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Ligand and structure based *in silico* studies to identify kinesin spindle protein (KSP) inhibitors as potential anticancer agents

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Abstract

Kinesin spindle protein (KSP) belongs to the kinesin superfamily of microtubulebased motor proteins. KSPis responsible for the establishment of the bipolar mitotic spindle which mediates cell division. Inhibition of KSP expedites the blockade of the normal cell cycle during mitosis through the generation of monoastralMTarrays that finally cause apoptotic cell death. As KSP is highly expressed in proliferating/cancer cells, it has gained considerable attention as a potential drug target for cancer chemotherapy. Therefore, this study envisaged to design novel KSP inhibitors by employing computational techniques/tools such as pharmacophore modeling, virtual database screening, molecular docking, and molecular dynamics. Initially, the pharmacophore models were generated from the dataset of highly potent KSP inhibitors and the pharmacophore models were validated against *in house*test set ligands. The validated pharmacophore model was then taken for database screening (Maybridgeand ChemBridge) to yield hits, which were further filtered for their drug-likeliness. The potential hits retrieved from virtual database screening were docked using CDOCKER to identify the ligandbinding landscape. The top-ranked hits obtained from molecular docking were progressed to molecular dynamics (AMBER) simulations to deduce the ligand binding affinity. This study identified MB-41570 and CB-10358 as potential hits and evaluated experimentally using *in vitro*KSP ATPase inhibition assays.

Key words: KSP inhibitors; pharmacophore modeling; CDOCKER; molecular dynamics;

KSP ATPase enzyme inhibition.

List of abbreviations

ADP	adenosine diphosphate
AMBER	assisted model building with energy refinement
ATP	adenosine triphosphate
CB	chembridge
CG	conjugate gradient
DS	discovery studio
GI	gastro intestinal
GAFF	generalized amber force field
KSP	kinesín spindle protein
MB	maybridge
MT	microtubule
MD	molecular dynamics
MM/GBSA	molecular mechanics/generalized Born and surface area continuum solvation
PDB	protein data bank
RoG	radius of gyration
RMSD	root mean square deviation
RMSF	root mean square fluctuations
SD	steepest descent

Introduction

Kinesins are microtubule (MT) based motor proteins that belong to the kinesin superfamily. In mitotic cells, they are involved in various key cellular events such as mitotic spindle assembly, re-modeling of MTs, the positioning of subcellular organelles, and chromosome segregation(Hirokawa, 1998).Kinesin spindle protein (also known as KSP, Eg5 or KIF11) is a plus-end-directed N-terminal MT motor and is responsible for the establishment of the bipolar organization of the mitotic spindle(Goodson, Kang, & Endow, 1994). The mitotic spindle segregates chromosome to process through mitosis. If a functional spindle fails to form, normal segregation of chromosomes does not take place. Consequently, checkpoint proteins inhibit cell division resulting in mitotic arrest (DeBonis et al., 2003) followed by cell death.KSP is highly expressed in proliferating human tissues and tumors of the colon, breast, uterus, ovary, and lungs(Compton, 2000). Therefore, KSP is a potential drug target in cancer chemotherapy. Inhibitors of KSP are mitosis-specific that preferably act on cells undergoing cell division(Sarli & Giannis, 2008). Hence, KSP inhibitorsmay likely to have fewer side effects (with the exception of neutropeniaa main side effect) while comparing with other agents affecting MT-based processes (Sakowicz et al., 2004; Song, Zhou, Wang, & Li, 2013).

The KSP inhibitors are structurally diverse and include (i) six-membered fused heterocycles (CK-0106023,monastrol, and dimethylenastron), (ii) five-membered nitrogencontaining heterocycles(dihydro-pyrazolothiophenes, dihydro-thiadiazoles,and thiazoles), (iii)biaryl compounds (GSK-1 and GSK-2) and (iv)natural products (adociasulfate-2)(El-Nassan, 2013). Few of the KSP inhibitors were studied under different stages of clinical trials.

SB-743921, a chromen-4-one derivative was evaluated in clinical trials(phase-II) to treat non-Hodgkin lymphoma and Hodgkin lymphoma(https://clinicaltrials.gov).Further, it has demonstrated 5-fold more potency than ispinesib in cell-based assays (Holen et al., 2011; Yin et al., 2015).ARQ-621 is an allosteric inhibitor evaluated against solid metastatic tumours and hematologic malignancies in a phase-Istudy (https://clinicaltrials.gov). Similarly, AZD-4877 was evaluated under phase-II clinical trial for the treatment of

advanced bladder cancer (https://clinicaltrials.gov). A pyrrole analogue, MK-0711 was studied in patients with advanced solid tumours under phase-I clinical trial (https://clinicaltrials.gov). Filanesib (also known as ARRY-520) is now under phase-III clinical studies in the management of multiple myeloma in combination with carfilzomib (https://clinicaltrials.gov)(Figure 1). Ispinesib (SB-715992) was investigated in phase-II studies in treating patients with metastatic or unresectable kidney cancer (https://clinicaltrials.gov) (Figure 2). Similarly, litronesib(known as LY2523355) is an ATP non-competitive, and allosteric inhibitor evaluated in patients with breast cancer (phase-II) (https://clinicaltrials.gov) and reported as a promising anticancer agent (Wakui et al., 2014)(Figure 2).

KSP encompasses 1057 amino acid residues and is divided into three major domains namely the motor domain (N-terminal), stalk and tail (C-terminal) domains (Figure 3). A short stretch of initial 20 amino acids precedes the motor domain (21-356 amino acids) followed by a neck linker (357-363), the internal stalk responsible for oligomerisation and the C-terminal tail (364-1057), which includes a phosphorylation-site (Thr927 in human KSP)(Blangy, Arnaud, & Nigg, 1997; Turner et al., 2001). The motor domain is the highly significant segment as it is responsible for the hydrolysis of adenosine triphosphate (ATP) to adenosine diphosphate (ADP) thus generating the required energy for movement of the motor along MTs.KSP utilizes the energy to propel spindle MTs in an antiparallel manner through its motor activity and sliding them apart. Thus, it contributes to the separation of centrosomes and the generation of bipolar spindle assembly during cell division(Civelekoglu-Scholey & Scholey, 2010; Saunders, Powers, Strome, & Saxton, 2007; Wordeman, 2010). As the motor domain of KSP plays a central role in cell division, many research groups determined the structure of the motor domain and studied the mode of action of small organic moleculesas KSP inhibitors(Barsanti et al., 2010; Fraley et al., 2006; Kaan, Ulaganathan, Rath, et al., 2010; Kaan, Ulaganathan, Hackney, & Kozielski, 2010; Wang et al., 2012). Recently, the Xray crystal structure of the KSP motor domain in complex with SB-743921 a second generation ispinesib analogue and ADP was reported in the protein data bank (PDB ID: 4BXN)(Figure 4). The inhibitor was found to bind at the allosteric binding-siteformed by helix $\alpha 2/loop L5/\alpha 3$ (Talapatra, Anthony, Mackay, & Kozielski, 2013).

Computational methods such as ligand-based (pharmacophore model and 3D QSAR), structure-based (molecular docking or molecular dynamics simulations) and integrated approaches are generally employed in designing therapeutically active new chemical entities. These approaches have been employed successfully to discover drugs such as dorzolamide (carbonic anhydrase inhibitor), captopril (angiotensin-converting enzyme inhibitor), saquinavir, ritonavir, indinavir and raltegravir (anti-HIV drugs) and tirofiban (fibrinogen antagonist)(Gu, Zhang, & Yuan, 2014; Hartman et al., 1992; Talele, Khedkar, & Rigby, 2010; Vijayakrishnan, 2009)as well as drug metabolism profiles(Ramesh & Bharatam, 2014). First,a quantitative 3D pharmacophore model with good correlation coefficient (r = 0.965) for KSP inhibitors was developed by Liu and co-workers. The studyreportedthe best hypothesis bearing four different chemical features (ring aromatic, hydrogen bond acceptor, hydrogen bond donor, and hydrophobic) using HypoGenmodule implemented in Catalyst software(Liu, You, & Chen, 2007). Subsequently, Jiangand his colleagues reported docking (structure-based) studies of 15 KSP inhibitors and proposed the significance of a 'minor pocket' that is situated away from the 'main pocket' in enhancing the binding affinity of ligands(Jiang, Chen, Wang, & You, 2007).

