theoretical treatment of these spectra will be published separately in due course. Some of these transitions, between 4400 – 8000 cm^{-1} , were previously reported in the HITRAN-2012 [85] and HITRAN-2016 [86] databases, while the second decade region, 8400 – 8900 cm^{-1} , is presented for the first time. It is also worth mentioning that the MARVEL dataset includes a considerable number of transitions which include new, unpublished energy levels. As only the upper energy levels of the transitions were reported in 05UuLiBeGr [53], 04UuLiBeGr_b [58], and 04UuLiBeGr_a [59], the corresponding transition wavenumbers were recovered from Liu (18Liu [55]) using the energy levels obtained from 97BrCrCrNa [54], 04BrNaPoSi_c [56], 04BrNaPoSi_a [57], and 04BrNaPoSi_b [60]. For this reason, the transitions in the MARVEL dataset which are referred to as from 05UuLiBeGr [53], 04UuLiBeGr_b [58], and 04UuLiBeGr_a [59] contain a larger number of assignments than reported in the original publications. The data of the first hexad region between 4500 – 5600 cm^{-1} were totally assigned based on the energy levels reported in 97BrCrCrNa [54]. The highly

accurate (an accuracy of 0.0005 cm^{-1}) set of H_2^{32}S transitions recorded using a FT spectrometer between 2200 – 4250 cm^{-1} was provided by Horneman (18Horneman [49]) and assigned using the energy levels reported in 98BrCrCrNa [48].

3.3. Sources from the HITRAN database

H_2^{32}S has been included in the HITRAN database [85–90] since 1991 [87]. The following are sources for the line positions and energy levels of the H_2^{32}S data in the HITRAN database, up to the 2016 release: 96UuMaKoAl [47], 83FlCaJo [43], 98BrCrCrNa [48], 13AzYuTeMa [42], 94YaKl [40], 95BeYaWiPo [35], 05UuLiBeGr [53], 04UuLiBeGr [58], 03DiNaHuZh [61], 01NaCa_b [66], 84LeFlCaJo [50], 82LaEdGiBo [45], 85LaEdGiBo [80], 94ByNaSmSi [81]. The variational ExoMol AYT2 linelist [91] was used to assign transitions in the ν_2 excited vibrational state in the 2012 release [85].

4. Results

The MARVEL website (<http://www.kkrk.chem.elte.hu/marvelonline>) has a version of MARVEL which can be run online. The variable NQN (number of quantum numbers) is 7 in the case of hydrogen sulfide, as illustrated in Table 3 which shows an extract of the *ortho* input file to MARVEL.

MARVEL automatically assigns the lowest energy state in a particular component of the spectroscopic network to 0. The ground rovibrational state is included in the *para* set of energy levels, however there needs to be a “magic number”, corresponding to the energy of the lowest *ortho* state, which is added to all the MARVEL *ortho*-symmetry energies. Here, this was taken as the ground vibrational state $(\nu_1, \nu_2, \nu_3) = (0, 0, 0)$ with the lowest rotational energy (see Section 2.2), $J = 1, K_a = 0, K_c = 1$, of 94KoJe [78], who determined the value of 13.74631 cm^{-1} . The output for the *ortho* energies in the supplementary data, and the extract in Table 6, all have this magic number added for the *ortho* principal spectroscopic network. There are a small number (4 *para*) of energy levels which are not joined to either of the principal components. If more experimental transitions became available in the future it would be possible to link these to the *para* principal network.

We collated and considered a total of 44 325 transitions from 33 experimental sources (25 474 *ortho* and 18 851 *para*). Of those 254 were found to be inconsistent (could not be validated to within 0.035 cm^{-1}) with others and so removed from the final data set, leaving a total of 44 071 transitions used as input into MARVEL (25 293 *ortho* and 18 778 *para*).

Fig. 1 gives a visual representation of NumTrans, the number of transitions which link the states of the MARVEL *ortho*- H_2^{32}S network, against corresponding upper state energies. Those upper states which are dark blue in colour, linked by only 1 transition, should be considered less reliable than those in red, which are supported by hundreds of different transitions, up to 404. The values of NumTrans for each level are given in the energy level files in the supplementary data.

5. Comparison to other derived energy levels

We used a number of sources to compare energy level values against those determined in this work. 94KoJe [78] contains a list of energy levels, which are taken or calculated from other sources [31,43,50,76,80,84]; see Table 7 for a breakdown by vibrational band. AYT2 [91] is a theoretical variational linelist for H_2^{32}S calculated as part of the ExoMol project [92,93], a database of theoretical linelists for molecules of astrophysical importance. AYT2 is appropriate up to temperatures of around 2000 K and designed for use in characterising the atmospheres of cool stars and exoplanets. The states file, available from <https://www.exomol.com>,

Table 6

Extract from the MARVEL output file for the *ortho* energy levels of H_2^{32}S . The full file is supplied as part of the supplementary information to this paper. All energies and uncertainties are in units of cm^{-1} . NumTrans gives the number of transitions which are linked to that particular energy level. Assignments are as given by Table 1.

Assignment	Energy	Uncertainty	NumTrans	Sym
0 0 0 1 0 1 ortho	13.746310	0.000001	177	B1
0 0 0 1 1 0 ortho	19.375630	0.000001	153	B2
0 0 0 2 1 2 ortho	38.297765	0.000002	271	B2
0 0 0 2 2 1 ortho	55.161605	0.000002	265	B1
0 0 0 3 0 3 ortho	71.424233	0.000002	282	B1
0 0 0 3 1 2 ortho	95.056325	0.000001	344	B2
0 0 0 3 2 1 ortho	107.368228	0.000001	331	B1
0 0 0 4 1 4 ortho	114.177613	0.000001	282	B2
0 0 0 3 3 0 ortho	117.392015	0.000001	265	B2
0 0 0 4 2 3 ortho	148.418340	0.000001	379	B1

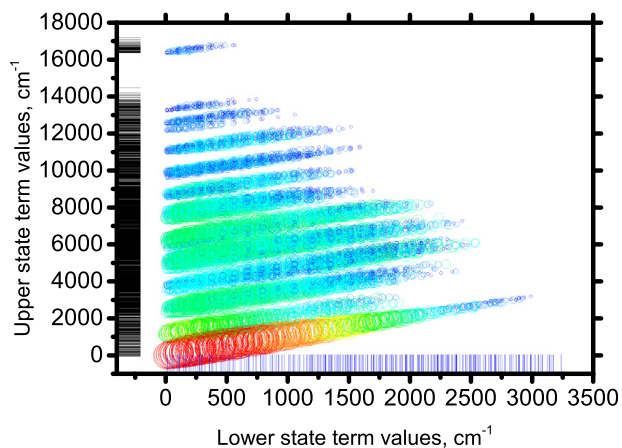


Fig. 1. The lower-state energies of the experimental *ortho*- H_2^{32}S transitions used in this work, against corresponding upper-state energies. The vertical bars along the horizontal-axis show the lower-state energies, while the horizontal bars along the vertical-axis give the upper-state energies. Each circle represents a particular transition, with the size proportional to the log of NumTrans, the number of transitions supporting the upper state. This value ranges from 1 (dark blue) to 404 (red). (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

gives the calculated energy levels, which can be compared against our empirical energy levels. For more information on the format of the ExoMol data files, please refer to [93]. 95FIGrRaSt [79] contains rovibrational energy levels from the $(\nu_1, \nu_2, \nu_3) = (2, 0, 3)$ and $(3, 0, 2)$ vibrational bands. The authors outline an experimental procedure used to deduce these levels but they do not provide the experimental transition data so we could not include the data from this source in our dataset. 87LeFICaAr [76] has a table of rovibrational energy levels for the $(\nu_1, \nu_2, \nu_3) = (2, 1, 0)$, $(1, 1, 1)$ and $(0, 1, 2)$ vibrational bands. 96UIMaKoAl [47] contains rovibrational energy levels from the $(\nu_1, \nu_2, \nu_3) = (0, 1, 0)$ vibrational band.