Till date, there are 49 solved X-ray crystal structures of KSP available in the PDB(http://www.rcsb.org/pdb/).Thus, these structures paved a way towards the determination of novel KSP inhibitors using structure-based drug design method. In 2012, Nagarajan and collaborators reported combined ligand- and structure-based virtual screening approaches toidentify novel KSP inhibitors. In their study, the interaction-based pharmacophore models (IBPs) were developed and validated using six different complexes of KSP crystal structures. The generated models were applied to filter the ChemDiv database (0.7 million) to identifyinitial hits. The IBP-based hits were subjected to docking (GOLD and GLIDE), biological screening (inhibition of tumor cell proliferation and basal Eg5 ATPase activity), and phenotypic analysis. These integrative approaches identified substituted imidazolidinecompound as a novel KSP inhibitor(Nagarajan, Skoufias, Kozielski, & Pae, 2012)(9, Figure 5). Similarly, Carbajales and co-workers analyzedKSP allosteric bindingsite from the 24 crystal structures of allostericinhibitor complexes and developed planar heterocycles based on the crystal structures of KSP (PDB ID: 2GM1). This structure-guided design led to the discovery of benzimidazole-based KSP inhibitors throughmulticomponent synthesis(10 and 11, Figure 5)(Carbajales et al., 2014). Recently (2015), Yokoyama and coworkers reported the crystal structure (PDB ID: 4WPN) and biochemical characterisation of the Eg5 motor domain in complex with a new type of allosteric inhibitor (PVZB-1194). This biphenyl-type inhibitor was proposed to bind within the novel $\alpha 4/\alpha 6$ allosteric pocket, whereas residues in the allosteric pocket formed by $\alpha 2/loop L5/\alpha 3$ are responsible for the binding of conventional KSP inhibitors (Yokoyama et al., 2015). In recent literature reports

(2016), chemical feature-based pharmacophore modelswere generated and validated. Subsequently, the best validated model was considered for virtual screening including the drug-like parameters to yield hits. Based on molecular interactions (docking) and computed electronic parameters (density functional theory calculations), authors reported six lead compounds as novel KSP inhibitors(Karunagaran, Subhashchandrabose, Lee, & Meganathan, 2016).

Currently, there are no specific KSP inhibitor drugs in the market. However, few KSP inhibitors (Figure 1 and 2) are at different stages of clinical trials. This prompts medicinal chemists to design and develop new KSP inhibitors through rational drug discovery approaches. Therefore, in the present study, we envisaged to develop novel KSP inhibitors by employing ligand- and structure-based drug design techniques. In addition, *in vitro* enzyme assay evaluations were carried out for the identified hits against the basal and MT-stimulated ATPase activity of KSP.

2. Materials and methods

2.1. Collection of training and test set

The selection of a suitable training set is crucialfor the generation of ligand-based pharmacophore models. Additionally, it is necessary to consider an appropriate test set of ligands that are not employed for the generation of models but may assistin the validation of the obtained pharmacophore models. In this study, a data set of highly active KSP inhibitors incorporating clinical/preclinical trial molecules (Cox et al., 2008; Gerecitano et al., 2009; Roecker et al., 2007; Rosen et al., 2010; Theoclitou et al., 2011; Woessner et al., 2009)as a training set (Figure1). Further, it was taken into an account that the selected ligands were pharmacologically screened under similar experimental conditions (*in vitro*), and structural diversity. Many of the reported active and inactive KSP inhibitors exhibited a stereocenter in their chemicalstructure. Hence, appropriate stereochemistry of ligands was carefully considered in the training as well as the test sets.

2.2. Generation of pharmacophore models

Pharmacophore modeling relates pharmacological activity with the 3D spatial arrangement of various chemical features in a set of active molecules.HipHop is a wellknown application for common feature-based pharmacophore model generation. It is incorporated in Discovery Studio (DS) software package as 'Common Feature PharmacophoreGeneration' protocol(DassaultSystèmes BIOVIA, Discovery Studio Modeling Environment, Release 4.5, San Diego: DassaultSystèmes, 2015).The present study includes the six most active ligands in the training set to build/construct suitablepharmacophore models.To generate the pharmacophore models, aconformational analysis was carried out for all molecules of thetraining set using the same software package. These conformations were generated by a fast conformational method with a maximum limit of 255,having an energy 20 kcal/molhigher than the global energy minimum(Neves, Dinis, Colombo, & Sá e Melo, 2009). Based on the presence of certain chemical features in the training set, hydrogen bond acceptor (A), hydrogen bond donor (D), hydrophobic (H), hydrophobic aliphatie (HY-Ali), positively ionizable (P) and ring aromatic (R) chemical functions were considered for feature analyses. The minimum inter-feature distance was set to the default value (2.97 Å).

2.3. Validation of pharmacophore model

The pharmacophore model usually consists of certain fixed parameters/filters that every new chemical entity has to satisfy inorder to be considered as an active inhibitor. The validation of pharmacophore model is performed by screening an *in house* test set consisting of both known active and inactive molecules. This is an indispensable step for the validation of the pharmacophore model. The test set was thoroughly verified to exclude high molecular weight compounds, prodrugs, and peptides. The *in house* test set constituted of 54 active KSP inhibitors (IC₅₀ = 0.1-100.0 nM) and 755 inactive molecules (IC₅₀>100.0 nM). From the 54 active KSP inhibitors, 36 ligands were identified as 'highly active' (IC₅₀ values < 20 nM) while the remaining 18 ligands were arbitrarily classified as 'moderately active' (IC₅₀ = 20 to 100 nM). Although, from the pharmacological point of view the ligands with IC₅₀ values 100 to 500 nM are also considered as active inhibitors, however in our study the cut off IC₅₀ value was set at 100 nM as one of the crucial parameter to identify highly potent inhibitors (Supplementary Table S1).

2.4. Virtual screening

Virtual 3D database screening can offer a valuable route to identify potential lead molecules for further drug development. Eventually, database searchprovides an advantage that theidentified or retrieved molecules can be procured handily or synthesized conveniently for pharmacological screeningprotocols when compared to other *de novo*drug design methodologies(Michaux et al., 2006). The approach uses 3D features of the pharmacophore

model asa query in the database screening to discover novelhit molecules. Twodatabase search methods (i) Fast/Flexible (diverse low-energy conformation) and Best/Flexible(best coverage of conformational space)are incorporated in DS. In general, Best/Flexible search method is employed for virtual database screening to obtain better results (Karunagaran, Subhashchandrabose, Lee, & Meganathan, 2016)and the same method was implemented in our study. Two different databases, namely Maybridge-HitDiscover (MB)and ChemBridge (CB) were searched to identify potential KSP targeting hits. Further, the molecules were filtered fortheir drug-like properties using Lipinski's rule of five[molecular weight less than 500, less than 5 hydrogen bond donor groups, less than 10 hydrogen bond acceptor groups and an octanol/water partition co-efficient (LogP) value of less than 5], Veber rule [10 or less than 10 rotatable bonds and polar surface area equal to or less than 140 Å²](Veber et al., 2002)andADMET (Absorption, Distribution, Metabolism, Excretion, andToxicity) filters using DS. Hit ligands which qualified for all of thesescreening protocols were takenforwardfor molecular docking studies.