We compared the pure rotational levels determined in this work both against those given in 94KoJe [78] and the calculated values from the AYT2 linelist [91], up to $J = 5$. These data are given in Table 8. The pure rotational levels are closer to those from the experimentally determined sources of 94KoJe than to those of variationally calculated AYT2, as would be expected.

The pure vibrational levels from this work are also compared with those from 94KoJe [78] and AYT2 [91], see Table 9. It should be noted here that there are some differences in labelling between these data sources and those used in this work. For example, the pure vibrational levels of 94KoJe [78] which are labelled $(\nu_1, \nu_2, \nu_3) = (2, 0, 2)$, $(1, 0, 2)$ and $(3, 0, 0)$, match the normal mode labelling of $(4, 0, 0)$, $(3, 0, 0)$ and $(1, 0, 2)$, respectively,

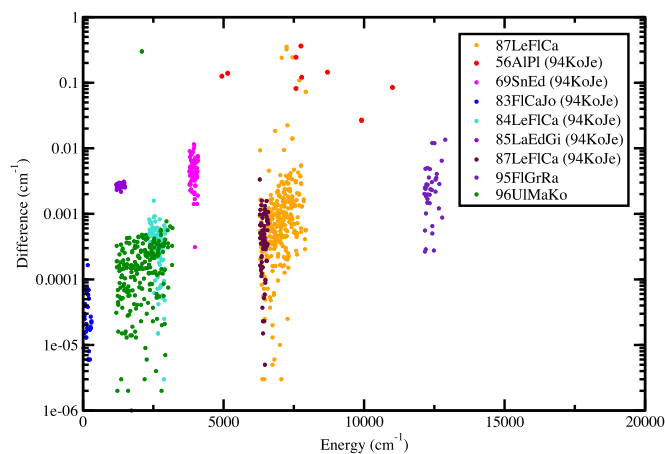


Fig. 2. Deviations in cm^{-1} between energy levels from this work and: 94KoJe [78] (see Table 7), 95FIGrRaSt [79], 87LeFICaAr [76], 96UIMaKoAl [47]. Note the logarithmic vertical-axis.

according to the labelling of 01TyTaSc [82] and the experimental sources used in this work.

We compared all the rovibrational energy levels given in 94KoJe [78], 95FIGrRaSt [79], 96UIMaKoAl [47], and 87LeFICaAr [76] against our set of MARVEL energy levels, taking the aforementioned labelling differences into account for 94KoJe [78] (see Table 7). These comparisons are illustrated in Fig. 2. Most levels are in good agreement, with only a few levels with differences between 0.1–0.4 cm^{-1} . These levels with the largest differences are from 87LeFICaAr [76] and 56AIPi [31]. All the levels which are compared to those of 87LeFICaAr [76] have several transitions which link them to other energy levels in our MARVEL dataset, and originate from different sources, which increases their reliability. This indicates that the lack of accuracy for these particular energies originates from 87LeFICaAr [76] and not the MARVEL dataset. The transition data from 56AIPi [31] was considered for use in the current study, but the data could not be validated with that from newer sources (see Table 5), and so the energy levels are also assumed to be unreliable.

As mentioned above, AYT2 [91] is a theoretical variational linelist for H_2^{32}S calculated as part of the ExoMol project [92,93]. Highly accurate experimental energy levels provide essential input for testing and improving theoretically calculated line positions such as those within AYT2. The ExoMol data format convention is to have a ‘states’ and a ‘trans’ file. The states file gives the calculated rovibrational energy levels, which can be compared against our empirical energy levels, and the trans file contains the Einstein- A coefficients for transitions between these states [93].

Table 7
Original sources of energy level data from 94KoJe [78].

Tag	Ref.	Vibrational bands	Comments
83FICaJo	[43]	(0, 0, 0)	Experimental source used in this work
85LaEdGiBo	[80]	(0, 1, 0)	Calculated values, compared to experimental values from 83FICaJo [43]
84LeFICaJo	[50]	(0, 2, 0) (1, 0, 0) (0, 0, 1)	Experimental source used in this work
69SnEd	[84]	(1, 1, 0) (0, 1, 1)	Experimental source used in this work
56AIPi	[31]	(0, 2, 1) (2, 0, 0) (1, 0, 1) (3, 0, 0) (2, 0, 1) (1, 0, 2) (0, 0, 3) (2, 1, 1) (3, 0, 1) (2, 0, 2) (1, 0, 3) (3, 1, 1)	Data source not considered reliable. All $J = 0$ levels only.
87LeFICaAr	[76]	(2, 1, 0) (1, 1, 1)	94KoJe [78] only contains levels from this source up to $J = 5$. See text and Fig. 2 for a full comparison.

Table 8

Comparison of pure rotational levels from this work with those of 94KoJe [78] (see Table 7) and AYT2 [91] up to $J = 5$. NumTrans gives the number of transitions linking a particular state. All energies and uncertainties are in cm^{-1} . The uncertainty refers to this work.