2.5. Molecular docking

Molecular docking is a structure-based in silico approach to identify the possible ligand binding landscape inside the binding-site of a macromolecular protein/receptor. Molecular docking was conducted using the CDOCKER module under the receptorligandinteractions protocol implemented in DS. It is a grid-based docking method thatemploys CHARMm force field. In this protocol, a macromolecule is kept under constraint while ligands are enabled for their conformational flexibility during the refinement step. A high-temperature molecular dynamics (MD) pursued by random rotations resulted in different conformations for initial ligand structure. The final refinement of random conformations was implemented through a grid-based simulated annealingfollowed by either grid-based or full forcefield minimization(Wu, Robertson, Brooks, & Vieth, 2003). To perform molecular docking, a crystal structure of humanKSP with bound inhibitor (SB-743921) was used (PDB ID: 4BXN)(Talapatra et al., 2013). Prior to the molecular docking, KSPwas prepared using 'prepare protein' implemented under 'Macromolecules' module of DS. All the water molecules, chloride ions, and cadmium ions present in co-crystallized structure of the protein were removed and the missing hydrogen atoms were added to the KSP structure (Figure 6A). The co-factor (ADP) present in the crystal structure was retained for moleculardocking protocol in which the pH for protonation was set at 7.4. To optimize the molecular docking

protocol, several docking runs were performed. During the optimization, various parameters were set for theactive-site radius (15Å), scoring function (CDOCKER energy), algorithm (CDOCKER), physiological state of the ligands (ionized state), root mean square deviation (< 1.5) to enhance the docking accuracy. By employing receptor-ligand interactions tool, the total protein was defined as a receptor and a binding-site sphere was generated with the radius of 15Å around the inhibitory ligand present in the protein structure (Figure 6B).

Further, the ligands were prepared for ionization change, Lipinski's filter, and 3D conformation generation. For the purpose of the validation of the molecular docking protocol, the bound ligand (SB-743921) was extracted from the co-crystallized complex and was docked into the definedbinding-site of the protein. Similarly, molecules present in the training set, test set and the hits retrieved from virtualscreeningwere also docked into the pre-defined binding-site of KSP (target protein) with default settings of DS. After completion of docking, the protein-ligand complexes were analyzed to investigate the type of interactions and CDOCKER energy values(Yang, Yang, Zou, Li, & Zhu, 2016).The final hits of molecular docking were selected based on their higher (most negative, thus favourable to binding)CDOCKER energy and selected complexes were subjected to molecular dynamics simulations.

2.6. Molecular dynamics

Molecular dynamics (MD) is a widely used structured-based *in silico* approach to determine the binding efficiency of ligands inside the binding-site of macromolecules(Zhao & Caflisch, 2015). All MD simulations were run in AMBER 14.0 software package in SANDER(Case et al.,2014). The protein-ligand complexes of 4BXN-SB743921 (native ligand in the crystal structure)(Talapatra et al., 2013) and the top-ranked hit molecules of molecular dockingfrom each database weresubjected to MD simulations. The ligand binding pose obtained from top-ranked molecular docking was considered as starting point to runMD simulations. Initially, the geometry of the protein and bound ligands from the docked complexes were separated using Chimera Software package(Pettersen et al., 2004). The ligands were defined with Generalized Amber Force Field (GAFF) and the partial atomic charges were derived from AM1-BCC method of Antechamber. The protein was defined with

AMBER99SB force field and was prepared in Leap module to add missing hydrogen and to generate the parameters for a solvated protein-ligand complex(Kholmurodov, Smith, Yasuoka, Darden, & Ebisuzaki, 2000). The complex was enclosed with TIP3P water molecules in a box which was extended up to 10 Å from the center of the molecule(Jorgensen, Chandrasekhar, Madura, Impey, & Klein, 1983). The solvated complex was electro-neutralized with counter ion Na⁺. Four Na⁺ions were added to neutralize the system. To remove the bad contacts, partial minimization was carried out with a restraint potential of 500 kcal mol⁻¹Å⁻². Partial minimization was run for 1000 steps of steepest descent (SD) followed by 500 steps of conjugate gradient (CG). Further, the complex was energetically minimized using 50 steps of SD followed by 150 steps of CG. After minimization, heating was conducted graduallyfrom 0 to 300 K for 5 psusing LangevinDynamics. Then the system was equilibrated at 300 K in a constant pressure and temperature ensemble for 500 ps. The pressure of the system was maintained at 1 bar using Berendsenbarostat. All the bonds involving hydrogens were constrained using SHAKEalgorithm(Ryckaert, Ciccotti, & Berendsen, 1977).MD was run without restrain at a temperature of 300 K and a pressure of 1 bar. A5 ns MD was run with a time step of 2fs and a distance cutoff of 12.0 Å for the non-bonded interactions(Berendsen, Postma, van Gunsteren, DiNola, & Haak, 1984). The trajectories were saved and analyzed for every 1 ps. Visualization of trajectories was carried out using Chimera software package. After MD simulations, binding free energies were computed.

2.7. MM/GBSA and other post dynamics analyses

Binding free energies were defined as bound and unbound states of protein-ligand complexes.After 5ns of MD simulation, the binding free energies were computed by MM/GBSA method[Eqn 1-5](Hou, Wang, Li, & Wang, 2011; Lyne, Lamb, & Saeh, 2006).Binding affinities were estimated by averaging snapshots taken from the MD trajectories. The binding free energies were averaged over 1000 snapshots of MD trajectories.

$\Delta G_{\text{bind}} = G_{\text{complex}} - G_{\text{receptor}} - G_{\text{ligand}}$	(1)
$\Delta G_{\text{bind}} = E_{\text{gas}} + G_{\text{sol}} - TS_{\dots}$	(2)
$E_{gas} = E_{int} + E_{vdw} + E_{ele}$	(3)
$G_{sol} = G_{GB} + G_{SA}$	(4)
$G_{SA} = \gamma SASA$	(5)

 E_{gas} : gas-phase energy that consists of the internal energy(E_{int}), coulomb energy (E_{ele}),

and thevan der Waals energies (E_{vdW}). E_{gas} was directly estimated from the FF99SB force field terms. The solvationfreeenergy (G_{sol}) was estimated from the energy contributionfrom the polar states (G_{GB}) and nonpolar states (G_{SA}). Thenonpolar solvation energy(G_{SA}) was determined from theSASA using a water probe radius of 1.4 Å, whereasthe polar solvation (G_{GB}) contribution was estimated bysolving the GB equation. The total entropy of the soluteand temperature was denoted by S and T, respectively.Post-MD analyses (root mean square deviation, root mean square fluctuation, and radius of gyration), the snapshot of the structures during simulation and average structure were carried out using CPPTRAJ and PTRAJ module of AMBER 14(Roe & Cheatham, 2013; Duan, Liu, & Zhang, 2016).

2.8. Chemical Synthesis and Characterization

2.8.1. General

Melting points of all the synthesized compounds were recorded in open capillaries using Electro-thermal (IA9300) digital melting point apparatus and are uncorrected. A Bruker FT-IR spectrophotometer with universal ATR sampling accessory was used to record all IR spectra. ¹H and ¹³C NMR spectra were recorded on NMR spectrometer (Bruker Avance IV) at 400 and 101 MHz, using CDCl₃with TMS as an internal standard and reported in parts per million (ppm). The progress of the reactions and the purity of the synthesized compounds were monitored by Thin Layer Chromatography (TLC) on precoated silica gel 60 F₂₅₄ (mesh) (E. Merck) and spots were visualized under UV light (long and short wavelength). The analytical grade (AR) chemicals and reagents were obtained from Merck and Sigma-Aldrich and were used as such without any further purification. The ChemBridge ligand (CB-10358) was procured directly from ChemBridge Corporation, San Diego, USA.

2.8.2. Synthesis of diethyl 2-amino-4,7-dihydrothieno[2,3-c]pyridine-3,6(5H)-dicarboxylate (3)

Ethyl 4-oxo-piperidine-1-carboxylate 1 (0.04 mol; 6.0 ml), ethyl cyano acetate 2 (0.04 mol; 4.52 ml), powdered sulphur (0.04 mol; 1.28 gm), and absolute ethanol (40 ml) were taken in a conical flask and warmed at 50-60 °C. Then diethylamine (4 ml) was added dropwise with constant stirring until all sulphur went into solution. The stirring was continued for an additional one hour until brownish-yellow coloured solid separated, then it was cooled to room temperature and filtered. The crude product was dried and recrystallized from ethanol to yield pure diethyl 2-amino-4,7-dihydrothieno[2,3-*c*]pyridine-3,6(5*H*)-

dicarboxylate (3).

Brownish-yellow solid, yield: 9.1 gm (87%), mp: 160-162 °C; IR (KBr): 3422, 3318 (NH₂), 1685 (ester-C=O), 1665 (piperidine-C=O) cm⁻¹;¹H NMR (400 MHz, CDCl₃): δ 1.24-1.29 (t, *J* = 7.02 Hz, 3H, CH₂C<u>H₃</u>), 1.30-1.35 (t, *J* = 7.10 Hz, 3H, CH₂C<u>H₃</u>), 2.80 (s, 2H, CH₂, piperidine), 3.64 (s, 2H, CH₂, piperidine), 4.12-4.18 (q, *J* = 7.06 Hz, 2H, C<u>H₂</u>CH₃), 4.21-4.27 (q, *J* = 7.13 Hz, 2H, C<u>H₂</u>CH₃), 4.38 (s, 2H, CH₂, piperidine), 5.5 (br s, 2H, NH₂) ppm; ¹³C NMR (100 MHz, CDCl₃, CPD): δ 14.58, 14.86, 27.22, 41.44, 42.78, 59.78, 61.71, 155.71, 162.51, 165.94 ppm.

2.8.3. Synthesis of diethyl 2-(2-chloroacetamido)-4,7-dihydrothieno[2,3-c]pyridine-3,6(5H)dicarboxylate (4)

The compound **3** (0.02 mol; 5.967 gm) was dissolved in dry *N*,*N*-dimethyl formamide (40 ml) and added triethylamine (0.02 mol; 2.8 ml) followed by the addition of a solution of chloroacetyl chloride (0.024 mol; 1.9 ml) in DMF (10 ml) slowly dropwise. The reaction mixture was stirred at room temperature for additional two hours until all starting material converted to product (monitored by TLC). Then the mixture was kept at room temperature to precipitate dark coloured solids. The product was isolated by filtration under vacuum, dried and recrystallized with absolute alcohol to obtain pure intermediate **4**.

Cream coloured solid, yield: 6.4 gm (85%), mp: 126-128 °C; IR (KBr): 3201 (NH), 1695 (ester-C=O), 1676 (piperidine-C=O), 1655 (amide-C=O); ¹H NMR (400 MHz, CDCl₃): δ 1.26-1.30 (t, J = 7.08 Hz, 3H, CH₂CH₃), 1.37-1.40 (t, J = 7.18 Hz, 3H, CH₂CH₃), 2.89-2.92 (m, 2H, CH₂, piperidine), 3.68-3.71 (m, 2H, CH₂, piperidine), 4.15-4.20 (q, J = 7.12 Hz, 2H, CH₂CH₃), 4.26 (s, 2H, CH₂-acetamido), 4.34-4.39 (q, J = 7.11 Hz, 2H, CH₂CH₃), 4.56 (s, 2H, CH₂, piperidine), 12.10 (br s, 1H, NH) ppm; ¹³C NMR (100 MHz, CDCl₃, CPD): δ 14.51, 14.88, 41/26, 42.35, 42.78, 61.22, 61.86, 112.79, 146.99, 155.67, 163.83, 165.86 ppm.

2.8.4. Synthesis of diethyl 2-(2-(4-methylpiperazin-1-yl)acetamido)-4,7-dihydrothieno[2,3c] pyridine-3,6(5H)-dicarboxylate (5)

To a suspension of the intermediate **4** (0.001 mol; 0.375 gm) in 3 ml of dry tetrahydrofuran (THF) added 1-methylpiperazine (0.0015 mol) dropwise and stirred at 60 °C for two hours. After completion of the reaction (as monitored by TLC), the excess THF was evaporated under reduced pressure, the resulting residue was triturated with water and filtered to obtain crude product. Recrystallization of crude compound with absolute alcohol resulted

in pure final compound 5 (Maybridgehit MB-41570).

Golden-yellow solid, yield: 350 mg (80%), mp: 95-97 °C; IR (KBr): 3229 (NH), 1659 (ester-C=O), 1562 (piperidine-C=O), 1516 (amide-C=O); ¹H NMR (400 MHz, CDCl₃): δ 1.26-1.29 (t, *J* = 7.01 Hz, 3H, CH₂C<u>H₃</u>), 1.36-1.40 (t, *J* = 7.20 Hz, 3H, CH₂C<u>H₃</u>), 2.45 (s, 3H, N-C<u>H₃</u>), 2.75 (s, 8H, 4CH₂-piperazine), 2.89 (s, 2H, CH₂, piperidine), 3.28 (s, 2H, CH₂-acetamido), 3.68-3.69 (m, 2H, CH₂, piperidine), 4.14-4.19 (q, *J* = 7.06 Hz, 2H, C<u>H₂CH₃</u>), 4.33-4.38 (q, *J* = 7.06 Hz, 2H, C<u>H₂CH₃</u>), 4.55 (s, 2H, CH₂, piperidine), 12.24 (br s, 1H, NH) ppm; ¹³C NMR (100 MHz, CDCl₃, CPD): δ 14.34, 14.67, 41.64, 45.45, 52.68, 54.61, 60.50, 60.74, 61.61, 147.34, 155.50, 165.34, 168.14 ppm.

2.8.5. Procurement and characterization of ChemBridgehit (CB-10358).

The ChemBridgehit "CB-10358" was procured directly from the drug database provider (ChemBridge Corporation, San Diego, USA) and confirmed its structure based on ¹H and ¹³C NMR.

¹H NMR (400 MHz, CDCl₃): δ 1.23-1.29 (t, *J* = 7.01 Hz, 3H, CH₂CH₃), 2.17-2.45 (m, 6H, piperidine), 2.79 (m, 1H, piperidine), 2.9-3.0 (m, 1H, piperidine), 3.18-3.22 (m, 2H, CH₂), 3.38-3.46 (m, 2H, CH₂, 1H, piperidine), 3.64-3.74 (br s, 6H, OCH₃), 4.15-4.20 (q, *J* = 7.06 Hz, 2H, CH₂CH₃), 6.20 (s, 1H, Ar-H), 6.93 (s, 2H, Ar-H), 11.46 (br s, 1H, NH) ppm; ¹³C NMR (100 MHz, CDCl₃, CPD): δ 14.07 (CH₃-CH₂), 25.29 (CH₂-piperidine), 31.98 (CH₂-piperidine), 32.09 (CH₂-CH₂), 34.98 (CH-piperidine), 50.45 (CH₂-piperidine), 52.74 (CH₂-piperidine), 53.04 (O-CH₃), 53.23 (O-CH₃), 55.34 (CH₂-CH₂), 61.31 (CH₂CH₃), 96.86, 98.15, 98.22 (Ar-C), 139.81 (Ar-C-N), 167.50 (C-OCH₃), 167.73 (C-OCH₃), 171.87 (C=O), 172.92 (C=O) ppm.

2.9. Pharmacological evaluation

2.9.1. Evaluation of hits against the steady-state basal and MT-stimulated KSP ATPase activities

The experiments were performedat 25 °C using a 96-well Sunrise photometer (Tecan, Mannesdorf, Switzerland) at a final volume of 100 μ L per well. Steady-state basal and MT-stimulated ATPase rates were measured using the pyruvate kinase/lactate dehydrogenaselinkedassay in buffer A25A [25 mmol/L potassium ACES (pH 6.9),2 mmol/L magnesium acetate, 2 mmol/L potassium EGTA, 0.1 mmol/L potassium EDTA, 1 mmol/L β -mercaptoethanol]. The amounts of KSP were optimized at 80-100 nM for the basal and 5 nM

for the MT-stimulated activity assays. The IC₅₀ values for the inhibition of the basal and MTstimulated ATPase activities of Eg5 were measured for K-858 (positive control) between 1 and 200 μ M. The ATP concentration was fixed at 1 mM. For the inhibition of the MTstimulated ATPase activity, the MT concentration was 1 μ M. The data were analyzed usingKaleidagraph 3.0 (Synergy SoftwareReading, PA) and Microsoft Excel to obtain IC₅₀ values (Talapatra et al., 2013; Nagarajan et al., 2012).