$J K_a K_c$	State	Sym	NumTrans	Uncertainty	This work	94KoJe	Difference	AYT2	Difference
0 0 0	<i>para</i>	A_1	58	0.000098	0	0	0	0	0
1 0 1	<i>ortho</i>	B_1	170	0.000001	13.746310	13.74631	0.00000	13.746340	0.000030
1 1 1	<i>para</i>	A_2	157	0.000002	15.090119	15.09011	-0.00001	15.090151	0.000032
1 1 0	<i>ortho</i>	B_2	147	0.000001	19.375630	19.37563	0.00000	19.375563	-0.000067
2 0 2	<i>para</i>	A_1	223	0.000001	38.016095	38.01607	-0.00003	38.016264	0.000169
2 1 2	<i>ortho</i>	B_2	258	0.000002	38.297765	38.29775	-0.00001	38.297961	0.000196
2 1 1	<i>para</i>	A_2	211	0.000001	51.140188	51.14016	-0.00003	51.140060	-0.000128
2 2 1	<i>ortho</i>	B_1	253	0.000002	55.161605	55.16158	-0.00003	55.161460	-0.000145
2 2 0	<i>para</i>	A_1	181	0.000001	58.368870	58.36884	-0.00003	58.368621	-0.000249
3 0 3	<i>ortho</i>	B_1	265	0.000002	71.424221	71.42426	0.00004	71.424760	0.000539
3 1 3	<i>para</i>	A_2	213	0.000001	71.465192	71.46515	-0.00004	71.465653	0.000461
3 1 2	<i>ortho</i>	B_2	329	0.000001	95.056313	95.05630	-0.00001	95.056267	-0.000046
3 2 2	<i>para</i>	A_1	272	0.000001	96.392529	96.39246	-0.00007	96.392428	-0.000101
3 2 1	<i>ortho</i>	B_1	313	0.000001	107.368216	107.36820	-0.00002	107.367927	-0.000289
4 0 4	<i>para</i>	A_1	205	0.000001	114.172246	114.17217	-0.00008	114.173085	0.000839
4 1 4	<i>ortho</i>	B_2	270	0.000001	114.177601	114.17758	-0.00002	114.178495	0.000894
3 3 1	<i>para</i>	A_2	250	0.000002	115.340656	115.34059	-0.00007	115.340038	-0.000618
3 3 0	<i>ortho</i>	B_2	247	0.000001	117.392003	117.39199	-0.00001	117.391407	-0.000596
4 1 3	<i>para</i>	A_2	306	0.000001	148.140653	148.14058	-0.00007	148.140740	0.000087
4 2 3	<i>ortho</i>	B_1	362	0.000001	148.418328	148.41831	-0.00002	148.418470	0.000142
5 0 5	<i>ortho</i>	B_1	264	0.000045	166.343488	166.34345	-0.00004	166.344887	0.001399
5 1 5	<i>para</i>	A_2	196	0.000077	166.344005	166.34417	0.00016	166.345606	0.001601
4 2 2	<i>para</i>	A_1	308	0.000001	170.335799	170.33574	-0.00006	170.335471	-0.000328
4 3 2	<i>ortho</i>	B_2	383	0.000002	173.967266	173.96726	-0.00001	173.966794	-0.000472
4 3 1	<i>para</i>	A_2	261	0.000001	182.648548	182.64849	-0.00006	182.647978	-0.000570
4 4 1	<i>ortho</i>	B_1	290	0.000002	195.661414	195.66142	0.00001	195.659965	-0.001449
4 4 0	<i>para</i>	A_1	219	0.000003	196.802189	196.80212	-0.00007	196.800713	-0.001476
5 1 4	<i>ortho</i>	B_2	356	0.000037	210.217262	210.21727	0.00001	210.217731	0.000469
5 2 4	<i>para</i>	A_1	270	0.000001	210.264821	210.26477	-0.00005	210.265229	0.000408
5 2 3	<i>ortho</i>	B_1	387	0.000001	243.343434	243.34344	0.00001	243.343211	-0.000223
5 3 3	<i>para</i>	A_2	333	0.000001	244.392517	244.39250	-0.00002	244.392149	-0.000368
5 3 2	<i>ortho</i>	B_2	363	0.000001	263.738931	263.73895	0.00002	263.738560	-0.000371
5 4 2	<i>para</i>	A_1	298	0.000001	271.106067	271.10604	-0.00003	271.104809	-0.001258
5 4 1	<i>ortho</i>	B_1	316	0.000002	277.337562	277.33758	0.00002	277.336633	-0.000929
5 5 1	<i>para</i>	A_2	229	0.000002	296.104442	296.10442	-0.00002	296.101282	-0.003160
5 5 0	<i>ortho</i>	B_2	278	0.000001	296.677577	296.67760	0.00003	296.674560	-0.003017

This format allows calculated linelists to be retrospectively updated using more reliable experimental energy levels in order to improve their accuracy, see [94] for an example.

To compare our energy levels with those from the AYT2 [91] linelist, a comparable states file was made; we labelled our states with the same A_1 , A_2 , B_1 , B_2 symmetry labels which are given in the AYT2 states file, see Table 2. We then matched all states with this same symmetry, ν_2 , $\nu_1 + \nu_3$, and J , and searched for the closest value within these given parameters. Fig. 3 gives the result of this comparison.

Work is underway to update the existing ExoMol AYT2 states file and linelist based on the energy levels obtained in this work, using only those levels based on the transitions we are most sure about. Care should be taken in general when using MARVEL energy levels with a low value of NumTrans (see Fig. 1). This, along with the uncertainty, gives an indication of the reliability of a particular energy level.

6. Conclusions

A total of 44 325 measured experimental rovibrational transitions of H_2^{32}S from 33 publications have been considered in this work. From this set, 3969 *ortho*- and 3467 *para*- H_2^{32}S energy levels have been determined using the Measured Active Rotational-Vibrational Energy Levels (MARVEL) technique. These results have been carefully compared with alternative compilations of energy levels.

A variational high-temperature linelist for H_2^{32}S has been computed as part of the ExoMol project, AYT2 [91]. Our new MARVEL energy levels will be used to improve the accuracy of this theoretical linelist.

A significant part of this work was performed by pupils from Highams Park School in London, as part of a project known as ORBYTS (Original Research By Young Twinkle Students) [95]. The MARVEL studies of $^{12}\text{C}_2\text{H}_2$ [26] and $^{48}\text{Ti}^{16}\text{O}$ [18] were also per-

Table 9

Comparison of pure vibrational levels ($J = 0$) from this work with those of 94KoJe [78] (see Table 7) and AYT2 [91]. NT stands for NumTrans, the number of transitions linking a particular state. All energies and uncertainties are in cm^{-1} . The uncertainty refers to this work. State is from this work and Sym is the corresponding symmetry label as used in AYT2 [91] (see Section 2.2).