3.Results and discussion

3.1. Common feature pharmacophore generation

Ten common feature-based pharmacophore models were generated using a training set of ligands. The scores of the obtained pharmacophore models ranged from 74.029to 68.245and each of the pharmacophore models had a minimum of four features(Table 1). These models vary in their orientation, vector directions, and composition. Essentially, the major structural requisite determined by DS consists of ring aromatic (R), hydrogen bond acceptor (A), positively ionizable (P) or hydrophobic (H) feature for favourable interactions. The Direct Hitbit mask is the ligand that directly match the pharmacophore and the value '1' indicates that the molecule maps to every feature in the pharmacophore. The Partial Hit bit mask is the ligand that partially match the pharmacophore and the value '0' indicates the complete matching with the pharmacophore. The positively ionizable (P) pharmacophoric feature and hydrogen bond acceptor (A) were observed as common features among all the ten pharmacophore models. This is in agreement with the presence of basic amino groups and hydrogen bond acceptor atoms in structural features of all the training set molecules. The topranked pharmacophore model (Hypo1) was chosen as the best model based on scores, fit values, the presence of the number of pharmacophoric features and feature diversity (Tables 1 and 2). The proposed modes of pharmacophore mapping of the training set ligands on this model were found to be good and in accordance with fit values. The best model (Hypo1) contains one ring aromatic (R), one positively ionizable (P), one hydrophobic (H), and two hydrogen bond acceptors (A_1 and A_2) features. Hypo1 was found to becompetentadequately to map the training set molecules and canbe applied effectively to recognize the potential KSP inhibitory ligands from drug databases. The distance between two hydrogen bond acceptors (A₁ and A₂) was determined as 3.510 Å. The hydrogen bond acceptor 1 (A₁) was located at a distance of 4.654 Å from the hydrophobic feature (H) and the feature P (6.080 Å), respectively. The hydrogen bond acceptor 2 (A_2) was distanced from the ring aromatic (R) feature at 3.653 Å and hydrophobic (H) feature at 5.048 Å. Three features (P, R, and H)

are positioned in atriangular fashionwherein the distance between 'P' and 'R' was found to be 7.947 Å, while the distance between 'P' and 'H' features was determined to be 8.364 Å (Figure 7A). The highly active molecule of the training set (**2**) mapped all the pharmacophoric features effectively on Hypo1 (Figure 7B).

3.2. Validation

The pharmacophore models were validated using a test set comprising of active and inactive KSP inhibitors (Table 3 and Supplementary Table S1). The test set constitutes a total of 809 ligands (active: 54andinactive: 755). The top-ranked pharmacophore model (Hypo1) was validatedcorrespondingly using a test set by defining Maximum Omitted Feature value as '0'. Hypo1 screened 48 compounds (highly active=34 and moderately active=14) as active out of 54 known active ligands (88.88%) and 13 compounds as inactive out of 755 known inactive ligands (1.72%) from the test set (Table 3). Hypo1was observed toidentify the active molecules and to distinguish the active from inactive molecules effectively. Finally, Hypo1 was selected as the best pharmacophore model of KSP inhibitors.

3.3. Database screening/searching

The validated pharmacophore model (Hypol) was subjected to searching in two different databases[Maybridge(MB) and ChemBridge(CB)]constituting 52610 and 74917 ligands, respectively. During the database searching,MaxOmit Feature '0' and fit value > 3 were applied so that molecules that match all the features would be screened. This screening process identified 389 and 632ligands from MB and CB,respectively (Figure 8). Further, molecules were filtered through (i) Lipinski'srule offive and (ii)Veberrule. The implementation of drug-likeliness filter reduced the hits of drug unlikeliness profiles and retrieved 340 and 603 hits from MBand CB databases as drug-like molecules. Further, these hit molecules were filtered to retain the hits with efficient ADMET profiles. The approach filtered 43 and 217 (Total: 260) ligands respectively as hits of necessary pharmacophoric, drug-likeliness, and ADMET profiles. These hits were subjected to molecular docking studies(Figure 8).

3.4. Molecular docking

Molecular docking was employed as a structure-based filter to assess the potential ligands interacting with binding-site residues as well as to predict the probable binding mode of ligands within the protein structure. All the hits retrieved from the two different

databaseswere subjected to molecular docking to analyzethe capability of binding ofligands inside the binding pocket of KSP.For each of the input ligands, CDOCKER generated numerousreasonable binding conformations and rankedthese on the basis of CDOCKERenergy values. Thus, the ligand-bound conformationbearing the most favourable energy was considered as the best orientation.

Before performing the molecular docking for all the hit molecules, SB-743921 (bound ligand of KSP) and training set (includes SB-743921) ligands were also docked into the crystal structure of KSP (PDB ID: 4BXN) as a part of the validation of the molecular docking protocol.Molecular docking reproduced almost the same binding mode of SB-743921 inside the defined binding pocket with a docking energy of -26.05 kcal/mol(Table 2). The docking pose of SB-743921 (pink coloured ball and stick model) showed favourable interactions with crucial residues within the binding-site of KSP as observed in the crystal structure(yellow coloured ball and stick model) pose (Figure 9). The chromen-4-one moiety of SB-743921 was surrounded by the crucial residues of Trp127, Try211, Glu215, and Ala218, whereas the chloro substituent demonstrated hydrophobic interactions with Leu214 and Lys216, respectively. Similarly, the methyl group on the aromatic ring exhibited hydrophobic interactions with the residues of Leu132 and Ala133 (Figure 9). In 2007, Roeckerand coworkers crystallized KSP with an inhibitor MKR (PDB: 2Q2Y) and revealed a weak hydrogen bond between the carbonyl oxygen of MKR with the hydroxyl group of an aromatic amino acid Tyr211 (3.26 Å)(Roecker et al., 2007).In 2010, Kaan and co-workers reported the phenolic hydroxyl group of (S)-enastron interacting with Glu118 of KSP through a hydrogen bond (2.69 Å). Further, a phenyl substituent of the same ligand interacted with aromatic residues (Trp127 and Tyr211) through a π - π stacking interaction and an acidic residue Glu116 exhibited vdW interactions thereby contributing to the higher ligand binding affinity (PDB ID:2X7C)(Kaan, Ulaganathan, Rath, et al., 2010). The same research group crystallized another KSP-inhibitor complex (PDB ID: 2WOG) and identified the existence of a hydrophobic interaction of STLC with the side chain of Glu116(Kaan, Ulaganathan, Hackney, et al., 2010). Recently (2013), the authors deposited the structure of the KSPispinesib complex (PDB ID: 4A5Y) in which a strong hydrogen bond prevailed between the free amino group of ispinesib with Glu116 (2.88 Å)(Yi et al., 2013). This has been further evidenced by another KSP structure crystallized with a β-carboline inhibitor (PDB ID:3K3B) which demonstrated the crucial involvement of two different hydrogen bonds for β-carboline NH- and phenolic OH group (substituent) with Glu116 (3.00 Å) and Glu118 (2.66 Å), respectively(Barsanti et al., 2010). The above literature clearly indicated the involvement of

the crucial residues Glu116, Glu118, Trp127 and Tyr211 in ligand binding and similar molecular interactions were observed for the docked SB-743921 ligand.

The CDOCKER energies ranged from -26.05 to -33.78 kcal/mol for the training set molecules (which are highly potent) suggesting favourable binding at the KSP bindingsite(Table 1). Further, molecular docking was performed for all the hit molecules resulting from database screening. Molecular docking experiments indicated the most favourable binding for the top-rankedhit molecules of Maybridge(MB-41570,CDOCKER energy: -42.70 kcal/mol)and ChemBridge (CB-10358, CDOCKER energy: -39.13 kcal/mol). It was observed that the two hits displayed higher CDOCKER energies indicating the most favourable binding. The KSP docked complexes of MB-41570 (Maybridge)(Figure 10)and CB-10358 (ChemBridge)(Figure 11) were found to have top-ranked CDOCKER energy values. These complexes were then subjected to MD simulations to further substantiate the CDOCKER energies of the hits. The CDOCKER energy values and molecular interactions of bound ligand, training set ligands, and top-ranked hits (Figure 12) from each database are presented in Table S2 of the supplementary material section. The top-ranked complex from each database and the corresponding fit values are also depicted (Supplementary Table S3).

3.5. Binding free energies

The MD studies were conducted for the top-ranked hits retrieved from molecular docking approach (Figure 9) as well as the co-crystallized ligand (SB-743921). After 5ns MD simulation, the binding free energies were computed to identify the inhibitory potential of the two hits and SB-743921. SB-743921 has shown very high *in vitro*potency against KSP (IC₅₀=0.1 nM) and the computed binding free energy of SB-743921 was used as a reference for the assessment of binding affinity of other hit molecules(Talapatra et al., 2013). The calculated binding free energy for SB-743921 at the binding-site of KSP was -31.54 kcal/mol.The binding free energies of hit molecules were found to be -42.89 and -47.53 kcal/molfor MB-41570 (Maybridge) and CB-10358 (ChemBridge)respectively. These two hit molecules therefore show stronger binding free energy values than the reference molecule SB-743921. Further, MB-41570 (Maybridge) and CB-10358 (ChemBridge) were predicted as the most promising candidatesfor KSP inhibition (Table 5). The van der Waals (vdW) and electrostatic energies of MB-41570 and CB-10358 were found to be -52.15±4.54kcal/mol and -51.73±3.53kcal/mol, respectively. The vdWinteractions energy has highly contributed to the ligand binding affinity of the identified hits.In the future, this study may help us to designand

develop various analogs of MB-41570 and CB-10358 asprobable potent KSP inhibitors forcancer chemotherapy.

3.6. Post molecular dynamics analysis

The stability of the protein-ligand complexes (i) 4BXN-SB743921, (ii) 4BXN-MB41570,and (iii) 4BXN-CB10358 was analyzed over 5ns MD simulation by computing RMSD, RMSF, and RoG.The ligand-protein complex of hit molecules showed a marginal variation as compared to reference during the entire simulation suggesting the ligand bound complex of hit molecules as stable as the reference. The RMSD values for the complexes4BXN-SB743921, 4BXN-MB41570,and 4BXN-CB10358 were determined as 2.45, 2.79, and 2.37,respectively(Table 6 and Figure 13).

The flexibility of amino acids for the complexes 4BXN-SB743921, 4BXN-MB41570,and 4BXN-CB10358 was analyzed by assessing the root mean square fluctuations (RMSF) from MD trajectories (Figure 14). The overall RMSF was found to be 0.92, 0.98 and 0.93,respectively for 4BXN-SB743921, 4BXN-MB41570and 4BXN-CB10358 complexes. The complex of 4BXN-MB41570and 4BXN-CB10358 indicatedslightly higher amino acid flexibility than the reference. However, the identical trend in the fluctuation of amino acid residues was observed in all three complexes. The highest magnitude of flexibility was observed around residues10-25, 35-45, and 230-245.

The radius of gyration (RoG) is defined as the moment of inertia of the group of atoms from their centroid of mass. Computation of RoG during the simulation sheds light on the stability of the system.Figure 15presents theRoG over 5ns simulations in the4BXN-SB743921, 4BXN-MB41570and 4BXN-CB10358 complexes. The complex of 4BXN-MB41570has shown highestRoG after the 0.5ns simulation. The simulation beyond 2nsresulted in a decreased RoG. The average RoG for the complexes of 4BXN-SB743921,4BXN-MB41570 and 4BXN-CB10358 was 20.97, 21.03and 20.87, respectively. The results suggested that the relatively close stability of the complexes during the MD simulations.

3.7. Synthesis and characterization of Maybridgehit

The Maybridgehit (MB-41570) was synthesized through a convenient synthetic route (Ismail et al., 2012) as shown in scheme-1. Structures of all the synthesized compounds were characterized based on their physicochemical and spectral (IR, ¹H NMR and ¹³C NMR) analysis. The spectra of the newly synthesized compounds along with their anticipated

structures are presented in the supplementary information. Ethyl 4-oxo-piperidine-1carboxylate 1 reacted with ethyl cyanoacetate2 and sulphur in absolute ethanol in the presence of diethylamine afforded the key starting material, diethyl 2-amino-4,7dihydrothieno [2,3-c]- pyridine-3,6(5H)-dicarboxylate (3) quantitatively. The FT-IR spectrum of **3**, we observed a reasonably sturdy and characteristic bands around 3422, 3318 cm⁻¹ accounting for N-H stretching of primary amine. Further, a couple of prominent carbonyl stretching frequencies were observed at 1685 and 1665 cm⁻¹ which was attributed to the ester-C=O and piperidine-C=O, respectively. The structure of **3** was elucidated by ^{1}H NMR spectrum (400 MHz), wherein the appearance of an informative broad singlet signal at δ 5.5 ppm attributing to NH₂protons. Further, appearance of additional triplet ($\delta = 1.30-1.35$ ppm) and quartet ($\delta = 4.21-4.27$ ppm) signals corresponding to CH₂-CH₃ and CH₂-CH₃, respectively indicated the formation of the desired starting material 3 via the well-known Gewald reaction. This has been further validated by the appearance of deshielded carbon signals at δ 165.94, 162.51, and 155.71 ppm attributable to C-NH₂, N-C=O, and C-C=O respectively in ¹³C NMR. Moreover, carbon signals in respect to that of ethyl groups appeared at δ 14.58, 14.86, 59.78, and 61.71 ppm in the shielded region confirmed the formation of 3. Reacting the amino group in 3 with chloroacetyl chloride in the presence of an organic base (triethylamine) resulted an intermediate, diethyl 2-(2-chloroacetamido)-4,7dihydrothieno[2,3-c]pyridine-3,6(5H)-dicarboxylate 4. Thisamide formation was further confirmed by the appearance of NH stretching band at 3201 cm⁻¹ and disappearance of primary amino stretching band in its FT-IR spectrum. Moreover appearance of additional C=O bond stretching at 1655 cm⁻¹has been assigned as carbonyl group of amide. The ¹H NMR of 4 showed a broad singlet in the shielded region at δ 12.1 ppm for amide -NH apart from the appearance of acetamido methylene proton at δ 4.26 ppm as a singlet. This has been further evidenced in the ¹³C NMR where the signal corresponding to acetamido carbonyl was appeared at δ 163.83 ppm apart from the appearance of acetamido methylene signal in the shielded region ($\delta = 41.26$ ppm). Nucleophilic displacement of chlorine in 4 with secondary heterocyclic amine such as N-methyl piperazine in tetrahydrofuran yielded the final compound 5. The ¹H NMR of 5 indicated the appearance of additional alicyclic proton in the shielded region as singlet at δ 2.75 ppm corresponding to eight protons of piperazinemoiety, apart from the appearance of one singlet signal at δ 2.45 ppm for N-CH₃ protons. Similarly, ¹³C NMR data corroborated well with the appearance of characteristic N-methyl carbon at δ 45.45 ppm apart from the carbon signals of piperazine moiety in the upfield region. All the above data confirmed the successful formation of the final compound 5 (Maybridgehit MB-

3.8. Pharmacological evaluation of hit compounds

3.8.1. Inhibition of hits against the steady-statebasal KSP and MT-stimulated KSP ATPase activities

In silico identified hits (Maybridge: MB-41570and ChemBridge: CB-10358) were evaluated against the basal and MT-stimulated KSP ATPase activities using standard reported protocols(Talapatra et al., 2013; Nagarajan et al., 2012). From the concentrationresponse curves obtained(seesupplementary information), it was observed that the two identified hitswere unable to displayinhibition against the basal or the MT-stimulated KSP ATPase.

4. Conclusion

In this study, pharmacophore models were developed using ligand-based approachesand validated against a test set of KSP inhibitors. The validated pharmacophore model (Hypo1) exhibited one ring aromatic (R), one positively ionizable (P), one hydrophobic (H), and two hydrogen bond acceptors (A1 and A2) assessmential features for KSP inhibition. This Hypo1 was used for virtual screening to identify potentKSP inhibitor scaffolds. Subsequently, thesescaffoldswere subjected to moleculardocking and molecular dynamics followed by the calculations of binding free energies. Finally, two ligands (MB-41570 and CB-10358) wereidentified as potentialand easily obtainable or synthesizable(Nankervis et al., 2011; Plater, Murdoch, Morphy, Rankovic, & Rees, 2000)KSP inhibitors. Since the two identified hits (MB41570 and CB10358) demonstrated similar interactions as compared to SB-743921 and exhibited higher binding energies, we synthesized these and evaluated them against the basal and the MT-stimulated KSP ATPase activities. Even though the identified hits had pharmacophoric feature similarities to SB-743921, they displayed no *in vitro* KSP enzyme inhibition. However, the pharmacophore model, interaction analysis, and the systematic protocol of the present study contributes to our understanding of KSP inhibition and may help future medicinal chemists in developing new KSP inhibitors by taking our research findings into account.