ν_1	ν_2	ν_3	State	Sym	NT	Uncertainty	This work	94KoJe	Difference	AYT2	Difference
0	0	0	para	A ₁	58	0.000098	0			0	0
0	1	0	para	A ₁	5	0.000477	1182.576991	1182.5742	-0.0028	1182.569618	-0.007373
0	2	0	para	A ₁	4	0.000629	2353.964679	2353.9655	0.0008	2353.907317	-0.057362
1	0	0	para	A ₁	3	0.000663	2614.407743	2614.4074	-0.0003	2614.394829	-0.012914
0	0	1	ortho	B ₂	2	0.000669	2628.454821	2628.4552	0.0004	2628.463320	0.008499
0	3	0	para	A ₁	3	0.000621	3513.789974			3513.705072	-0.084902
1	1	0	para	A ₁	3	0.000390	3779.166566	3779.1710	0.0044	3779.189348	0.022782
0	1	1	ortho	B ₂	2	0.000625	3789.269211	3789.2720	0.0028	3789.269878	0.000667
0	4	0	para	A ₁	1	0.001000	4661.672219			4661.605794	-0.066425
1	2	0	para	A ₁	2	0.000707	4932.699369			4932.688937	-0.010432
0	2	1	ortho	B ₂	2	0.000707	4939.104010	4939.2300	0.1260	4939.129851	0.025841
2	0	0	para	A ₁	2	0.000707	5144.986319	5145.1200	0.1337	5145.031868	0.045549
1	0	1	ortho	B ₂	2	0.000707	5147.220560	5147.3600	0.1394	5147.166622	-0.053938
0	0	2	para	A ₁	2	0.000707	5243.101919			5243.158956	0.057037
1	3	0	para	A ₁	2	0.000928	6074.581067			6074.566059	-0.015008
0	3	1	ortho	B ₂	2	0.000707	6077.594560			6077.626636	0.032076
2	1	0	para	A ₁	2	0.000707	6288.146119	6288.1428	-0.0033	6288.134723	-0.011396
1	1	1	ortho	B ₂	3	0.000637	6289.172875	6289.1739	0.0010	6289.128284	-0.044591
1	2	1	ortho	B ₂	4	0.000632	7420.092707			7420.077786	-0.014921
1	0	2	para	A ₁	1	0.002000	7576.381719	7576.3000	-0.0817	7576.413281	0.031562
2	0	1	ortho	B ₂	2	0.001414	7576.544710	7576.3000	-0.2447	7576.596211	0.051501
3	0	0	para	A ₁	2	0.001414	7752.263219	7751.9000	-0.3632	7752.343205	0.079986
0	0	3	ortho	B ₂	2	0.001414	7779.321260	7779.2000	-0.1213	7779.352004	0.030744
1	3	1	ortho	B ₂	1	0.002000	8539.561310			8539.565999	0.004689
1	1	2	para	A ₁	2	0.001414	8697.141469			8697.133905	-0.007564
2	1	1	ortho	B ₂	2	0.001481	8697.154984	8697.3000	0.1450	8697.179341	0.024357
1	4	1	ortho	B ₂	1	0.005000	9647.167310			9647.098855	-0.068455
2	2	1	ortho	B ₂	1	0.005000	9806.667310			9806.712978	0.045668
1	2	2	para	A ₁	1	0.005000	9806.733119			9806.748170	0.015051
2	0	2	para	B ₂	1	0.005000	9911.023119	9911.0500	0.0269	9911.102285	0.079166
3	0	1	ortho	A ₁	1	0.005000	9911.023310	9911.0500	0.0267	9911.112478	0.089168
3	1	1	ortho	A ₁	1	0.005000	11008.695310	11008.7800	0.0847	11008.774494	0.079184
3	0	2	para	A ₁	1	0.001000	12149.460119			12149.552318	0.092199
1	0	4	para	A ₁	1	0.015000	12524.637119			12524.834491	0.197372
4	0	1	ortho	B ₂	1	0.015000	12525.214310			12525.346292	0.131982
3	1	2	para	B ₂	1	0.015000	13222.762119			13222.790827	0.028708
2	1	3	ortho	A ₁	1	0.015000	13222.772310			13222.798559	0.026249

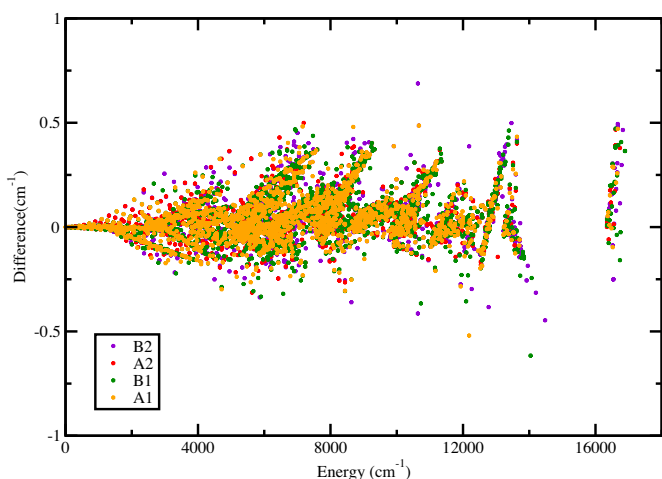


Fig. 3. Deviations, in cm^{-1} , between the MARVEL energy levels from this work and the variationally calculated linelist AYT2 [91]. Different colors represent different rovibrational symmetries (see Table 2 in Section 2.2).

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Supplementary material

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.jqsrt.2018.07.012.

References

- [1] Salama F, Allamandola L, Witteborn F, Cruikshank D, Sandford S, Bregman J. The 2.5–5.0 μm spectra of Io: Evidence for H_2S and H_2O frozen in SO_2 . *Icarus* 1990;83:66–82. doi:10.1016/0019-1035(90)90006-U.
- [2] Schulze-Makuch D, Grinspoon DH, Abbas O, Irwin LN, Bullock MA. A sulfur-based survival strategy for putative phototrophic life in the Venusian atmosphere. *Astrobiology* 2004;4:11–18. doi:10.1089/153110704773600203.

formed as part of the ORBYTS project and further studies on other key molecules will be published in due course, involving high-school students from the UK, Australia, and Hungary. A paper discussing our experiences of performing original research in collaboration with school children can be found elsewhere [96].