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Authors hereby declare that there are no financial/commercial conflicts of interest.

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Figure Legends

Figure 1. The training set employed for the generation of pharmacophore models.

Figure 2. KSP inhibitors under different stages of clinical development.

Figure3. Pictorial map of KSP representing the three major domains (motor, stalk, and tail).

Figure4. Three-dimensional (3D) arrangement of KSP (ribbon, coloured in grey), inhibitor

(SB-743921; magenta sticks) and ADP (shaded in blue) in the crystal structure (PDB ID:

4BXN).

Figure 5. Novel KSP inhibitors designed through molecular modeling approaches.

Figure 6.KSP for molecular docking (A) and location of the binding-site sphere (B).

Figure 7. (A) Best pharmacophore model (Hypo1) and (B) Mapping of 2 upon Hypo1.

Figure 8.Flowchart of database screening, drug-like filtrations, molecular docking and molecular dynamics in identifying novel KSP hits.

Figure 9. The orientation of the docked pose (pink coloured ball and stick model) with the cognate ligand (yellow coloured ball and stick model) in the binding-site of KSP.

Figure 10. Interactions of MB-41570 with KSP (Maybridge database).

Figure 11. Interactions of CB-10358 with KSP (ChemBridge database).

Figure 12.Top-ranked hits from each database after molecular docking/MD-MMGBSA. Figure 13. Root mean square deviation (RMSD) for the complexes of 4BXN-SB743921 (black), 4BXN-MB41570 (red) and 4BXN-CB10358 (green).

Figure 14. Root mean square fluctuations (RMSF) of C-a atoms for 4BXN-SB743921 (black), 4BXN-MB41570 (red) and 4BXN-CB10358 (green) complexes.

Figure 15. Radiation of gyration (RoG) for the complex of 4BXN-SB743921 (black), 4BXN-MB41570 (red) and 4BXN-CB10358 (green).

Scheme 1: Synthetic route for the synthesis of Maybridge hit: Reagents and conditions (a) S, diethylamine, EtOH, 60-70 °C, 2h; (b)ClCH₂COCl, *N*,*N*-DMF, rt, 2h; (c) *N*-methyl-piperazine, THF, 50-60 °C, 2 h.

Table Legends

Table 1. Description of common feature pharmacophore models.

Table 2. A training set of molecules for the generation of ligand-based pharmacophore model.

Table 3. Validation of pharmacophore model (Hypo1) using test set.

Table 4.Statistical analysis of top-ranked pharmacophore model (hypo1).

Table 5. MM/GBSA binding free energies of a reference molecule and the best-docked hits in complexation with kinesin spindle protein.





























Hypos	Features	Scores	Direct Hit	Partial Hit	Max Fit
Rank					
01	RPHAA	74.029	111111	000000	5
02	PHHAA	73.637	111111	000000	5
03	RPHAA	73.610	111111	000000	5
04	PHHAA	69.774	111111	000000	5
05	RRPA	68.876	111111	000000	4
06	RRPA	68.876	111111	000000	4
07	RRPA	68.526	111111	000000	4
08	RRPA	68.276	111111	000000	4
09	RRPA	68.276	111111	000000	5)4
10	RRPA	68.245	111111	000000	4
			AA		
			\mathbf{O}		

C

No.	Ligands	Features	IUPAC	KSP	Fit	CDOCKER
				Inhibition	Value	energy
				(IC ₅₀ ,		
				nM)		
1	SB-743921 (1)	RPHAD	(<i>R</i>)- <i>N</i> -(3-aminopropyl)- <i>N</i> -	0.1	2.09	-26.05
			(1-(3-benzyl-7-chloro-4-oxo			
			-4H-chromen-2-yl)-methyl-			
			propyl)-4-methyl			
			benzamide			
2	2	RPHA	(R)-1-(4-(3-(dimethylamino	0.2	4.99	-32.63
			-propyl)-1-(2-fluoro-5-		6	
			methy-l-phenyl)-4-phenyl-			\int
			4,5-dihydro-1 <i>H</i> -pyrazol-3-	(21	<u>ک</u>
			yl)ethan-1-one	$\langle \langle \rangle$	\mathcal{D}	
3	ARQ-621 (3)	RPHAD	(R)-N-(3-aminopropyl)-3-	1.2	1.88	-33.68
			chloro-N-(1-(7-chloro-4-	7/10		
			oxo-3-(phenylamino)-3,4-	$\langle \langle \rangle$		
			dihydro-quinazolin-2-yl)but	\gg		
			-3-yn-1-yl)-2-fluoro-			
			benzamide			
4	AZD-4877	RPHA	N-(3-aminopropyl)-N-(1-(5-	2	0.88	-33.47
	(4)		benzyl-methyl-4-oxo-4,5-			
			dihydroisothizolo[5,4-d]-			
		\square	pyrimidin-6-yl)-2-methyl-			
			propyl)-4-methylbenzamide			
5	MK-0731 (5)	RPHAD	(S)-4-(2,5-difluorophenyl)-	2.2	0.69	-4.58
		\sim	<i>N</i> -((3 <i>S</i> ,4 <i>S</i>)-3-fluoro-1-			
			methyl-piperidin-4-yl)-2-			
	$\left(\left(\begin{array}{c} \\ \end{array} \right) \right)$		(hydroxymethyl)-N-methyl-			
(\sim		2-phenyl-2,5-dihydro-1H-			
	\bigcirc		pyrrole-1-carboxamide			
6	Filanesib (6)	RPHA	(<i>S</i>)-2-(3-aminopropyl)-5-	6	2.44	-33.78
V			(2,5-difluorophenyl)-N-			
Ŧ			methoxy-N-methyl-2			
			phenyl-1,3,4-thiadiazole-			
			3(2H)-carboxamide			

	Category of KSP inhibition profile	Biological response (IC ₅₀)	Total No. of Ligands	Screening by Hypo1
Active (Total: 54)	Highly active	< 20 nM	36	34 (94.44%)
	Moderately active	20-100 nM	18	14 (77.77%)
(Total: 755)	Weakly active	> 100 nM	755	13 (1.72%)
	Active (Total: 54) Inactive (Total: 755)	Active Highly active (Total: 54) Moderately active (Total: 755) Weakly active	Active Highly active < 20 nM (Total: 54) Moderately 20-100 nM active (Total: 755) Weakly active > 100 nM	Active Highly active < 20 nM 36 (Total: 54) Moderately 20-100 nM 18 active (Total: 755) Weakly active > 100 nM 755

Parameters	Significance
Total number of molecules in database (D)	809
Total number of active molecules in database (A)	54
Total Hits (Ht)	51
Active Hits (Ha)	48
% Yield of actives [(Ha/Ht) X 100]	94.1
% Ratio of actives [(Ha/A) X100]	88.9
Enrichment factor (E)	14.1
[(Ha X D)/(Ht X A)]	(
False Negatives [A - Ha]	6
False Positives [Ht - Ha]	3
Goodness of Hit Score (GH) ^a	0.9244

^a[(Ha/4HtA) (3A+Ht)) X (1- ((Ht- Ha)/(D- A))];GH Score 0.9244 indicates good model.