- [3] Tennyson J, Yurchenko SN. Laboratory spectra of hot molecules: data needs for hot super-earth exoplanets. *Mol Astrophys* 2017;8:1–18. doi:10.1016/j.molap.2017.05.002.
- [4] Irwin PGJ, Toledo D, Garland R, Teanby NA, Fletcher LN, Orton GA, Bezard B. Detection of hydrogen sulfide above the clouds in Uranus's atmosphere. *Nat Astron* 2018;2:420–7. doi:10.1038/s41550-018-0432-1.
- [5] Pace NR. A molecular view of microbial diversity and the biosphere. *Science* 1997;276(5313):734–40. doi:10.1126/science.276.5313.734.
- [6] Kaltenecker L, Sasselov D. Detecting planetary geochemical cycles on exoplanets: Atmospheric signatures and the case of SO₂. *Astrophys J* 2010;708:1162. doi:10.1088/0004-637X/708/2/1162.
- [7] King SB. Potential biological chemistry of hydrogen sulfide (H₂S) with the nitrogen oxides. *Free Radical Biol Med* 2013;55(Supplement C):1–7. doi:10.1016/j.freeradbiomed.2012.11.005.
- [8] Bazhanov N, Ansar M, Ivancic T, Garofalo RP, Casola A. Hydrogen sulfide: a novel player in airway development, pathophysiology of respiratory diseases, and antiviral defenses. *Am J Respiratory Cell Molec Bio* 2017;57:403–10. doi:10.1165/rcmb.2017-0114TR.
- [9] Ciaffoni L, Cummings BL, Denzer W, Peverall R, Procter SR, Ritchie GAD. Line strength and collisional broadening studies of hydrogen sulphide in the 1.58 μm region using diode laser spectroscopy. *Appl Phys B* 2008;92:627. doi:10.1007/s00340-008-3119-y.
- [10] Thaddeus P, Kutner ML, Penzias AA, Wilson RW, Jefferts KB. Interstellar hydrogen sulfide. *Astron Astrophys* 1972;176:73. doi:10.1086/181023.
- [11] Furtenbacher T, Császár AG, Tennyson J. MARVEL: measured active rotational-vibrational energy levels. *J Mol Spectrosc* 2007;245:115–25. doi:10.1016/j.jms.2007.07.005.
- [12] Furtenbacher T, Császár AG. The role of intensities in determining characteristics of spectroscopic networks. *J Molec Struct (THEOCHEM)* 2012;1009:123–9. doi:10.1016/j.molstruc.2011.10.057.
- [13] Furtenbacher T, Császár AG. MARVEL: measured active rotational-vibrational energy levels. II. algorithmic improvements. *J Quant Spectrosc Radiat Transf* 2012;113:929–35. doi:10.1016/j.jqsrt.2012.01.005.
- [14] Polovtseva ER, Lavrentiev NA, Voronina SS, Naumenko OV, Fazliev AZ. Information system for molecular spectroscopy. 5. Ro-vibrational transitions and energy levels of the hydrogen sulfide molecule. *Atmos Oceanic Optics* 2012;25:157–65. doi:10.1134/S1024856012020133.
- [15] Császár AG, Furtenbacher T. Spectroscopic networks. *J Mol Spectrosc* 2011;266:99–103. doi:10.1016/j.jms.2011.03.031.
- [16] Árendás P, Furtenbacher T, Császár AG. On spectra of spectra. *J Math Chem* 2016;54:806–22. doi:10.1007/s10910-016-0591-1.
- [17] Furtenbacher T, Szabó I, Császár AG, Bernath PF, Yurchenko SN, Tennyson J. Experimental energy levels and partition function of the ¹²C molecule. *Astrophys J Suppl* 2016;224:44. doi:10.3847/0067-0049/224/2/44.
- [18] McKemmish LK, Masseron T, Sheppard S, Sandeman E, Schofield Z, Furtenbacher T, Császár AG, Tennyson J, Sousa-Silva C. MARVEL Analysis of the measured high-resolution spectra of ⁴⁸Ti¹⁶O. *Astrophys J Suppl* 2017;228:15. doi:10.3847/1538-4365/228/2/15.
- [19] Tennyson J, Bernath PF, Brown LR, Campargue A, Carleer MR, Császár AG, Gamache RR, Hodges JT, Jenouvrier A, Naumenko OV, Polyansky OL, Rothman LS, Toth RA, Vandaele AC, Zobov NF, Daumont L, Fazliev AZ, Furtenbacher T, Gordon IE, Mikhailenko SN, Shirin SV. IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part I. Energy levels and transition wavenumbers for H₂¹⁷O and H₂¹⁸O. *J Quant Spectrosc Radiat Transf* 2009;110:573–96. doi:10.1016/j.jqsrt.2009.02.014.
- [20] Tennyson J, Bernath PF, Brown LR, Campargue A, Carleer MR, Császár AG, Daumont L, Gamache RR, Hodges JT, Naumenko OV, Polyansky OL, Rothman LS, Toth RA, Vandaele AC, Zobov NF, Furtenbacher T, Gordon IE, Mikhailenko SN, Voronin BA. IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part II. Energy levels and transition wavenumbers for HD¹⁶O, HD¹⁷O, and HD¹⁸O. *J Quant Spectrosc Radiat Transf* 2010;111:2160–84. doi:10.1016/j.jqsrt.2010.06.012.
- [21] Tennyson J, Bernath PF, Brown LR, Campargue A, Carleer MR, Császár AG, Daumont L, Gamache RR, Hodges JT, Naumenko OV, Polyansky OL, Rothman LS, Vandaele AC, Zobov NF, Al Derzi AR, Fábri C, Fazliev AZ, Furtenbacher T, Gordon IE, Lodi L, Mizus II. IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part III. Energy levels and transition wavenumbers for H₂¹⁶O. *J Quant Spectrosc Radiat Transf* 2013;117:29–80. doi:10.1016/j.jqsrt.2012.10.002.
- [22] Tennyson J, Bernath PF, Brown LR, Campargue A, Császár AG, Daumont L, Gamache RR, Hodges JT, Naumenko OV, Polyansky OL, Rothman LS, Vandaele AC, Zobov NF, Dénes N, Fazliev AZ, Furtenbacher T, Gordon IE, Hu SM, Szidarovszky T, Vasilenko IA. IUPAC critical evaluation of the rotational-vibrational spectra of water vapor. Part IV. Energy levels and transition wavenumbers for D₂¹⁶O, D₂¹⁷O and D₂¹⁸O. *J Quant Spectrosc Radiat Transf* 2014;142:93–108. doi:10.1016/j.jqsrt.2014.03.019.
- [23] Furtenbacher T, Tennyson J, Naumenko O.V., Polyansky O.L., Zobov N.F., Császár A.G.. The 2018 update of the IUPAC database of water energy levels. *J Quant Spectrosc Radiat Transf* (In preparation).
- [24] Furtenbacher T, Szidarovszky T, Mátyus E, Fábri C, Császár AG. Analysis of the rotational-vibrational states of the molecular ion H₃⁺. *J Chem Theory Comput* 2013;9:5471–8. doi:10.1021/ct4004355.
- [25] Furtenbacher T, Szidarovszky T, Fábri C, Császár AG. MARVEL analysis of the rotational-vibrational states of the molecular ions H₂D⁺ and D₂H⁺. *Phys Chem Chem Phys* 2013;15:10181–93. doi:10.1039/c3cp44610g.