	Molecules	Energy comp	oonents (kcal/mol)			
		ΔE_{vdW}	ΔE_{elec}	ΔG_{gas}	ΔG_{solv}	ΔG _{bind}
1.	SB-743921	-44.40±3.95	-187.56±10.06	-231.96±12.03	200.42±9.10	-31.54±4.33
	(Reference)					
2.	MB-41570	-52.15±4.54	-3.34±4.33	-55.50±7.17	12.60±5.85	-42.89±4.29
	(Maybridge)				\square	*
3.	CB-10358	-51.73±3.53	-25.77±7.40	-77.50±7.06	77.50±7.06	-47.53±4.09
	(ChemBridge)					
			~			
			$\langle E \rangle$	\sum		
				γ_{r}		
				7~		
				γ		
				7~		
				7~		
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				7~		
				7~		

	Molecules	RMSD	RMSF	RoG	
l.	SB-743921	2.45	0.92	20.87	$\langle \rangle$
	(Reference)				
2.	MB-41570	2.79	0.98	21.03	\rightarrow
	(Maybridge)			C X	\checkmark
3.	CB-10358	2.37	0.93	20.87	
	(ChemBridge)				
				\bigcirc	
(

Ligand and structure based *in silicostudies* to identify kinesin spindle protein (KSP) inhibitors as potential anticancer agents

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Supplementary information

Table S1 Test set employed in the validation

S.	Compd Id	Structure	IC ₅₀	Reference	Category
No		> >	(nM)		
1	24	O N-N O N-N O N	0.4	(Garbaccio et al., 2007)	Active
2	28	H ₂ N	0.5	(Garbaccio et al., 2007)	Active



























S.	Ligands	CDOCKE	Interactions af	ter molecular	Interactions af	íter
No		R energy	docking		molecular dyn	amics
•			Type of	Key residues	Type of	Key
			interactions		interactions	residues
			(Distance in		(Distance in	
			Å)		Å)	\bigcirc \lor
1	SB-743921	-26.05	-NH: H-bond	Gly117		<u> </u>
	(PDB ID:		(2.81);	(O=C);		\rightarrow
	4BXN)		-Cl: halogen	Ala218 (-		
			(1.35)	NH)	20	
2	SB-743921	-26.05	-Cl:	Leu214,	ring-C=O: H-	-
	(1)		hydrophobic;	Lys216;	bond (1.83);	NH:Arg104;
			-CH ₃ of tolyl	Leu132,	-CH ₂ : H-bond	
			moiety:	Ala133	(2.50);	O=C:Glu10
			alkylinteractio		NH ₂ :Salt	3;
			ns	\searrow	bridge;	Asp115;
			\sim		π -Orbitals:	Ala203,
		\square			hydrophobic;	Pro122,
					alkyl	
					interactions	Leu199,
		$\langle \bigcirc \rangle$ \checkmark				Leu125 and
						Val125
3	2	-32.63	-NH: H-bond	Glu116		
	\sim		(2.11);	(O=C);		
\sim	\bigcirc		-N:C-H bond	Glu118;		
	\geq		(2.92);	Glu116		
V			-СН3:С-Н	(O=C);		
			bond (2.73);	Glu118;		
			π -Orbitals:			
			hydrophobic;	Ala218		
			π-alkyl			

Table S2CDOCKER energy and molecular interactions of bound ligand, training set ligands and top-ranked hits from each database

			interaction			
4	ARQ-621	-33.68	-NH ₂ : H-bond	Glu116		
	(3)		(2.78 and	(O=C);		
			1.83);			
			-Methylene:	Glu116		
			C-H bond	(O=C);		\land
			(2.76);			
			-Methylene:			$\langle \rangle \rangle^{\vee}$
			C-H bond	Gly117;		
			(2.92);			\searrow
			π -Orbitals:	Leu214;		, ,
			hydrophobic;	Pro137 and	\mathcal{C}	
			alkyl	Ala133;	$\langle \mathcal{O} \rangle$	
			interactions;	Arg149))	
			π -cation:			
			electrostatic			
5	AZD-4877	-33.47	-NH: H-bond	Glu118		
	(4)		(1.90);	(O=C);		
			-Methylene:	Glu118		
			C-H bond	(O=C);		
	\sim		(2.48);			
		\rightarrow	-Methylene:	Gly117;		
			C-H bond			
	\mathcal{C}		(2.80);	Glu118		
\sim	\bigcirc		-Methylene:	(O=C);		
	\sim		C-H bond			
\lor			(2.83);	Ala218,		
			π -Orbitals:	Leu214 and		
			hydrophobic;	Pro137;		
			alkyl	Pro137		
			interactions			

6	MK-0731	-4.58	OH:H bond	Arg119;		
	(5)		(2.20)	Trp127,Glu1		
			-F: halogen	18 and		
			(3.66, 2.77	Gly117;		
			and 3.05);	Gly117 and		
			-CH ₃ :C-H	Glu116;		\land
			bond (2.46	Ala218,		
			and 2.99);	Pro137 and	~	$\langle \rangle \rangle^{\vee}$
			π -Orbitals:	Ala133		$\langle \rangle$
			hydrophobic			\succ
7	Filanesib (6)	-33.78	-F: halogen	Glu118		
			(2.79);	(O=C);	2	
			-OCH ₃ :C-H		\mathcal{O}	
			bond (2.63	Glu116)	
			and 2.73);	(O=C);		
			-NCH ₃ :C-H	Leu214;		
			bond (2.59			
			and 2.90);	Pro137 and		
			π -Orbitals:	Ala133;		
			hydrophobic;	Trp127		
			π - π stacking			
8	MB-41570	-42.70	-Ester CH ₂ : C-	Glu116;	-Ester C=O: H	Glu103;
	(Maybridge)		H bond (2.99);	Glu118;	bond (2.82);	Lys192,
			N-CH ₃ : C-H	Tyr211;	N-CH ₃ : C-H	Leu125,
			bond(2.89);	Ala133,	bond (2.59);	Trp122;
	\mathcal{C}		π - sulphur;	Gly117,		
	\bigcirc		vdW	Ala218,	π - sulphur;	
	\geq		interactions;	Leu214,	Piperazine	Tyr196;
				Glu128, and	CH ₂ : C-H	
				Phe144;	bond (2.90);	Pro122;
					Ester	
					methylene: C-	Glu103 and
			alkyl		H bond (2.73	Trp112;

			interactions	Arg119 and	& 2.65);	
				Val210	alkyl	Arg104 and
					interactions	Val195
9	CB-10358	-39.19	-OCH3: C-H	Glu116;	-CH ₂ : C-H	O=C of
	(ChemBridg		bond (3.06);		bond (2.84);	Glu103;
	e)		Piperidine-	Gly117	Piperidine-	\land
			CH ₂ :C-H		СН₂: С-Н	Gly102
			bond (2.85);	Tyr211;	bond (3.00);	$\langle \rangle >$
			π - π stacking;	Trp127,	-Ester CH ₃ :	Glu101;
			vdW	Glu128, and	C-H bond	\rightarrow
			interactions;	Ala133;	(2.45);	Try196;
				Leu214,	π - π stacking;	Ala203 and
			π-alkyl	Pro137,	π-alkyl	Leu199
			interactions;	and Ala218	interactions	

 Table S3 Top-ranked complex from each database and their fit values

S. No.	Ligand ID	Structure	CDOCKER energy	Fit value
			(kcal/mol)	
1	CB-10358	No No	-39.19	3.12
	(ChemBridge			
	$(\bigcirc)^{\sim}$			
2	MB-41570	N	-42.70	3.28
	(Maybridge)			

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Spectral characterization of the hits (Maybridge and ChemBridge)



Figure S1 FT-IR spectrum of starting compound 3



Figure S3 ¹³C NMR of compound 3


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Figure S4 FT-IR spectrum of compound 4

Figure S4 FT-IR

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Figure S6¹³C NMR of compound 4



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Figure S9 ¹³C NMR of compound 5 (Maybridge hit)



Figure S11¹³C NMR spectrum of ChemBridge hit

KSP ATPase inhibition profiles of Hits



ATPase activity in the presence of increasing Maybridge hit (red dots) and ChemBridge hit (blue dots) concentrations. **(B)** Inhibition of the MT-stimulated Eg5 ATPase activities in the presence of increasing Maybridge hit (red dots) and ChemBridge hit (blue dots) concentrations. The results indicate that both the hits have no inhibition effect on neither the basal nor the MT-stimulated Eg5 ATPase activities. The data points were fitted using Kaleidagraph 4.0. All measurements were performed in triplicate.