- [26] Chubb KL, Joseph M, Franklin J, Choudhury N, Furtenbacher T, Császár AG, Gaspar G, Oguoko P, Kelly A, Yurchenko SN, Tennyson J, Sousa-Silva C. MARVEL analysis of the measured high-resolution spectra of C₂H₂. *J Quant Spectrosc Radiat Transf* 2018;204:42–55. doi:10.1016/j.jqsrt.2017.08.018.
- [27] Al Derzi AR, Furtenbacher T, Yurchenko SN, Tennyson J, Császár AG. MARVEL analysis of the measured high-resolution spectra of ¹⁴NH₃. *J Quant Spectrosc Radiat Transf* 2015;161:117–30. doi:10.1016/j.jqsrt.2015.03.034.
- [28] Furtenbacher T, Coles P.A., Tennyson J., Császár A.G.. Updated MARVEL energy levels for ammonia. *J Quant Spectrosc Radiat Transf*; To be submitted.
- [29] Fábri C, Mátyus E, Furtenbacher T, Nemes L, Mihály B, Zoltáni T, Császár AG. Variational quantum mechanical and active database approaches to the rotational-vibrational spectroscopy of ketene, H₂CCO. *J Chem Phys* 2011;135:094307.
- [30] Herman M, Lievin J. Acetylene- from intensity alternation in spectra to ortho and para molecule. *J Chem Educ* 1982;59:17. doi:10.1021/ed059p17.
- [31] Allen Jr HC, Plyler EK. Infrared spectrum of hydrogen sulfide. *J Chem Phys* 1956;25:1132–6. doi:10.1063/1.1743164.
- [32] Bunker PR, Jensen P. *Molecular Symmetry and Spectroscopy*. 2nd ed. Ottawa: NRC Research Press; 1998.
- [33] Miani A, Tennyson J. Can ortho-para transitions for water be observed? *J Chem Phys* 2004;120:2732–9.
- [34] Helminger P, Cook RL, De Lucia FC. Microwave spectrum and centrifugal distortion effects of H₂S. *J Chem Phys* 1972;56:4581–4. doi:10.1063/1.1677906.
- [35] Belov SP, Yamada KMT, Winniewisser G, Poteau L, Bocquet R, Demaison J, Polyansky O, Tretyakov MY. Terahertz rotational spectrum of H₂S. *J Mol Spectrosc* 1995;173:380–90. doi:10.1006/jmsp.1995.1242.
- [36] Cupp RE, Keilpf RA, Gallagher JJ. Hyperfine structure in the millimeter spectrum of hydrogen sulfide electric spectroscopy on asymmetric-top molecules. *Phys Rev* 1968;171:60–9. doi:10.1103/PhysRev.171.60.
- [37] Huiszoon C. A high resolution spectrometer for the shorter millimeter wavelength region. *Rev Sci Instrum* 1971;42:477–81. doi:10.1063/1.1685135.
- [38] Cazzoli G, Puzzarini C. The rotational spectrum of hydrogen sulfide: The H₂³²S and H₂³⁴S isotopologues revisited. *J Mol Spectrosc* 2014;298(Supplement C):31–7. doi:10.1016/j.jms.2014.02.002.
- [39] Burenin AV, Fevral'skikh TM, Melnikov AA, Shapin SM. Microwave spectrum of the hydrogen sulfide molecule H₂³²S in the ground state. *J Mol Spectrosc* 1985;109:1–7. doi:10.1016/0022-2852(85)90045-1.
- [40] Yamada KMT, Klee S. Pure rotational spectrum of H₂S in the far-infrared region measured by FTIR spectroscopy. *J Mol Spectrosc* 1994;166:395–405. doi:10.1006/jmsp.1994.1204.
- [41] Cazzoli G, Puzzarini C. Sub-doppler resolution in the THz frequency domain: 1 μkHz accuracy at 1 THz by exploiting the lamb-dip technique. *J Phys Chem A* 2013;117:13759–66. doi:10.1021/jp407980f.
- [42] Azzam AAA, Yurchenko SN, Tennyson J, Martin MA, Piralí O. Terahertz spectroscopy of hydrogen sulfide. *J Quant Spectrosc Radiat Transf* 2013;130:341–51. doi:10.1016/j.jqsrt.2013.05.035.
- [43] Flaud JM, Camy-Peyret C, Johns JWC. The far-infrared spectrum of hydrogen-sulfide - the (000) rotational-constants of H₂³²S, H₂³³S and H₂³⁴S. *Can J Phys* 1983;61:1462–73. doi:10.1139/p83-188.
- [44] Ulenikov ON, Bekhtereva ES, Gromova OV, Glushkov PA, Scherbakov AP, Horne-man VM, Sydow C, Maul C, Bauerecker S. Extended analysis of the high resolution FTIR spectra of H₂³²S (M = 32.33,34,36) in the region of the bending fundamental band: the ν₂ and bands: line positions, strengths, and pressure broadening widths. *J Quant Spectrosc Radiat Transf* 2018;216:76–98. doi:10.1016/j.jqsrt.2018.05.009.
- [45] Lane WC, Edwards TH, Gillis JR, Bonomo FS, Murcay FJ. Analysis of ν₂ of H₂S. *J Mol Spectrosc* 1982;95:365–80. doi:10.1016/0022-2852(82)90136-9.
- [46] Strow LL. Measurement and analysis of the ν₂ band of H₂S: comparison among several reduced forms of the rotational hamiltonian. *J Mol Spectrosc* 1983;97:9–28. doi:10.1016/0022-2852(83)90334-X.
- [47] Ulenikov ON, Malikova AB, Koivusaari M, Alanko S, Anttila R. High resolution vibrational rotational spectrum of H₂S in the region of the ν₂ fundamental band. *J Mol Spectrosc* 1996;176:229–35. doi:10.1006/jmsp.1996.0082.
- [48] Brown LR, Crisp JA, Crisp D, Naumenko OV, Smirnov MA, Sinita LN, Perrin A. The absorption spectrum of H₂S between 2150 and 4260 cm⁻¹: Analysis of the positions and intensities in the first (2ν₂, ν₁, and ν₃) and second (3ν₂, ν₁/ν₂, and ν₂/ν₃) triad regions. *J Mol Spectrosc* 1998;188:148–74. doi:10.1006/jmsp.1997.7501.
- [49] Horneman VM. Private communication.
- [50] Lechuga-Fossat L, Flaud JM, Camy-Peyret C, Johns JWC. The spectrum of natural hydrogen-sulfide between 2150 cm⁻¹ and 2950 cm⁻¹. *Can J Phys* 1984;62:1889–923. doi:10.1139/p84-233.
- [51] Gillis JR, Edwards TH. Analysis of 2ν₂, ν₁, and ν₃ of H₂S. *J Mol Spectrosc* 1981;85:55–73. doi:10.1016/0022-2852(81)90309-X.
- [52] Ulenikov ON, Onopenko GA, Koivusaari M, Alanko S, Anttila R. High resolution fourier transform spectrum of H₂S in the 3300–4080 cm⁻¹ region. *J Mol Spectrosc* 1996;176:236–50. doi:10.1006/jmsp.1996.0083.
- [53] Ulenikov ON, Liu AW, Bekhtereva ES, Gromova OV, Hao LY, Hu SM. High-resolution fourier transform spectrum of H₂S in the region of the second hexad. *J Mol Spectrosc* 2005;234:270–8. doi:10.1016/j.jms.2005.09.010.
- [54] Brown LR, Crisp JA, Crisp D, Naumenko OV, Smirnov MA, Sinita LN. The first hexad of interacting states of H₂S molecule. *SPE* 1997;3090:111–13. doi:10.1117/12.267745.
- [55] Liu AW. Private communication.
- [56] Brown LR, Naumenko OV, Polovtseva ER, Sinita LN. Hydrogen sulfide absorption spectrum in the 5700–6600 cm⁻¹ spectral region. *Proceedings 14th Sym-*

- posium on High-Resolution Molecular Spectroscopy 2004;5311:59–67. doi:10.1117/12.545192.
- [57] Brown LR, Naumenko OV, Polovtseva ER, Sinita LN. Absorption spectrum of H₂S between 7200 and 7890 cm⁻¹. Proc SPIE 2004;5396:5396–7. doi:10.1117/12.548211.
- [58] Ulenikov ON, Liu AW, Bekhtereva ES, Gromova OV, Hao LY, Hu SM. On the study of high-resolution rovibrational spectrum of H₂S in the region of 7300–7900 cm⁻¹. J Mol Spectrosc 2004;226:57–70. doi:10.1016/j.jms.2004.03.014.
- [59] Ulenikov ON, Liu AW, Bekhtereva ES, Grebneva SV, Deng WP, Gromova OV, Hu SM. High resolution fourier transform spectrum of H₂S in the region of 8500–8900 cm⁻¹. J Mol Spectrosc 2004;228:110–19. doi:10.1016/j.jms.2004.07.011.
- [60] Brown LR, Naumenko OV, Polovtseva ER, Sinita LN. Hydrogen sulfide absorption spectrum in the 8400–8900 cm⁻¹ spectral region. Eleventh International Symposium on Atmospheric and Ocean Optics/Atmospheric Physics 2004;5743:1–7. doi:10.1117/12.606253.
- [61] Ding Y, Naumenko O, Hu SM, Zhu Q, Bertseva E, Campargue A. The absorption spectrum of H₂S between 9540 and 10 000 cm⁻¹ by intracavity laser absorption spectroscopy with a vertical external cavity surface emitting laser. J Mol Spectrosc 2003;217:222–38. doi:10.1016/S0022-2852(02)00037-1.
- [62] Naumenko O, Campargue A. H₂S : First observation of the (70±, 0) local mode pair and updated global effective vibrational hamiltonian. J Mol Spectrosc 2001;210:224–32. doi:10.1006/jmsp.2001.8460.
- [63] Großkloß R, Rai SB, Stuber R, Demtroder W. Diode laser overtone spectroscopy of hydrogen sulfide. Chem Phys Lett 1994;229:609–15. doi:10.1016/0009-2614(94)01079-X.
- [64] Vaittinen O, Biennier L, Campargue A, Flaud JM, Halonen L. Local mode effects on the high-resolution overtone spectrum of H₂S around 12 500 cm⁻¹. J Mol Spectrosc 1997;184:288–99. doi:10.1006/jmsp.1997.7319.
- [65] Campargue A, Flaud JM. The overtone spectrum of H₂S near 13 200 cm⁻¹. J Mol Spectrosc 1999;194:43–51. doi:10.1006/jmsp.1998.7754.
- [66] Naumenko O, Campargue A. Local mode effects in the absorption spectrum of H₂S between 10 780 and 11 330 cm⁻¹. J Mol Spectrosc 2001;209:242–53. doi:10.1006/jmsp.2001.8417.
- [67] Cours T, Rosmus P, Tyuterev VG. Ab initio dipole moment functions of H₂S² and intensity anomalies in rovibrational spectra. J Chem Phys 2002;117:5192–208. doi:10.1063/1.1499487.
- [68] Waschull J, Kuhnemann F, Sumpf B. Self-, air, and helium broadening in the ν₂ band of H₂S. J Mol Spectrosc 1994;165:150–8. doi:10.1006/jmsp.1994.1117.
- [69] Sumpf B, Meusel I, Kronfeldt HD. Self- and air-broadening in the ν₁ and ν₃ bands of H₂S. J Mol Spectrosc 1996;177:143–5. doi:10.1006/jmsp.1996.0126.
- [70] Sumpf B, Meusel I, Kronfeldt HD. Noble gas broadening in fundamental bands of H₂S. J Mol Spectrosc 1997;184:51–5. doi:10.1006/jmsp.1997.7290.
- [71] Sumpf B. Experimental investigation of the self-broadening coefficients in the ν₁ / ν₃ band of SO₂ and the 2ν₂ band of H₂S. J Mol Spectrosc 1997;181:160–7. doi:10.1006/jmsp.1996.7168.
- [72] Pickett HM, Poynter RL, Cohen EA, Delitsky ML, Pearson JC, Muller HSP. Submillimeter, millimeter, and microwave spectral line catalog. J Quant Spectrosc Radiat Transf 1998;60:883–90. doi:10.1016/S0022-4073(98)00091-0.
- [73] Helminger P, De Lucia FC, Kirchhoff WH. Microwave spectra of molecules of astrophysical interest IV. Hydrogen sulfide. J Phys Chem Ref Data 1973;2:215–23. doi:10.1063/1.3253117.
- [74] Kissel A, Sumpf B, Kronfeldt HD, Tikhomirov BA, Ponomarev YN. Molecular-gas-pressure-induced line-shift and line-broadening in the ν₂-band of H₂S. J Mol Spectrosc 2002;216:345–54. doi:10.1006/jmsp.2002.8630.
- [75] Rothman LS, Rinsland CP, Goldman A, Massie ST, Edwards DP, Flaud JM, Perrin A, Camy-Peyret C, Dana V, Mandin JY, Schroeder J, McCann A, Gamache RR, Wattson R, Yoshino K, Chance KV, Jucks KW, Brown LR, Nemtchinov V, Varanasi P. The HITRAN molecular spectroscopic database and HAWKS (HITRAN atmospheric workstation): 1996 edition. J Quant Spectrosc Radiat Transf 1998;60:665–710. doi:10.1016/S0022-4073(98)00078-8.
- [76] Lechuga-Fossat L, Flaud JM, Camy-Peyret C, Arcas P, Cuisenier M. The H₂S spectrum in the 1.6 μm spectral region. Mol Phys 1987;61:23–32. doi:10.1080/00268978700100961.
- [77] Azzam AAA. A linelist for the hydrogen sulphide molecule. University College London; 2013. The thesis is available on-line at <http://discovery.ucl.ac.uk/1404058/>.
- [78] Kozin IN, Jensen P. Fourfold clusters of rotational energy levels for H₂S studied with a potential energy surface derived from experiment. J Mol Spectrosc 1994;163:483–509. doi:10.1006/jmsp.1994.1041.
- [79] Flaud JM, Grosskloss R, Rai SB, Struber R, Demtroder W, Tate DA, Wang Lg, Gallagher TF. Diode laser spectroscopy of H₂S² around 0.82 μm. J Mol Spectrosc 1995;172:275–81. doi:10.1006/jmsp.1995.1175.
- [80] Lane WC, Edwards TH, Gillis JR, Bonomo FS, Murcay FJ. Analysis of ν₂ of H₂S² and H₂S³. J Mol Spectrosc 1985;111:320–6. doi:10.1016/0022-2852(85)90008-6.
- [81] Bykov AD, Naumenko OV, Smirnov MA, Sinita LN, Brown LR, Crisp J, Crisp D. The infrared-spectrum of H₂S from 1 to 5 μm. Can J Phys 1994;72:989–1000. doi:10.1139/p94-130.
- [82] Tyuterev VG, Tashkun SA, Schwenke DW. An accurate isotopically invariant potential function of the hydrogen sulphide molecule. Chem Phys Lett 2001;348:223–34. doi:10.1016/S0009-2614(01)01093-4.
- [83] Miller RE, Leroy GE, Hard TM. Analysis of the pure rotational absorption spectra of hydrogen sulfide and deuterium sulfide. J Chem Phys 1969;50:677–84. doi:10.1063/1.1671116.
- [84] Snyder LE, Edwards TH. Simultaneous analysis of the (110) and (011) bands of hydrogen sulfide. J Mol Spectrosc 1969;31:347–61. doi:10.1016/0022-2852(69)90365-8.
- [85] Rothman LS, Gordon IE, Babikov Y, Barbe A, Benner DC, Bernath PF, Birk M, Bizzocchi L, Boudon V, Brown LR, Campargue A, Chance K, Cohen EA, Coudert LH, Devi VM, Drouin BJ, Fayt A, Flaud JM, Gamache RR, Harrison JJ, Hartmann JM, Hill C, Hodges JT, Jacquemart D, Jolly A, Lamouroux J, Roy RJL, Li G, Long DA, Lyulin OM, Mackie CJ, Massie ST, Mikhailenko S, Müller HSP, Naumenko OV, Nikitin AV, Orphal J, Perevalov V, Perrin A, Polovtseva ER, Richard C, Smith MAH, Starikova E, Sung K, Tashkun S, Tennyson J, Toon GC, Tyuterev VG, Wagner G. The HITRAN 2012 molecular spectroscopic database. J Quant Spectrosc Radiat Transf 2013;130:4–50. doi:10.1016/j.jqsrt.2013.07.002.
- [86] Gordon IE, Rothman LS, Hill C, Kochanov RV, Tan Y, Bernath PF, Birk M, Boudon V, Campargue A, Chance KV, Drouin BJ, Flaud JM, Gamache RR, Hodges JT, Jacquemart D, Perevalov VI, Perrin A, Shine KP, Smith MAH, Tennyson J, Toon GC, Tran H, Tyuterev VG, Barbe A, Császár AG, Devi VM, Furtenbacher T, Harrison JJ, Hartmann JM, Jolly A, Johnson TJ, Karman T, Kleiner I, Kyuberis AA, Loos J, Lyulin OM, Massie ST, Mikhailenko SN, Moazzen-Ahmadi N, Müller HSP, Naumenko OV, Nikitin AV, Polyansky OL, Rey M, Rotger M, Sharpe SW, Sung K, Starikova E, Tashkun SA, Vander Auwera J, Wagner G, Wilzewski J, Wcislo P, Yu S, Zak EJ. The HITRAN 2016 molecular spectroscopic database. J Quant Spectrosc Radiat Transf 2017;203:3–69. doi:10.1016/j.jqsrt.2017.06.038.
- [87] Rothman L, Gamache RR, Tipping RH, Rinsland CP, Smith MAH, Benner DC, Devi VM, Flaud JM, Camy-Peyret C, Perrin A, Goldman A, Massie ST, Brown LR, Toth RA. The HITRAN molecular database - editions of 1991 and 1992. J Quant Spectrosc Radiat Transf 1992;48:469–507. doi:10.1016/0022-4073(92)90115-K.
- [88] Rothman LS, Barbe A, Benner DC, Brown LR, Camy-Peyret C, Carleer MR, Chance K, Clerbaux C, Dana V, Devi VM, Flaud JM, Gamache RR, Goldman A, Jacquemart D, Jucks KW, Lafferty WJ, Mandin JY, Massie ST, Nemtchinov V, Newnham DA, Perrin A, Rinsland CP, Schroeder J, Smith KM, Smith MAH, Tang K, Toth RA, Auwera JV, Varanasi P, Yoshino K. The HITRAN molecular spectroscopic database: edition of 2000 including updates through 2001. J Quant Spectrosc Radiat Transf 2003;82:5–44. doi:10.1016/S0022-4073(03)00146-8.
- [89] Rothman LS, Jacquemart D, Barbe A, Benner DC, Birk M, Brown LR, Carleer MR, Chackerian C, Chance K, Coudert LH, Dana V, Devi VM, Flaud JM, Gamache RR, Goldman A, Hartmann JM, Jucks KW, Maki AG, Mandin JY, Massie ST, Orphal J, Perrin A, Rinsland CP, Smith MAH, Tennyson J, Tolchenov RN, Toth RA, Auwera JV, Varanasi P, Wagner G. The HITRAN 2004 molecular spectroscopic database. J Quant Spectrosc Radiat Transf 2005;96:139–204.
- [90] Rothman LS, Gordon IE, Barbe A, Benner DC, Bernath PF, Birk M, Boudon V, Brown LR, Campargue A, Champion JP, Chance K, Coudert LH, Dana V, Devi VM, Fally S, Flaud JM, Gamache RR, Goldman A, Jacquemart D, Kleiner I, Lacombe N, Lafferty WJ, Mandin JY, Massie ST, Mikhailenko SN, Miller CE, Moazzen-Ahmadi N, Naumenko OV, Nikitin AV, Orphal J, Perevalov VI, Perrin A, Predoi-Cross A, Rinsland CP, Rotger M, Simeckova M, Smith MAH, Sung K, Tashkun SA, Tennyson J, Toth RA, Vandaele AC, Auwera JV. The HITRAN 2008 molecular spectroscopic database. J Quant Spectrosc Radiat Transf 2009;110:533–72.
- [91] Azzam AAA, Yurchenko SN, Tennyson J, Naumenko OV. Exomol line lists XVI: A hot line list for H₂S. Mon Not R Astron Soc 2016;460:4063–74. doi:10.1093/mnras/stw1133.
- [92] Tennyson J, Yurchenko SN. Exomol: molecular line lists for exoplanet and other atmospheres. Mon Not R Astron Soc 2012;425:21–33. doi:10.1111/j.1365-2966.2012.21440.x.
- [93] Tennyson J, Yurchenko SN, Al-Refaie AF, Barton EJ, Chubb KL, Coles PA, Diamantopoulou S, Gorman MN, Hill C, Lam AZ, Lodi L, McKemmish LK, Na Y, Owens A, Polyansky OL, Rivlin T, Sousa-Silva C, Underwood DS, Yachmenev A, Zak E. The exomol database: molecular line lists for exoplanet and other hot atmospheres. J Mol Spectrosc 2016;327:73–94. doi:10.1016/j.jms.2016.05.002.
- [94] Barber RJ, Strange JK, Hill C, Polyansky OL, Mellau GC, Yurchenko SN, Tennyson J. Exomol line lists—III. An improved hot rotation-vibration line list for HCN and HNC. Mon Not R Astron Soc 2014;437:1828–35. doi:10.1093/mnras/stt2011.
- [95] McKemmish LK, Chubb KL, Rivlin T, Baker JS, Gorman MN, Heward A, Dunn W, Tessenyi M. Bringing pupils into the ORBYTS of research. Astron Geophys 2017;58(5):5.11. doi:10.1093/astroge/atx169.
- [96] Sousa-Silva C, McKemmish LK, Chubb KL, Baker J, Barton EJ, Gorman MN, Rivlin T, Tennyson J. Original research by young twinkle students (ORBYTS): when can students start performing original research? Phys Educ 2018;53:015020. doi:10.1088/1361-6552/aa8f2a.